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#### Authors for correspondence:

Yong Guo e-mail: guoyong\_122@zzu.edu.cn Yanbing Zhang e-mail: zhanqyb@zzu.edu.cn

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THE ROYAL SOCIETY PUBLISHING

# Semisynthesis and insecticidal activity of some novel fraxinellone-based thioethers containing 1,3,4-oxadiazole moiety

Yong Guo<sup>1</sup>, Xiaoguang Wang<sup>1</sup>, Jiangping Fan<sup>1</sup>, Qian Zhang<sup>1</sup>, Yi Wang<sup>2</sup>, Yi Zhao<sup>1</sup>, Mengxing Huang<sup>1</sup>, Ming Ding<sup>1</sup> and Yanbing Zhang<sup>1</sup>

<sup>1</sup>Key Laboratory of Advanced Drug Preparation Technologies, Ministry of Education, School of Pharmaceutical Sciences, Zhengzhou University, Zhengzhou, 450001, Henan Province, People's Republic of China

<sup>2</sup>College of Agriculture, Shanxi Agriculture University, Taigu 030801, Shanxi Province, People's Republic of China

(D) YG, 0000-0003-2227-8276

Two series of novel fraxinellone-based thioethers containing 1,3,4-oxadiazole moiety were prepared as insecticidal agents against the oriental armyworm, *Mythimna separata* Walker. The structural assignment was based on the spectroscopic and X-ray analysis data. Among all the target compounds, compounds **4b**, **4k**, **5b**, **5j** and **5k** exhibited more potent insecticidal activity with final mortality rates (FMRs) of more than 65%, especially **4k** with the FMR of 75.9%, when compared with toosendanin. Some interesting results of structure–activity relationships are also discussed.

# 1. Introduction

The oriental armyworm (*Mythimna separata* Walker, Lepidoptera: Noctuidae), a typical long-distance migratory insect, is one of the most serious pests of cereal crops in countries including China, India, Australia and New Zealand [1,2]. Seasonal outbreaks of this pest can cause significant economic damage to cereal crops in China and other countries [3]. A recent outbreak of *M. separata* has been reported in northeast and central China during 2012, which caused losses of approximately 10 million acres of crops [4]. Synthetic chemical pesticides play a crucial role in agriculture with the characteristics of high-efficiency,

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Figure 1. Chemical structures of fraxinellone (1) and its derivatives (I–VII).

quick-fix and broad-spectrum insecticides. Although they have been extensively used to control insect pest outbreaks, the overuse and improper application of synthetic chemical pesticides over the years has resulted in enhancement of pest resistance, environmental problems and negative impacts on human health [5–7]. Hence the discovery and development of effective, selective and eco-friendly pesticides is necessary in the future.

Fraxinellone (1, figure 1), a naturally occurring degraded limonoid, isolated from Fagaropsis glabra, [8] Dictamnus albus, [9] Melia azadarach [10] and Dictamnus dasycarpus, [11] exhibits a variety of interesting activities both in the fields of medicinal chemistry and agrochemistry, such as anti-inflammatory [12], vascular relaxing activity [13] and insecticidal activity [14-16]. The total synthesis of fraxinellone can be easily achieved, which has been reported early in 1972 [17]. Previously, we have studied the insecticidal activity of some fraxinellone-based hydrazones and esters [18,19] (I-IV, figure 1) modified at the C-4 or C-10 position in the A ring of fraxinellone, and N-phenylpyrazole fraxinellone hybrid compounds [20] (V, figure 1), and found some compounds against M. separata displayed higher insecticidal activity than positive control toosendanin. To the best of our knowledge, little attention has been paid to the introduction of active N-heterocyclic moieties on the furyl-ring of fraxinellone as insecticidal agents. 1,3,4-Oxadiazoles are an important class of N-heterocyclic compounds with a wide range of biological activities [21] including antimicrobial, analgesic, anticancer activities, especially insecticidal and herbicidal activities [22,23]. In a continuation of our programme aimed at the development of fraxinellone-based insecticidal agents, herein we prepared two series of novel fraxiellone-based thioethers containing 1,3,4-oxadiazole moiety (VI and VII, figure 1) as insecticidal agents against M. separata.

