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A BIBLIOGRAPHY OF THERMOPHYSICAL PROPERTIES OF ARGON FROM 0 TO 300°K

L. A. HALL, J. G. HUST, AND A. L. GOSMAN



U. S. DEPARTMENT OF COMMERCE
NATIONAL BUREAU OF STANDARDS

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A BIBLIOGRAPHY OF THERMOPHYSICAL PROPERTIES OF ARGON
FROM 0 TO 300°K¹

L. A. Hall, J. G. Hust, and A. L. Gosman

A bibliography of 450 references is presented for mechanical, thermodynamic, and transport properties of argon from 0 to 300°K. Each article has been reviewed and coded with regard to properties studied, type of article (i.e., experimental, theoretical, etc.), and method of presentation of the data. The temperature and pressure ranges for each property under consideration are also given. An index was prepared according to property with 4 sub-categories: solid, liquid, gas up to 200°K, and gas above 200°K.

1. INTRODUCTION

The Compilation Unit of the Cryogenic Data Center has in its mission the critical evaluation of quantitative information from the world's literature related to the thermophysical properties of materials at cryogenic temperatures. At the outset of the study of a particular material, copies of the documents concerned with the properties are obtained and reviewed. As the task of document accumulation continues, a concerted effort is made to complete a systematic and thorough literature search on the selected topic. For argon, the collection of documents began over five years ago in conjunction with the data compilation presented in the "Compendium."² This bibliography on the properties of argon is the publication of the results, which provides a methodical organization of the literature on argon.

The initial literature search was conducted by the use of various abstracting journals, and in particular of Chemical Abstracts. Copies of each of the pertinent documents were obtained, and each document was reviewed and coded. In addition, all pertinent articles, which were referenced in these documents, were also obtained, reviewed, and coded. The search of Chemical Abstracts has been periodically up-dated.

¹ This bibliography is a result of a study made under contract with the National Aeronautics and Space Administration.

² A Compendium of the Properties of Materials at Low Temperature (Phase I.) Part I. Properties of Fluids, V. J. Johnson, editor, Wright Air Development Division Tech. Rept. 60-56 (1960) 560 p.

2. FORMAT FOR LISTING CITATION AND DOCUMENT CONTENTS

The citations have been arranged alphabetically by first author and numbered. Late additions were inserted in alphabetical sequence and given the number of the preceding article followed by an A, B, etc. Only information from the article which concerns the properties of argon was noted in this bibliography.

The information given for each citation includes and is ordered as follows:

1. Author(s)
2. Title (original language) and translated title, if original is in a language other than English
3. Reference (If the same article is published in more than one place, each reference is cited.)
4. Chemical Abstract number or ASTIA number, when they are known
5. Properties studied for argon, state of substance, temperature and pressure ranges as available
6. Substances, other than argon, which are discussed in the article (Where no fluids are listed, the contents of the document are restricted to properties of argon only.)
7. Designation as to primary character of article
 - a. experimental
 - b. theoretical
 - c. compilation¹
 - d. correlation
 - e. reference book²
8. Form in which data is reported
 - a. tabular - tables (number of values)
 - b. graphical
 - c. equations
9. Apparatus, if described or illustrated
10. Original language, if other than English.

¹ In compilation, the bibliography number of the original article from which the data was obtained is listed.

² The amount of data in reference books is not given.

3. INDEX OF PROPERTIES

The bibliography is indexed according to property with sub-indexes for the state of the substance; i.e., solid, liquid, gas up to 200°K, and gas above 200°K. The letters E, T, C, and R following each citation number refer to the type of data; i.e., E = experimental, T = theoretical, C = compilation, correlation, calculation, and R = review, discussion, reference work. A few reference books were coded by property only and are designated by *R following the citation number.

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+ The properties for which the literature search has been conducted, including a complete search in Chemical Abstracts (1917 to 1962).

PROPERTY INDEX

1. DENSITY, P-V-T DATA, COMPRESSIBILITY FACTOR

SOLID

95 E 101 E 102 C 103 E 105 R 112 T 193 E 279 C 346 E 350 E
413 I

L I Q U I D

18 C 49 E 87 C 104 E 126 T 153 C 174 T 193 E 238 R 279 C
346 E 376 R 377 E 384 E 385 E 405 T 413 T

GAS (UP TO 200 DEGREES K)

24 T	48 C	49 E	87 C	88 T	134 E	144 T	153 C	174 T	175 E
177 E	180 C	193 E	204 C	236 E	237 T	238 R	239 T	244 C	247 C
253 C	289 C	290 E	293 T	322 E	394 E	401 C	413 T		

GAS (ABOVE 200 DEGREES K)

PROPERTY INDEX (CONT.)

2. EQUATION OF STATE, VIRIAL COEFFICIENTS

2 T	22 T	23 T	24 T	25 T	26 T	27 T	31 C	32 T	34 T
55 T	60 E	67 T	80 T	88 T	89A T	92 T	105 R	110 C	147 T
160 T	164 T	165 T	167*R	171 T	172 T	178 C	195 T	197 T	198 T
199 T	204 C	205 E	216 T	231 T	234 T	236 E	250 R	254 T	265 E
274 E	278 R	289 C	290 E	292 T	328 C	334 R	364 T	364A T	365 T
381 E	385 E	394 E	395 T	399 T	402 E	423 T	425 T		

SOLID

32 T	92 T	105 R	160 T	165 T	198 T	199 T	423 T		
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LIQUID

2 T	231 T	236 E	385 E	399 T					
-----	-------	-------	-------	-------	--	--	--	--	--

GAS (UP TO 200 DEGREES K)

24 T	25 T	31 C	60 E	67 T	88 T	164 T	178 C	204 C	205 E
234 T	236 E	254 T	289 C	290 E	292 T	334 R	364 T	364A T	365 T
381 E	395 T	399 T							

GAS (ABOVE 200 DEGREES K)

24 T	25 T	26 T	31 C	34 T	60 E	67 T	88 T	164 T	171 T
172 T	205 E	236 E	250 R	254 T	274 E	289 C	290 E	292 T	334 R
364 T	364A T	365 T	381 E	395 T	399 T	402 E			

PROPERTY INDEX (CONT.)

3. EXPANSIVITY AND COMPRESSIBILITY

15 E	16 C	28 T	32 T	33 T	89A T	100 E	103 E	105 R	106 T
107 T	147 T	160 T	189*R	218 T	219 T	229 E	232 E	239 T	241 E
265 E	285 E	317B T	322 E	329 E	351 E	356 E	357 E	376 R	377 E
383 E	385 E	386 E	396*R	399 T	413 T	422 T	425 T	426 T	

SOLID

15 E 16 C 28 T 32 T 33 T 89A T 100 E 103 E 105 R 106 T
160 T 229 E 351 E 356 E 357 E 413 T 422 T

L I Q U I D

15 E 229 E 241 E 376 R 377 E 383 E 385 E 386 E 399 T 413 T

GAS (UP TO 200 DEGREES K)

265 E 322 E 399 T 413 T

GAS (ABOVE 200 DEGREES K)

232 E 265 E 285 E 322 E 329 E 399 T 413 T

4. VAPOR PRESSURE

SOLID

41 E 45 E 70 E 85 E 86 E 98*R 99 R 115 R 127 T 133 E
 167*R 243 C 279 C 308 E 317A T 362 C 391 T 396*R 414 C 420 C

L I Q U I D

39 T	40 E	42 E	45 E	69 E	70 E	82 E	86 E	98*R	99 R
108 T	127 T	133 E	138 R	146 R	147 T	150 T	154 T	167*R	182 E
183 C	235 T	244 C	255 C	279 C	286 E	287 E	308 E	322 E	328 C
362 C	366A R	385 E	395 T	396*R	420 C				

5. MELTING PRESSURE

49 E 50 E 75 E 92 T 145 T 229 E 270 E 313 E 319 E 337 E
347 E 348 E

PROPERTY INDEX (CONT.)

6. SATURATION DENSITY

13 E	35 E	49 E	56 C	83 E	86 E	163 E	229 E	236 E	243 C
256 E	258 E	265 E	317B T	383 E					

SOLID (SOLID-LIQUID TRANSITION)

49 E	229 E								
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LIQUID (SOLID-LIQUID TRANSITION)

49 E	229 E	317B T							
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LIQUID (LIQUID-VAPOR TRANSITION)

13 E	35 E	56 C	83 E	86 E	236 E	243 C	256 E	258 E	265 E
383 E									

GAS (LIQUID-VAPOR TRANSITION)

83 E	86 E	163 E	236 E	243 C	256 E	258 E	265 E		
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7. LATENT HEATS

28 T	32 T	49 E	50 E	62 T	72 E	85 E	86 E	98*R	119 E
122 E	125 R	126 T	132 E	133 E	137 E	147 T	152 T	155 T	183 C
189*R	199 T	203 T	204A C	215A T	226 T	229 E	231 T	235 T	236 E
243 C	248 T	255 C	257 E	279 C	300 C	314 T	315 T	316 T	318 C
323*R	328 C	330 T	355 T	363*R	391 T	392 T	420 C		

FUSION

49 E	50 E	62 T	72 E	98*R	122 E	125 R	132 E	133 E	147 T
183 C	189*R	204A C	215A T	229 E	243 C	255 C	315 T	323*R	363*R

SUBLIMATION

28 T	32 T	85 E	98*R	189*R	199 T	314 T	316 T	391 T	420 C
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VAPORIZATION

86 E	98*R	119 E	125 R	126 T	132 E	133 E	137 E	147 T	152 T
155 T	183 C	189*R	203 T	204A C	226 T	231 T	235 T	236 E	243 C
248 T	255 C	257 E	279 C	300 C	318 C	323*R	328 C	330 T	355 T
363*R	392 T	420 C							

PROPERTY INDEX (CONT.)

8. FIXED POINTS

1 T	62 T	70 E	72 E	73 E	74 T	79 C	82 E	83 E	91 T
92 T	98*R	125 R	133 E	137 E	138 R	138A E	146 R	147 T	153 C
159 T	162 R	182 E	183 C	196 C	204 C	215A T	234 T	235 T	236 E
240 T	243 C	255 C	256 E	256A R	265 E	270 E	273 E	279 C	286 E
287 E	295 C	298 R	301 E	302 T	303 T	308 E	318 C	328 C	344 T
358 T	362 C	363*R	395 T	396*R					

MELTING POINT (NORMAL)

62 T	72 E	73 E	98*R	125 R	159 T	183 C	243 C	255 C	279 C
286 E	303 T	318 C	363*R	395 T	396*R				

BOILING POINT (NORMAL)

70 E	74 T	98*R	125 R	137 E	138 R	138A E	146 R	183 C	235 T
243 C	255 C	279 C	286 E	287 E	302 T	318 C	328 C	363*R	396*R

TRIPLE POINT

70 E	72 E	73 E	91 T	92 T	98*R	133 E	137 E	138 R	138A E
146 R	182 E	204 C	256A R	270 E	273 E	279 C	301 E	363*R	396*R

CRITICAL POINT

1 T	79 C	82 E	83 E	91 T	98*R	125 R	146 R	147 T	153 C
162 R	196 C	207 C	215A T	234 T	236 E	240 T	243 C	255 C	256 E
265 E	279 C	286 E	287 E	295 C	298 R	308 E	318 C	328 C	344 T
358 T	362 C	363*R	395 T						

PROPERTY INDEX (CONT.)

9. SPECIFIC HEAT

SOLID

4 E 28 T 45 E 72 E 89A T 105 R 119 E 123 E 128 E 132 E
 133 E 135 T 160 T 99 T 277 C 316 T 413 T 415 E 422 T

L I Q U I D

11 E	18 C	45 E	72 E	104 E	119 E	120 T	122 E	125 R	126 T
132 E	133 E	193 E	194 E	212 C	238 R	241 E	257 E	279 C	304 T
317 T	373 T	377 E	383 E	385 E	386 E	394 E	399 T	413 T	

GAS (UP TO 200 DEGREES K)

11 E 118 E 143 T 166 E 168 C 178 C 193 E 194 E 212 C 220 T
224 T 236 E 238 R 257 E 266 C 279 C 304 T 317 T 394 E 395 T
399 T 413 T

GAS (ABOVE 200 DEGREES K)

PROPERTY INDEX (CONT.)

10. VELOCITY OF SOUND

10 T	14 E	15 E	33 T	100 E	104 E	141 E	149 E	151 C	167*R
218 T	219 T	228 E	235A E	236 C	241 E	243 C	266 C	281 E	283 T
352A E	360 E	361 E	376 R	377 E	378 E	379 E	380 E	381 E	383 E
385 E	386 E	390 E	413 T						

SOLID

15 E	33 T	100 E	149 E	413 T
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LIQUID

10 T	104 E	141 E	151 C	241 E	283 T	376 R	377 E	378 E	379 E
380 E	383 E	385 E	386 E	390 E	413 T				

GAS (UP TO 200 DEGREES K)

236 E	266 C	380 E	381 E	413 T
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GAS (ABOVE 200 DEGREES K)

14 E	228 E	235A E	236 E	243 C	266 C	281 E	352A E	360 E	361 E
380 E	413 T								

11. JOULE-THOMSON COEFFICIENTS, INVERSION CURVE

22 T	98*R	135B C	172 T	222 T	223 T	236 E	266 C	269 C	321 E
385 E	401 C								

LIQUID

385 E

GAS (UP TO 200 DEGREES K)

135B C	236 E	266 C	321 E
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GAS (ABOVE 200 DEGREES K)

135B C	172 T	236 E	266 C	269 C	321 E	401 C
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PROPERTY INDEX (CONT.)

12. ENTROPY, ENTHALPY, INTERNAL ENERGY, FUGACITY,
GIBB'S FUNCTION, HELMHOLTZ FUNCTION

22 T	28 T	74 T	78 T	89 C	98*R	112 T	143 T	148 T	159 T
167*R	173 E	174 T	185 C	189*R	204A C	224 T	236 E	237 T	238 R
239 T	244 C	247 C	255 C	266 C	267 C	268 C	269 C	316 T	317 T
323*R	363*R	399 T	401 C	413 T	425 T				

SOLID

28 T	112 T	148 T	316 T	317 T	413 T
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LIQUID

174 T	204A C	238 R	317 T	399 T	413 T
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GAS (UP TO 200 DEGREES K)

78 T	143 T	174 T	224 T	236 E	237 T	238 R	266 C	399 T	413 T
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GAS (ABOVE 200 DEGREES K)

89 C	143 T	173 E	174 T	185 C	224 T	236 E	237 T	255 C	266 C
267 C	268 C	269 C	399 T	401 C	413 T				

PROPERTY INDEX (CONT.)

13. THERMAL CONDUCTIVITY

3 T	5 T	9 C	52 T	57 E	59 E	65 C	76 T	90 C	97 E
105 R	118 E	121 E	136 R	157 R	167*R	168 C	169 T	184 T	186 E
189*R	200 E	201 E	202 E	212 C	213 E	214 E	215 T	230 E	242 R
243 C	244 C	245 T	249A C	254 T	255 C	261 R	261A R	263 E	271 E
284 C	288 E	295 C	296 C	331 R	332 R	333 E	340 R	341 R	342 E
358 T	359 T	372 E	374 E	393 T	396*R	397 C	398 R	406 E	407 E
408 E	411 E	416 E	418 R	419 E	424 T				

SOLID

105 R	215 T	230 E	406 E	407 E	408 E				
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LIQUID

5 T	57 E	184 T	186 E	212 C	214 E	242 R	331 R	374 E	419 E
424 T									

GAS (UP TO 200 DEGREES K)

3 T	52 T	57 E	65 C	76 T	118 E	121 E	136 R	157 R	168 C
169 T	186 E	200 E	212 C	214 E	242 R	244 C	245 T	249A C	254 T
284 C	288 E	295 C	296 C	359 T	374 E	393 T	418 R	419 E	

GAS (ABOVE 200 DEGREES K)

3 T	9 C	52 T	59 E	65 C	90 C	97 E	118 E	121 E	157 R
168 C	169 T	186 E	200 E	201 E	202 E	212 C	213 E	214 E	242 R
243 C	244 C	245 T	249A C	254 T	255 C	261 R	263 E	271 E	284 C
288 E	295 C	296 C	332 R	333 E	340 R	341 R	342 E	359 T	372 E
393 T	398 R	411 E	416 E						

PROPERTY INDEX (CONT.)

14. VISCOSITY

3 T	5 T	6 T	8 T	9 C	17 R	36 T	43 T	44 E	52 T
53 T	65 C	77 T	79 C	93 T	114 T	118 E	121 E	124 C	125 R
126 T	129 E	129A E	134 E	135A E	157 R	167*R	168 C	169 T	170 T
187 E	189*R	191 E	195 T	196 C	206 E	207 E	208 E	209 E	210 C
211 C	212 C	225 E	240 T	242 R	244 C	249 E	251 R	254 T	255 C
259 T	261 R	264 E	271 E	279 C	282 T	288 E	297 R	309 E	310 E
320 E	325 C	326 C	327 E	335 E	336 E	339 E	345 C	358 T	359 T
366 C	366B T	367 R	382 E	387 E	387A E	388 E	389 E	396*R	400 C
403 T	417 E								

LIQUID

5 T	6 T	36 T	44 E	124 C	125 R	126 T	135A E	196 C	211 C
212 C	242 R	251 R	259 T	279 C	282 T	320 E	326 C	327 E	366 C
387 E	417 E								

GAS (UP TO 200 DEGREES K)

3 T	52 T	53 T	65 C	77 T	114 T	118 E	121 E	129 E	129A E
134 E	157 R	168 C	169 T	191 E	195 T	210 C	211 C	212 C	225 E
240 T	242 R	254 T	279 C	288 E	325 C	336 E	345 C	359 T	366 C
382 E	388 E	400 C							

GAS (ABOVE 200 DEGREES K)

3 T	8 T	9 C	17 R	43 T	52 T	53 T	65 C	77 T	79 C
114 T	118 E	121 E	129 E	129A E	134 E	157 R	168 C	169 T	170 T
187 F	191 E	195 T	206 E	207 E	208 E	209 E	211 C	212 C	225 E
240 T	242 R	244 C	249 E	254 T	255 C	261 R	264 E	271 E	279 C
288 E	297 R	309 E	310 E	325 C	335 E	336 E	339 E	345 C	359 T
366 C	367 R	387A E	388 E	389 E	400 C				

PROPERTY INDEX (CONT.)

15. PRANDTL NUMBER

167*R 168 C 212 C 271 E

LIQUID

212 C

GAS (UP TO 200 DEGREES K)

168 C 212 C

GAS (ABOVE 200 DEGREES K)

168 C 212 C 271 E

16. SELF-DIFFUSION COEFFICIENT

3 T 7 C 9 C 37 E 52 T 66 E 170 T 254 T 259 T 358 T
409 E 410 C

SOLID

37 E

LIQUID

66 E 259 T

GAS (UP TO 200 DEGREES K)

3 T 52 T 254 T 409 E

GAS (ABOVE 200 DEGREES K)

3 T 7 C 9 C 52 T 254 T 409 E 410 C

PROPERTY INDEX (CONT.)

17. SURFACE TENSION

10 T 156A T	13 E 189*R	46 T 203 T	61 T 217 T	84 E 231 T	126 T 305 T	139 T 318 C	147 T 344 T	155 T 354 E	156 T 405 T
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18. DIELECTRIC CONSTANT, CLAUSIUS-MOSSOTTI FUNCTION

47 E 190 E	51 C 252 C	54 E 258A T	71 E 260 E	92A T 261A R	116 E 272 E	117 E 275 C	158 E 294 E	188 E 421 E	189*R
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LIQUID

252 C 260 E

GAS (ABOVE 200 DEGREES K)

47 E 272 E	54 E 421 E	71 E 260 E	92A T 258A T	116 E 117 E	117 E 158 E	158 E 188 E	188 E 190 E	190 E 258A T
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19. REFRACTIVE INDEX

29 E 306 E	58 E 350 E	96 E 352 E	116 E 365A E	117 E 365B T	140 E 192 E	192 E 261A R	262 E 262 E	275 C 365B T
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SOLID

192 E 352 E

LIQUID

192 E

GAS (ABOVE 200 DEGREES K)

29 E	58 E	96 E	116 E	117 E	140 E	262 E	365A E	365B T
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PROPERTY INDEX (CONT.)

20. LATTICE CHARACTERISTICS
(CRYSTAL STRUCTURE, LIQUID STRUCTURE, DEBYE TEMPERATURE)

28 T	32 T	33 T	91 T	102 C	105 R	112 T	148 T	161 E	165 T
181 C	199 T	277 C	349 E						

21. GRUNEISEN'S PARAMETER

101 E	106 T
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22. CORRESPONDING STATES

30 C	33 T	38 T	48 C	65 C	67 T	76 T	77 T	78 T	79 C
87 C	88 T	105 R	109 C	110 C	111 T	111A T	112 T	135B C	147 T
151 C	153 C	154 T	171 T	222 T	236 E	237 T	240 T	245 T	246 T
253 C	270 E	277 C	280 C	289 C	291 T	295 C	300 C	312 T	317A T
317B T	318 C	345 C	366B T	375 C	392 T	423 T			

23. INTERMOLECULAR POTENTIAL

12 T	89A T	102 C	170 T	254 T	302 T	303 T	314 T	316 T	353 T
404 T									

24. DOCUMENTS NOT APPEARING IN THE CROSS-INDEX

63 E	64 E	94 T	113 E	276 E	370 E	371 T
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4. BIBLIOGRAPHY OF REFERENCES

(LISTED ALPHABETICALLY BY FIRST AUTHOR)

1. ALDER, B. J. AND JURA, G.
NOTE ON THE CRITICAL TEMPERATURE OF SOLIDS.
J. CHEM. PHYS. VOL. 20, 1491-2 (1952)

CRITICAL TEMPERATURE, VOLUME AND PRESSURE (SOLID)
NEON, HELIUM
THEORETICAL - EQUATION, TABLE (3 VALUES)
LETTER TO EDITOR!
2. ALTAR, W.
A STUDY OF THE LIQUID STATE.
J. CHEM. PHYS. VOL. 5, 577-86 (1937)
C.A. 31, 5639 9

EQUATION OF STATE (LIQUID)(84 TO 148 DEGREES K AND 0 TO 48 ATM)
THEORETICAL - EQUATION, TABLE (11 VALUES)
3. AMDUR, I.
LOW TEMPERATURE TRANSPORT PROPERTIES OF GASES II. NEON, ARGON,
KRYPTON AND XENON.
J. CHEM. PHYS. VOL. 16, 190-4 (1948)
C.A. 42, 3231 G

VISCOSITY, THERMAL CONDUCTIVITY (GAS)(110 TO 290 DEGREES K), SELF-
DIFFUSION COEFFICIENT (GAS)(110 TO 290 DEGREES K AND 1 ATM)
NEON, KRYPTON, XENON
THEORETICAL - EQUATIONS, TABLES (30 VALUES)
4. ANDERSON, A.
UNIV. OF OXFORD PH. D. THESIS (1960)

HEAT CAPACITY (SOLID) (1.2 TO 20 DEGREES K)
ARGON, KRYPTON
EXPERIMENTAL
A COPY WAS NOT AVAILABLE FOR CODING
5. ANDRADE, E. N.
VISCOSITY AND THERMAL CONDUCTIVITY OF LIQUID ARGON.
NATURE VOL. 170, 794 (1952)
C.A. 47,5751 A

VISCOSITY, THERMAL CONDUCTIVITY (LIQUID)
THEORETICAL - EQUATIONS
LETTER TO EDITOR!
6. ANDRADE, E. N. DA C.
THE VISCOSITY OF LIQUIDS.
PROC. PHYS. SOC. (LONDON) VOL. 52, 748-58 (1940)
C.A. 35,945 8

VISCOSITY (LIQUID)(84 DEGREES K)
METHANE, 4 OTHER HYDROCARBONS, 7 METALS, 3 HALOGENS, 7 HALIDES
THEORETICAL - EQUATION, 1 TABULAR VALUE

7. ANDRUSSOW, L.
 UBER DIE DIFFUSION IN GASEN III. BEZIEHUNG ZWISCHEN DER SELBSTDIFFUSION
 UND DER ZAHIGKEIT, DIFFUSION IN EDELGASEN UND WASSERSTOFF.
 (DIFFUSION IN GASES III. RELATION BETWEEN SELF DIFFUSION AND
 VISCOSITY. DIFFUSION IN INERT GASES AND HYDROGEN.)
Z. PHYSIK. CHEM. VOL. 199, 314-30 (1952)
- SELF DIFFUSION COEFFICIENTS (273 DEGREES K AND 1 ATM)
 HELIUM, NEON, ARGON, KRYPTON, XENON
 CORRELATIONS
 GERMAN
8. ANDRUSSOW, L.
 UBER ZAHIGKEIT UND DIFFUSION IN DER GASPHASE. IV. BERECHNUNG DER
 ZAHIGKEIT REINER GASE UND DEREN GEMISCHE TEMPERATURKOEFFIZIENT DER
 ZAHIGKEIT.
 (CONCERNING VISCOSITY AND DIFFUSION IN THE GAS PHASE. IV. CALCULATION
 OF THE VISCOSITY OF PURE GASES AND THEIR MIXTURES. TEMPERATURE
 COEFFICIENT OF VISCOSITY.)
Z. PHYSIK. CHEM. VOL. 199, 330-44 (1952)
C.A. 46, 10733 H
- VISCOSITY (GAS) (273 TO 1273 DEGREES K)
 HYDROGEN, HELIUM, AMMONIA, WATER, OXYGEN, METHANE, NEON, NITROGEN
 THEORETICAL - EQUATIONS, GRAPHS
 GERMAN
9. ANDRUSSOW, L.
 WARMELEITFAHIGKEIT, VISKOSITAT UND DIFFUSION IN DER GASPHASE VIII.
 (SPECIFIC HEAT, VISCOSITY AND DIFFUSION IN THE GAS PHASE VIII.)
Z. ELEKTROCHEM. VOL. 57, 124-30 (1953)
- TEMPERATURE DEPENDENT EXPONENTS FOR EXPRESSIONS FOR VISCOSITY,
 SELF-DIFFUSION COEFFICIENTS, THERMAL CONDUCTIVITY AND SPECIFIC HEAT
 (GAS) (273 TO 1273 DEGREES K)
 HELIUM, NEON, ARGON, KRYPTON, XENON, CARBON DIOXIDE, NITROGEN
 CALCULATIONS - EQUATIONS
 GERMAN
10. AUERBACH, R.
 OBERFLACHENSPANNUNG UND SCHALLGESCHWINDIGKFT.
 (SURFACE TENSION AND VELOCITY OF SOUND.)
EXPERIMENTIA VOL. 4, 473-4 (1948)
- RELATION BETWEEN SURFACE TENSION AND VELOCITY OF SOUND (87 DEGREES K)
 OXYGEN, NITROGEN, HELIUM, HYDROGEN AND 52 OTHER GASES, LIQUIDS AND
 SOLIDS
 THEORETICAL - EQUATION, 1 TABULAR VALUE
 GERMAN
11. BAGATSKII, M. I., VORONEL, A. V. AND GUSAK, V. G.
 MEASUREMENT OF THE SPECIFIC HEAT ($V=CONSTANT$) OF ARGON IN THE
 IMMEDIATE VICINITY OF THE CRITICAL POINT.
SOVIET PHYS. JETP VOL. 16, NO. 2, 517-8 (FEB 1963)
ZHUR. EKSPL. I TEORET. FIZ. VOL. 43, 728-9 (1962) (IN RUSSIAN)
- SPECIFIC HEAT ($V=CONSTANT$) (LIQUID, GAS) (130 TO 156 DEGREES K)
 EXPERIMENTAL - GRAPHS

12. BAHADUR, J. AND MADAN, M. P.
 THE THIRD VIRIAL COEFFICIENT AND THE FORCE BETWEEN MOLECULES.
 PROC. NATL. INST. SCI. INDIA PT. A VOL. 26, 64-8 (1960)
 C.A. 54, 21900 1
 INTERMOLECULAR POTENTIAL (200 TO 700 DEGREES K)
 KRYPTON, CARBON MONOXIDE
 THEORETICAL - TABLES (4 VALUES)
13. BALY, E. C. C. AND DONNAN, F. G.
 THE VARIATION WITH TEMPERATURE OF THE SURFACE ENERGIES AND DENSITIES
 OF LIQUID OXYGEN, NITROGEN, ARGON, AND CARBON MONOXIDE.
 J. CHEM. SOC. (LONDON) VOL. 81, NO. 93, 907-23 (1902)
 SURFACE TENSION, SURFACE ENERGY, SATURATION DENSITY (LIQUID)
 (84 TO 89 DEGREES K)
 OXYGEN, NITROGEN, CARBON MONOXIDE
 EXPERIMENTAL - TABLES (20 VALUES), GRAPHS
 APPARATUS
14. BANCROFT, D.
 MEASUREMENT OF VELOCITY OF SOUND IN GASES.
 AM. J. PHYS. VOL. 24, 355-8 (1956)
 VELOCITY OF SOUND (GAS) (273 AND 373 DEGREES K)
 OXYGEN, CARBON DIOXIDE
 EXPERIMENTAL - ONE TABULAR VALUE, EQUATION
 APPARATUS
15. BARKER, J. R. AND DOBBS, E. R.
 MEASUREMENT OF THE ELASTICITY OF SOLID ARGON WITH AN ULTRASONIC
 INTERFEROMETER.
 PHIL. MAG. VOL. 46, 1069-79 (1955)
 VELOCITY OF SOUND (SOLID) (65 TO 80 DEGREES K), CALCULATED VALUES
 FOR ADIABATIC AND ISOTHERMAL COMPRESSIBILITY, POISSON'S RATIO (SOLID,
 LIQUID) (0 TO 84 DEGREES K)
 EXPERIMENTAL - TABLE OF SMOOTHED VALUES, GRAPHS
 APPARATUS
16. BARKER, J. R., DOBBS, E. R. AND JONES, G. O.
 MEASUREMENT OF THE ELASTICITY OF SOLID ARGON BY ULTRASONIC METHODS.
 PHIL. MAG. (7) VOL. 44, 1182-4 (1953)
 ADIABATIC COMPRESSIBILITY (SOLID) (60 AND 78 DEGREES K)
 TWO CALCULATED VALUES
17. BATEMAN, J. S.
 A REVIEW OF THERMAL CONDUCTIVITY AND VISCOSITY DATA FOR AMMONIA,
 ETHYLENE, NITROGEN, CARBON DIOXIDE AND ARGON.
 P. 169-81 IN PROC. CONG. THERMODYNAMIC AND TRANSPORT PROPERTIES FLUIDS
 LONDON, 1957 (PUB. 1958)
 C.A. 52, 16821 I
 VISCOSITY (GAS) (273 TO 2173 DEGREES K)
 NITROGEN, AMMONIA, CARBON DIOXIDE, ETHYLENE
 REVIEW

18. BAUER, E., MAGAT, M. AND SURDEN, M.
REDUCED TEMPERATURE AND GENERAL PROPERTIES OF PURE LIQUIDS.
TRANS. FARADAY SOC. VOL. 33, 81-7 (1937)
C.A. 31, 2888 1

P-V-T DATA, SPECIFIC HEAT (V=CONSTANT) (83 TO 151 DEGREES K)
18 FLUIDS
CORRELATION - EQUATIONS, GRAPHS
19. BAXTER, G. P.
THIRTY-FOURTH ANNUAL REPORT OF THE COMMITTEE ON ATOMIC WEIGHTS.
DETERMINATIONS PUBLISHED DURING 1927.
J. AM. CHEM. SOC. VOL. 50, NO. 3, 603-17 (1928)

DENSITY (GAS) (273 DEGREES K AND 253 TO 760 MM HG)
NITROGEN, NEON, CHLORINE, ACETYLENE, 7 METALS
COMPILED - DATA FROM 20
20. BAXTER, G. P. AND STARKWEATHER, H. W.
THE DENSITY, COMPRESSIBILITY AND ATOMIC WEIGHT OF ARGON.
PROC. NATL. ACAD. SCI. U. S. VOL. 14, 57-63 (1928)
C.A. 22, 1503 G

DENSITY (GAS) (273 DEGREES K AND 253 TO 760 MM HG)
OXYGEN, NEON, NITROGEN
EXPERIMENTAL - TABLES (23 VALUES)
APPARATUS
21. BAXTER, G. P. AND STARKWEATHER, H. W.
THE DENSITY, COMPRESSIBILITY AND ATOMIC WEIGHT OF ARGON. II.
PROC. NATL. ACAD. SCI. U. S. VOL. 15, 441-4 (1929)
C.A. 23, 4112 9

DENSITY (GAS) (273 DEGREES K AND 253 TO 760 MM HG), ATOMIC WEIGHT
EXPERIMENTAL - TABLE (38 VALUES)
22. BEATTIE, J. A.
THE COMPUTATION OF THE THERMODYNAMIC PROPERTIES OF REAL GASES AND
MIXTURES OF REAL GASES.
CHEM. REV. VOL. 44, 141-92 (1949)

ENTHALPY, ENTROPY, INTERNAL ENERGY, JOULE-THOMSON COEFFICIENT,
HELMHOLTZ FUNCTION, EQUATION OF STATE, HEAT CAPACITIES (V=CONSTANT)
(P=CONSTANT) (GAS)
PURE GASES AND GAS MIXTURES
THEORETICAL - EQUATIONS
23. BEATTIE, J. A.
A NEW EQUATION OF STATE FOR FLUIDS IV. AN EQUATION EXPRESSING THE
VOLUME AS AN EXPLICIT FUNCTION OF THE PRESSURE AND TEMPERATURE.
PROC. NATL. ACAD. SCI. U. S. VOL. 16, 14-9 (1930)

EQUATION OF STATE (GAS)
THEORETICAL - EQUATION

24. BEATTIE, J. A. AND BRIDGEMAN, O. C.
 A NEW EQUATION OF STATE FOR FLUIDS.
 PROC. AM. ACAD. ARTS SCI. VOL. 63, 229-308 (1929)
 C.A. 23,3385 1
- EQUATION OF STATE, DENSITY(GAS)(124 TO 673 DEGREES K AND 0.5 TO 10 ATM)
 HELIUM, NEON, HYDROGEN, NITROGEN, OXYGEN, AIR, CARBON DIOXIDE, METHANE
 THEORETICAL - EQUATIONS, TABLE (200 VALUES)
25. BEATTIE, J. A. AND BRIDGEMAN, O. C.
 A NEW EQUATION OF STATE FOR FLUIDS. II. APPLICATION TO HELIUM, NEON,
 ARGON, HYDROGEN, NITROGEN, OXYGEN, AIR AND METHANE.
 J. AM. CHEM. SOC. VOL. 50, 3133-9 (1928)
- EQUATION OF STATE (GAS)(123 TO 673 DEGREES K)
 HELIUM, NEON, HYDROGEN, NITROGEN, OXYGEN, AIR, CARBON DIOXIDE, METHANE
 THEORETICAL - EQUATION
26. BEATTIE, J. A. AND JULIEN, H. P.
 DENSITIES OF GASEOUS ARGON.
 IND. ENG. CHEM. VOL. 46, NO. 8, 1668-9 (1954)
 C.A. 48,13308 H
- EQUATION OF STATE, DENSITY (GAS)(233 TO 328 DEGREES K AND 0 TO
 3600 LB/SQ IN)
 THEORETICAL - EQUATION, TABLE (360 VALUES)
27. BEATTIE, J. A. AND STOCKMAYER, W. H.
 EQUATIONS OF STATE.
 REPTS. PROGR. IN PHYS. VOL. 7, 195-299 (1940)
- EQUATION OF STATE (GAS), VIRIAL COEFFICIENTS
 HELIUM, NEON, NITROGEN, HYDROGEN, OXYGEN, AIR, METHANE AND 10 OTHERS
 THEORETICAL - EQUATION, TABLE (5 VALUES), GRAPHS
28. BEAUMONT, R. H., CHIHARA, H. AND MORRISON, J. A.
 THERMODYNAMIC PROPERTIES OF KRYPTON. VIBRATIONAL AND OTHER PROPFRTIES
 OF SOLID ARGON AND SOLID KRYPTON.
 PROC. PHYS. SOC. (LONDON) VOL. 78, 1462-81 (1961)
 C.A. 56,10996 G
- DEBYE TEMPERATURES (SOLID)(0 TO 40 DEGREES K), HEAT OF SUBLIMATION
 AT 0 DEGREES K, SPECIFIC HEAT (V=CONSTANT)(SOLID)(0 TO 9 DEGREES K),
 DIFFERENCES BETWEEN SPECIFIC HEATS AT CONSTANT VOLUME AND CONSTANT
 PRESSURE (SOLID)(10 TO 50 DEGREES K), EXPANSIVITY (SOLID)(20 TO
 80 DEGREES K), ENTHALPY OF FORMATION OF VACANCIES (SOLID)(45 TO
 83 DEGREES K), ZERO POINT ENERGY, STATIC LATTICE ENERGY
 KRYPTON
 THEORETICAL - EQUATIONS, GRAPH, TABLES

29. BENNETT, C. E.
 OPTICAL DISPERSION AND MOLAR REFRACTION AT ZERO FREQUENCY FOR
 COMPRESSED NITROGEN, ARGON, AND CARBON DIOXIDE MEASURED AS FUNCTIONS
 OF DENSITY
 PHYS. REV. VOL. 58, 263-6 (AUG 1940)
- REFRACTIVE INDEX, LORENTZ-LORENZ FUNCTION (GAS)(273 AND 303 DEGREES K
 AND 305 TO 1405 CM HG)
 NITROGEN, CARBON DIOXIDE
 EXPERIMENTAL - TABLE (6 VALUES)
30. BENSON, S. W.
 CRITICAL DENSITIES AND RELATED PROPERTIES OF LIQUIDS.
 J. PHYS. AND COLLOID CHEM. VOL. 52, 1060-74 (1948)
- THEORY OF CORRESPONDING STATES, RELATIONS BETWEEN CRITICAL CONSTANTS
 AND FIXED POINT PARAMETERS
 OTHER HEAVY FLUIDS
 COMPILATION
31. BERGEON, R., KIEFER, J. AND VODAR, B.
 EQUATION D'ETAT DE L'ARGON AUX TRES HAUTES PRESSIONS ET LA
 COMPRESSIBILITE DANS L'ONDE DE CHOC.
 (EQUATION OF STATE OF ARGON AT VERY HIGH PRESSURES AND ITS
 COMPRESSIBILITY IN A SHOCK WAVE.)
 J. PHYS. RADIUM VOL. 16, 813-4 (1955)
 C.A. 50,11752 C
- EQUATION OF STATE, SHOCK WAVE PHENOMENA (86 TO 1240 DEGREES K)
 COMPARISON - GRAPHS, TABLE
 FRENCH
32. BERNARDES, N.
 THEORY OF SOLID NEON, ARGON, KRYPTON AND XENON AT 0 DEGREES K.
 PHYS. REV. VOL. 112, NO. 5, 1534-9 (1958)
 ASTIA AD 154 247
- HEAT OF SUBLIMATION, EQUATION OF STATE, ISOTHERMAL COMPRESSIBILITY,
 DEBYE TEMPERATURE (SOLID)(0 DEGREES K)
 NEON, KRYPTON, XENON
 THEORETICAL - EQUATIONS, GRAPHS
33. BERNARDES, N.
 QUANTUM MECHANICAL LAW OF CORRESPONDING STATES FOR VAN DER WAALS
 SOLIDS AT 0 DEGREES K.
 PHYS. REV. VOL. 120, 807-13 (1960)
- VELOCITY OF SOUND (SOLID), ISOTHERMAL COMPRESSIBILITY (SOLID)
 (0 DEGREE K AND 0 TO 20000 ATM) CORRESPONDING STATES, DEBYE
 TEMPERATURE
 NEON, HYDROGEN, XENON, KRYPTON
 THEORETICAL - EQUATIONS , TABLE, GRAPH
34. BIRD, R. B., SPOTZ, E. L. AND HIRSCHFELDER, J. O.
 THE THIRD VIRIAL COEFFICIENT FOR NON-POLAR GASES.
 J. CHEM. PHYS. VOL. 18, NO. 10, 1395-402 (1950)
- SECOND AND THIRD VIRIAL COEFFICIENT (GAS)(273 TO 423 DEGREES K)
 METHANE, NITROGEN, HYDROGEN, HELIUM, DEUTERIUM, ETHANE, CARBON DIOXIDE
 THEORETICAL - EQUATION, TABLE (14 VALUES)

35. BLAGOI, YU. P. AND RUDENKO, N. S.
DENSITY OF LIQUEFIED GAS SOLUTIONS NITROGEN-OXYGEN AND ARGON-OXYGEN.
(IN RUSSIAN)
IZVEST. VYSSHikh UCHEB. ZAVEDENII. FIZ. NO. 6, 145-51 (1958)
- SATURATED DENSITY (LIQUID) (84 TO 90 DEGREES K)
OXYGEN AND NITROGEN
EXPERIMENTAL - TABLE (3 VALUES), GRAPH
APPARATUS
RUSSIAN
36. BLOMGREN, G. E.
PARTITION FUNCTIONS FOR NORMAL LIQUIDS AND MOLTEN SALTS.
ANN. N. Y. ACAD. SCI. VOL. 79, 781-9 (1960)
- PARTITION FUNCTION, VISCOSITY (LIQUID) (85 TO 150 DEGREES K)
METHANE, NITROGEN
THEORETICAL - EQUATIONS, GRAPH
37. BOATO, G.
DIFFUSION COEFFICIENT IN SOLID ARGON AND THERMOCONDUCTIVITY OF SOLID
ARGON.
UNIVERSITA' DI GENOVA, ISTITUTO DI FISICA SPERIMENTALE, ITALY, FINAL
TECH. REPT. (MAR 1963)
DDC AD 405 692
- SELF-DIFFUSION COEFFICIENT (SOLID) (65 TO 83.8 DEGREES K)
EXPERIMENTAL - GRAPH
APPARATUS
38. BOATO, G. AND CASANOVA, G.
A SELF-CONSISTENT SET OF MOLECULAR PARAMETERS FOR NEON, ARGON, KRYPTON
AND XENON.
PHYSICA VOL. 27, 571-89 (1961)
- MOLECULAR PARAMETERS, CORRESPONDING STATES THEORY
NEON, KRYPTON, XENON
THEORETICAL - EQUATIONS, TABLE (3 VALUES), GRAPHS
39. BOATO, G., CASANOVA, G., SCOLES, G. AND VALLAURI, M. E.
VAPOUR PRESSURE OF ISOTOPIC LIQUIDS. III. SOME CORRECTIONS TO
PREVIOUS PAPERS.
NUOVO CIMENTO VOL. 20, NO. 1, 87-93 (1961)
- VAPOR PRESSURE (LIQUID) (84 TO 118 DEGREES K)
NEON
THEORETICAL - EQUATION, TABLE (30 VALUES), GRAPHS
40. BOATO, G., CASANOVA, G. AND VALLAURI, M. E.
VAPOUR PRESSURE OF ISOTOPIC LIQUIDS II. NEON AND ARGON ABOVE BOILING
POINT.
NUOVO CIMENTO VOL. 16, 505-19 (1960)
- VAPOR PRESSURE (LIQUID) (88 TO 118 DEGREES K)
NEON
EXPERIMENTAL - TABLE (25 VALUES), GRAPHS, EQUATION
APPARATUS

41. BOATO, G., SCOLES, G. AND VALLAURI, M. E.
 VAPOUR PRESSURE OF ISOTOPIC SOLIDS BY A STEADY FLOW METHOD. ARGON
 BETWEEN 72 DEGREES K AND TRIPLE POINT.
 NUOVO CIMENTO VOL. 23, 1041-53 (1962)
 C.A. 57,4246 C
- VAPOR PRESSURE (SOLID)(72 TO 83 DEGREES K)
 EXPERIMENTAL - TABLE (25 VALUES), EQUATION, GRAPHS
 APPARATUS
42. BOATO, G., SCOLES, G. AND VALLAURI, M. E.
 VAPOUR PRESSURE OF ISOTOPIC LIQUIDS I. ARGON, NITROGEN, OXYGEN BELOW
 BOILING POINT.
 NUOVO CIMENTO VOL. 14, 735-47 (1959)
- VAPOR PRESSURE (LIQUID)(84 TO 87 DEGREES K)
 NITROGEN, OXYGEN
 EXPERIMENTAL - TABLES (15 VALUES), GRAPHS, EQUATION
 APPARATUS
43. BONILLA, C. F., WANG, S. J. AND WEINER, H.
 THE VISCOSITY OF STEAM, HEAVY-WATER VAPOR, AND ARGON AT ATMOSPHERIC
 PRESSURE UP TO HIGH TEMPERATURES.
 TRANS. ASME VOL. 78, 1285-9 (1956)
- VISCOSITY (GAS)(273 TO 2073 DEGREES K)
 STEAM, HEAVY WATER VAPOR
 THEORETICAL - EQUATION, TABLE (20 VALUES), GRAPH
44. BOON, J. P. AND THOMAES, G.
 THE VISCOSITY OF LIQUEFIED GASES.
 PHYSICA VOL. 29, 208-14 (1963)
- VISCOSITY (LIQUID) (84 TO 89 DEGREES K)
 KRYPTON, OXYGEN, METHANE
 EXPERIMENTAL - GRAPH .
45. BORN, F.
 UBER DAMPFDRUCKMESSUNGEN AN REINEM ARGON.
 (VAPOR PRESSURE MEASUREMENTS ON PURE ARGON.)
 ANN. PHYSIK VOL. 69, 473-504 (1922)
 C.A. 17,1174 9
- VAPOR PRESSURE (SOLID,LIQUID)(65 TO 90 DEGREES K), SPECIFIC
 HEAT (P=CONSTANT)(SOLID,LIQUID)(10 TO 83 DEGREES K)
 EXPERIMENTAL - TABLES (28 VALUES), GRAPH, EQUATIONS
 APPARATUS
 GERMAN
46. BRADLEY, R. S.
 THE MOLECULAR THEORY OF SURFACE ENERGY. THE SURFACE ENERGY OF THE
 LIQUEFIED INERT GASES.
 PHIL. MAG. VOL. 11, 846-9 (1931)
 C.A. 25, 3216 7
- SURFACE ENERGY (84 DEGREES K)
 HELIUM
 THEORETICAL - EQUATION, ONE TABULAR VALUE
-

47. BRAUNMUHL, H-J. VON
 UBER DIE TEMPERATURABHANGIGKEIT DER DIELEKTRIZITATSKONSTANTE EINIGER
 GASE.
 (ON THE TEMPERATURE DEPENDENCE OF THE DIELECTRIC CONSTANT OF SEVERAL
 GASES.)
 PHYSIK. Z. VOL. 28, 141-9 (1927)
 C.A. 21,3782 9
- DIELECTRIC CONSTANT (GAS)(296 TO 383 DEGREES K)
 CARBON MONOXIDE, HYDROGEN, HYDROGEN SULFIDE, HYDROGEN CHLORIDE, CARBON
 DIOXIDE AND NITROUS OXIDE
 EXPERIMENTAL - TABLE (12 VALUES), GRAPH
 GERMAN
48. BREBACH, W.J. AND THODOS, G.
 DENSITIES OF THE LIQUID AND DENSE PHASE REGIONS.
 CHEM. AND ENG. DATA SER. VOL. 3, 338-41 '1958)
 C.A. 53, 12778 C
- P-V-T DATA (GAS)(75 TO 640 DEGREES K AND 48 TO 8000 ATM), LAW OF
 CORRESPONDING STATES
 OXYGEN, CARBON MONOXIDE, METHANE
 CORRELATION - GRAPHS
49. BRIDGMAN, P. W.
 THE MELTING CURVES AND COMPRESSIBILITIES OF NITROGEN AND ARGON.
 PROC. AM. ACAD. ARTS SCI. VOL. 70, 1-32 (1935);
 C.A. 29, 3889 4
- MELTING PRESSURE (84 TO 184 DEGREES K), SATURATED LIQUID AND SOLID
 DENSITIES (106 TO 193 DEGREES K), P-V-T DATA (GAS, LIQUID) (100 TO
 330 DEGREES K AND 700 TO 15000 KG/SQ CM), CALCULATED VALUES OF HEAT
 OF FUSION
 NITROGEN
 EXPERIMENTAL - TABLE (7 VALUES), GRAPH
 APPARATUS
50. BRIDGMAN, P. W.
 THE MELTING PARAMETERS OF NITROGEN AND ARGON UNDER PRESSURE, AND THE
 NATURE OF THE MELTING CURVE.
 PHYS. REV. VOL. 46, 930-3 (1934)
- MELTING PRESSURE, HEAT OF FUSION (SOLID)(84 TO 193 DEGREES K AND
 1 TO 6000 ATM)
 NITROGEN
 EXPERIMENTAL - TABLE (87 VALUES)
51. BRINDLEY, G. W.
 THE DIELECTRIC CONSTANT OF HELIUM AND ARGON.
 PHIL. LIT. SOC., SCI. SECT. VOL. 2, NO. 1, 1-7 (1929)
 C.A. 24, 3426 6
- DIELECTRIC CONSTANT
 HELIUM
 CALCULATED VALUE COMPARED WITH VALUE FROM THEORETICAL EXPRESSION FOR
 H-LIKE ATOMS
 REVIEW - EQUATIONS