## 2. Experimental

#### 2.1. Instrument and materials

The intermediate 2-mercapto-5-aryl-1,3,4-oxadiazoles  $\mathbf{a}-\mathbf{k}$  (scheme 1) were synthesized as previously reported [24]. Other reagents were of analytically grade and purchased from commercial resources. Fraxinellone (1) was isolated from *Dictamnus dasycarpus* and its purity was more than 99% as measured with reverse phase high-performance liquid chromatography (RP-HPLC). Analytical thin-layer chromatography (TLC) and preparative thin-layer chromatography (PTLC) were prepared by silica gel plates using silica gel GF<sub>254</sub> (Qingdao Haiyang Chemical Co., Ltd, Qingdao, China). Melting points were determined on a digital melting-point apparatus and were uncorrected (Beijing Tech Instrument Co., Ltd). Optical rotation was measured using an Autopol III automatic polarimeter (Rudolph Research

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Scheme 1. Synthetic route for the preparation of fraxiellone-based thioethers containing 1,3,4-oxadiazole moiety (4a-k and 5a-k).

Analytical, NJ, USA). Infrared spectra (IR) were recorded on a PE-1710 FT-IR spectrometer (Perkin-Elmer, Waltham, MA, USA). NMR spectra were obtained in CDCl<sub>3</sub> on a Bruker Avance (400 MHz) spectrometer using tetramethylsilane (TMS) as the internal standard (Bruker, Bremerhaven, Germany). High-resolution mass spectra (HR-MS) were carried out with LTQ FT Ultra instrument (Thermo Fisher Scientific Inc., Waltham, MA, USA).

#### 2.1.1. Data for **1**

White solid, m.p. 113–115°C; IR cm<sup>-1</sup>: 3148, 2930, 1741, 1671, 1607, 1202, 1022; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.46 (s, 1H, H-2'), 7.43 (t, *J* = 1.6 Hz, 1H, H-5'), 6.34 (d, *J* = 1.2 Hz, 1H, H-4'), 4.88 (s, 1H, H-8), 2.13–2.31 (m, 2H, H-4), 2,11(s, 3H, H-10), 1.71–1.87 (m, 3H, H-5, 6), 1.42–1.48 (m, 1H, H-6), 0.86 (s, 3H, H-9); MS (ESI), *m/z* (%) 233.05 ([M + H]<sup>+</sup>, 58).

## 2.2. General procedure for synthesis of compounds 2 and 3

To a stirred suspension solution of AlCl<sub>3</sub> (1.0 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) at RT, chloroacetyl chloride (1.1 mmol) was added. The mixture was then stirred for 10 min, and a solution of compound **1** (1.0 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (5 ml) was added dropwise to the above mixture. When the reaction was complete according to TLC analysis, the reaction mixture was poured into ice water (15 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (40 ml × 3). The combined organic phase was washed with saturated brine (40 ml), dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, concentrated *in vacuo*, and then purified by PTLC to give the pure products **2** (35% yield) and **3** (45% yield).

#### 2.2.1. Data for **2**

White solid, yield: 35%, m.p. 96–98°C;  $[\alpha]^{20}_{D} = 45$  (*c* 4.0 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2946, 2918, 2872, 1757, 1687, 1472, 1203, 975; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.55 (d, *J* = 1.6 Hz, 1H, H-5'), 6.79 (d, *J* = 2.0 Hz, 1H,

H-4'), 5.66 (s, 1H, H-8), 4.60–4.72 (m, 2H,  $-CH_2Cl$ ), 2.19–2.26 (m, 2H, H-4), 2.13 (s, 3H, H-10), 1.77–1.83 (m, 2H, H-5, 6), 1.58–1.63 (m, 2H, H-5, 6), 0.84 (s, 3H, H-11); HRMS (ESI): Calcd for C<sub>16</sub>H<sub>18</sub>O<sub>4</sub>Cl ([M+H]<sup>+</sup>), 309.0888; found, 309.0888.

#### 2.2.2. Data for **3**

White solid, yield: 45%, m.p. 106–108°C;  $[\alpha]^{20}_{D} = -20$  (*c* 3.5 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2956, 2923, 2870, 1737, 1671, 1496, 1235, 908; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.68 (s, 1H, H-2'), 7.27 (s, 1H, H-4'), 4.89 (s, 1H, H-8), 4.59 (s, 2H, –CH<sub>2</sub>Cl), 2.23–2.34 (m, 2H, H-4), 2.14 (s, 3H, H-10), 1.72–1.86 (m, 3H, H-5, 6), 1.47–1.53 (m, 1H, H-6), 0.84 (s, 3H, H-11); HRMS (ESI): Calcd for C<sub>16</sub>H<sub>18</sub>O<sub>4</sub>Cl ([M + H]<sup>+</sup>), 309.0888; found, 309.0888.

#### 2.3. General procedure for synthesis of compounds 4a-k and 5a-k

A mixture of the corresponding 2-mercapto-5-aryl-1,3,4-oxadiazole (0.3 mmol), **2** or **3** (0.2 mmol, 61.6 mg),  $K_2CO_3$  (0.3 mmol, 41.5 mg) and KI (0.05 mmol, 8.3 mg) in acetone (10 ml) was stirred at room temperature. After the reaction was complete according to TLC analysis, the solvent was removed and the residue was dissolved in CH<sub>2</sub>Cl<sub>2</sub> and filtered. The filtrate was concentrated *in vacuo* and purified by PTLC to give pure products **4a–k** and **5a–k**. The example data of **4a–d** and **5a–d** are described as follows, whereas the data of other compounds **4e–k** and **5e–k** are shown in the electronic supplementary material.

#### 2.3.1. Data for 4a

White solid, yield: 54%, m.p. 76–79°C;  $[\alpha]^{20}_{D} = 18 (c \ 3.4 \text{ mg ml}^{-1}, \text{acetone})$ ; IR cm<sup>-1</sup>: 2935, 1750, 1673, 1587, 1473, 1414, 1355, 1204, 1129, 1046; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.97 (dd, J = 7.6, 1.2 Hz, 2H, –Ph), 7.60 (d, J = 1.6 Hz, 1H, H-5'), 7.47–7.53 (m, 3H, –Ph), 6.83 (d, J = 1.2 Hz, 1H, H-4'), 5.67 (s, 1H, H-8), 4.67–4.87 (m, 2H, –COCH<sub>2</sub>S–), 2.13–2.26 (m, 5H, H-4, 10), 1.71–1.77 (m, 2H, H-5, 6), 1.55–1.61 (m, 2H, H-5, 6), 0.86 (s, 3H, H-11).

#### 2.3.2. Data for **4b**

White solid, yield: 68%, m.p. 66–68°C;  $[\alpha]^{20}_{D} = 11$  (*c* 3.6 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2923, 1752, 1674, 1588, 1495, 1472, 1414, 1204, 1129, 1048; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.97–8.01 (m, 1H, –Ph), 7.60 (d, *J* = 1.6 Hz, 1H, H-5'), 7.50–7.55 (m, 1H, –Ph), 7.28 (d, *J* = 8.0 Hz, 1H, –Ph), 7.21 (d, *J* = 8.4 Hz, 1H, –Ph), 6.83 (d, *J* = 1.6 Hz, 1H, H-4'), 5.67 (s, 1H, H-8), 4.67–4.88 (m, 2H, –COCH<sub>2</sub>S–), 2.13–2.26 (m, 5H, H-4, 10), 1.73–1.77 (m, 2H, H-5, 6), 1.55–1.63 (m, 2H, H-5, 6), 0.86 (s, 3H, H-11).