52. BROKAW, R. S.
 ALIGNMENT CHARTS FOR TRANSPORT PROPERTIES VISCOSITY, THERMAL CONDUCTIVITY AND DIFFUSION COEFFICIENTS FOR NONPOLAR GASES AND GAS MIXTURES AT LOW DENSITY.
 LEWIS RESEARCH CENTER, CLEVELAND, OHIO.
 NATL. AERONAUT. SPACE ADMIN. TECH. REPT. NO. R-81 (1960) 33 P.
- VISCOSITY, THERMAL CONDUCTIVITY, DIFFUSION COEFFICIENT (VAPOR)(44 TO 22400 DEGREES K)
 AIR, HYDROGEN, NITROGEN, OXYGEN, FLUORINE, NEON AND 45 OTHER GASES THEORETICAL - EQUATIONS, NOMOGRAPHS
53. BROMLEY, L. A. AND WILKE, C. R.
 VISCOSITY BEHAVIOR OF GASES
 IND. ENG. CHEM. VOL. 43, 1641-8 (1951)
- VISCOSITY (GAS)(37 TO 24800 DEGREES K)
 AIR, HELIUM, HYDROGEN, NEON, NITROGEN AND 37 OTHER GASES THEORETICAL - EQUATIONS, NOMOGRAPH
54. BRYAN, A. B.
 THE DIELECTRIC CONSTANTS OF ARGON AND NEON.
 PHYS. REV. VOL. 34, 615-7 (1929)
 C.A. 24,1259 7
- DIELECTRIC CONSTANT (GAS)(297 DEGREES K)
 NEON
 EXPERIMENTAL - ONE TABULAR VALUE
 APPARATUS
55. BUCKINGHAM, R. A.
 THE CLASSICAL EQUATION OF STATE OF GASEOUS HELIUM, NEON AND ARGON.
 PROC. ROY. SOC. (LONDON) VOL. A168, 264-83 (1938)
 C.A. 33, 3646 3
- EQUATION OF STATE (GAS)
 HELIUM, NEON
 THEORETICAL - EQUATION
56. BURCH, R. J. AND CHRISTAKOS, J.
 PHYSICAL PROPERTIES OF MATERIALS AT LOW TEMPERATURE. THE DENSITIES OF LIQUID OXYGEN, NITROGEN, ARGON AND THEIR LIQUID MIXTURES.
 AIR REDUCTION COMPANY CENTRAL RESEARCH DEPARTMENT, CRD-61-153 (DEC 1961) 54 P.
- SATURATED DENSITY (LIQUID)(84 TO 150 DEGREES K AND 1 TO 280 ATM)
 OXYGEN, NITROGEN, AIR
 COMPILATION - DATA FROM 49, 258, 265, 383
57. BURTON, J. T. A. AND ZIEBLAND, H.
 THERMAL CONDUCTIVITIES OF NITROGEN AND ARGON IN THE LIQUID AND GASEOUS STATES.
 GR. BRIT. MINISTRY OF SUPPLY, REPT. NO. ERDE 11/R/57 (OCT 1957)
 ASTIA AD 145 956
- THERMAL CONDUCTIVITY (LIQUID, GAS) (90 TO 200 DEGREES K)
 NITROGEN
 EXPERIMENTAL - TABLE (100 VALUES), GRAPH

58. BURTON, W.
THE REFRACTIVE INDEX AND DISPERSION OF LIGHT IN ARGON AND HELIUM.
PROC. ROY. SOC. (LONDON) VOL. A80, 390-405 (1908)
- REFRACTIVE INDEX (GAS) (273 DEGREES K AND 1 ATM)
HELIUM
EXPERIMENTAL - TABLE (25 VALUES)
APPARATUS
59. CALLEAR, A. B. AND ROBB, J. C.
AN EXPERIMENTAL METHOD OF MEASURING THE THERMAL CONDUCTIVITY OF GASES.
TRANS. FARADAY SOC. VOL. 51, 630-8 (1955)
C.A. 49, 14390 F
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K AND 2 TO 59 MM HG)
AIR, HYDROGEN, CARBON DIOXIDE, WATER
EXPERIMENTAL - TABLE (12 VALUES)
APPARATUS
60. CATH, P. G. AND ONNES, H. K.
SUR LA MESURE DES TEMPERATURES TRES BASSES. XXX. COMPARAISON DES
THERMOMETRES A HELIUM, A ARGON, A NEON, A OXYGENE ET A AZOTE AU
THERMOMETRE A HYDROGENE. CORRECTIONS POUR RAMENER LES INDICATIONS DE
CES THERMOMETRES A L'ECHELLE INTERNATIONALE KELVIN. LE DEUXIEME
COEFFICIENT DU VIRIEL POUR L'HELIOUM, L'ARGON, LE NEON, L'OXYGENE ET
L'AZOTE AU-DESSOUS DE 0 DEGRE C.
(MEASUREMENT OF VERY LOW TEMPERATURES. III. COMPARISON OF HELIUM,
NITROGEN, ARGON, OXYGEN, AND NEON THERMOMETERS WITH THE HYDROGEN
THERMOMETER. CORRECTIONS WHICH WILL REDUCE THE INDICATIONS OF THESE
THERMOMETERS TO THE INTERNATIONAL SCALE OF KELVIN. THE SECOND VIRIAL
COEFFICIENTS FOR HELIUM, ARGON, NEON, OXYGEN AND NITROGEN BELOW 0
DEGREES C.)
COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 156A (1922)
ARCH. NEERL. SCI. (IIIA) VOL. 6, 1-30 (1922)
- SECOND VIRIAL COEFFICIENT (GAS) (83 TO 273 DEGREES K)
OXYGEN, NITROGEN, HYDROGEN
EXPERIMENTAL - TABLE (20 VALUES)
APPARATUS
FRENCH
61. CHANG, S., REE, T., EYRING, H. AND MATZNER, I.
STATISTICAL THEORY OF SURFACE TENSION.
P. 88-92 IN PROGRESS IN INTERNATIONAL RESEARCH ON THERMODYNAMIC AND
TRANSPORT PROPERTIES, 2 ND SYMP. THERMOPHYS. PROPERTIES, PRINCETON,
N. J., 1962, ACADEMIC PRESS, N. Y. (1962)
C.A. 57, 2837 D
- SURFACE TENSION (84 TO 90 DEGREES K)
NITROGEN, METHANE
THEORETICAL - EQUATIONS, TABLE (4 VALUES)

62. CHOWDRI, A. G. AND AULUCK, F. C.
 A NOTE ON FUSION AND THE HOLE-THEORY OF LIQUIDS.
 PHIL. MAG. VOL. 37, 809-11 (1946)
 C.A. 42, 1471 A
 MELTING POINT, LATENT HEAT OF FUSION
 THEORETICAL
63. CHRISTIAN, R. H. AND SHREFFLER, R. G.
 EQUATION OF STATE OF ARGON.
 J. APPL. PHYS. VOL. 25, 1341-2 (1954)
 C.A. 49,19 I
 SHOCK WAVE PHENOMENA
 EXPERIMENTAL - GRAPH
 'LETTER TO EDITOR'
64. CHRISTIAN, R. H. AND YARGER, F. L.
 EQUATION OF STATE OF GASES BY SHOCK-WAVE MEASUREMENTS. I.
 EXPERIMENTAL METHOD AND THE HUGONIOT OF ARGON.
 J. CHEM. PHYS. VOL. 23, 2042-4 (1955)
 C.A. 50,3027 F
 VELOCITY OF SHOCK WAVE (GAS)(294 TO 301 DEGREES K)
 EXPERIMENTAL - TABLE (8 VALUES)
 APPARATUS
65. CINI-CASTAGNOLI, G., PIZZELLA, G., AND RICCI, F. P.
 TRANSPORT PROPERTIES IN THE LIQUID STATE AND THE CORRESPONDING STATE
 PRINCIPLE.
 NUOVO CIMENTO VOL. 11, NO. 3, 466-7 (1959)
 C.A. 54, 1005 F
 VISCOSITY (GAS)(196 TO 223 DEGRES K), THERMAL CONDUCTIVITY (GAS)
 (126 TO 169 DEGREES K), CORRESPONDING STATES
 OXYGEN, NITROGEN, METHANE, CARBON MONOXIDE
 CORRELATION - GRAPHS, DATA FROM 327, 419
66. CINI-CASTAGNOLE, G. AND RICCI, F. P.
 SELF-DIFFUSION IN LIQUID ARGON.
 J. CHEM. PHYS. VOL. 32, NO. 1, 19-20 (1960)
 SELF-DIFFUSION COEFFICIENT (LIQUID)(84.5 DEGREES K AND 70 CM HG)
 EXPERIMENTAL - TABLE (5 VALUES)
67. CLAITOR, L. C.
 THE SECOND VIRIAL COEFFICIENT OF THE MAJOR ATMOSPHERIC GASES AT LOW
 TEMPERATURES.
 TEXAS AGRICULTURAL AND MECHANICAL COLLEGE, COLLEGE STATION, MASTER'S
 THESIS (1948) 83 P.
 SECOND VIRIAL COEFFICIENT (GAS)(87 TO 273 DEGRES K), THEORY OF
 CORRESPONDING STATES
 OXYGEN, NITROGEN, AIR
 THEORETICAL - EQUATION, ONE TABULAR VALUE, GRAPH

68. CLARK, A. L. AND KATZ, L.
 THE RESONANCE METHOD OF MEASURING THE RATIO OF THE SPECIFIC HEATS OF A GAS. C (P=CONSTANT)/ C (V=CONSTANT). PART IV.
 CAN J. RESEARCH VOL. 21A, 1-17 (1943)
- SPECIFIC HEAT RATIO (GAS)(297 DEGREES K AND 1 TO 20 ATM)
 NITROGEN, CARBON DIOXIDE, HYDROGEN, HELIUM
 EXPERIMENTAL - TABLE (11 VALUES), GRAPH
 APPARATUS
69. CLARK, A. M., DIN, F. AND ROBB, J.
 THE LIQUID-VAPOUR EQUILIBRIUM OF THE BINARY SYSTEM ARGON-OXYGEN.
 PROC. ROY. SOC. (LONDON) VOL. A221, 517-34 (1954)
- VAPOR PRESSURE (LIQUID)(90 TO 110 DEGREES K)
 OXYGEN
 EXPERIMENTAL - TABLE (5 VALUES), EQUATION
 APPARATUS
70. CLARK, A. M., DIN, F., ROBB, J., MICHELS, A., WASSENAAR, T. AND ZWIETERING, TH.
 THE VAPOUR PRESSURE OF ARGON.
 PHYSICA VOL. 17, NO. 10, 876-84 (1951)
 116TH PUBLICATION OF THE VAN DER WAALS FUND, AMSTERDAM, NEDERLAND
- VAPOR PRESSURE (SOLID,LIQUID)(70 TO 150 DEGREES K), TRIPLE POINT,
 BOILING POINT
 EXPERIMENTAL - TABLE (38 VALUES), EQUATIONS, GRAPHS
 APPARATUS
71. CLAY, J. AND VAN DER MAESEN, F.
 THE ABSOLUTE DIELECTRIC CONSTANT OF GASES AT PRESSURES OF 0 TO 80 ATM.
 AT 25 DEGREES C.
 PHYSICA VOL. 15, 467-80 (1949)
 C. A. 44, 3318
- DIELECTRIC CONSTANT (GAS)(298 DEGREES K AND 1 TO 80 ATM)
 AIR, HELIUM, CARBON DIOXIDE
 EXPERIMENTAL - TABLE (28 VALUES)
 APPARATUS
72. CLUSIUS, K.
 ATOMWARMEN UND SCHMELZWARMEN VON NEON, ARGON, UND KRYPTON.
 (ATOMIC HEAT AND HEAT OF FUSION OF NEON, ARGON AND KRYPTON.)
 Z. PHYSIK. CHEM. (LEIPZIG) VOL. B31, 459-74 (1926)
 C.A. 30, 3707 7
- SPECIFIC HEAT(P=CONSTANT)(SOLID)(10 TO 80 DEGREES K), SPECIFIC HEAT
 (V=CONSTANT)(SOLID)(10 TO 17 DEGREES K), SPECIFIC HEAT (SATURATION)
 (LIQUID)(86 TO 89 DEGREES K), HEAT OF FUSION (81 TO 84 DEGREES K),
 MELTING POINT, TRIPLE POINT
 NEON, KRYPTON
 EXPERIMENTAL - TABLES (35 VALUES)
 GERMAN
-

73. CLUSIUS, K.
 WIRD DIE SCHMELZSCHARFE DURCH DIE ISOTOPEN BEEINFLUSST. DIE TRIPelpunktsdrucke DER GASE CO, A, N₂O, HCl UND HBr.
 (IS THE SHARPNESS OF THE MELTING POINT INFLUENCED BY THE ISOTOPES.
 THE TRIPLE POINT PRESSURES OF THE GASES CARBON MONOXIDE, ARGON,
 NITROUS OXIDE, HYDROGEN CHLORIDE, AND HYDROGEN BROMIDE.)
 Z. PHYSIK. CHEM. (LEIPZIG) VOL. B49, 1-8 (1941)
- MELTING POINT, TRIPLE POINT
 CARBON MONOXIDE, NITROUS OXIDE, HYDROGEN CHLORIDE AND HYDROGEN BROMIDE
 EXPERIMENTAL
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 GERMAN
74. CLUSIUS, K. AND FRANK, A.
 ZUR ENTROPIE DES ARGONS.
 (ON THE ENTROPY OF ARGON.)
 Z. ELEKTROCHEM. VOL. 49, NO. 4, 308-9 (1943)
 C.A. 37,6508 5
- ENTROPY (87.29 DEGREES K), BOILING POINT
 THEORETICAL - EQUATION
 GERMAN
75. CLUSIUS, K. AND WEIGAND, K.
 DIE SCHMELZKURVEN DER GASE A, KR, X, CH₄, CH₃D, CD₄, C₂H₄, C₂H₆, COS
 UND PH₃ BIS 200 ATM. DRUCK.
 (THE MELTING CURVES OF THE GASES ARGON, KRYPTON, XENON, METHANE,
 DEUTEROMETHANE, ETHANE, ETHENE, CARBONYL SULFIDE AND PHOSPHINE AT
 200 ATMOSPHERES PRESSURE.)
 Z. PHYSIK. CHEM. VOL. B46, 1-37 (1940)
 C.A. 35,947 3
- MELTING PRESSURE (83.78 TO 88.68 DEGREES K AND 3.3 TO 198 ATM)
 KRYPTON, XENON, METHANE AND 6 OTHER HYDROCARBONS
 EXPERIMENTAL TABLE (25 VALUES)
 APPARATUS
 GERMAN
76. CODEGONE, C.
 CONDUTTIVITA TERMICA E GRANDEZZE TERMODINAMICHE DEI GAS E DEI VAPORI.
 (THERMAL CONDUCTIVITY AND MANY THERMODYNAMIC PROPERTIES OF GAS AND
 VAPOR.)
 TERMOTECNICA (MILAN) VOL. 6, 507-12 (1952)
- THERMAL CONDUCTIVITY (GAS)(151 TO 188 DEGREES K), THEORY OF
 CORRESPONDING STATES
 HYDROGEN, NITROGEN, HELIUM, AIR, OXYGEN, METHANE AND 7 OTHER GASES
 THEORETICAL - EQUATION, GRAPHS
 ITALIAN
77. CODEGONE, C.
 LA VISCOSITA DEI GAS E DEI VAPORI.
 (VISCOSITY OF GASES AND VAPORS.)
 RICERCA SCI. VOL. 22, 1416-9 (1952)
- VISCOSITY (GAS)(166 TO 1057 DEGREES K), THEORY OF CORRESPONDING STATES
 HYDROGEN, HELIUM, NITROGEN, AIR, OXYGEN, METHANE AND 8 OTHER GASES
 THEORETICAL - EQUATION, GRAPH
 ITALIAN

78. CODEGONE, C.
 DRUCK UND ENTROPIE DER GESATTIGTEN DAMPFE.
 (PRESSURE AND ENTROPY OF SATURATED VAPORS.)
 ALGEM. WARMETECH. VOL. 9, 58-9 (1959)
 C.A. VOL. 54, 5197 G
- ENTROPY (GAS) (3 TO 151 DEGREES K), CORRESPONDING STATES THEORY
 AIR, HYDROGEN, HELIUM, METHANE AND 6 OTHER COMPOUNDS
 THEORETICAL
 GERMAN
79. COREMANS, J. M. J. AND BEENAKKER, J. J. M.
 THE INFLUENCE OF THE DENSITY ON THE VISCOSITY COEFFICIENT OF GASES.
 PHYSICA VOL. 26, 653-63 (1960)
 C.A. 55,13966 I
- VISCOSITY (GAS)(273 TO 348 DEGREES K), CRITICAL CONSTANTS, THEORY OF
 CORRESPONDING STATES
 NITROGEN, METHANE, HYDROGEN, DEUTERIUM, HELIUM, CARBON DIOXIDE
 CORRELATION - TABLE (10 VALUES), GRAPHS
80. CORNER, J.
 THE CONSTANTS OF THE BEATTIE-BRIDGEMAN EQUATION.
 TRANS. FARADAY SOC. VOL. 37, 358-61 (1941)
 C.A. 35,7252 2
- BEATTIE-BRIDGEMAN EQUATION CONSTANTS
 OXYGEN, HELIUM, NEON, NITROGEN, CARBON MONOXIDE, HYDROGEN AND CARBON
 DIOXIDE
 THEORETICAL - ONE TABULAR VALUE
81. CRAGOE, C. S.
 SLOPES OF PV ISOTHERMS OF HELIUM, NEON, ARGON, HYDROGEN, NITROGEN AND
 OXYGEN AT 0 DEGREES C.
 J. RESEARCH NATL. BUR. STANDARDS VOL. 26, 495-536 (1941)
 C.A. 35,6168 3
- PVT DATA (GAS)(273 DEGREES K AND 1 TO 100 ATM), DENSITY (GAS)
 (273 DEGREES K AND 0.5 TO 1 ATM)
 OXYGEN, HYDROGEN, NITROGEN, HELIUM, NEON
 CORRELATION - EQUATIONS, TABLES, DATA FROM 179, 290
82. CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC GASES AND OF THEIR BINARY MIXTURES. IV.
 REMARKS ON THE PREPARATION OF ARGON. V. VAPOUR PRESSURES ABOVE -140
 DEGREES C., CRITICAL TEMPERATURE AND CRITICAL PRESSURE OF ARGON.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 115 (1910)
 KONINKL. NED. AKAD. WETENSCHAP. PROC. VOL. 13, 54-65 (1910)
- VAPOR PRESSURE (133 TO 151 DEGREES K) CRITICAL TEMPERATURE AND
 PRESSURE
 EXPERIMENTAL - TABLE (8 VALUES), GRAPH
-

83. CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC GASES AND OF THEIR BINARY MIXTURES.VI.
 COEXISTING LIQUID AND VAPOUR DENSITIES OF ARGON, CALCULATION OF THE
 CRITICAL DENSITY OF ARGON.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 118A (1910)
 VERSLAG. GEWONE VERGADER. AFDEL. NATUURK. KONINKL. NED. AKAD.
 WETENSCHAP. VOL. 18, 390-6 (1910)
 PROC. ACAD. SCI. AMSTERDAM VOL. 13, 607-13 (1911)
- SATURATED DENSITY (LIQUID, GAS)(133 TO 148 DEGREES K AND 22 TO
 42 ATM), CRITICAL DENSITY, RECTILINEAR DIAMETER
 EXPERIMENTAL - GRAPH, TABLES (8 VALUES)
84. CROMMELIN, C. A.
 THE CAPILLARY CONSTANTS FOR LIQUID CARBON MONOXIDE AND LIQUID ARGON.
 A CORRECTION.
 PROC. CHEM. SOC. (LONDON) VOL. 30, NO. 429, 248 (1914)
- SURFACE TENSION (84 TO 90 DEGREES K)
 CARBON MONOXIDE
 CORRECTION OF BIBLIOGRAPHY REFERENCE NUMBER 13 - TABLE (7 VALUES)
85. CROMMELIN, C. A.
 ISOHERMEN VAN EENATOMIGE STOFFEN EN HUNNE BINNAIRE MENGSELS. XVI.
 HERNIEUWDE BEPALING VAN DE DAMPSPANNINGEN VAN VAST ARGON TOT -205
 GRAD C.
 (ISOTHERMS FOR MONATOMIC SOLIDS AND FOR THEIR BINARY MIXTURES. XVI.
 NEW DETERMINATIONS OF VAPOR PRESSURE OF SOLID ARGON TO -205 DEGREES
 C.)
 VERSLAG GEWONE VERGADER. AFDEL. NATUURK. KONINKL. NED. AKAD.
 WETENSCHAP. VOL. 22, 1212-5 (1914)
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 140A (1914)
- VAPOR PRESSURE, HEAT OF SUBLIMATION (SOLID)(68 TO 84 DEGREES K)
 EXPERIMENTAL - TABLES (16 VALUES), EQUATION
 DUTCH
86. CROMMELIN, C. A.
 ISOTHERMALS OF MONATOMIC GASES AND THEIR BINARY MIXTURES XV. THE
 VAPOR PRESSURE OF SOLID AND LIQUID ARGON FROM THE CRITICAL POINT DOWN
 TO -206 DEGREES C.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 138C (1913)
 VERSLAG GEWONE VERGADER. WIS- EN NATUURK. AFDEEL. 510-9 (OCT 1913)
 PROC. ACAD. SCI. AMSTERDAM VOL. 16, 477-85 (1913)
- VAPOR PRESSURE (SOLID)(67 TO 84 DEGREES K), VAPOR PRESSURE (LIQUID)
 (84 TO 151 DEGREES K), HEAT OF VAPORIZATION, SATURATED DENSITY
 (LIQUID, GAS)(90 TO 148 DEGREES K)
 EXPERIMENTAL - TABLE (20 VALUES), EQUATION
87. DAMASIUS, G. AND THODOS, G.
 REDUCED STATE CORRELATION FOR THE ENSKOG MODULUS OF SUBSTANCES OF
 SIMPLE MOLECULAR STRUCTURE.
 IND. ENG. CHEM. FUNDAMENTALS VOL. 2, NO. 1, 73-7 (1963)
- DENSITY (LIQUID,GAS)(75 TO 347 DEGREES K AND 4 TO 2400 ATM), ENSKOG
 MODULUS, THEORY OF CORRESPONDING STATES
 CORRELATION - GRAPHS, EQUATION

88. DANON, F. AND PITZER, K. S.
CORRESPONDING STATES THEORY FOR ARGON AND XENON.
J. PHYS. CHEM. VOL. 66, 583-5 (1962)

COMPRESSIBILITY FACTOR (GAS) (158 TO 219 DEGREES K AND 43 TO 435 ATM),
SECOND VIRIAL COEFFICIENT (GAS) (135 TO 468 DEGREES K), CORRESPONDING
STATES PRINCIPLE
XENON
THEORETICAL - TABLE (14 VALUES), GRAPHS
89. DANZE, J.
POUVOIR ENERGETIQUE DES COMBUSTIBLES.
(ENERGY VALUE OF FUELS.)
REV. UNIVERSELLE MINES VOL. 13, NO. 1, 3-17 (1937)
C.A. 31, 2773 5

ENTROPY (GAS) (298 DEGREES K AND 1 ATM)
HELUM, HYDROGEN, OXYGEN, NITROGEN, METHANE
COMPILATION - SOURCE NOT STATED
FRENCH
- 89A. DAVIES, R. O. AND PARKE, S.
A GENERALIZATION OF GRUNEISEN'S THEORY OF SOLIDS AND ITS APPLICATION.
TO SOLID ARGON.
PHIL. MAG. VOL. 4, 341-58 (1959)
C.A. 54, 32

SPECIFIC HEAT (P=CONSTANT), ISOTHERMAL COMPRESSIBILITY (SOLID) (5 TO
80 DEGREES K), GRUNEISEN'S EQUATION OF STATE, INTERATOMIC POTENTIAL
THEORETICAL - EQUATIONS, GRAPHS
90. DAYNES, H. A.
CHAPTER II. THERMAL CONDUCTIVITY OF GASES IN RELATION TO GAS
COMPOSITION.
P. 10-25 IN GAS ANALYSIS BY MEASUREMENT OF THERMAL CONDUCTIVITY,
CAMBRIDGE UNIVERSITY PRESS, LONDON (1933) 349 P.