#### 2.3.3. Data for **4c**

White solid, yield: 72%, m.p. 81–83°C;  $[\alpha]^{20}_{D} = 10 (c 4.2 \text{ mg ml}^{-1}, \text{acetone})$ ; IR cm<sup>-1</sup>: 2927, 1753, 1673, 1587, 1478, 1414, 1355, 1275, 1204, 1129, 1046; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.91 (dd, J = 7.6, 1.6 Hz, 1H, –Ph), 7.60 (d, J = 1.6 Hz, 1H, H-5'), 7.52–7.54 (m, 1H, –Ph), 7.44 (td, J = 7.6, 2.0 Hz, 1H, –Ph), 7.37 (td, J = 7.6, 1.2 Hz, 1H, –Ph), 6.83 (d, J = 1.6 Hz, 1H, H-4'), 5.67 (s, 1H, H-8), 4.67–4.88 (m, 2H, –COCH<sub>2</sub>S–), 2.15–2.26 (m, 2H, H-4), 2.13 (s, 3H, H-10), 1.71–1.77 (m, 2H, H-5, 6), 1.54–1.58 (m, 2H, H-5, 6), 0.86 (s, 3H, H-11).

#### 2.3.4. Data for **4d**

Pale yellow solid, yield: 73%, m.p. 64–66°C;  $[\alpha]^{20}_{D} = 5 (c \ 3.7 \text{ mg ml}^{-1}, \text{ acetone})$ ; IR cm<sup>-1</sup>: 2931, 1752, 1673, 1587, 1475, 1413, 1355, 1259, 1203, 1129, 1046; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.84 (d, J = 8.4 Hz, 2H, -Ph), 7.63 (d, J = 8.4 Hz, 2H, -Ph), 7.60 (d, J = 1.6 Hz, 1H, H-5'), 6.83 (d, J = 1.6 Hz, 1H, H-4'), 5.66 (s, 1H, H-8), 4.67–4.88 (m, 2H, -COCH<sub>2</sub>S–), 2.08–2.26 (m, 2H, H-4, 10), 1.71–1.77 (m, 2H, H-5, 6), 1.54–1.58 (m, 2H, H-5, 6), 0.86 (s, 3H, H-11).

#### 2.3.5. Data for **5a**

White solid, yield: 65%, m.p. 81–83°C;  $[\alpha]^{20}_{D} = -2$  (*c* 3.2 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2940, 2920, 1744, 1676, 1474, 1205, 1133, 1048, 1003; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.98 (d, *J* = 6.4 Hz, 2H, –Ph), 7.70 (s, 1H, H-2'), 7.47–7.53 (m, 3H, –Ph), 7.29 (s, 1H, H-4'), 4.89 (s, 1H, H-8), 4.73 (s, 2H, –COCH<sub>2</sub>S–), 2.17–2.34 (m, 2H, 2H, 2H) (m, 2H) (m

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Figure 2. Comparison of partial <sup>1</sup>H NMR spectra of compounds 1–3, 4a and 5a.

H-4), 2.14 (s, 3H, H-10), 1.83–1.91 (m, 2H, H-5, 6), 1.71–1.78 (m, 1H, H-5), 1.46–1.53 (m, 1H, H-6), 0.85 (s, 3H, H-11).

#### 2.3.6. Data for **5b**

White solid, yield: 87%, m.p.  $111-112^{\circ}$ C;  $[\alpha]^{20}_{D} = -19$  (*c* 3.2 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2935, 1750, 1682, 1618, 1495, 1472, 1388, 1206, 1161, 1047; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.97–8.01 (m, 1H, –Ph), 7.71 (s, 1H, H-2'), 7.48–7.55 (m, 1H, –Ph), 7.21–7.30 (m, 3H, H-4' and –Ph), 4.90 (s, 1H, H-8), 4.73 (s, 2H, –COCH<sub>2</sub>S–), 2.17–2.34 (m, 2H, H-4), 2.14 (s, 3H, H-10), 1.84–1.87 (m, 2H, H-5, 6), 1.73–1.77 (m, 1H, H-5), 1.47–1.54 (m, 1H, H-6), 0.85 (s, 3H, H-11).

#### 2.3.7. Data for **5c**

White solid, yield: 79%, m.p. 106–108°C;  $[\alpha]^{20}_{D} = -12$  (*c* 2.1 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2930, 1746, 1675, 1599, 1506, 1474, 1466, 1265, 1208, 1169, 1037; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.91 (dd, *J* = 7.6, 1.6 Hz, 1H, -Ph), 7.70 (s, 1H, H-2'), 7.53 (d, *J* = 8.0 Hz, 1H, -Ph), 7.44 (td, *J* = 8.0, 2.0 Hz, 1H, -Ph), 7.38–7.41 (m, 1H, -Ph), 7.29 (s, 1H, H-4'), 4.90 (s, 1H, H-8), 4.73 (s, 2H, -COCH<sub>2</sub>S-), 2.17–2.34 (m, 2H, H-4), 2.14 (s, 3H, H-10), 1.84–1.87 (m, 2H, H-5, 6), 1.73–1.77 (m, 1H, H-5), 1.46–1.54 (m, 1H, H-6), 0.86 (s, 3H, H-11).

#### 2.3.8. Data for **5d**

Pale yellow solid, yield: 89%, m.p. 131–132°C;  $[\alpha]^{20}_{D} = -16$  (*c* 3.2 mg ml<sup>-1</sup>, acetone); IR cm<sup>-1</sup>: 2961, 2930, 2862, 1746, 1675, 1600, 1507, 1466, 1393, 1317, 1207, 1169, 1020; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$ : 7.86 (dd, *J* = 7.6, 1.6 Hz, 1H, –Ph), 7.72 (dd, *J* = 8.0, 1.2 Hz, 1H, –Ph), 7.70 (s, 1H, H-2'), 7.42 (td, *J* = 7.6, 1.2 Hz, 1H,





Figure 3. X-ray crystal structure of compound 4e.



Figure 4. X-ray crystal structure of compound 5d.

–Ph), 7.35 (td, *J* = 7.6, 1.6 Hz, 1H, –Ph), 7.30 (s, 1H, H-4'), 4.90 (s, 1H, H-8), 4.74 (s, 2H, –COCH<sub>2</sub>S–), 2.17–2.34 (m, 2H, H-4), 2.14 (s, 3H, H-10), 1.84–1.87 (m, 2H, H-5, 6), 1.73–1.77 (m, 1H, H-5), 1.46–1.54 (m, 1H, H-6), 0.86 (s, 3H, H-11).

## 2.4. X-ray crystallography

The structures of compounds **4e** and **5d** were unambiguously confirmed by X-ray crystallography. Crystallographic data (excluding structure factors) of compounds **4e** and **5d** were deposited at the Cambridge Crystallographic Data Centre (CCDC) with deposition numbers of CCDC 1552786 and 1552787, respectively.

## 2.5. Biological assay

Growth inhibitory activity of compounds 1–3, 4a–k and 5a–k against *M. separata* was evaluated by leafdipping method as described previously [19,25]. For each compound, 30 pre-third-instar larvae of same size and level of health (10 larvae per group) were chosen as the tested pests. Solutions of compounds 1–3, 4a–k and 5a–k and toosendanin (used as a positive control) were prepared in acetone at the concentration of 1 mg ml<sup>-1</sup>. The larvae of tested groups were fed with compound-coated leaves (fresh corn leaf discs (1 × 1 cm) were dipped into the corresponding solution for 3 s, then taken out and dried at RT), whereas the blank control group (CK) was fed with acetone alone. Several treated leaf discs were kept in each dish.