THERMAL CONDUCTIVITY (GAS) (273 AND 373 DEGREES K)
HELUM, NEON, OXYGEN, NITROGEN, HYDROGEN, CARBON MONOXIDE AND 6 OTHERS
COMPILATION - DATA FROM 121, 118, 340, 342, 397
91. DE BOER, J.
QUANTUM THEORY OF CONDENSED PERMANENT GASES I. THE LAW OF
CORRESPONDING STATES.
PHYSICA VOL. 14, 139-48 (1948)

CRITICAL CONSTANTS, TRIPLE POINT CONSTANTS, DEBYE-TEMPERATURE GRAPH
(SOLID)
HELUM, HYDROGEN, NEON, NITROGEN, KRYPTON, XENON
THEORETICAL - TABLE (6 VALUES)
92. DE BOER, J. AND BLAISSE, B. S.
QUANTUM THEORY OF CONDENSED PERMANENT GASES II. THE SOLID STATE AND
THE MELTING LINE.
PHYSICA VOL. 14, 149-64 (1948)

EQUATION OF STATE (SOLID), MELTING PRESSURE, TRIPLE POINT
HYDROGEN, HELUM, NEON, NITROGEN
THEORETICAL - EQUATION, TABLES (25 VALUES), GRAPHS

- 92A. DE BOER, J., VAN DER MAESEN, F. AND TEN SELDAM, C.A.
THE MOLECULAR POLARISATION OF COMPRESSED NON-POLAR GASES.
PHYSICA VOL. 19, 265-78 (1953)
- CLAUSIUS-MOSSOTTI (GAS) (160 TO 298 DEGREES K)
THEORETICAL - EQUATION, GRAPH
93. DE CARVALHO, H. G.
VARIATION OF VISCOSITY OF GASES WITH TEMPERATURE.
ANALIS. ASSOC. QUIM. BRASIL VOL. 4, NO. 2, 79-82 (1945)
C.A. 40, 4929 5
- VISCOSITY (GAS)
NEON, HELIUM, KRYPTON, XENON
THEORETICAL - EQUATION, CALCULATED VALUES ARE COMPARED WITH OBSERVED
RESULTS
-CODED FROM ABSTRACT-
94. DEITZ, V.
ZERO-POINT ENERGY OF MOLECULAR CRYSTALS.
J. CHEM. PHYS. VOL. 2, 296 (1934)
- ZERO POINT ENERGY
HELUM, NEON, HYDROGEN, NITROGEN, OXYGEN AND 5 OTHFR GASES
THEORETICAL - EQUATION, TABLE (1 VALUE)
'LETTER TO EDITOR'
95. DE SMEIDT, J. AND KEESOM, W. H.
THE CRYSTAL STRUCTURE OF ARGON. RESEARCHES ON THE STRUCTURE OF
NITROGEN AND OXYGEN AT THE TEMPERATURE OF LIQUID HYDROGEN.
COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 178B (1925)
PHYSICA VOL. 5, 344 (1925)
- DENSITY (SOLID) (20 DEGREES K)
NITROGEN, OXYGEN
EXPERIMENTAL - ONE VALUE
96. DE WIJN, H. W. AND HEINEKEN, F. W.
THE LORENTZ-LORENZ FUNCTIONS OF ARGON, NITROGEN, AND CARBON DIOXIDE UP
TO 50 ATMOSPHERES AT A WAVELENGTH OF 12 MM.
PHYSICA VOL. 25, 615-25 (1959)
- LORENTZ-LORENZ FUNCTION (GAS) (295 DEGREES K AND AT DENSITIES FROM
2 TO 45 AMAGATS)
NITROGEN, CARBON DIOXIDE
EXPERIMENTAL - TABLE (30 VALUES), GRAPH
97. DICKINS, B. G.
THE EFFECT OF ACCOMMODATION ON HEAT CONDUCTION THROUGH GASES.
PROC. ROY. SOC. (LONDON) VOL. A143, 517-40 (1934)
C.A. 28, 2604 4
- THERMAL CONDUCTIVITY (GAS) (273 TO 285 DEGREES K), ACCOMMODATION
COEFFICIENT
OXYGEN, HYDROGEN, HELIUM, CARBON MONOXIDE, NITROGEN, AIR, AND 4 OTHERS
EXPERIMENTAL - TABLE (2 VALUES), EQUATION, GRAPH

98. DIN, F.
ARGON.
THERMODYNAMIC FUNCTIONS OF GASES VOL. 2, 146-201 (1956)
BUTTERWORTHS SCIENTIFIC PUBLICATIONS, LONDON
C.A. 51, 11028 C
- DENSITY, SPECIFIC HEAT ($P=CONSTANT$) ($V=CONSTANT$) (SATURATION),
ENTROPY, ENTHALPY, JOULE-THOMSON EFFECT, MELTING AND BOILING
POINTS, LATENT HEATS, CRITICAL CONSTANTS, TRIPLE POINT CONSTANTS
VAPOR PRESSURE
REFERENCE BOOK
99. DITCHBURN, R. W. AND GILMOUR, J. C.
THE VAPOR PRESSURE OF MONATOMIC VAPORS.
REV. MOD. PHYS. VOL. 13, 310-27 (1941)
- VAPOR PRESSURE (SOLID, LIQUID)
HELIUM, NEON, KRYPTON, XENON AND 25 OTHER ELEMENTS
REVIEW OF EXPERIMENTAL TECHNIQUES AND A TABLE OF CONSTANTS FOR A
VAPOR PRESSURE EQUATION
100. DOBBS, E. R.
EQUATION OF STATE AND ELASTICITY OF SOLID ARGON.
J. CHEM. PHYS. VOL. 24, 477-8 (1956)
- VELOCITY OF SOUND, ADIABATIC COMPRESSIBILITY (SOLID)
(10 TO 80 DEGREES K)
EXPERIMENTAL - TABLE (8 VALUES)
'LETTER TO EDITOR'
101. DOBBS, E. R., FIGGINS, B. F., HEASTIE, R., JONES, G. O., WALKER, P. A.
PROPERTIES OF THE CONDENSED INERT GASES.
P. 516-8 IN LOW TEMPERATURE PHYSICS AND CHEMISTRY, PROCEEDINGS 5TH
INTERN. CONF. ON LOW TEMP. PHYS. AND CHEM., MADISON, WISC.,
AUG 26-31, 1957. UNIVERSITY OF WISCONSIN PRESS, MADISON (1958)
- DENSITY, GRUNEISENS PARAMETER (SOLID) (20 TO 80 DEGREES K)
EXPERIMENTAL - GRAPH, TABLE (4 VALUES)
102. DOBBS, E. R., FIGGINS, B. F. AND JONES, G. O.
PROPERTIES OF SOLID ARGON.
NUOVO CIMENTO SUPPL. VOL. 9, NO. 1, 32-5 (1958)
- DENSITY (SOLID) (20 TO 84 DEGREES K), INTERATOMIC POTENTIALS, DEBYE
TEMPERATURE (SOLID)
COMPARISON OF PUBLISHED DATA - GRAPHS
103. DOBBS, E. R., FIGGINS, B. F., JONES, G. O., PIERCEY, D. C., AND
RILEY, D. P.
DENSITY AND EXPANSIVITY OF SOLID ARGON.
NATURE VOL. 178, 483 ONLY (1956)
C.A. 51, 7087 F
- DENSITY, EXPANSIVITY (SOLID) (20 TO 80 DEGREES K)
EXPERIMENTAL - TABLE (8 VALUES), GRAPH
'LETTER TO EDITOR'

104. DOBBS, E. R. AND FINEGOLD, L.
 MEASUREMENT OF THE VELOCITY OF SOUND IN LIQUID ARGON AND LIQUID
 NITROGEN AT HIGH PRESSURES.
 J. ACOUST. SOC. AM. VOL. 32, 1215-20 (1960)
 C.A. 55, 26587 G
- VELOCITY OF SOUND (LIQUID)(87.2 AND 90.1 DEGREES K AND 0 TO 135 ATM),
 DENSITY, RATIO OF SPECIFIC HEATS (LIQUID)(90.1 DEGREES K AND 1.4 TO
 135.3 ATM)
 NITROGEN
 EXPERIMENTAL - TABLE (70 VALUES), EQUATIONS, GRAPHS
 APPARATUS
105. DOBBS, E. R. AND JONES, G. O.
 THEORY AND PROPERTIES OF SOLID ARGON.
 REPTS. PROGR. IN PHYS. VOL. 20, 516-64 (1957)
- MOLECULAR STRUCTURE, LAW OF CORRESPONDING STATES, LATTICE ENERGY,
 THERMAL CONDUCTIVITY, SPECIFIC HEAT ($V=CONSTANT$)($P=CONSTANT$), DENSITY,
 EXPANSIVITY, ISOTHERMAL COMPRESSIBILITY, EQUATION OF STATE (SOLID)
 REVIEW - EQUATIONS, TABLES, GRAPHS
106. DOMB, C. AND ZUCKER, I. J.
 THEORETICAL CALCULATIONS ON SOLID ARGON.
 NATURE VOL. 178, 484 (SEPT 1956)
- THERMAL EXPANSION, GRUNEISEN'S PARAMETER (SOLID)(20 TO 80 DEGREES K)
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107. DUCLAUX, J.
 LES ETATS CORRESPONDANTS DES GAZ.
 (THE CORRESPONDING STATES OF GASES.)
 COMPT. REND. VOL. 252, 1703-4 (1961)
- ISOETHERMAL COMPRESSIBILITY
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108. DUCLAUX, J.
 THEORIES DES GAZ ET EQUATION D'ETAT XII. COMPRESSIBILITE ET
 LIQUEFACTION DE L'ARGON.
 (GAS THEORY AND EQUATION OF STATE XII. COMPRESSIBILITY AND
 LIQUEFACTION OF ARGON.)
 J. PHYS. RADIIUM VOL. 13, 199-205 (1952)
 C.A. 46, 10737 A
- VAPOR PRESSURE (83.93 TO 143.5 DEGREES K)
 THEORETICAL - EQUATIONS, TABLE
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-

109. DUCLAUX, J.
 THEORIE DES GAZ REELS. XIII. COURBE DE CONDENSATION ET DONNEES CRITIQUES.
 (THEORY OF REAL GASES. XIII. CONDENSATION CURVE AND CRITICAL CONSTANTS.)
 J. CHIM. PHYS. VOL. 49, 522-6 (1952)
- CORRESPONDING STATES
 OXYGEN, NITROGEN
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110. DUCLAUX, J.
 THEORIE DES GAZ. XV. ETATS CORRESPONDANTS ARGON, OXYGENE, NEON.
 (THEORY OF GASES XV. CORRESPONDING STATES OF ARGON, OXYGEN, AND NEON.)
 J. CHIM. PHYS. VOL. 50, 113-6 (1953)
- CORRESPONDING STATES (173 TO 373 DEGREES K), 4TH VIRIAL COEFFICIENT
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 CORRELATION - GRAPH
 FRENCH
111. DUCLAUX, J.
 PHYSIQUE DES GAZ. COMPRESSIBILITE DES GAZ EN FONCTION DE LA TEMPERATURE.
 (PHYSICS OF GASES. COMPRESSIBILITIES OF GASES AS FUNCTION OF THE TEMPERATURE.)
 COMPT. REND. VOL. 226, 1124-6 (1948)
 C.A. 43, 3677 H
- CORRESPONDING STATES THEORY
 OXYGEN, NITROGEN
 THEORETICAL
 FRENCH
- 111A. DUCLAUX, J.
 THEORIE DES GAS. XXVIII. LA LOI DES ETATS CORRESPONDANTS.
 (THEORY OF GASES. XXVIII. THE LAW OF CORRESPONDING STATES.)
 J. CHIM. PHYS. VOL. 60, 374-78 (MAR 1963)
- CORRESPONDING STATES THEORY (GAS)
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112. DUGDALE, J. S. AND MACDONALD, D. K. C.
 THE INFLUENCE OF ZERO-POINT ENERGY ON THE THERMODYNAMIC PROPERTIES OF THE LOW BOILING POINT ELEMENTS.
 PHIL. MAG. (7) VOL. 45, 811-17 (AUG 1954)
- INTERNAL ENERGY, DENSITY, DEBYE TEMPERATURE, LAW OF CORRESPONDING STATES (SOLID)
 KRYPTON, NEON, XENON, HYDROGEN, HELIUM, DEUTERIUM
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113. EISENSTEIN, A. AND GINGRICH, N. S.
 THE DIFFRACTION OF X-RAYS BY ARGON IN THE LIQUID, VAPOR AND CRITICAL
 REGIONS.
 PHYS. REV. VOL. 62, 261-70 (1942)
- X-RAY DIFFRACTION STUDY (LIQUID, GAS)(84 TO 168 DEGREES K AND 0.8 TO
 51 ATM)
 EXPERIMENTAL - GRAPHS
114. DUTTA, A.
 BOILING POINT AND VISCOSITY OF GASES.
 NATURE VOL. 152, 445-6 (1943)
 C.A. 38, 286 7
- VISCOSITY (GAS)(87 TO 870 DEGREES K)
 AIR, HYDROGEN, NITROGEN, OXYGEN, CARBON MONOXIDE
 THEORETICAL - EQUATION, GRAPH
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115. EGERTON, A. C.
 NOTE ON VAPOUR PRESSURES OF MONATOMIC SUBSTANCES.
 PHIL. MAG. VOL. 48, 1048-54 (1924)
 C.A. 19, 1221 4
- VAPOR PRESSURE (SOLID)(65 TO 84 DEGREES K)
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116. ESSEN, L. AND FROOME, K. D.
 DIELECTRIC CONSTANT AND REFRACTIVE INDEX OF AIR AND ITS PRINCIPAL
 CONSTITUENTS AT 24,000 MC/S.
 NATURE VOL. 167, 512-3 (1951)
 C.A. 45, 7397 I
- DIELECTRIC CONSTANT, REFRACTIVE INDEX (GAS)(273 DEGREES K AND 1 ATM)
 EXPERIMENTAL - TABLE (2 VALUES), EQUATIONS
117. ESSEN, L. AND FROOME, K. D.
 THE REFRACTIVE INDICES AND DIELECTRIC CONSTANTS OF AIR AND ITS
 PRINCIPAL CONSTITUENTS AT 24,000 MC/S.
 PROC. PHYS. SOC. (LONDON) VOL. B64, 862-75 (1951)
 C. A. 45, 7397 AND C. A. 46, 10728
- REFRACTIVE INDEX, DIELECTRIC CONSTANT (GAS)(273 DEGREES K AND 1 ATM)
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118. EUCKEN, A.
 UBER DAS WARMELEITVERMOGEN, DIE SPEZIFISCHE WARME UND DIE INNERE REIBUNG DER GASE.
 (CONCERNING THE THERMAL CONDUCTIVITY, THE SPECIFIC HEAT AND VISCOSITY OF GASES.)
 PHYSIK. Z. VOL. 14, 324-32 (1913)
- THERMAL CONDUCTIVITY, VISCOSITY, SPECIFIC HEAT ($V=$ CONSTANT)
 (273.1 AND 90.6 DEGREES K)
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119. EUCKEN, A.
 UBER DAS THERMISCHE VERHALTEN EINIGER KOMPRIMIERTER UND KONDENSIERTER GASE BEI TIEFEN TEMPERATUREN.
 (ON THE THERMAL BEHAVIOR OF A COMPRESSED AND CONDENSED GAS AT LOW TEMPERATURES.)
 VERHANDL. DEUT. PHYSIK. GES. VOL. 18, 4-17 (1916)
- SPECIFIC HEAT ($P=$ CONSTANT) ($V=$ CONSTANT) (SOLID, LIQUID) (17 TO 94 DEGREES K), HEAT OF VAPORIZATION
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120. EUCKEN, A.
 UBER DEN WARMEINHALT EINATOMIGER FLUSSIGKEITEN.
 (ON THE HEAT CONTENT OF MONATOMIC FLUIDS.)
 VERHANDL. DEUT. PHYS. GES. VOL. 18, 18-27 (1916)
- SPECIFIC HEAT (LIQUID)
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 THEORETICAL
 GERMAN
121. EUCKEN, A.
 UBER DIE TEMPERATURABHANGIGKEIT DER WARMELEITFAHIGKEIT EINIGER GASE.
 (CONCERNING THE TEMPERATURE DEPENDENCE OF THERMAL CONDUCTIVITY OF SEVERAL GASES.)
 PHYSIK. Z. VOL. 12, 1101-7 (1911)
- THERMAL CONDUCTIVITY, VISCOSITY (GAS) (90 TO 373 DEGREES K)
 HELIUM, HYDROGEN, NITROGEN, OXYGEN, AIR, CARBON DIOXIDE
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122. EUCKEN, A. AND HAUCK, F.
 DIE SPEZIFISCHEN WARMEN P = KONSTANT UND V = KONSTANT EINIGER STOFFE IM FESTEN, FLUSSIGEM UND HYPERKRITISCHEN GEBIET ZWISCHEN 80 UND 320 GRAD ABS.
 (THE SPECIFIC HEATS P = CONSTANT AND V = CONSTANT OF MATERIALS IN SOLID, LIQUID AND HYPERCRITICAL REGION BETWEEN 80 AND 320 DEGREES ABS.)
Z. PHYSIK. CHEM. VOL. 134, 161-77 (1928)
C.A. 22, 4288 6
 SPECIFIC HEAT (V=CONSTANT) (LIQUID) (90 TO 190 DEGREES K), SPECIFIC HEAT (SATURATION) (90 TO 140 DEGREES K), HEAT OF FUSION
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123. EUCKEN, A. AND VEITH, H.
 DIE MOLWARME DES METHANS IN FESTEN METHAN-KRYPTON MISCHUNGEN.
 (THE MOLECULAR HEAT OF METHANE IN SOLID METHANE-KRYPTON MIXTURES.)
Z. PHYSIK. CHEM. (LEIPZIG) VOL. B34, 275-99 (1936)
 SPECIFIC HFAT (V= CONSTANT) (SOLID) (10 TO 80 DEGREES K)
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124. EYRING, H., FULLER, E. J. AND REE, T.
 SIGNIFICANT STRUCTURES FOR TRANSPORT AND THERMODYNAMIC PROPERTIES OF LIQUIDS
AM. CHEM. SOC., DIV. PETROL. CHEM., PREPRINTS, VOL. 5, NO. 3, 73-81 (1960)
C.A. 55, 20398 A
 VISCOSITY (LIQUID)
 METHANE, BENZENE
 CALCULATED VALUES
 -CODED FROM ABSTRACT-
125. EYRING, H., HENDERSON, D., AND REE, T.
 THERMODYNAMIC AND TRANSPORT PROPERTIES OF LIQUIDS.
P. 340-51 IN PROGRESS IN INTERNATIONAL RESEARCH ON THERMODYNAMIC AND TRANSPORT PROPERTIES, 2 ND SYMP. THERMOPHYS. PROPERTIES, PRINCETON, N. J. 1962, ACADEMIC PRESS, N. Y. (1962)
C.A. 57, 2016 D
 SPECIFIC HEAT (V=CONSTANT), VISCOSITY (LIQUID)(84 TO 150 DEGREES K), CRITICAL CONSTANTS, MELTING POINT, BOILING POINT, ENTROPY OF MELTING AND VAPORIZATION
 NEON, KRYPTON, XENON, HYDROGEN, METHANE, CHLORINE, MERCURY, SODIUM
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126. EYRING, H. AND REE, T.
 SIGNIFICANT LIQUID STRUCTURES. VI. THE VACANCY THEORY OF LIQUIDS.
PROC. NATL. ACAD. SCI. VOL. 47, 526-37 (1961)
 SPECIFIC HEAT (V=CONSTANT), ENERGY OF VAPORIZATION, SURFACE TENSION, DENSITY, VISCOSITY (LIQUID)
 THEORETICAL - EQUATIONS, GRAPHS

127. FIESCHI, R. AND TERZI, N.
 QUANTUM EFFECTS IN THE LIQUID STATE BY MEANS OF A PHENOMENOLOGICAL
 CELL MODEL. THE VAPOUR PRESSURE RATIO OF NEON AND ARGON ISOTOPES.
 PHYSICA VOL. 27, 453-64 (1961)
- VAPOR PRESSURE RATIO OF ISOTOPES (SOLID,LIQUID)(83 TO 100 DEGREES K)
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 THEORETICAL - EQUATIONS, TABLE (9 VALUES), GRAPH
128. FIGGINS, B. F.
 THE SPECIFIC HEATS OF SOLID ARGON AND AN EQUIMOLAR ARGON-KRYPTON
 MIXTURE.
 PROC. PHYS. SOC. (LONDON) VOL. 76, 732-6 (1960)
- SPECIFIC HEAT (P=CONSTANT)(SOLID)(16 TO 34 DEGREES K)
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 EXPERIMENTAL - TABLE (16 VALUES), GRAPH
129. FILIPPOVA, G. P. AND ISHKIN, I. P.
 THE VISCOSITY OF AIR, NITROGEN AND ARGON AT LOW TEMPERATURES AND AT
 PRESSURES UP TO 150 ATMOSPHERES.
 INZHENER. FIZ. ZHUR. AKAD. NAUK. BELORUS S.S.R. VOL. 4, NO. 3, 105-9
 (1961)
 TRANSLATION BY FOREIGN TECHNOLOGY DIVISION, AIR FORCE SYSTEMS
 COMMAND, FTD-TT-62-113
 ASTIA AD 281 125
 THE VISCOSITY OF AIR, NITROGEN AND ARGON AT LOW TEMPERATURES AND AT
 C.A. 55, 22967 H
- VISCOSITY (GAS)(90 TO 273 DEGREES K AND 35 TO 150 ATM)
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- 129A. FILIPPOVA, F. P. AND ISHKIN, I. P.
 THE VISCOSITY OF AIR AND ARGON AT TEMPERATURES BETWEEN 0 AND -183
 DEGREES C AND PRESSURES FROM 0 TO 150 ATM. (IN RUSSIAN)
 KISLOROD VOL. 12, NO. 2, 38 (1959)
- VISCOSITY (GAS) (90 TO 273 DEGREES K AND 1 TO 150 ATM)
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 EXPERIMENTAL - TABLE (32 VALUES)
 RUSSIAN
130. FISCHER, F. AND FRÖBOESE, V.
 UBER DIE FRAKTIONIERTE KRYSTALLISATION UND DAS ATOMGEWICHT DES ARGONS.
 (CONCERNING THE FRACTIONAL CRYSTALIZATION AND THE ATOMIC WEIGHT OF
 ARGON.)
 BER. DEUTSCH. CHEM. GES. VOL. 44, 92-104 (1911)
- DENSITY (GAS)(290 TO 293 DEGREES K AND 744 TO 765 MM HG)
 EXPERIMENTAL - TABLE (13 VALUES)
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131. FISCHER, F. AND HAHNEL, O.
 UBER DIE REINDARSTELLUNG VON ARGON UND STICKSTOFF.
 (CONCERNING THE PREPARATION OF ARGON AND NITROGEN.)
 BER. DEUTSCH. CHEM. GES. VOL. 43, 1435-42 (1910)
- DENSITY (GAS) (287 TO 289 DEGREES K AND 1 ATM)
 EXPERIMENTAL
 APPARATUS
 GERMAN
132. FLUBACHER, P., LEADBETTER, A. J. AND MORRISON, J. A.
 THERMODYNAMIC PROPERTIES OF ARGON.
P. 695-7 IN PROCEEDINGS OF THE INTERNATIONAL CONFERENCE ON LOW TEMPERATURE PHYSICS., 7TH, TORONTO, CANADA, 1960 (1961)
 C.A. 56, 9510 F
- HEAT CAPACITY (SOLID, LIQUID) (2 TO 86 DEGREES K)
 HEAT OF VAPORIZATION AND FUSION
 EXPERIMENTAL - GRAPHS
133. FLUBACHER, P., LEADBETTER, A. J. AND MORRISON, J. A.
 A LOW-TEMPERATURE ADIABATIC CALORIMETER FOR CONDENSED SUBSTANCES.
 THERMODYNAMIC PROPERTIES OF ARGON.
 PROC. PHYS. SOC. (LONDON) VOL. 78, 1449-61 (1961)
- SPECIFIC HEAT (SATURATION)(SOLID, LIQUID)(2 TO 86 DEGREES K), VAPOR PRESSURE (SOLID,LIQUID)(66 TO 86 DEGREES K), HEATS OF FUSION AND VAPORIZATION, TRIPLE POINT
 EXPERIMENTAL - TABLES (130 VALUES), GRAPH, EQUATION
134. FLYNN, G. P., HANKS, R. V., LEMAIRE, N. A. AND ROSS, J.
 THE VISCOSITY OF NITROGEN, HELIUM, NEON AND ARGON FROM -78.5 TO 100 DEGREES C BELOW 200 ATMOSPHERES.
 BROWN UNIV., METSLAF CHEM. LAB., PROVIDENCE, R. I., TECH. REPT. NO. BRN-3-P (NOV 1962) NASA N63 11774
- VISCOSITY, DENSITY (GAS) (195 TO 373 DEGREES K AND 29 TO 186 ATM)
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135. FOREMAN, A. J. E. AND LIDIARD, A. B.
 VACANCY CONTRIBUTION TO THE SPECIFIC HEAT OF SOLID ARGON.
 PHIL. MAG. (8) VOL. 8, NO. 85, 97-103 (1963)
- SPECIFIC HEAT (V=CONSTANT)(SOLID)(40 TO 84 DEGREES K)
 THEORETICAL - EQUATION, GRAPH
- 135A. FORSTER, S.
 VISCOSITY MEASUREMENTS IN LIQUID NEON, ARGON, AND NITROGEN.
 CRYOGENICS VOL. 3, NO. 3, 176-77 (SEPT 1963)
- VISCOSITY (LIQUID) (85 TO 113 DEGREES K)
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- 135B. FRANCIS, P. G. AND LUCKHURST, G. R.
 JOULE-TOMSON COEFFICIENTS AND THE PRINCIPLE OF CORRESPONDING STATES.
 TRANS. FARADAY SOC. VOL. 59, 667-72 (1963)
- JOULE-TOMSON COEFFICIENT (GAS) (140 TO 300 DEGREES K), CORRESPONDING
 STATES THEORY
 NITROGEN, OXYGEN, METHANE
 CORRELATION - GRAPH
136. FRANCK, E. U.
 WARMELDUNG IN HOCHVERDICHTETEN GASSEN.
 (THERMAL CONDUCTION IN HIGHLY COMPRESSED GASES.)
 CHEM. ING. TECH. VOL. 25, 238-44 (1953)
 C. A. 47, 7274 G
- THERMAL CONDUCTIVITY (GAS) (93 TO 193 DEGREES K)
 AIR, HELIUM, HYDROGEN, NITROGEN, OXYGEN, METHANE, WATER AND 5 ORGANIC
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137. FRANK, A. AND CLUSIUS, K.
 PRAZISIONSMESSUNGEN DER VERDAMPFUNGSWARME DER GASE O₂, H₂S, PH₃, A,
 COS, CH₄, AND CH₃D.
 (PRECISION MEASUREMENTS OF THE HEAT OF VAPORIZATION OF GASES OXYGEN,
 HYDROGEN SULPHIDE, PHOSPHENE, ARGON, CARBON OXISULPHIDE, METHANE AND
 DEUTEROMETHANE.)
 Z. PHYSIK. CHEM. (LEIPZIG) VOL. B42, 395-421 (1939)
 C.A. 33, 5734 8
- HEAT OF VAPORIZATION, TRIPLE POINT PRESSURE, BOILING POINT
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138. FREEMAN, M. P.
 I. SECOND- AND THIRD-ORDER INTERACTION OF GAS MOLECULES AND A SURFACE,
 II. SOLID SOLUTIONS OF KRYPTON IN XENON AND THE VAPOR PRESSURE OF
 ARGON, KRYPTON AND XENON.
 UNIVERSITY MICROFILMS (ANN ARBOR, MICH.) PUBL. NO. 17124, 151 P.
 DISSERTATION ABSTRACTS VOL. 16, 2039 (1956)
 C.A. 51, 9245 B
- VAPOR PRESSURE, BOILING POINT, TRIPLE POINT
 KRYPTON, XENON
 REVIEW
 -CODED FROM ABSTRACT-
- 138A. FREEMAN, M. P. AND HALSEY, G. D.
 THE SOLID SOLUTION KRYPTON - XENON FROM 90 TO 120 DEGREES K., THE
 VAPOR PRESSURES OF ARGON, KRYPTON AND XENON.
 J. PHYS. CHEM. VOL. 60, 1119-25 (1956)
- TRIPLE POINT TEMPERATURE AND PRESSURE, BOILING POINT
 EXPERIMENTAL - THREE TABULAR VALUES, VAPOR PRESSURE EQUATION