Table 1. 🤆	Growth	inhibitory	activity	of compounds	1-3, 4a	— <b>k</b> and	5a–k	against <i>M</i> .	separata	on leaves	s treated at	a concentratio	n
of 1 ma ml	-1												

	corrected mortality rate (% $\pm$ s.d.)						
compound	10 days	20 days	34 days				
1	16.7(±3.3)	24.1(±3.3)	44.8(±3.3)				
2	20.0(±5.8)	27.6(±5.8)	41.4(±3.3)				
3	23.3(±3.3)	31.0(±3.3)	48.3(±5.8)				
4a	13.3(±3.3)	24.1(±3.3)	37.9(±5.8)				
4b	30.0(±0)	48.3(±5.8)	72.4(±6.7)				
4c	20.0(土0)	37.9(±5.8)	55.2(±3.3)				
4d	23.3(±3.3)	41.4(±3.3)	51.7(±3.3)				
4e	16.7(±3.3)	31.0(±3.3)	51.7(±3.3)				
4f	16.7(±3.3)	31.0(±3.3)	48.3(±0)				
4g	23.3(±3.3)	31.0(±6.7)	44.8(±3.3)				
4h	16.7(±3.3)	20.7(±3.3)	41.4(±3.3)				
4i	13.3(土3.3)	24.1(±3.3)	37.9(±0)				
4j	23.3(±6.7)	41.4(±3.3)	62.1(±3.3)				
4k	33.3(±3.3)	51.7(±3.3)	75.9(±3.3)				
5a	13.3(土3.3)	20.7(±3.3)	37.9(±5.8)				
5b	26.7(±3.3)	41.4(±3.3)	69.0(±0)				
5c	26.7(±3.3)	37.9(±5.8)	62.1(±3.3)				
5d	23.3(±3.3)	34.5(±3.3)	55.2(±3.3)				
5e	16.7(±3.3)	27.6(土0)	48.3(±5.8)				
5f	20.0(±5.8)	31.0(±6.7)	48.3(±5.8)				
5g	16.7(±6.7)	27.6(土0)	41.4(±3.3)				
5h	13.3(±3.3)	17.2(土0)	34.5(±3.3)				
5i	16.7(±3.3)	24.1(±3.3)	44.8(±3.3)				
5j	26.7(±3.3)	41.4(±3.3)	65.5(±3.3)				
5k	30.0(±5.8)	55.2(±3.3)	69.0(±5.8)				
toosendanin	16.7(±3.3)	34.5(±3.3)	51.7(±3.3)				
blank control	0(±0)	3.3(±3.3)	3.3(±3.3)				

Once the treated leaves were consumed, the corresponding ones were added to the dish. The experiment was carried out at  $25 \pm 2^{\circ}$ C; relative humidity (RH) 65–80%, and on 12 h/12 h (light/dark) photoperiod. After 48 h, untreated fresh leaves were added to all dishes until the adult emergence. The corrected mortality rate values of the tested compounds were calculated by the following formula: corrected mortality rate (%) =  $(T - C) \times 100/(1 - C)$ ; Where *T* is the mortality rate in the treated group, and *C* is the mortality rate of CK.

## 3. Result and discussion

### 3.1. Synthesis

As shown in scheme 1, compounds  $\mathbf{a}-\mathbf{k}$  were synthesized based on previously reported literature [24]. Using different hydrazides as starting materials, compounds  $\mathbf{a}-\mathbf{k}$  were prepared by cyclization reaction of different hydrazides with carbon disulfide in the presence of KOH in EtOH at reflux temperature,



Figure 5. The representative abnormal larvae pictures of 5e (XSN-16), 4e (XSN-17), 4b (XSN-21), 5c (YG-158), 5d (YG-160), 5j (YG-163) and 5k (YG-165) during the larval period (CK: blank control group).

followed by acidification with 5% HCl. When fraxinellone reacted with chloroacetyl chloride in the presence of AlCl<sub>3</sub>, the corresponding 2'-chloroacetylfraxinellone (**2**) and 5'-chloroacetylfraxinellone (**3**) were obtained. Subsequently, different 2-mercapto-5-aryl-1,3,4-oxadiazoles (**a**–**k**) reacted with compound **2** or **3** in the presence of K<sub>2</sub>CO<sub>3</sub>/KI in anhydrous acetone at RT to smoothly acquire desired compounds **4a–k** and **5a–k**, respectively. The structures of all target compounds **4a–k** and **5a–k** were fully characterized by melting points, IR, optical rotation and <sup>1</sup>H NMR. Additionally, comparison of partial <sup>1</sup>H NMR spectra of compounds **1–3**, **4a** and **5a** was illustrated in figure 2. It is obvious that the chemical shifts of H-4' and H-8 of 2'-substituted fraxinellone are very different from 5'-substituted fraxinellone. When chloroacetyl substituted on 2'-position of fraxinellone, the proton of H-4' was shifted from 6.34 [d, J = 1.2 Hz, **1**, figure 2 (1)] to 5.66 [s, **2**, figure 2 (2)] ppm; By contrast, when chloroacetyl substituted on 5'-position of fraxinellone, the proton of H-8 was shifted from 4.88 [s, **1**, figure 2 (4)] ppm, while the proton of H-8 was slightly shifted from 4.88 [s, **1**, figure 2 (4)] ppm, while the proton of H-8 was slightly shifted from 4.88 [s, **1**, figure 2 (4)] ppm, while the proton of H-4' and H-8 of compounds **4a** and **5a** were similar to compounds **4a** and **5a** were similar to compounds **4a** and **5a** were similar to compounds **4**.

The structures of compounds **4e** and **5d** were further confirmed by X-ray crystallography (figures 3 and 4). As shown in figure 3, to the compound **4e**, 2-mercapto-5-(4-fluorophenyl)-1,3,4-oxadiazole (**e**) linked with chloroacetyl was at the 2'-position on the furyl ring. On the contrary, to the compound **5d** (figure 4), 2-mercapto-5-(2-bromophenyl)-1,3,4-oxadiazole (**d**) linked with chloroacetyl was at the 5'-position on the furyl ring of fraxinellone.

#### 3.2. Insecticidal activity

The growth inhibitory activity of compounds 1–3, 4a–k and 5a–k against *M. separata* was tested at 1 mg ml<sup>-1</sup>. Toosendanin, a commercial insecticide derived from *Melia azedarach*, was used as the positive control at 1 mg ml<sup>-1</sup>, and corn leaves treated with acetone alone were used as a blank control. As shown in table 1, compounds 4b, 4c, 4j, 4k, 5b–d, 5j, 5k exhibited higher insecticidal activity than toosendanin and their precursor fraxinellone. For example, the final mortality rates (FMRs) of compounds 4b, 4c, 4j, 4k, 5b–d, 5j, 5k were 72.4%, 55.2%, 62.1%, 75.9%, 69.0%, 62.1%, 55.2%, 65.5% and 69.0%, respectively. In particular, compound 4k showed the most potent insecticidal activity, which was about 24% higher than toosendanin. The symptoms for the tested *M. separata* during the different periods of larval, pupation and adult were recorded by the same methods as our previous reports [19,25]. For example, in treated



Figure 6. The representative malformed pupae pictures of 4b (XSN-21), 5c (YG-158), 5d (YG-160), 5i (YG-161), 4a (YG-166), 4f (YG-167) and 4d (YG-170) during the pupation period (CK: blank control group).



Figure 7. The representative malformed moth pictures of 4b (XSN-21), 5b (YG-157), 5j (YG-163), 5k (YG-165), 4d (YG-170), 4j (YG-173) and 4k (YG-174) during the emergence period (CK: blank control group).

groups, due to overfeeding of treated leaves in the beginning, some larvae died slowly with thin and wrinkled bodies (figure 5). This phenomenon maybe results from these fraxinellone derivatives effects to nutritional or digestive interference [26]. During the pupation stage, some of the larvae did not successfully moult to normal pupae, and died (figure 6). In the last stage of emergence, many malformed moths appeared with shrunken or immature wings (figure 7). These results suggest that the fraxinellone derivatives containing the 1,3,4-oxadiazole probably affected the insect moulting hormone, which was crucial for the growth of *M. separata*. On the other hand, as displayed in figure 8, the percentages of FMRs of compounds **4b**, **4j**, **4k**, **5b**, **5c**, **5j**, **5k** and toosendanin at three different growth stages of *M. separata* were investigated. We found that at least 50% of FMRs for compounds **4b**, **4k**, **5c**, **5j**, **5k** and toosendanin were at the larval period except for compounds **4j** and **5b**.