139. FRENKEL, Y. I. AND GUBANOV, A.
 THE RELATION BETWEEN SURFACE TENSION OF AMORPHOUS BODIES AND
 TEMPERATURE. (IN RUSSIAN)
 ZHUR. EKSPTL. I TEORET. FIZ. VOL. 16, 435-50 (1946)
 C.A. 41, 325 F
- SURFACE TENSION (LIQUID)
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 -CODED FROM ABSTRACT-
140. FROOME, K. D.
 THE REFRACTIVE INDICES OF WATER VAPOUR, AIR, OXYGEN, NITROGEN AND
 ARGON AT 72 KMC/S.
 PROC. PHYS. SOC. (LONDON) VOL. 68B, 833-5 (1955)
- REFRACTIVE INDEX (GAS)(273 DEGREES K AND 1 ATM)
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141. GALT, J. K.
 SOUND ABSORPTION AND VELOCITY IN LIQUEFIED ARGON, OXYGEN, NITROGEN
 AND HYDROGEN.
 J. CHEM. PHYS. VOL. 16, 505-7 (1948)
 C.A. 42, 4813 C
- VELOCITY OF SOUND (LIQUID)(85 DEGREES K)
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142. GOBLYN, J. H.
 REFLECTIONS ON STATIC THERMODYNAMICS.
 CHALEUR ET IND. VOL. 11, 31-9 (1930)
 C.A. 24, 1790 1
- P-V DATA
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 -CODED FROM ABSTRACT-
143. GOFF, J. A., GRATCH, S. AND VAN VOORHIS, S. W.
 ZERO PRESSURE THERMODYNAMIC PROPERTIES OF SOME MONATOMIC GASES.
 TRANS. AM. SOC. MECH. ENG. VOL. 72, 725-39 (1950)
- ENTROPY, ENTHALPY, SPECIFIC HEAT ($P=CONSTANT$)(GAS)(100 TO 5000
 DEGREES K AND 0 ATM)
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144. GOMBAS, P. AND KUNVARI, O.
 UBER DIE ZUSTANDSGLEICHUNG DER EDELGASATOME NE, AR, KR, UND XE AM
 ABSOLUTEN NULLPUNKT DER TEMPERATUR.
 (THE EQUATION OF STATE OF THE NOBLE GAS ATOMS NEON, ARGON, KRYPTON AND
 XENON AT ABSOLUTE ZERO.)
 ACTA PHYS. ACAD. SCI. HUNGARY VOL. 5, 339-45 (1955)
 UCRL TRANSLATION NO. 721 (L)
 C.A. 50, 8274 H
- P-V DATA (0 DEGREES ABSOLUTE) FROM AN EQUATION DERIVED BY USE OF THE
 ELECTRON EXCHANGE AND CORRELATION OF A MODIFIED STATISTICAL ATOM MODEL
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 -CODED FROM ABSTRACT-
145. GOODWIN, R. D. AND WEBER, L. A.
 A COMPARISON OF TWO MELTING-PRESSURE EQUATIONS CONSTRAINED TO THE
 TRIPLE POINT USING DATA FOR ELEVEN GASES AND THREE METALS.
 NATL. BUR. STANDARDS TECH. NOTE 183 (OCT 1963)
- MELTING PRESSURE
 HYDROGEN, DEUTERIUM, TRITIUM, NEON, OXYGEN, NITROGEN, AND 7 OTHER
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 THEORETICAL - EQUATIONS, GRAPHS
146. GRATCH, S.
 VAPOR PRESSURE, SPECIFIC VOLUME, P-V-T DATA FOR HYDROGEN, NITROGEN,
 OXYGEN, CARBON MONOXIDE, CARBON DIOXIDE, AIR, HELIUM, ARGON AND
 MERCURY.
 TRANS. AM. SOC. MECH. ENGRS. VOL. 70, 631-40 (1948)
- P-V-T DATA, VAPOR PRESSURE, BOILING POINT, TRIPLE POINT, CRITICAL
 CONSTANTS
 HYDROGEN, NITROGEN, OXYGEN, CARBON MONOXIDE, CARBON DIOXIDE, AIR,
 HELIUM AND MERCURY
 REVIEW - WITH BIBLIOGRAPHY (154 REFERENCES)
147. GUGGENHEIM, E. A.
 THE PRINCIPLE OF CORRESPONDING STATES.
 J. CHEM. PHYS. VOL. 13, 253-61 (1945)
 C.A. 39, 3986 1
- CORRESPONDING STATES PRINCIPLE, VAPOR PRESSURE, ENTROPY OF VAPORIZATION AND FUSION, COEFFICIENT OF THERMAL EXPANSION, SURFACE TENSION, HEAT CAPACITY (P=CONSTANT), BOYLE POINT, CRITICAL CONSTANTS
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148. GUGGENHEIM, E. A. AND MC GLASHAN, M. L.
 INTERACTION BETWEEN ARGON ATOMS.
 PROC. ROY. SOC. (LONDON) VOL. A255, 456-76 (1960)
- CRYSTAL STRUCTURE, ENTROPY, ENTHALPY (SOLID) (20 TO 80 DEGREES K)
 THEORETICAL - EQUATIONS, GRAPHS

149. GUPTILL, E. W., HOYT, C. K. AND ROBINSON, D. K.
 THE VELOCITY AND ATTENUATION OF SOUND IN SOLID ARGON.
 CAN. J. PHYS. VOL. 33, 397 (1955)
 C.A. 49, 12069 D
 VELOCITY OF SOUND (SOLID)(64 AND 78 DEGREES K)
 EXPERIMENTAL - TWO TABULAR VALUES
 'LETTER TO EDITOR'
150. HAGGENMACHER, J. E.
 AN EQUATION FOR THE LINE OF SATURATION OF LIQUIDS AND VAPORS.
 J. AM. CHEM. SOC. VOL. 68, 1123-6 (1946)
 C.A. 40, 5312 8
 VAPOR PRESSURE (112 TO 151 DEGREES K)
 HELIUM, HYDROGEN, NEON, NITROGEN, OXYGEN AND 40 OTHER COMPOUNDS
 THEORETICAL - EQUATION
151. HAMANN, S. D.
 A CORRESPONDING STATES TREATMENT OF THE SPEED OF SOUND IN SIMPLE
 LIQUIDS.
 AUSTRALIAN J. CHEM. VOL. 13, 325-31 (1960)
 C.A. 55, 7951 I
 VELOCITY OF SOUND (LIQUID)(84 TO 87 DEGREES K), CORRESPONDING STATES
 THEORY
 NITROGEN, OXYGEN, METHANE, HELIUM, HYDROGEN
 CORRELATION - GRAPH
152. HAMMICK, D. L.
 LATENT HEATS OF VAPORIZATION AND EXPANSION.
 PHIL. MAG. (6) VOL. 44, 590-4 (1922)
 LATENT HEAT OF VAPORIZATION
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153. HAMRIN, C. E. AND THODOS, G.
 DENSITY. REDUCED STATE CORRELATION FOR THE INERT GASES.
 AM. INST. CHEM. ENGRS. J. VOL. 4, 480-4 (1958)
 C.A. 56, 8025 D
 DENSITY (LIQUID, GAS)(84 TO 800 DEGREES K), CRITICAL CONSTANTS,
 CORRESPONDING STATES
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154. HAMRIN, C. E. AND THODOS, G.
 VAPOR PRESSURES OF THE INERT GASES.
 J. CHEM. PHYS. VOL. 35, 899-902 (1961)
 C.A. 56, 8026 B
 VAPOR PRESSURE (100 TO 150 DEGREES K), THEORY OF CORRESPONDING STATES
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155. HARASIMA, A.
 HEAT OF VAPORIZATION AND SURFACE ENERGY OF SEVERAL LIQUIDS.
 PROC. PHYS. MATH. SOC. JAPAN VOL. 23, 978-83 (1941)
 C.A. 41, 6097 D
 HEAT OF VAPORIZATION, SURFACE ENERGY (SATURATION)
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156. HARASIMA, A.
 CALCULATION OF THE SURFACE ENERGIES OF SEVERAL LIQUIDS.
 PROC. PHYS. MATH. SOC. JAPAN VOL. 22, 825-40 (1940)
 C.A. 35, 2047 I
 SURFACE ENERGY
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- 156A. HARASIMA, A.
 SURFACE TENSION OF LIQUIDS.
 PROC. PHYS-MATH. SOC. JAPAN VOL. 23, 983-91 (1941)
 SURFACE TENSION (LIQUID) (85 TO 90 DEGREES K)
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157. HAWKINS, G. A.
 BRIEF REVIEW OF AVAILABLE DATA ON THE DYNAMIC VISCOSITY AND THERMAL CONDUCTIVITY FOR TWELVE GASES.
 TRANS. AM. SOC. MECH. ENGRS. VOL. 70, 655-9 (1948)
 VISCOSITY, THERMAL CONDUCTIVITY (GAS)(90 TO 1973 DEGREES K)
 METHANE, CARBON MONOXIDE, HELIUM, HYDROGEN, NITROGEN, OXYGEN AND 5 OTHER GASES
 REVIEW
158. HECTOR, L. G. AND WOERNLEY, D. L.
 THE DIELECTRIC CONSTANTS OF EIGHT GASES.
 PHYS. REV. VOL. 69, 101-5 (1946)
 C.A. 40, 2366 7
 DIELECTRIC CONSTANT (GAS)(297 DEGREES K AND 1 ATM)
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159. HENKEL, J. H.
 MELTING POINT OF CRYSTALLINE ARGON.
 ABSTRACTED IN BULL. AM. PHYS. SOC. VOL. 1, 258 (1956) PAPER A 10.
 C.A. 52, 12473 I
 MELTING POINT, FREE ENERGY (HELMHOLTZ FUNCTION)
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 -CODED FROM ABSTRACT-

160. HENKEL, J. H.
 EQUATION OF STATE AND THE THERMAL DEPENDENCE OF THE ELASTIC
 COEFFICIENTS OF CRYSTALLINE ARGON.
 J. CHEM. PHYS. VOL. 23, NO. 4, 681-7 (1955)
- EQUATION OF STATE, SPECIFIC HEATS ($V=CONSTANT$) ($P=CONSTANT$), ELASTIC
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161. HENSHAW, D. G.
 ATOMIC DISTRIBUTION IN LIQUID AND SOLID NEON AND SOLID ARGON BY
 NEUTRON DIFFRACTION.
 PHYS. REV. VOL. 111, 1470-5 (1958)
- ATOMIC DISTRIBUTION (SOLID) (4.2 DEGREES K) CRYSTAL STRUCTURE
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162. HERZ, W.
 ZUR KENNTNIS DER SATTIGUNGSDRUCKE.
 (INFORMATION ON SATURATION PRESSURE.)
 Z. ELEKTROCHEM. VOL. 36, 300-1 (1930)
 C.A. 24, 3935 7
- CRITICAL CONSTANTS
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163. HERZ, W.
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 Z. ELEKTROCHEM. VOL. 33, 348-9 (1927)
 C.A. 21, 3505 7
- SATURATED DENSITY (GAS) (90 TO 148 DEGREES K)
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164. HERZFELD, K. F.
 THE SECOND VIRIAL COEFFICIENT OF ARGON.
 PHYS. REV. VOL. 52, 374 (1937)
 C.A. 31, 7301 4
- SECOND VIRIAL COEFFICIENT (GAS) (173 TO 673 DEGREES K)
 THEORETICAL - EQUATION
165. HERZFELD, K. F. AND MAYER, M. G.
 ON THE THEORY OF FUSION.
 PHYS. REV. VOL. 46, NO. 11, 995-1001 (1934)
- EQUATION OF STATE (SOLID), CRYSTAL STRUCTURE
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 THEORETICAL - EQUATIONS

166. HEUSE, W.
 DIE SPEZIFISCHE WARME VON ARGON UND EINIGEN MEHRATOMIGEN GASEN.
 (THE SPECIFIC HEAT OF ARGON AND SOME POLYATOMIC GASES.)
 ANN. PHYSIK VOL. 59, 86-94 (1919)
- SPECIFIC HEAT ($P=$ CONSTANT) (GAS) (93 TO 291 DEGREES K)
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 (CONCERNING A NEW DETERMINATION OF THE CRITICAL VALUE OF THE
 EXPANSION AND VOLTAGE COEFFICIENTS OF HELIUM, HYDROGEN AND NITROGEN.)
 ANN. PHYSIK (5) VOL. 2, 1012-30 (1929)
- P-V-T DATA (GAS) (273 TO 373 DEGREES K)
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167. HILSENARTH, J., BECKETT, C. W., BENEDICT, W. S., FANO, L., HOGE, H.J.,
 MASI, J. F., NUTTALL, R. L., TOULOUKIAN, Y. S. AND WOOLLEY, H. W.
 TABLES OF THERMAL PROPERTIES OF GASES.
 NATL. BUR. STANDARDS CIRC. NO. 564 (1955)
 REPRINTED AS TABLES OF THERMODYNAMIC AND TRANSPORT PROPERTIES OF AIR,
 ARGON, CARBON DIOXIDE, CARBON MONOXIDE, HYDROGEN, NITROGEN, OXYGEN AND
 STEAM.
 PERGAMON PRESS, OXFORD (1960) 478 P.-
- COMPRESSIBILITY FACTOR, DENSITY (70 TO 5000 DEGREES K AND 0.01 TO
 100 ATM), SPECIFIC HEAT ($P=$ CONSTANT, ENTHALPY, ENTROPY, SOUND VELOCITY
 (100 TO 3000 DEGREES K AND 0.01 TO 100 ATM), SPECIFIC HEAT RATIO (100
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 REFERENCE BOOK - TABLES
168. HILSENARTH, J. AND TOULOUKIAN, Y. S.
 THE VISCOSITY, THERMAL CONDUCTIVITY, AND PRANDTL NUMBER FOR AIR,
 OXYGEN, NITROGEN, NITRIC OXIDE, HYDROGEN, CARBON MONOXIDE,
 CARBON DIOXIDE, WATER, HELIUM AND ARGON.
 TRANS. AM. SOC. MECH. ENGRS. VOL. 76, 967-85 (1954)
 C.A. 48, 11135 C
- VISCOSITY, THERMAL CONDUCTIVITY, PRANDTL NUMBER, SPECIFIC HEAT
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169. HIRSCHFELDER, J. O., BIRD, R. B. AND SPOTZ, E. L.
THE TRANSPORT PROPERTIES FOR NON-POLAR GASES.
J. CHEM. PHYS. VOL. 16, NO. 10, 968-81 (1948)
- VISCOSITY, THERMAL CONDUCTIVITY (GAS) (80 TO 1500 DEGREES K)
AIR, HYDROGEN, NITROGEN, OXYGEN, METHANE, NEON, HELIUM, CARBON
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170. HIRSCHFELDER, J. O., BIRD, R. B. AND SPOTZ, E. L.
THE TRANSPORT PROPERTIES OF GASES AND GASEOUS MIXTURES. II.
CHEM. REV. VOL. 44, 205-31 (1949)
- VISCOSITY (GAS) (293 TO 523 DEGREES K), SELF-DIFFUSION COEFFICIENTS,
FORCE CONSTANTS
AIR, HYDROGEN, NITROGEN, METHANE, OXYGEN, CARBON MONOXIDE, HELIUM,
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171. HIRSCHFELDER, J. O., BUEHLER, R. J., MC GEE, H. A. AND SUTTON, J. R.
A GENERALIZED EQUATION OF STATE FOR BOTH GASES AND LIQUIDS. I.
WISCONSIN, UNIV. NAVAL RESEARCH LAB., DEPT. OF CHEMISTRY, TECH. REPT.
WIS-OOR-15 (OCTOBER 5, 1956)
IND. ENG. CHEM. VOL. 50, 375-85 (1958)
- EQUATION OF STATE, DENSITY, P-V-T DATA (GAS) (273 TO 423 DEGREES K)
CORRESPONDING STATES PRINCIPLE
NITROGEN AND 9 OTHER LIQUIDS AND GASES
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172. HIRSCHFELDER, J. O. AND ROSEVEARE, W. E.
INTERMOLECULAR FORCES AND THE PROPERTIES OF GASES.
J. PHYS. CHEM. VOL. 43, 15-35 (1939)
- EQUATION OF STATE, JOULE-THOMSON COEFFICIENT (273 TO 573 DEGREES K
AND 0 TO 200 ATM)
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173. HO, L. T., VANDERSLICE, J. T., FALLON, R. J., SEIGEL, A. E., AND
SLAWSKY, Z. I.
DETERMINATION OF ISENTROPIC PRESSURE-DENSITY CURVES FOR ARGON FROM A
RAPID DYNAMIC PROCESS.
PHYS. FLUIDS VOL. 4, 947-54 (1961)
- ENTROPY, DENSITY (GAS) (300 DEGREES K AND 403 TO 903 ATM)
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174. HO, L. T., VANDERSLICE, J. T., FALLON, R. J., SEIGEL, A. E., AND SLAWSKY, Z. I.
 PRESSURE-DENSITY ISENTROPES FOR ARGON AT TEMPERATURES BETWEEN 150 AND -140 DEGREES C AND AT PRESSURES UP TO 1950 ATMOSPHERES.
PHYS. FLUIDS VOL. 4, 784-6 (1961)
 C.A. 56, 76 I
 ENTROPY, DENSITY (GAS,LIQUID)(133 TO 423 DEGREES K AND 5 TO 2105 ATM)
 THEORETICAL - EQUATIONS, TABLE (380 VALUES)
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175. HOLBORN, L. AND OTTO, J.
 UBER DIE ISOTHERMEN VON HELIUM, STICKSTOFF UND ARGON UNTERHALB 0 GRAD. C.
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Z. PHYSIK. VOL. 30, 320-8 (1924)
 P-V-T DATA (GAS)(173 TO 223 DEGREES K AND 20 TO 60 ATM)
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176. HOLBORN, L. AND OTTO, J.
 UBER DIE ISOTHERMEN EINIGER GASE BIS 400 GRAD UND IHRE BEDEUTUNG FUR DAS GASTHERMOMETER.
 (ON ISOTHERMS OF SEVERAL GASES UP TO 400 DEGREES AND THEIR IMPORTANCE FOR THE GAS THERMOMETER.)
Z. PHYSIK VOL. 23, 77-94 (1924)
 P-V-T DATA (GAS)(300 TO 400 DEGREES K AND 0 TO 80 M HG)
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177. HOLBORN, L. AND OTTO, J.
 UBER DIE ISOTHERMEN EINIGER GASE ZWISCHEN 400 UND -183 GRAD.
 (ON THE ISOTHERMS OF VARIOUS GASES BETWEEN 400 AND -183 DEGREES C.)
Z. PHYSIK. VOL. 33, 1-11 (1925)
WISS-ABHANDL. PHYSIK. TECH. REICHSANSTALT 9(1), 163-73 (1942)
 P-V-T DATA (GAS)(173 TO 673 DEGREES K AND 0 TO 80 M HG)
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178. HOLBORN, L., SCHEEL, K. AND HENNING, F.
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 (HEAT TABLES - RESULTS OF THE THERMAL INVESTIGATIONS OF PHYSIKALISCH-TECHNISCHEN REICHSANSTALT.)
VIEWEG AND SON, BRAUNSCHWEIG (1919) 72 P.
 P-V-T DATA(GAS)(273 TO 573 DEGREES K AND 0 TO 75 M HG), SPECIFIC HEAT (P=CONSTANT)(V=CONSTANT)(93 TO 288 DEGREES K AND 1' ATM), SECOND AND THIRD VIRIAL COEFFICIENTS (0 TO 200 DEGREES K)
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179. HOLBORN, L. AND SCHULTZE, H.
 UBER DIE DRUCKWAGE UND DIE ISOTHERMEN VON LUFT, ARGON UND HELIUM
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 (CONCERNING THE PRESSURE BALANCE AND THE ISOTHERMS OF AIR, ARGON AND
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 ANN. PHYSIK VOL. 47, 1089-1111 (1915)
- P-V-T DATA (GAS) (273 TO 473 DEGREES K AND 14 TO 75 M HG)
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180. HOLBORN, L., SCHULTZ, H. AND OTTO, J.
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 P. 152 IN HANDBUCH DER EXPERIMENTAL PHYSIK VOL. 8, PART 2, (1929)
- P-V DATA (GAS) (173 TO 673 DEGREES K AND 19 TO 187 ATM)
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181. HOLM, M. W.
 DEBYE CHARACTERISTIC TEMPERATURES. TABLE AND BIBLIOGRAPHY
 PHILLIPS PETROL. CO., IDAHO FALLS, IDAHO, REPT. NO. IDU-16399
 (AUG 1957)
 ASTIA AD 144 739
- DEBYE CHARACTERISTIC TEMPERATURE (85 DEGREES K)
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182. HOLST, G. AND HAMBURGER, L.
 INVESTIGATION OF THE EQUILIBRIUM LIQUID-VAPOUR OF THE SYSTEM
 ARGON-NITROGEN.
 PROC. ACAD. SCI. AMSTERDAM VOL. 18, 872-94 (1916)
 Z. PHYSIK. CHEM. (LEIPZIG) VOL. 91, 513-47 (1916)
- VAPOR PRESSURE (83 TO 90 DEGREES K), TRIPLE POINT
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183. HONIG, R. E. AND HOOK, H. O.
 VAPOR PRESSURE DATA FOR SOME COMMON GASES.
 RCA REV. VOL. 21, 360-8 (SEPT 1960)
 C.A. 55, 5066 H
- VAPOR PRESSURE (82 TO 88 DEGREES K AND 0 TO 1000 MM HG), MELTING AND
 BOILING POINTS, HEATS OF FUSION AND VAPORIZATION
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184. HORROCKS, J. K. AND MC LAUGHLIN, E.
THERMAL CONDUCTIVITY OF SIMPLE MOLECULES IN THE CONDENSED STATE.
TRANS. FARADAY SOC. VOL. 56, 206-12 (1960)

THERMAL CONDUCTIVITY (LIQUID)(84 AND 87 DEGREES K)
NITROGEN, CARBON MONOXIDE, METHANE, BENZENE, CARBON TETRACHLORIDE
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185. HUFF, V. N., GORDON, S. AND MORRELL, V. E.
GENERAL METHOD AND THERMODYNAMIC TABLES FOR COMPUTATION OF
EQUILIBRIUM. COMPOSITION AND TEMPERATURE OF CHEMICAL REACTIONS.
NATIONAL ADVISORY COMMITTEE FOR AERONAUTICS REPT. NO. 1037 (1951) 57 P

SPECIFIC HEAT (P=CONSTANT), ENTROPY, ENTHALPY (IDEAL GAS)(298 TO
6000 DEGREES K)
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186. IKENBERRY, L. D. AND RICE, S. A.
KINETIC THEORY OF DENSE FLUIDS. XIV. EXPERIMENTAL AND THEORETICAL
STUDIES OF THERMAL CONDUCTIVITY IN LIQUID ARGON, KRYPTON, XENON AND
METHANE.
J. CHEM. PHYS. VOL. 39, 1561-71 (1963)
C.A. 59, 8136 B

THERMAL CONDUCTIVITY(LIQUID,GAS)(90 TO 234 DEGREES K AND 1 TO 500 ATM)
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187. ISHIDA, Y.
DETERMINATION OF VISCOSITIES AND OF THE STOKES-MILLIKAN LAW CONSTANT
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PHYS. REV. VOL. 21, 550-63 (1923)
C.A. 18, 2986 2

VISCOSITY (GAS)(296 DEGREES K)
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188. JELATIS, J. G.
MEASUREMENTS OF DIELECTRIC CONSTANT AND DIPOLE MOMENT OF GASES BY THE
BEAT-FREQUENCY METHOD.
J. APPL. PHYS. VOL. 19, 419-25 (1948)
C.A. 42, 6593 G

DIELECTRIC CONSTANT (GAS)(273 DEGREES K AND 1 ATM)
OXYGEN, HELIUM, NEON, ETHYL CHLORIDE AND CARBON OXYSULFIDE
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189. JOHNSON, V. J.
 A COMPENDIUM OF THE PROPERTIES OF MATERIALS AT LOW TEMPERATURE.
 PART 1. PROPERTIES OF FLUIDS.
 NATL. BUR. STANDARDS, CRYOGENIC ENG. LAB., WADD TECH. REPT. 60-56
 (1960) 489 P.
- DENSITY, EXPANSIVITY, THERMAL CONDUCTIVITY, SPECIFIC HEAT, ENTHALPY,
 DIELECTRIC CONSTANTS, SURFACE TENSION, VISCOSITY, PHASE TRANSITION
 HEATS
 HELIUM, HYDROGEN, NEON, NITROGEN, OXYGEN, AIR, FLUORINE, METHANE,
 CARBON MONOXIDE
 REFERENCE WORK
190. JOHNSTON, D. R., OUDEMANS, G. J. AND COLE, R. H.
 DIELECTRIC CONSTANTS OF IMPERFECT GASES. I. HELIUM, ARGON, NITROGEN,
 AND METHANE.
 J. CHEM. PHYS. VOL. 33, NO. 5, 1310-7 (1960)
 C. A. 55, 15032 (1961)
- CLAUSIUS-MOSSOTTI FUNCTION(GAS)(242 TO 306 DEGREES K AND 1 TO 100 ATM)
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 EXPERIMENTAL - TABLE OF CONSTANTS (6 VALUES), GRAPHS, EQUATION
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191. JOHNSTON, H. L. AND GRILLY, E. R.
 VISCOSITIES OF CARBON MONOXIDE, HELIUM, NEON, AND ARGON BETWEEN 80 AND
 300 DEGREES K. COEFFICIENTS OF VISCOSITY.
 J. PHYS. CHEM. VOL. 46, 948-63 (1942)
- VISCOSITY (GAS)(80 TO 296 DEGREES K)
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 EXPERIMENTAL - TABLE (45 VALUES)
192. JONES, G. O. AND SMITH, B. L.
 THE REFRACTIVE INDICES OF LIQUID AND SOLID ARGON.
 PHIL. MAG. VOL. 5, 355-8 (1960)
 C.A. 55, 15027 G
- REFRACTIVE INDEX (SOLID, LIQUID)(74 TO 95 DEGREES K), LORENTZ-LORENZ
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 EXPERIMENTAL - GRAPH, EQUATION
193. JONES, G. O. AND WALKER, P. A.
 THE DIAGRAM OF STATE AND SPECIFIC HEATS OF LIQUID ARGON.
 CONFERENCE DE PHYSIQUE DES BASSES TEMPERATURES, PARIS, 2-8 (SEPT 1955)
- P-V-T DATA (SOLID, LIQUID, GAS)(80 TO 150 DEGREES K AND 0 TO 500 ATM),
 SPECIFIC HEATS (P=CONSTANT)(V=CONSTANT)(LIQUID, GAS)(110 TO
 170 DEGREES K AND 10 TO 90 ATM)
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-

194. JONES, G. O. AND WALKER, P. A.
 SPECIFIC HEATS OF FLUID ARGON NEAR THE CRITICAL POINT.
 PROC. PHYS. SOC. (LONDON) VOL. B69, 1348-9 (1956)
 C.A. 51, 14350 G
 SPECIFIC HEATS (P=CONSTANT)(LIQUID,GAS)(130 TO 180 DEGREES K AND
 40 TO 90 ATM)
 EXPERIMENTAL - GRAPHS
 'LETTER TO EDITOR'
195. JONES, J. E.
 ON THE DETERMINATION OF MOLECULAR FIELDS. I. FROM THE VARIATION OF THE
 VISCOSITY OF A GAS WITH TEMPERATURE. II. FROM THE EQUATION OF STATE
 OF A GAS.
 PROC. ROY. SOC. (LONDON) VOL. A106, 441-76 (1924)
 VISCOSITY (GAS)(90 TO 456 DEGREES K), SECOND VIRIAL COEFFICIENT
 CARBON DIOXIDE
 THEORETICAL - EQUATIONS, TABLE (40 VALUES)
196. JOSSI, J. A., STIEL, L. I. AND THODOS, G.
 THE VISCOSITY OF PURE SUBSTANCES IN THE DENSE GASEOUS AND LIQUID
 PHASES.
 A. I. CH. E. JOURNAL VOL. 8, NO. 1, 59-63 (1962)
 VISCOSITY (LIQUID, GAS) (AT DENSITIES FROM 0.01 TO 1.5 G/CC),
 CRITICAL CONSTANTS
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 CORRELATION - GRAPH, EQUATIONS
197. KAGANER, M. G.
 ISOTHERMAL JOULE EFFECT AND THE EQUATION OF STATE OF GASES WITH
 NON-POLAR MOLECULES. (IN RUSSIAN)
 ZHUR. FIZ. KHIM. VOL. 30, 2691-704 (1956)
 VIRIAL EQUATIONS OF STATE, CLASSICAL EVALUATION OF VIRIAL COEFFICIENTS
 AND MOLECULAR PARAMETER EVALUATION
 THEORETICAL - GRAPHS, TABLES
 RUSSIAN
198. KALININ, V. A.
 EQUATION OF STATE OF SOLID ARGON.
 SOVIET PHYS. JETP VOL. 7, 158-9 (1958)
 ZHUR. EKSPTL. I TEORET. FIZ. VOL. 34, 229-30 (1958)
 C.A. 52, 7805 C
 EQUATION OF STATE (SOLID)(65 AND 80 DEGREES K)
 THEORETICAL - EQUATION, GRAPH
199. KANE, G.
 THE EQUATION OF STATE OF FROZEN NEON, ARGON, KRYPTON, AND XENON.
 J. CHEM. PHYS. VOL. 7, 603-13 (1939)
 C.A. 33, 8067 5
 EQUATION OF STATE, DEBYE FUNCTION, SPECIFIC HEAT (P=CONSTANT)(SOLID)
 (20 TO 70 DEGREES K), HEAT OF SUBLIMATION
 NEON, KRYPTON, XENON
 THEORETICAL - EQUATIONS, TABLES (33 VALUES), GRAPHS

200. KANNULUIK, W. G. AND CARMAN, E. H.
 THE THERMAL CONDUCTIVITY OF RARE GASES.
 PROC. PHYS. SOC. (LONDON) VOL. 365, 701-9 (1952)
 C.A. 46, 10736 I
- THERMAL CONDUCTIVITY (GAS) (90 TO 579 DEGREES K)
 HELIUM, NEON, KRYPTON, XENON
 EXPERIMENTAL - TABLE (6 VALUES)
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201. KANNULUIK, W. G. AND DONALD, H. B.
 THE PRESSURE DEPENDENCE OF THE THERMAL CONDUCTIVITY OF POLYATOMIC
 GASES AT 0 DEGREES C.
 AUSTRALIAN J. SCI. RESEARCH VOL. A3, 417-27 (1950)
- THERMAL CONDUCTIVITY (GAS) (275 DEGREES K AND 1 TO 47 ATM)
 NEON, AIR, METHANE, CARBON DIOXIDE, NITROUS OXIDE
 EXPERIMENTAL - TABLE (9 VALUES), GRAPH
 APPARATUS
202. KANNULUIK, W. G. AND MARTIN, L. H.
 THE THERMAL CONDUCTIVITY OF SOME GASES AT 0 DEGREE.
 PROC. ROY. SOC. (LONDON) VOL. A144, 496-513 (1934)
 C.A. 28, 4656 2
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K)
 HELIUM, NEON, HYDROGEN, AIR, OXYGEN, CARBON MONOXIDE
 EXPERIMENTAL - ONE TABULAR VALUE
203. KASSEL, L. S. AND MUSKAT, M.
 SURFACE ENERGY AND HEAT OF VAPORIZATION OF LIQUIDS.
 PHYS. REV. VOL. 40, 627-32 (1932)
 C.A. 26, 3941 4
- SURFACE ENERGY, HEAT OF VAPORIZATION
 HELIUM, NEON, NITROGEN, OXYGEN, CHLORINE
 THEORETICAL - EQUATIONS, TABLES (2 VALUES)
204. KEESOM, W. H.
 THERMODYNAMISCHE ONDERZOEKINGEN (MET UITZONDERING VAN DICHTHEDEN VAN
 VERZADIGDE VIOEISTOF EN DAMP, VAN DE THERMOMETRIE EN DE MANOMETRIE EN
 VAN DE INWENDIGE WRIJVING.)
 (THERMODYNAMIC INVESTIGATIONS (EXCEPTING DENSITIES OF SATURATED LIQUID
 AND VAPOR, OF THERMOMETRY AND MANOMETRY, AND VISCOSITY.))
 ONNES FESTSCHR. 89-163 (1922)
 ABSTRACTED IN PHYSIK. BER. VOL. 4, 613 (1923)
 C.A. 18, 2457 3
- P-V-T DATA (GAS) (123 TO 293 DEGREES K AND 13 TO 62 ATM), VIRIAL
 COEFFICIENTS (68 TO 151 DEGREES K), CRITICAL CONSTANTS, TRIPLE POINT
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- 204A. KELLEY, K. K. AND KING, E. G.
CONTRIBUTIONS TO THE DATA ON THEORETICAL METALLURGY.
XIV. ENTROPIES OF THE ELEMENTS AND INORGANIC COMPOUNDS.
U.S. BUR. MINES BULL. NO. 592 (1961) (SUPERSEDES BULL. NO. 477, 1950)
ENTROPY (LIQUID), ENTROPY OF FUSION AND VAPORIZATION
COMPILATION - DATA FROM 74, 122
205. KERR, E.C.
I. SECOND VIRIAL COEFFICIENT OF ARGON AT LOW TEMPERATURE AND LOW
PRESSURE. II. HEAT CAPACITY OF LIQUID NITRIC OXIDE ABOVE ITS NORMAL
BOILING POINT.
UNIVERSITY MICROFILMS (ANN ARBOR, MICH.) PUBL. NO. 21480
DISSERTATION ABSTRACTS VOL. 17, 1232 (1957)
C.A. 51, 14353 C
SECOND VIRIAL COEFFICIENT (GAS)(80 TO 300 DEGREES K AND 280 TO 854
MM HG), EQUATION OF STATE
NITRIC OXIDE
EXPERIMENTAL - TABLE (23 VALUES)
206. KESTIN, J. AND LEIDENFROST, W.
AN ABSOLUTE DETERMINATION OF THE VISCOSITY OF ELEVEN GASES OVER A
RANGE OF PRESSURES.
PHYSICA VOL. 25, 1033-62 (1959)
C.A. 54, 20387 B
VISCOSEITY, DENSITY (GAS)(293 TO 298 DEGREES K AND 0 TO 33 ATM)
AIR, CARBON DIOXIDE, DEUTERIUM, HELIUM, HYDROGEN, NEON, NITROGEN,
OXYGEN, KRYPTON, WATER, XENON
EXPERIMENTAL - TABLE (25 VALUES), GRAPH
APPARATUS
207. KESTIN, J. AND PILARCZYK, K.
MEASUREMENT OF THE VISCOSITY OF FIVE GASES AT ELEVATED PRESSURES BY
THE OSCILLATING DISK METHOD.
TRANS. AM. SOC. MECH. ENG. VOL. 76, 987-99 (1954)
C.A. 48, 11135 A
VISCOSEITY (GAS)(293 DEGREES K AND 15 TO 1009 PSIA)
HYDROGEN, AIR, NITROGEN, HELIUM
EXPERIMENTAL - GRAPH, EQUATION
APPARATUS
208. KESTIN, J. AND WANG, H. E.
THE VISCOSITY OF FIVE GASES. A RE-EVALUATION.
TRANS. AM. SOC. MECH. ENG. VOL. 80, 11-7 (1958)
C.A. 52, 5068 I
VISCOSEITY (GAS)(298 DEGREES K AND 1 TO 70 ATM)
AIR, NITROGEN, HELIUM, HYDROGEN
EXPERIMENTAL - TABLE (13 VALUES), GRAPH, EQUATION

209. KESTIN, J. AND WHITELAW, J. H.
 A RELATIVE DETERMINATION OF THE VISCOSITY OF SEVERAL GASES BY THE
 OSCILLATING DISK METHOD.
 PHYSICA VOL. 29, 335-56 (1963)
 BROWN UNIV., PROVIDENCE, R. I., REPT. NO. BRN-2-P (SEPT 1962)
 ASTIA AD 287 471

VISCOSITY (GAS) (298 TO 543 DEGREES K AND 1 TO 140 ATM)
 HELIUM, NITROGEN, NEON, CARBON DIOXIDE
 EXPERIMENTAL - TABLE (45 VALUES), GRAPH

210. KEYES, F. G.
 SUTHERLAND VISCOSITY CONSTANT AND ITS RELATION TO THE MOLECULAR
 POLARIZATION.
 Z. PHYSIK. CHEM. VOL. 130, 709-14 (1927)
 C.A. 22, 1715 3

VISCOSITY (GAS) (100 TO 150 DEGREES K)
 HYDROGEN, HELIUM, NITROGEN, OXYGEN, CARBON MONOXIDE, METHANE, KRYPTON,
 XENON, ETHYLENE
 CORRELATION - GRAPHS, EQUATION

211. KEYES, F. G.
 A SUMMARY OF VISCOSITY AND HEAT-CONDUCTION DATA FOR HELIUM, ARGON,
 HYDROGEN, OXYGEN, NITROGEN, CARBON MONOXIDE, CARBON DIOXIDE, WATER
 AND AIR.
 TRANS. AM. MECH. ENGRS. VOL. 73, 589-95 (1951)
 C.A. 45, 7400 E

VISCOSITY (LIQUID, GAS) (55 TO 1873 DEGREES K)
 HELIUM, HYDROGEN, OXYGEN, NITROGEN, CARBON DIOXIDE, WATER AND AIR
 CORRELATION - EQUATION

212. KEYES, F. G.
 THE HEAT CONDUCTIVITY, VISCOSITY, SPECIFIC HEAT AND PRANDTL NUMBERS
 FOR THIRTEEN GASES.
 SQUID TECHNICAL REPT. NO. 37, MASSACHUSETTS INSTITUTE OF TECHNOLOGY,
 CAMBRIDGE MIT-2-R (1952)
 ASTIA ATI-167 173

VISCOSITY, THERMAL CONDUCTIVITY, SPECIFIC HEAT ($P=CONSTANT$), PRANDTL
 NUMBER (LIQUID, GAS) (73 TO 1273 DEGREES K)
 AIR, METHANE, OXYGEN, HYDROGEN, NITROGEN, CARBON MONOXIDE, HELIUM AND
 5 OTHER GASES
 CORRELATION - EQUATIONS, TABLE (80 CALCULATED VALUES)

213. KEYES, F. G.
 THERMAL CONDUCTIVITY OF GASES.
 TRANS. AM. SOC. MECH. ENG. VOL. 76, 809-16 (1954)
 C.A. 48, 10398 B

THERMAL CONDUCTIVITY (GAS) (363 TO 623 DEGREES K AND 0 TO 19 ATM)
 HELIUM, NEON, HYDROGEN, METHANE AND 7 OTHER GASES
 EXPERIMENTAL - TABLE (10 VALUES)
 CORRELATION WITH AVAILABLE VISCOSITY DATA, PRESSURE-COEFFICIENT
 EQUATIONS
 APPARATUS

214. KEYES, F. G.
 THERMAL CONDUCTIVITY OF GASES.
 TRANS. AM. SOC. MECH. ENG. VOL. 77, 1395-6 (1955)
 C.A. 50, 636 D
- THERMAL CONDUCTIVITY(LIQUID,GAS)(87 TO 273 DEGREES K AND 1 TO 11 ATM)
 NITROGEN, OXYGEN, METHANE, XENON- KRYPTON, CARBON DIOXIDE
 EXPERIMENTAL - TABLE (16 VALUES)
215. KEYES, R. W.
 LAWS OF CORRESPONDING STATES FOR THE THERMAL CONDUCTIVITY OF MOLECULAR SOLIDS.
 J. CHEM. PHYS. VOL. 31, 452-4 (1959)
 C.A. 54, 2916 F
- THERMAL CONDUCTIVITY (SOLID)(2 TO 75 DEGREES K)
 NEON, KRYPTON
 THEORETICAL - EQUATION, GRAPHS
- 215A. KIGOSHI, K.
 FUSION AND STATE EQUATION OF SIMPLE LIQUIDS. I. DERIVATION OF THE EXPRESSIONS FOR STATE EQUATION AND FREE ENERGY. II. CRITICAL DATA AND FUSION PARAMETERS.
 BULL. CHEM. SOC. JAPAN VOL. 23, 236-44 (1950)
 C.A. 46, 9917 C
- ENTROPY OF FUSION, CHANGE OF VOLUME OF MELTING, CRITICAL TEMPERATURE NITROGEN, NEON, HYDROGEN
 THEORETICAL - EQUATIONS, TABLES (4 VALUES)
216. KIHARA, T.
 VIRIAL COEFFICIENTS AND MODELS OF MOLECULES IN GASES.
 REV. MOD. PHYS. VOL. 25, 831-43 (1953)
- SECOND AND THIRD VIRIAL COEFFICIENTS (GAS)
 NEON, KRYPTON, XENON, HYDROGEN, NITROGEN, METHANE AND 4 OTHER GASES
 THEORETICAL - EQUATIONS, GRAPHS
 COMPILATION - DATA FROM 177
217. KIRKWOOD, J. G. AND BUFF, F. P.
 THE STATISTICAL MECHANICAL THEORY OF SURFACE TENSION.
 J. CHEM. PHYS. VOL. 17, 338 43 (1949)
- SURFACE TENSION (90 DEGREES K)
 THEORETICAL - EQUATION, ONE TABULAR VALUE
218. KISHIMOTO, T. AND NOMOTO, O.
 ON MOLECULAR SOUND VELOCITY OF LIQUIDS. I. CONSIDERATION ON THE RAO'S RELATION BETWEEN THE TEMPERATURE COEFFICIENT OF SOUND VELOCITY IN AND THE COEFFICIENT OF THERMAL EXPANSION OF LIQUIDS.
 J. PHYS. SOC. JAPAN VOL. 9, 59-65 (1954)
 C.A. 49, 23 A
- EXPANSIVITY, COMPRESSIBILITY(ADIABATIC)(ISOTHERMAL), VELOCITY OF SOUND OXYGEN, NITROGEN, HYDROGEN, NEON, HELIUM, CARBON MONOXIDE, METHANE, NITROUS OXIDE AND NITRIC OXIDE
 THEORETICAL - EQUATIONS, GRAPHS

219. KISHIMOTO, T. AND NOMOTO, O.
 ON MOLECULAR SOUND VELOCITY OF LIQUIDS. II. DEPENDENCE OF THE THERMAL EXPANSION COEFFICIENT AND THE TEMPERATURE COEFFICIENT OF COMPRESSIBILITY OF LIQUIDS ON THE BOILING POINT.
 J. PHYS. SOC. JAPAN VOL. 9, 66-72 (1954)
 C.A. 49, 23 D
 EXPANSION COEFFICIENT, COMPRESSIBILITY (ADIABATIC)(ISOTHERMAL), VELOCITY OF SOUND
 NITROGEN, OXYGEN, 30 COMPOUNDS, 7 METALS
 THEORETICAL - EQUATIONS, TABLE (3 VALUES)
220. KITTLE, C.
 ULTRASONIC PROPAGATION IN LIQUIDS II. THEORETICAL STUDY OF THE FREE VOLUME MODEL OF THE LIQUID STATE.
 J. CHEM. PHYS. VOL. 14, 614-24 (1946)
 THE DIFFERENCE IN SPECIFIC HEATS AT CONSTANT PRESSURE AND CONSTANT VOLUME (GAS) (87 DEGREES K)
 OXYGEN, NITROGEN, HYDROGEN, METHANE AND 7 ORGANIC COMPOUNDS
 THEORETICAL - EQUATIONS, 1 TABULAR VALUE
221. KOEHLER, W. F.
 THE RATIO OF THE SPECIFIC HEATS OF GASES BY A METHOD OF SELF-SUSTAINED OSCILLATIONS.
 J. CHEM. PHYS. VOL. 18, NO. 4, 465-72 (1950)
 C.A. 44, 9204 B
 SPECIFIC HEAT RATIO (GAS)(298 DEGREES K AND 1 ATM)
 HELIUM, HYDROGEN, NITROGEN, OXYGEN, METHANE, AND 4 OTHER COMPOUNDS
 EXPERIMENTAL - ONE VALUE
222. KOEPPE, W.
 ON THE INVERSION CURVE AT LOW TEMPERATURES AND THE THEOREM OF CORRESPONDING STATES.
P. 156-63 IN PROGRESS IN REFRIGERATION SCIENCE AND TECHNOLOGY, VOL. 1,
 PROC. 10TH INTERN. CONGR. OF REFRIG., COPENHAGEN (1959)
 PERGAMON PRESS, NEW YORK(1960)
 INVERSION CURVE BY THE THEORY OF CORRESPONDING STATES
 HELIUM, HYDROGEN, NEON, NITROGEN, METHANE, OXYGEN, AIR, CARBON MONOXIDE AND 7 OTHER GASES
 THEORETICAL - GRAPH
223. KOEPPE, W.
 VERLAUF DER INVERSIONSKURVE BEI TIEFEN TEMPERATURN.
 (PATH OF THE INVERSION CURVE AT LOW TEMPERATURES.)
 MONATSBER. DEUT. AKAD. WISS. BERLIN VOL. 2, 78-86 (1960)
 INVERSION CURVE
 HELIUM, NEON, DEUTERIUM, NITROGEN, OXYGEN, CARBON MONOXIDE, METHANE, HYDROGEN
 THEORETICAL - GRAPH
 GERMAN

224. KOLSKY, H. G.
 THE THERMODYNAMIC PROPERTIES OF 54 ELEMENTS CONSIDERED AS IDEAL MONATOMIC GASES.
 CALIFORNIA UNIV., LOS ALAMOS SCIENTIFIC LAB., LOS ALAMOS, NEW MEXICO,
 REPT. NO. LA 2110, (1957) 138 P.
 ENTHALPY, SPECIFIC HEAT ($P=CONSTANT$), ENTROPY, FREE ENERGY (GIBBS FUNCTION) (IDEAL GAS) (10 TO 8000 DEGREES K)
 HYDROGEN, NITROGEN, FLUORINE, OXYGEN, ARGON, NEON, HELIUM AND 46 OTHER ELEMENTS
 THEORETICAL - EQUATIONS, TABLE (560 VALUES)
225. KOPSCHE, W.
 UBER DEN KOEFFIZIENTEN DER INNEREN REIBUNG VON WASSERSTOFF UND ARGON BEI NIEDEREN TEMPERATUREN.
 (CONCERNING THE COEFFICIENT OF VISCOSITY OF HYDROGEN AND ARGON AT LOW TEMPERATURES.)
 VEREINIGTEN FRIEDRICHSH-UNIVERSITY, HALLE (1909)
 VISCOSITY (GAS) (90 TO 288 DEGREES K)
 HYDROGEN
 EXPERIMENTAL - TABLE (80 VALUES)
 APPARATUS
 GERMAN
226. KORDES, E.
 UNTERSUCHUNGEN UBER DAS HETEROGENE GLEICHGEWICHT FLUSSIGKEIT - DAMPF.
 (RESEARCH ON THE HETEROGENEOUS EQUILIBRIUM, LIQUID - VAPOR.)
 Z. ELEKTROCHEM. VOL. 58, 424-31 (1954)
 C.A. 49, 2140 E
 HEAT OF VAPORIZATION (87 TO 147 DEGREES K)
 HYDROGEN, CARBON MONOXIDE, OXYGEN, AND 10 OTHER LIQUIDS
 THEORETICAL - EQUATION, TABLE (8 VALUES)
 GERMAN
227. LACAM, A. AND NOURY, J.
 SUR LE RAPPORT GAMMA DES CHALEURS SPECIFIQUES DE L'ARGON SOUS PRESSION.
 (THE SPECIFIC HEAT RATIO, GAMMA, OF ARGON UNDER PRESSURE.)
 COMPT. REND. VOL. 236, 589-90 (1953)
 SPECIFIC HEAT RATIO (GAS) (298 DEGREES K AND 110 TO 910 ATM)
 EXPERIMENTAL - TABLE (22 VALUES), GRAPH
 FRENCH
228. LACAM, A. AND NOURY, J.
 VITESSE DES ULTRASONS ET RAPPORT GAMMA DES CHALEURS SPECIFIQUES RELATIVES A L'ARGON ET A L'AZOTE SOUS PRESSION.
 (SPEED OF SOUND AND RATIO OF SPECIFIC HEATS RELATIVE TO ARGON AND NITROGEN UNDER PRESSURE.)
 COMPT. REND. VOL. 236, 2039-41 (1953;
 C.A. 49, 9340 H
 VELOCITY OF SOUND, SPECIFIC HEAT RATIO (GAS) (298 DEGREES K AND 10 TO 124 ATM)
 NITROGEN
 EXPERIMENTAL - TABLE (24 VALUES), GRAPH
 FRENCH