Figure 8. The percentages of FMRs during three different growth periods of compounds 4b, 4j, 4 k, 5b, 5c, 5j, 5k and toosendanin.

Finally, we also discovered some interesting results of structure–activity relationships of the tested compounds.

When 2'-chloroacetylfraxinellone (2) or 5'-chloroacetylfraxinellone (3) linked with 2-mercapto-5-(pyridinyl)-1,3,4-oxadiazoles exhibited more potent insecticidal activity than toosendanin. For example, the FMRs of compounds 4j, 4k, 5j and 5k were 62.1%, 75.9%, 65.5% and 69.0%, respectively. In other fraxinellone-based thioethers, introduction of electron-donating groups on the phenyl of 4a/5a resulted in less active compounds (e.g. 41.4% for 4h, 37.9% for 4i, 34.5% for 5h and 44.8% for 5i). When halogen atoms were introduced on the *para*-position of phenyl of 4a/5a gave potent compounds. For instance, the FMRs of compounds 4b-d and 5b-d were greater than or equal to toosendanin (51.7%), especially the FMR of 4b was 72.4%. In our previous research, we found that introduction of heterocycle, fluorophenyl or *o*-chlorophenyl fragments on the 1,3,4-oxadiazole ring at the C-3 position of sarisan could afford more potent compounds [27]. In this paper, introduction of 3/4-pyridinyl or *o*-fluoro/chlorophenyl units on the 1,3,4-oxadiazole ring to the compound 2 or 3 also obtained the promising compounds 4j, 5j, 4k, 5k, 4b, 5b, 4c and 5c, respectively. Hence, this suggested that we could introduce the 2-mercapto-5-(3/4-pyridinyl/*o*-fluoro/chlorophenyl)-1,3,4-oxadiazoles activity units into other insecticidal lead compounds in the future.

## 4. Conclusion

In summary, we have prepared two series of novel fraxinellone-based thioethers containing 1,3,4oxadiazole and evaluated for their insecticidal activity against a cereal crop-threatening agricultural insect pest, *M. separata*. The structures of key compounds **4e** and **5d** were assigned by X-ray crystallography. Among all target compounds, compounds **4b**, **4k**, **5b**, **5j** and **5k** exhibited more potent insecticidal activity with FMRs of more than 65%. The results suggested that the introduction of 3/4pyridinyl or *o*-fluoro/chlorophenyl units on the 1,3,4-oxadiazole ring to the compound **2** or **3** could afford more promising compounds. This will lay the foundations for further structural modification and application of fraxinellone as novel pesticidal agents in agriculture.

Ethics. The study was approved by the Research Ethics Committee of Zhengzhou University, Henan Province, PR China.

Data accessibility. Electronic supplementary material is available at the Dryad Digital Repository: https://doi.org/10. 5061/dryad.424tb [28]. Crystallographic data (excluding structure factors) of compounds **4e** and **5d** were deposited at the Cambridge Crystallographic Data Centre (CCDC) with deposition numbers of CCDC 1552786 and 1552787, respectively. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, U.K. [fax, +44 (0)1223 336033; or e-mail, deposit@ccdc.cam.ac.uk].

Authors' contributions. Y.G. and Y. Zhang designed the study, and Y.G. performed part of the experiments and wrote the manuscript. X.W., J.F., Q.Z., Y.W., Y. Zhao, M.H. and M.D. carried out the experiments. All authors gave their final approval for publication.

Competing interests. We declare we have no competing interests.

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