229. LAHR, P. H. AND EVERSOLE, W. G.
 COMPRESSION ISOTHERMS OF ARGON, KRYPTON, AND XENON THROUGH THE
 FREEZING ZONE.
J. CHEM. ENG. DATA VOL. 7, NO. 1, 42-7 (1962)
- MELTING PRESSURE, SATURATED DENSITY, COMPRESSIBILITY (SOLID, LIQUID)
 (137 TO 360 DEGREES K AND 2389 TO 18040 ATM), ENTROPY AND ENTHALPY
 OF FUSION (SOLID) (83 TO 378 DEGREES K AND 0 TO 18000 ATM)
 KRYPTON, XENON
 EXPERIMENTAL - TABLE (39 VALUES), GRAPHS, EQUATIONS
 APPARATUS
230. LAWRENCE, D.J., STEWART, A.T. AND GUPTILL, E.W.
 THERMAL CONDUCTIVITY OF SOLID ARGON AT 80 DEGREES K.
CAN. J. PHYS. VOL. 37, NO. 9, 1069-72 (1959)
- THERMAL CONDUCTIVITY (SOLID) (80 DEGREES K)
 EXPERIMENTAL - ONE TABULAR VALUE, GRAPH
 APPARATUS
231. LEBOWITZ, J. L., FRISCH, H. L. AND REISS, H.
 THE EQUATION OF STATE OF A FLUID.
P. 164-70 IN PROGRESS IN REFRIGERATION SCIENCE AND TECHNOLOGY, VOL. 1,
PROC. 10TH INTERN. CONGR. OF REFRIG., COPENHAGEN (1959)
PERGAMON PRESS, NEW YORK (1960)
ASTIA AD 241 256
- EQUATION OF STATE (LIQUID), SURFACE TENSION, HEAT OF VAPORIZATION
 THEORETICAL - EQUATIONS, GRAPH
232. LEDUC, A.
 COMPRESSIBILITE ET DILATABILITE DES GAZ.
 (COMPRESSIBILITY AND EXPANSIVITY OF GAS.)
ANN. DE PHYS. VOL. 9, 1-28 (1918)
C.A. 12, 1432
- EXPANSIVITY (281 TO 305 DEGREES K)
 NEON
 EXPERIMENTAL
 FRENCH
 -CODED FROM ABSTRACT-
233. LEDUC, A.
 THE SPECIFIC HEATS OF GASES AND VAPORS. A CRITICAL REVIEW OF METHODS
 AND RESULTS.
CHEM. REV. VOL. 6, 1-16 (1929)
- SPECIFIC HEAT ($V=CONSTANT$) ($P=CONSTANT$), SPECIFIC HEAT RATIO (GAS)
 HYDROGEN, NITROGEN, OXYGEN, CARBON MONOXIDE, METHANE AND 12 COMPOUNDS
 REVIEW - METHODS. TABLES OF VALUES
234. LENNARD-JONES, J. E. AND DEVONSHIRE, A. F.
 CRITICAL PHENOMENA IN GASES - I.
PROC. ROY. SOC. (LONDON) VOL. A163, 53-70 (1937)
- EQUATION OF STATE (151 DEGREES K), CRITICAL CONSTANTS
 NEON, HELIUM, KRYPTON
 THEORETICAL - EQUATION

235. LENNARD-JONES, J. E. AND DEVONSHIRE, A. F.
 CRITICAL PHENOMENA IN GASES. II. VAPOUR PRESSURES AND BOILING POINTS.
 PROC. ROY. SOC. (LONDON) VOL. A165, 1-11 (1938)
- VAPOR PRESSURE, BOILING POINT, ENTROPY OF VAPORIZATION (151 DEGREES K)
 NEON, NITROGEN
 THEORETICAL - EQUATIONS
- 235A. LESTZ, S. S.
 ACOUSTIC ISOTHERMS FOR NITROGEN, ARGON, AND KRYPTON.
 J. CHEM. PHYS. VOL. 38, 2830-34 (1963)
 C.A. 59, 1115 C
- VELOCITY OF SOUND (GAS) (273 AND 303 DEGREES K AND 1 TO 12 ATM),
 CALCULATED VALUES FOR SPECIFIC HEATS ($P=$ CONSTANT AND $V=$ CONSTANT)
 (GAS) (273 AND 303 DEGREES K AND 1 TO 12 ATM)
 NITROGEN
 EXPERIMENTAL - TABLES (75 VALUES), EQUATIONS, GRAPHS
236. LEVELT, J. M. H.
 MEASUREMENTS OF THE COMPRESSIBILITY OF ARGON IN THE GASEOUS AND
 LIQUID PHASE.
 AMSTERDAM UNIVERSITY, NETHERLANDS, PH. D. THESIS (1958) 120 P.
- P-V-T DATA, DENSITY (GAS)(118 TO 248 DEGREES K AND 6 TO 1028 ATM),
 SATURATED DENSITY (LIQUID, GAS) (133 TO 148 DEGREES K), EQUATION OF
 STATE COEFFICIENTS (LIQUID, GAS) (133 TO 248 DEGREES K), HEAT OF
 VAPORIZATION (133 TO 150 DEGREES K), ENTROPY, ENTHALPY, INTERNAL
 ENERGY, SPECIFIC HEAT ($P=$ CONSTANT)($V=$ CONSTANT), VELOCITY OF SOUND,
 JOULE-THOMSON COEFFICIENT (GAS) (133 TO 273 DEGREES K AND 1 TO
 1050 ATM), CRITICAL POINT, PRINCIPLE OF CORRESPONDING STATES
 EXPERIMENTAL - TABLES (2400 VALUES), EQUATIONS, GRAPHS
 APPARATUS
237. LEVELT, J. M. H.
 THE REDUCED COMPRESSIBILITY FACTOR, INTERNAL ENERGY AND ENTROPY OF
 ARGON AND XENON.
 WISCONSIN UNIV., MADISON, REPT. NO. WIS-OOR-24 (AUG 1959) 32 P.
 ASTIA AD 226 217
 PHYSICA VOL. 26, NO. 6, 361-77 (1960)
 C.A. 55, 8972 C
- LAW OF CORRESPONDING STATES, REDUCED COMPRESSIBILITY FACTOR, REDUCED
 INTERNAL ENERGY, REDUCED INTERNAL ENTROPY (GAS)(143 TO 423 DEGREES K)
 XENON
 THEORETICAL - EQUATIONS, TABLES (1800 VALUES)
238. LEVELT, J. M. H. AND COHEN, E. G. D.
 THE THERMODYNAMIC PROPERTIES OF DENSE GASES AND LIQUIDS.
 BULL. INST. INTERN. FROID ANNEXE 1960-1, 129-32 (1960)
- COMPRESSIBILITY FACTOR, INTERNAL ENERGY, SPECIFIC HEAT
 ($V=$ CONSTANT)(LIQUID, GAS)
 XENON
 DISCUSSION
-

239. LEVITT, L. S.
 EXTREME PRESSURES. II. VOLUME-TEMPERATURE RELATIONSHIP FOR GASES.
J. CHEM. PHYS. VOL. 34, NO. 4, 1440-3 (APR 1961)
 C.A. 55, 16051 E
- DENSITY (GAS)(0 TO 323 DEGREES K AND 2000 TO 5500 ATM)
 DERIVED PROPERTY RELATIONS AT HIGH PRESSURES - ENTROPY, SPECIFIC HEAT
 $(P=\text{CONSTANT})(V=\text{CONSTANT})$, THERMAL EXPANSION
 NITROGEN, WATER
 THEORETICAL - EQUATION, GRAPH
240. LICHT, W. AND STECHERT, D. G.
 THE VARIATION OF THE VISCOSITY OF GASES AND VAPORS WITH TEMPERATURE.
J. PHYS. CHEM. VOL. 48, 23-47 (1944)
- VISCOSITY (GAS)(15 TO 1500 DEGREES K), CRITICAL CONSTANTS, PRINCIPLE
 OF CORRESPONDING STATES
 HELIUM, HYDROGEN, METHANE, NITROGEN, OXYGEN AND 20 OTHER GASES
 THEORETICAL - EQUATIONS, NOMOGRAPH
241. LIEPMANN, H. W.
 UBER SCHALLGESCHWINDIGKEITSMESSUNGEN IN FLUSSIGEM ARGON.
 (VELOCITY OF SOUND MEASUREMENTS IN LIQUID ARGON.)
HELV. PHYS. ACTA VOL. 12, 421-42 (1939)
 C.A. 34, 11 4
- VELOCITY OF SOUND, SPECIFIC HEAT ($V=\text{CONSTANT}$), ADIABATIC AND
 ISOTHERMAL COMPRESSIBILITIES (LIQUID)(84 TO 87 DEGREES K)
 EXPERIMENTAL - TABLE (40 VALUES), EQUATIONS, GRAPH
 APPARATUS
 GERMAN
242. LILEY, P. E.
 SURVEY OF RECENT WORK ON THE VISCOSITY, THERMAL CONDUCTIVITY, AND
 DIFFUSION OF GASES AND LIQUEFIED GASES BELOW 500 DEGREES K.
 THERMOPHYSICAL PROPERTIES RESEARCH CENTER, PURDUE UNIV., LAFAYETTE,
 IND.,
 TPRC REPT. 13 (JUN 1961) 33 P.
- VISCOSITY, THERMAL CONDUCTIVITY (LIQUID,GAS)(50 TO 500 DEGREES K)
 HELIUM, HYDROGEN, METHANE, CARBON MONOXIDE, AIR, FLUORINE, NEON,
 NITROGEN, OXYGEN AND 19 OTHER FLUIDS
 BIBLIOGRAPHY
243. LINDE AIR PRODUCTS COMPANY
 LINDE RARE GASES.
 LINDE AIR PRODUCTS CO., UNION CARBIDE AND CARBON CORP., NEW YORK 17,
 N. Y. (1957)
- VELOCITY OF SOUND, HEAT CONDUCTIVITY (GAS)(273 DEGREES K),
 SATURATED DENSITY (LIQUID, GAS)-(87 DEGREES K), VAPOR PRESSURE
 (SOLID), HEATS OF FUSION AND VAPORIZATION, CRITICAL CONSTANTS,
 BOILING AND MELTING POINTS
 NEON, HELIUM, KRYPTON, XENON
 COMPILATION - SOURCES NOT GIVEN

244. LOBO, W. E.
 TECHNICAL DATA (PERTAINING TO AIR, ITS LIQUEFACTION AND DISTILLATION.)
 M. W. KELLOGG COMPANY, FINAL REPT. TO NATL. DEFENSE RESEARCH COMM.,
 OFFICE OF SCIENTIFIC RESEARCH AND DEVELOPMENT, REPT. OSRD NO. 4206,
 (OCT 1944) 52 P.
- DENSITY (GAS)(173 TO 368 DEGREES K), VAPOR PRESSURE (83 TO 158
 DEGREES K), THERMAL CONDUCTIVITY (GAS)(148 TO 1033 DEGREES K),
 VISCOSITY (GAS)(200 TO 1400 DEGREES K), P-V-T DATA (GAS)(173 TO
 368 DEGREES K), ENTHALPY (0 TO 468 DEGREES K)
 AIR, NITROGEN, OXYGEN, CARBON DIOXIDE AND HELIUM
 COMPILATION - GRAPHICAL DATA, SOURCES NOT GIVEN
245. LOSENICKY, Z.
 ON THE APPLICATION OF THE PRINCIPLE OF CORRESPONDING STATES TO THE
 THERMAL CONDUCTIVITY OF THE GASEOUS STATE.
 CZECHOSLOV. J. PHYS. VOL. 9, 399-400 (1959)
- THERMAL CONDUCTIVITY (GAS)(90 TO 900 DEGREES K), PRINCIPLE OF
 CORRESPONDING STATES
 AIR, HELIUM, NEON, NITROGEN, OXYGEN, CARBON MONOXIDE, METHANE,
 HYDROGEN, AND 6 OTHER COMPOUNDS
 THEORETICAL - GRAPH
246. LUNBECK, R. J.
 HET PRINCIPE VAN OVEREENSTEMMENDE TOESTANDEN IN DE QUANTUMMECHANICA.
 (THE PRINCIPLE OF CORRESPONDING STATES IN QUANTUM MECHANICS.)
 DOCTORIAL DISSERTATION, AMSTERDAM (1950)
- CORRESPONDING STATES THEORY
 THEORETICAL - EQUATIONS, GRAPHS, TABLES
 DUTCH
247. LYDERSEN, A. L., GREENKORN, R. A. AND HOUGEN, O. A.
 GENERALIZED THERMODYNAMIC PROPERTIES OF PURE FLUIDS.
 WISCONSIN UNIV. ENG. EXPT. STA. REPT. NO. 4, (OCT 1955) 99 P.
- ENTHALPY FUNCTION, ENTROPY, COMPRESSIBILITY FACTOR, REDUCED DENSITY,
 FUGACITY COEFFICIENTS, VAPORIZATION EQUILIBRIUM CONSTANTS, INTERNAL
 ENERGY (REDUCED TEMPERATURE = 0.5 TO 2 AND REDUCED PRESSURE = 0.1 TO
 1.2)
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248. MAC LEOD, D. B.
 A CALCULATION OF THE LATENT HEAT OF VAPORISATION BASED ON A REVISED
 EQUATION OF STATE.
 TRANS. FARADAY SOC. VOL. 41, 122-6 (1945)
 C.A. 39, 3711 7
- HEAT OF VAPORIZATION (89.8 TO 147.8 DEGREES K)
 HYDROGEN, BENZENE, PENTANE, CARBON TETRACHLORIDE
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249. MAKITA, T.
 THE VISCOSITY OF ARGON, NITROGEN AND AIR AT PRESSURES UP TO
 800 KG/SQ CM.
 REV. PHYS. CHEM. JAPAN VOL. 27, 16-21 (1957)
 C.A. 52, 9696 H
- VISCOSITY (GAS) (298 TO 423 DEGREES K AND 0 TO 800 KG/SQ CM)
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- 249A. MAKITA, T.
 TRANSPORT PHENOMENA OF GASES UNDER HIGH PRESSURES. II. THERMAL
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 KOATSU GASU KYOKAISHI VOL. 23, NO. 10, 489-501 (OCT 1959)
- THERMAL CONDUCTIVITY (GAS) (153 TO 348 DEGREES K AND 1 TO 2000 ATM)
 METHANE, HELIUM, HYDROGEN, NITROGEN
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250. MARGENAU, H.
 THE SECOND VIRIAL COEFFICIENT FOR GASES. A CRITICAL COMPARISON
 BETWEEN THEORETICAL AND EXPERIMENTAL RESULTS.
 PHYS. REV. VOL. 36, 1782-90 (DEC 1930)
 C.A. 25, 3886 4
- SECOND VIRIAL COEFFICIENT (200 TO 500 DEGREES K)
 HELIUM, NEON, HYDROGEN, NITROGEN, OXYGEN, CARBON DIOXIDE, AMMONIA,
 WATER
 DISCUSSION - EQUATIONS, GRAPHS
251. MARKHAM, J. J., MEYER, R. T. AND LINDSAY, R. B.
 ABSORPTION OF SOUND IN FLUIDS.
 REV. MOD. PHYS. VOL. 23, 353-411 (1951)
- SOUND ABSORPTION, VISCOSITY (LIQUID) (85 DEGREES K)
 REVIEW
252. MARYOTT, A. A. AND SMITH, F. R.
 TABLE OF DIELECTRIC CONSTANTS OF PURE LIQUIDS.
 NATL. BUR. STANDARDS CIRC. NO. 514 (1951)
- DIELECTRIC CONSTANT (LIQUID) (82 DEGREES K)
 HYDROGEN, HELIUM, OXYGEN, NITROGEN, METHANE, FLUORINE, DEUTERIUM AND
 73 ORGANIC AND INORGANIC COMPOUNDS
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253. MASLAN, F. D. AND LITTMAN, T. M.
 COMPRESSIBILITY CHART FOR HYDROGEN AND INERT GASES.
 IND. ENG. CHEM. VOL. 45, 1566-8 (1953)
- COMPRESSIBILITY FACTOR (GAS) (120 TO 1000 DEGREES K AND 9 TO 6000 ATM)
 THEORY OF CORRESPONDING STATES
 HELIUM, NEON, HYDROGEN
 CORRELATION - GRAPH

254. MASON, E. A., AND RICE, W. E.
 THE INTERMOLECULAR POTENTIALS FOR SOME SIMPLE NONPOLAR MOLECULES.
J. CHEM. PHYS. VOL. 22, NO. 5, 843-51 (1954)
 C.A. 48, 11855 I
- INTERMOLECULAR POTENTIAL, 2ND VIRIAL COEFFICIENT, VISCOSITY, THERMAL CONDUCTIVITY, SELF-DIFFUSION COEFFICIENTS (GAS)(150 TO 600 DEGREES K) NEON, METHANE, KRYPTON, XENON, NITROGEN
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255. MATHESON COMPANY, INC.
MATHESON GAS DATA BOOK
 THE MATHESON COMPANY, INC. (1961) 420 P.
 EAST RUTHERFORD, NEW JERSEY
- VAPOR PRESSURE (90 TO 150 DEGREES K), DENSITY (GAS)(233 TO 323 DEGREES K AND 0 TO 231 ATM), ENTHALPY, ENTROPY (GAS)(250 TO 700 DEGREES K AND 0.1 TO 100 ATM), SPECIFIC HEAT ($P=CONSTANT$) (GAS) (294 DEGREES K), SPECIFIC HEAT ($V=CONSTANT$), SPECIFIC HEAT RATIO (GAS) (288 DEGREES K), VISCOSITY (GAS)(294 DEGREES K), THERMAL CONDUCTIVITY (GAS)(273 DEGREES K), FREEZING AND BOILING POINTS, LATENT HEATS OF FUSION AND VAPORIZATION, CRITICAL CONSTANTS
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256. MATHIAS, E.
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 (STUDY OF DENSITY CURVES AT LOW TEMPERATURES IN THE CRYOGENIC LABORATORY OF LEIDEN.)
 P. 165-98 IN HET NATUURKUNDIG LABORATORIUM DER RIJKSUNIVERSITEIT TE LEIDEN IN DE JAREN 1904-22 (THE PHYSICAL LABORATORY AT THE UNIVERSITY OF LEIDEN IN THE YEARS 1904-22) EDUARD IJDO, LEIDEN (1922)
 ABSTRACTED IN PHYSIK. BER. VOL. 4, 701-2 (1923)
 C.A. 18, 2625 7
- SATURATED DENSITY (LIQUID, GAS)(90 TO 148 DEGREES K), CRITICAL CONSTANTS, RECTILINEAR DIAMETER
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- 256A. MATHIAS, E., AND CROMMELIN, C. A.
 RAPPORT SUR LE DIAMETRE RECTILIGNE.
 (REPORT ON THE RECTILINEAR DIAMETER).
 PROC. INTERN. CONGR. REFRIG. 7TH CONGR. AMSTERDAM 96-102 (1936)
 C.A. 30, 7938 4
- TRIPLE POINT
 CARBON MONOXIDE, HELIUM, KRYPTON, NEON
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 FRENCH

257. MATHIAS, E., CROMMELIN, C. A., AND ONNES, H. K.
 LA CHALEUR DE VAPORISATION ET LA DIFFERENCE M₁-M DES CHALEURS
 SPECIFIQUES A L'ETAT DE SATURATION POUR L'ARGON, L'OXYGENE, L'AZOTE ET
 L'HYDROGENE.
 (THE HEAT OF VAPORIZATION AND THE DIFFERENCE M₁-M OF THE SPECIFIC HEAT
 AT THE STATE OF SATURATION FOR ARGON, OXYGEN, NITROGEN AND HYDROGEN.)
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 162A (1923)
 ANN. PHYS. (9E) VOL. 19, 239-47 (1923)
 COMPT. REND. VOL. 174, 1395-7 (1922)
 C.A. 16, 4120 8
- HEAT OF VAPORIZATION, DIFFERENCE OF SPECIFIC HEATS (SATURATION)
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258. MATHIAS, E., ONNES, H. K., AND CROMMELIN, C. A.
 ON THE RECTILINEAR DIAMETER FOR ARGON.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 131A (1912)
 PROC. ACAD. SCI. AMSTERDAM VOL. 15, 667-73 AND 960-5 (1912)
 VERSLAG GEWONE VERGADER WIS- EN NATUURK. AFDEEL. KON. AKAD.
 WETENSCHAP. 70U-6 AND 893-8 (1912)
 ANN. PHYS. VOL. 17, 442-55 (1922)
- SATURATED DENSITY (LIQUID, GAS)(90 TO 138 DEGREES K), RECTILINEAR
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- 258A. MAZUR, P., AND JANSEN, L.
 ON THE THEORY OF MOLECULAR POLARIZATION IN GASES. II. EFFECT OF
 MOLECULAR INTERACTIONS ON THE CLAUSIUS-MOSSOTTI FUNCTION FOR SYSTEMS
 OF SPHERICAL NONPOLAR MOLECULES.
 PHYSICA VOL. 21, 208-18 (1955)
- DIELECTRIC CONSTANT (GAS) (160 TO 298 DEGREES K)
 THEORETICAL - EQUATIONS, GRAPH
259. MC LAUGHLIN, E.
 VISCOSITY AND SELF-DIFFUSION IN LIQUIDS.
 TRANS. FARADAY SOC. VOL. 55, 28-38 (1959)
 C.A. 54, 25 B
- SELF-DIFFUSION COEFFICIENT (LIQUID)(84 DEGREES K), VISCOSITY (LIQUID)
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260. MC LENNAN, J. C., JACOBSEN, R. C., AND WILHELM, J. O.
 DIELECTRIC CONSTANTS OF LIQUEFIED GASES.
 TRANS. ROY. SOC. CAN. (3) VOL. 24, 37-46 (1930)
 C.A. 25, 2340 4
- DIELECTRIC CONSTANT (LIQUID)(82 TO 88 DEGREES K)
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261. MICHELS, A.
 VISCOSITY AND HEAT CONDUCTION OF GASES (AND LIQUIDS) IN THE
 NEIGHBORHOOD OF THE CONDENSATION REGION.
 PROC. INTERN. SYMP. TRANSPORT PROCESSES IN STATISTICAL MECHANICS,
 BRUSSELS 1956, P. 365-75 (1958)
 C.A. 53, 13712 H
- THERMAL CONDUCTIVITY (GAS)(273 TO 348 DEGREES K AND 0 TO 2000 ATM)
 VISCOSITY (GAS)(273 TO 348 DEGREES K AND 1 TO 1400 ATM)
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 DISCUSSION - GRAPHS
- 261A. MICHELS, A.
 SOME REMARKS ON EXPERIMENTS IN THE DENSE STATE.
 NUOVO CIMENTO SUPPL. VOL. 9, NO. 1, 152-62 (1958)
- THERMAL CONDUCTIVITY, DIFFERENCE IN SPECIFIC HEATS AT CONSTANT
 PRESSURE AND VOLUME (GAS) (273 TO 348 DEGREES K), LORENTZ-LORENZ
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262. MICHELS, A., AND BOTZEN, A.
 REFRACTIVE INDEX AND LORENTZ-LORENZ FUNCTION OF ARGON UP TO 2300
 ATMOSPHERES AT 25 DEGREES C.
 PHYSICA VOL. 15, 769-73 (1949)
 C.A. 44, 3758 D
- REFRACTIVE INDEX (GAS)(298 DEGREES K AND 1 TO 2369 ATM), LORENTZ-
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263. MICHELS, A., BOTZEN, A., FRIEDMAN, A. S., AND SENGERS, J. V.
 THE THERMAL CONDUCTIVITY OF ARGON BETWEEN 0 AND 75 DEGREES C AT
 PRESSURES UP TO 2500 ATMOSPHERES.
 PHYSICA VOL. 22, 121-8 (1956)
 C.A. 50, 16328 I
- THERMAL CONDUCTIVITY, P-V-T DATA (GAS)(273 TO 348 DEGREES K AND
 1 TO 2400 ATM)
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264. MICHELS, A., BOTZEN, A., AND SCHUURMAN, W.
 THE VISCOSITY OF ARGON AT PRESSURES UP TO 2000 ATMOSPHERES.
 PHYSICA VOL. 20, 1141-8 (1954)
 C.A. 49, 9990 H
- VISCOSITY (GAS)(273 TO 348 DEGREES K AND 9 TO 2000 ATM)
 EXPERIMENTAL - TABLE (85 VALUES), GRAPH, EQUATION
-

265. MICHELS, A., LEVELT, J. M. AND DE GRAAFF, W.
 COMPRESSIBILITY ISOTHERMS OF ARGON AT TEMPERATURES BETWEEN -25
 AND -155 DEGREES C AND AT DENSITIES UP TO 640 AMAGAT. (PRESSURES UP
 TO 1050 ATMOSPHERES).
 PHYSICA VOL. 24, 659-71 (1958)
 C.A. 53, 3811 E
 ISOTHERMAL COMPRESSIBILITY (GAS) (118 TO 248 DEGREES K AND 6 TO
 1028 ATM), SATURATION DENSITY (LIQUID, GAS)(120 TO 151 DEGREES K),
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266. MICHELS, A., LEVELT, J. M. AND WOLKERS, G. J.
 THERMODYNAMIC PROPERTIES OF ARGON AT TEMPERATURES BETWEEN 0 AND -140
 DEGREES C AND AT DENSITIES UP TO 640 AMAGAT. (PRESSURES UP TO 1050
 ATMOSPHERES.).
 PHYSICA VOL. 24, 769-94 (1958)
 C.A. 53, 6717 F
 ENERGY, ENTROPY, ENTHALPY, FREE ENERGY, INTERNAL ENERGY, SPECIFIC
 HEATS (V=CONSTANT)(P=CONSTANT), VELOCITY OF SOUND, JOULE-THOMSON
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267. MICHELS, A., LUNBECK, R. J. AND WOLKERS, G. J.
 THERMODYNAMICAL PROPERTIES OF ARGON AS FUNCTION OF DENSITY AND
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 PHYSICA VOL. 15, 689-95 (1949)
 ENTROPY, INTERNAL ENERGY, ENTHALPY, FREE ENERGY, SPECIFIC HEAT (V=
 CONSTANT)(P=CONSTANT)(GAS)(273 TO 423 DEGREES K)
 COMPILATION - TABLE (900 VALUES)
268. MICHELS, A., LUNBECK, R. J. AND WOLKERS, G. J.
 THERMODYNAMICAL FUNCTIONS OF ARGON.
 PHYSICA VOL. 16, 224 ONLY (1950)
 ERRATUM TO MICHELS, LUNBECK AND WOLKERS, PHYSICA VOL. 15, 689 (1949)
269. MICHELS, A., LUNBECK, R. J. AND WOLKERS, G. J.
 THERMODYNAMICAL PROPERTIES OF ARGON AS FUNCTION OF PRESSURE AND
 TEMPERATURE BETWEEN 0 AND 2000 ATMOSPHERES AND 0 AND 150 DEGREES C.
 APPL. SCI. RESEARCH VOL. A2, 345-50 (1950)
 C.A. 45, 3799 H
 P-V-T DATA, ENTROPY, INTERNAL ENERGY, HELMHOLTZ FUNCTION, ENTHALPY,
 SPECIFIC HEAT (P=CONSTANT), JOULE-THOMSON COEFFICIENT (GAS)(273 TO
 423 DEGREES K AND 0 TO 2000 ATM)
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-

270. MICHELS, A. AND PRINS, C.
 THE MELTING LINES OF ARGON, KRYPTON AND XENON UP TO 1500 ATMOSPHERES,
 REPRESENTATION OF THE RESULTS BY A LAW OF CORRESPONDING STATES.
 PHYSICA VOL. 28, 101-16 (1962)
 C.A. 57, 7931 H
- MELTING LINE (SOLID)(84 TO 118 DEGREES K), TRIPLE POINT, THEORY OF
 CORRESPONDING STATES
 KRYPTON, XENON, NEON
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271. MICHELS, A., SENGERS, J. V. AND VAN DE KLUNDERT, L. J. M.
 THE THERMAL CONDUCTIVITY OF ARGON AT ELEVATED DENSITIES.
 PHYSICA VOL. 29, 149-60 (1963)
 178TH PUBLICATION OF THE VAN DER WAALS FUND.
- THERMAL CONDUCTIVITY, PRANDTL NUMBER, VISCOSITY (GAS)(273 TO
 348 DEGREES K AND 1 TO 2400 ATM)
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272. MICHELS, A., TEN SELDAM, C. A. AND OVERDIJK, S. D. J.
 THE DIELECTRIC CONSTANT OF ARGON AT 25 AND 125 DEGREES C FOR PRESSURES
 UP TO 2700 ATMOSPHERES.
 PHYSICA VOL. 17, 781-7 (1951)
 C. A. 46, 1828
- DIELECTRIC CONSTANT (GAS)(298 TO 398 DEGREES K AND 4 TO 2666 ATM)
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273. MICHELS, A., WASSENAAR, T., SLUYTERS, K. AND DE GRAAFF, A.
 THE TRIPLE POINTS OF CARBON DIOXIDE AND OF ARGON AS FIXED POINTS FOR
 THE CALIBRATION OF THERMOMETERS.
 PHYSICA VOL. 23, 89-94 (1957)
 154TH PUBLICATION OF THE VAN DER WAALS FUND.
- TRIPLE POINT
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 EXPERIMENTAL - ONE TABULAR VALUE
274. MICHELS, A., WIJKER, H. AND WIJKER, HK.
 ISOTHERMS OF ARGON BETWEEN 0 AND 150 DEGREES C AND PRESSURES UP TO
 2900 ATMOSPHERES.
 PHYSICA VOL. 15, NO. 7, 627-33 (1949)
 C.A. 44, 394 D
- P-V-T DATA (GAS)(273 TO 423 DEGREES K AND 18 TO 2888 ATMOSPHERES),
 VIRIAL COEFFICIENTS (273 TO 423 DEGREES K)
 EXPERIMENTAL - TABLES (200 VALUES), EQUATION
275. MILLER, J. G.
 DIELECTRIC CONSTANT AND REFRACTIVITY.
 TRANS. AM. SOC. MECH. ENG. VOL. 70, 645-9 (1948)
- DIELECTRIC CONSTANT, REFRACTIVE INDEX
 HELIUM, HYDROGEN, NITROGEN, OXYGEN, METHANE AND 3 OTHER GASES
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276. MOLES, E.
 DAS LITERGEWICHT UND DAS ATOMGEWICHT DES ARGONS.
 (THE LITER WEIGHT AND THE ATOMIC WEIGHT OF ARGON.)
 BER. DEUT. CHEM. GES. VOL. 60, 134-8 (1927)
 C.A. 21, 841 1
- ATOMIC WEIGHT
 EXPERIMENTAL - ONE TABULAR VALUE
 GERMAN
277. MURPHY, G. W. AND RICE, O. K.
 CORRESPONDING STATES IN THE FROZEN RARE GASES.
 J. CHEM. PHYS. VOL. 14, 518-25 (1946)
 C.A. 40, 6911 2
- CORRESPONDING STATES THEORY, DEBYE TEMPERATURE, SPECIFIC HEAT
 (P=CONSTANT)(SOLID)(10 TO 80 DEGREES K)
 NEON, KRYPTON AND XENON
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278. MYERS, A. L. AND PRAUSNITZ, J. M.
 SECOND VIRIAL COEFFICIENTS AND KIHARA PARAMETERS FOR ARGON.
 PHYSICA VOL. 28, 303-4 (1962)
 C.A. 57, 6632 F
- KIHARA POTENTIAL, SECOND VIRIAL COEFFICIENTS
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 'LETTER TO EDITOR'
279. NASON, W. C.
 SOME PHYSICAL PROPERTIES OF AIR AND ITS COMPONENTS.
 AERO MEDICAL LAB., WRIGHT FIELD, DAYTON, OHIO, CONTRACT NO. W33-038
 AC 2031
- HEAT CAPACITY (P=CONSTANT), VISCOSITY(LIQUID,GAS)(84 TO 300 DEGREES K)
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280. NELSON, L. C. AND OBERT, E. F.
 GENERALIZED P-V-T PROPERTIES OF GASES.
 TRANS. AM. SOC. MECH. ENGRS. VOL. 76, 1057-66 (1954)
 C.A. 48, 13307 B
- LAW OF CORRESPONDING STATES, P-V-T PROPERTIES, COMPRESSIBILITY FACTOR
 HYDROGEN, DEUTERIUM, HELIUM, NITROGEN, OXYGEN, NEON, METHANE,
 CARBON MONOXIDE, AIR, AND 20 OTHER GASES
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281. NIEMEYER, O.
 DAS VERHALTNIS DER SPECIFISCHEN WARMEN VON ARGON UND SEINE ANDERUNG
 MIT DER TEMPERATUR.
 (THE STUDY OF THE SPECIFIC HEATS OF ARGON AND THEIR CHANGE WITH
 TEMPERATURE.)
 VEREINIGTEN FRIEDRICH'S UNIVERSITY, HALLE, PH. D. THESIS (1902)
- VELOCITY OF SOUND, SPECIFIC HEAT RATIO (GAS) (273 TO 287 DEGREES K
 AND 1 ATM)
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 GERMAN
282. NISSAN, A. H.
 VISCOSITY-TEMPERATURE FUNCTION OF LIQUIDS.
 PHIL. MAG. VOL. 32, 441-56 (1941)
- VISCOSITY (LIQUID) (83 TO 87 DEGREES K)
 OXYGEN, NITROGEN, AND APPROXIMATELY 75 OTHER ELEMENTS AND COMPOUNDS
 THEORETICAL - EQUATION
283. NOMOTO, O. AND KISHIMOTO, T.
 ON MOLECULAR SOUND VELOCITY OF LIQUIDS III. CONSIDERATION ON RAO'S
 RELATION ON THE DEPENDENCE OF MOLECULAR SOUND VELOCITY ON CRITICAL
 CONSTANTS.
 J. PHYS. SOC. JAPAN VOL. 9, 73-7 (1954)
 C.A. 49, 23 E
- VELOCITY OF SOUND (LIQUID) (83 TO 87 DEGREES K)
 HELIUM, HYDROGEN, NITROGEN, OXYGEN, METHANE AND 29 ORGANIC LIQUIDS
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284. NUTTALL, R. L.
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 NATL. BUR. STANDARDS, HEAT AND POWER DIV.
 REPRINTED IN TABLES OF THERMAL PROPERTIES OF GASES, HILSENARTH ET AL.,
 NATL. BUR. STANDARDS CIRC. NO. 564 (1955)
- THERMAL CONDUCTIVITY (GAS) (90 TO 600 DEGREES K)
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285. OISHI, J.
 0 AND 100 DEGREE ISOTHERMS OF HELIUM, HYDROGEN, NEON, ARGON, AIR AND
 CARBON DIOXIDE AT PRESSURES BELOW 2 ATMOSPHERES AND THE ABSOLUTE
 TEMPERATURE OF 0 DEGREE C.
 J. SCI. RESEARCH INST. (TOKYO) VOL. 43, 220-31 (1949)
 C.A. 43, 8772 D
- P-V-T DATA, COMPRESSIBILITY FACTOR, EXPANSION COEFFICIENT (GAS)
 (273 AND 373 DEGREES K AND 0.8 AND 1.2 ATM)
 HELIUM, HYDROGEN, NEON, AIR, AND CARBON DIOXIDE
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286. OLSZEWSKI, K.
 THE LIQUEFACTION AND SOLIDIFICATION OF ARGON.
 TRANS. ROY. SOC. (LONDON) VOL. A186, 253-7 (1895)
- VAPOR PRESSURE (134 TO 145 DEGREES K), CRITICAL CONSTANTS, BOILING AND FREEZING POINTS
 EXPERIMENTAL - TABLES (17 VALUES)
 APPARATUS
287. OLSZEWSKI, K.
 DIE UBERFUHRUNG DES ARGONS IN DEN FLUSSIGEN UND FESTEN ZUSTAND.
 (THE CONVERSION OF ARGON IN THE LIQUID AND SOLID STATE.)
 Z. PHYSIK. CHEM. VOL. 16, 380-4 (1895)
- VAPOR PRESSURE (134 TO 145 DEGREES K), CRITICAL CONSTANTS, BOILING POINT
 NITROGEN, HYDROGEN, OXYGEN, METHANE, CARBON MONOXIDE
 EXPERIMENTAL - TABLE (17 VALUES)
 GERMAN
288. O'NEAL, C. AND BROKAW, R. S.
 RELATION BETWEEN THERMAL CONDUCTIVITY AND VISCOSITY FOR SOME NONPOLAR GASES.
 PHYS. FLUIDS VOL. 5, NO. 5, 567-74 (1962)
 NASA N62 13935
 C.A. 57, 9233 C
- RATIO OF THERMAL CONDUCTIVITY TO VISCOSITY (GAS)(100 TO 275 DEGREES K)
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289. ONNES, H. K. AND CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC SUBSTANCES AND THEIR BINARY MIXTURES IX.
 THE BEHAVIOR OF ARGON WITH REGARD TO THE LAW OF CORRESPONDING STATES.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 120A (1911)
 VERSLAG. AFDEEL. NATUURK. AKAD. WETENSCHAP. AMSTERDAM 1177-87 (1911)
- P-V-T DATA (GAS)(124 TO 300 DEGREES K AND DENSITIES FROM 20 TO 336 AMAGATS), VIRIAL COEFFICIENTS (GAS) (123 TO 300 DEGREES K), LAW OF CORRESPONDING STATES
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290. ONNES, H. K. AND CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC GASES AND OF THEIR BINARY MIXTURES.VII.
 ISOTHERMS OF ARGON BETWEEN 20 AND -150 DEGREES C.
 PROC. ACAD. SCI. AMSTERDAM VOL. 13, 614-25 (1911)
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 118B (1911)
- P-V-T DATA, DENSITY, (GAS)(123 TO 300 DEGREES K AND 11 TO 61 ATM), EQUATION OF STATE, VIRIAL COEFFICIENTS (GAS)(152 TO 300 DEGREES K)
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291. ONNES, H. K. AND CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC SUBSTANCES AND OF THEIR BINARY MIXTURES. X. THE BEHAVIOR OF ARGON WITH RESPECT TO THE LAW OF CORRESPONDING STATES.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 121B (1911)
 VERSLAG. AFDEEL. NATUURK. KON. AKAD. WETENSCHAP. AMSTERDAM 68-72 (1911)
- CORRESPONDING STATES
 ISOPENTANE
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 CONTINUATION OF REFERENCE NUMBER 289
292. ONNES, H. K. AND CROMMELIN, C. A.
 ISOTHERMALS OF MONATOMIC SUBSTANCES AND OF THEIR BINARY MIXTURES. XIII. THE EMPIRICAL REDUCED EQUATION OF STATE FOR ARGON.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 128 (1912)
 VERSLAG GEWONE VERGADER. AFDEEL. NATUURK. NED. AKAD. WETENSCHAP. 256-63 (1912)
- EQUATION OF STATE, VIRIAL COEFFICIENTS (GAS) (124 TO 293 DEGREES K)
 THEORETICAL - TABLE (110 VALUES), GRAPHS
293. ONNES, H. K. AND CROMMELIN, C. A.
 ISOTHERMS OF MONATOMIC SUBSTANCES AND OF THEIR BINARY MIXTURES. XIV.
 CALCULATION OF SOME THERMAL QUANTITIES FOR ARGON.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 131C (1912)
 VERSLAG GEWONE VERGADER. WIS- EN NATUURK. AFDEEL. 893-8 (1912)
- P-V-T DATA (151 TO 293 DEGREES K AND 20 TO 340 ATM)
 THEORETICAL - EQUATIONS
294. OUDEMANS, G. J. AND COLE, R. H.
 DIELECTRIC CONSTANT AND PAIR INTERACTIONS IN GASEOUS HELIUM AND ARGON.
 J. CHEM. PHYS. VOL. 31, 843-4 (1959)
 C.A. 54, 5194 (1960)
- CLAUSIUS-MOSSOTTI FUNCTION (GAS) (1 TO 100 ATM)
 HELIUM
 EXPERIMENTAL - GRAPH
 'LETTER TO EDITOR'
295. OWENS, E. J.
 REDUCED THERMAL CONDUCTIVITY CORRELATION FOR THE INERT GASES.
 NORTHWESTERN UNIV., EVANSTON, ILL., MASTER'S THESIS (1956)
- THERMAL CONDUCTIVITY (15 TO 9000 DEGREES K), CRITICAL CONSTANTS,
 CORRESPONDING STATES PRINCIPLE
 CORRELATION - GRAPH, TABLE (20 VALUES)
296. OWENS, E. J. AND THODOS, G.
 THERMAL CONDUCTIVITY - REDUCED-STATE CORRELATION FOR THE INERT GASES.
 A. I. CH. E. JOURNAL VOL. 3, NO. 4, 454-61 (1957)
 C.A. 52, 3438 I
- THERMAL CONDUCTIVITY (GAS) (75 TO 4530 DEGREES K AND 0 TO 1920 ATM)
 HELIUM, NEON, XENON, KRYPTON
 CORRELATION - GRAPH, EQUATION

297. PARTINGTON, J. R.
 TEMPERATURABHANGIGKEIT DER INNEREN REIBUNG VON LUFT, ARGON, UND
 CHLORWASSERSTOFF.
 (THE VARIATION ON THE VISCOSITY OF AIR, ARGON AND HYDROGEN CHLORIDE
 WITH TEMPERATURE.)
 PHYSIK. Z. VOL. 34, 289-92 (1933)
 C.A. 27, 3649 6
- VISCOSITY (GAS)(296 TO 996 DEGREES K)
 AIR, HYDROGEN CHLORIDE
 REVIEW - EQUATION
298. PICKERING, S. F.
 A REVIEW OF THE LITERATURE RELATING TO THE CRITICAL CONSTANTS OF
 VARIOUS GASES.
 SCIENTIFIC PAPERS OF THE BUREAU OF STANDARDS VOL. 21, 597-629 (1926)
 J. PHYS. CHEM. VOL. 28, 97-123 (1924)
 C.A. 18, 1409 2 AND C.A. 21, 1036 2
- CRITICAL CONSTANTS
 AIR, CARBON MONOXIDE, HELIUM, HYDROGEN, METHANE, NEON, NITROGEN,
 OXYGEN, AND 25 OTHER GASES
 REVIEW - DATA FROM 82, 83, 258, 286, 308
299. PICKERING, S. F.
 COMPRESSIBILITIES OF GASES.
 NATL. BUR. STANDARDS MISCELLANEOUS PUBL. NO. 71 (NOV 21, 1925)
 C.A. 20, 853 4
- COMPRESSIBILITY FACTOR (GAS)(216 TO 473 DEGREES K AND 1 TO 99 ATM)
 HELIUM, AIR, HYDROGEN, METHANE, NEON, NITROGEN, OXYGEN
 COMPILATION - GRAPHICAL DATA FROM 179, 289, 308
300. PITZER, K. S.
 CORRESPONDING STATES FOR PERFECT LIQUIDS.
 J. CHEM. PHYS. VOL. 7, 583-90 (1939)
- VAPOR PRESSURE, DENSITY, CORRESPONDING STATES THEORY, ENTROPY OF
 VAPORIZATION
 NEON, METHANE, XENON, KRYPTON
 CORRELATION - EQUATION, GRAPHS
301. POOL, R. A. H., SHIELDS, B. D. C. AND STAVELEY, L. A. K.
 THE TRIPLE POINT OF ARGON AS A THERMOMETRIC FIXED POINT.
 NATURE VOL. 181, 831 (1958)
- TRIPLE POINT TEMPERATURE AND PRESSURE
 EXPERIMENTAL - 2 VALUES
 'LETTER TO EDITOR'
-

302. POPOVICI, S. AND POP, M.
 SUR LE POINT D'EBULLITION ET LES FORCES INTERMOLECULAIRES.
 (THE BOILING POINT AND INTERMOLECULAR FORCES.)
 COMPT. REND. VOL. 244, 195-8 (1957)
 C.A. 51, 7084
- BOILING POINT, INTERMOLECULAR FORCES
 HELIUM, NEON, FLUORINE, AND 45 OTHER LIQUIDS
 THEORETICAL - ONE VALUE FOR RATIO OF BOILING POINT TO MOLECULAR WEIGHT
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303. POPOVICI, S. AND POP, M.
 SUR LE POINT DE FUSION ET LES FORCES INTERMOLECULAIRES.
 (THE MELTING POINT AND INTERMOLECULAR FORCES.)
 COMPT. REND. VOL. 246, 2609-11 (1958)
- MELTING POINT, INTERMOLECULAR FORCES
 HELIUM, NEON, FLUORINE, OXYGEN, NITROGEN AND 34 OTHER SOLIDS
 THEORETICAL - ONE VALUE FOR RATIO OF MELTING POINT TO MOLECULAR WEIGHT
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304. PRIGOGINE, I. AND RAULIER, S.
 CHALEUR SPECIFIQUE A VOLUME CONSTANT DES LIQUIDES MONATOMIQUES.
 (THE SPECIFIC HEAT OF MONATOMIC LIQUIDS AT CONSTANT VOLUME.)
 PHYSICA VOL. 9, NO. 4, 396-404 (1942)
 C.A. 37, 6188 2
- SPECIFIC HEAT (V=CONSTANT) (LIQUID, GAS) (90 TO 190 DEGREES K)
 MERCURY
 THEORETICAL - EQUATIONS, TABLE (7 VALUES)
 FRENCH
305. PRIGOGINE, I. AND SARAGA, L.
 SUR LA TENSION SUPERFICIELLE ET LE MODELE CELLULAIRE DE L'ETAT LIQUIDE.
 (SURFACE TENSION AND THE CELL MODEL OF THE LIQUID STATE.)
 J. CHIM. PHYS. VOL. 49, 399-407 (1952)
 C.A. 51, 17300 C
- SURFACE TENSION, SURFACE ENTROPY, SURFACE ENERGY (LIQUID)
 THEORETICAL - EQUATIONS, GRAPHS, ONE TABULAR VALUE
 FRENCH
306. QUARDER, B.
 UBER LICHTBRECHUNG IN GASSEN IM SPEKTRALBEREICH VON 5782 BIS 2442 A.
 (THE REFRACTION OF LIQUID IN GASES IN THE SPECTRAL REGION FROM 5782 TO 2442 ANGSTROMS.)
 ANN. PHYSIK VOL. 74, 255-74 (1924)
- REFRACTIVE INDEX (GAS)
 AIR
 EXPERIMENTAL - TABLE (15 VALUES)
 GERMAN

307. RAMSAY, W. AND TRAVERS, M. W.
 THE PREPARATION AND SOME OF THE PROPERTIES OF PURE ARGON.
 PROC. ROY. SOC. (LONDON) VOL. 64, 183-92 (1898)
 DENSITY (GAS) (285 TO 292 DEGREES K AND 535 TO 767 MM HG),
 REFRACTIVITY
 EXPERIMENTAL - TABLE (16 VALUES)
308. RAMSAY, W. AND TRAVERS, M. W.
 II. ARGON AND ITS COMPANIONS.
 PHIL. TRANS. ROY. SOC. LONDON SER. A. VOL. 197, 47-91 (1901)
 Z. PHYSIK. CHEM. VOL. 38, 641-89 (1901)
 VAPOR PRESSURE (SOLID, LIQUID) (77 TO 155.7 DEGREES K), P-V-T DATA
 (GAS) (284 AND 510 DEGREES K), CRITICAL POINT
 KRYPTON, XENON
 EXPERIMENTAL - TABLE (20 VALUES)
 APPARATUS
309. RANKINE, A. O.
 ON THE VISCOSITIES OF THE GASES OF THE ARGON GROUP.
 PROC. ROY. SOC. (LONDON) VOL. A83, 516-25 (1910)
 PHYSIK. Z. VOL. 11, 497-502 (1910)
 VISCOSITY (GAS) (285 DEGREES K)
 HELIUM, KRYPTON, XENON, NEON, AIR
 EXPERIMENTAL - ONE TABULAR VALUE, GRAPH
 APPARATUS
310. RAYLEIGH
 ON SOME PHYSICAL PROPERTIES OF ARGON AND HELIUM.
 PROC. ROY. SOC. (LONDON) VOL. 59, 198-209 (1896)
 DENSITY, VISCOSITY (GAS) (273 DEGREES K)
 HELIUM
 EXPERIMENTAL - TWO TABULAR VALUES
 APPARATUS
311. RAYLEIGH AND RAMSAY, W.
 A NEW CONSTITUENT OF THE ATMOSPHERE.
 PHIL. TRANS. VOL. 186, 238-41 (1895)
 DENSITY (GAS) (286 TO 291 DEGREES K AND 1 ATM)
 EXPERIMENTAL - TABLE (6 VALUES)
312. REID, R. C. AND VALBERT, J. R.
 DERIVATIVE COMPRESSIBILITY FACTORS.
 IND. ENG. CHEM. FUNDAMENTALS VOL. 1, NO. 4, 292-8 (1962)
 COMPRESSIBILITY FACTOR (REDUCED TEMPERATURE = 0.8 TO 4 AND REDUCED
 PRESSURE = 0.2 TO 9)
 CORRESPONDING STATE'S PRINCIPLE
 ALL GASES
 THEORETICAL - EQUATION

313. RICE, O. K.
 THE SOLID-LIQUID EQUILIBRIUM IN ARGON.
 J. CHEM. PHYS. VOL. 6, 472-5 (1938)
 MELTING PRESSURE (SOLID) (83 TO 193 DEGREES K)
 EXPERIMENTAL - GRAPHS, TABLE (10 VALUES)
314. RICE, O. K.
 THE INTERATOMIC POTENTIAL CURVE AND THE EQUATION OF STATE FOR ARGON.
 J. AM. CHEM. SOC. VOL. 63, 3-11 (1941)
 INTERATOMIC POTENTIAL (SOLID), HEAT OF SUBLIMATION (11 AND 16
 DEGREES K)
 THEORETICAL - EQUATIONS, GRAPH
315. RICE, O. K.
 A NOTE ON THE ENTROPY OF FUSION FOR ARGON.
 J. CHEM. PHYS. VOL. 9, 121 (1941)
 ENTROPY OF FUSION, PARTITION FUNCTION
 THEORETICAL - EQUATIONS
316. RICE, O. K.
 THE THERMODYNAMIC PROPERTIES AND POTENTIAL ENERGY OF SOLID ARGON.
 J. CHEM. PHYS. VOL. 12, NO. 7, 289-95 (1944)
 SPECIFIC HEAT (P=CONSTANT), ENTROPY, POTENTIAL ENERGY (SOLID) (3 TO
 83 DEGREES K), INTERATOMIC POTENTIAL, HEAT OF SUBLIMATION (0 DEGREE K)
 THEORETICAL - TABLE (40 VALUES), GRAPH
317. RICE, O. K.
 THE THERMODYNAMIC PROPERTIES OF LIQUID ARGON.
 J. CHEM. PHYS. VOL. 14, 324-38 (1946)
 C.A. 40, 4271 4
 ENTROPY (SOLID, LIQUID) (83 TO 193 DEGREES K), SPECIFIC HEAT (V=CONSTANT)
 (LIQUID, GAS) (83 TO 320 DEGREES K), SPECIFIC HEAT (SATURATION) (83 TO
 101 DEGREES K), POTENTIAL ENERGY EQUATION, PARTITION FUNCTION
 THEORETICAL - EQUATIONS, GRAPHS, TABLE (30 VALUES)
- 317A. RIEDEL, L.
 EINE NEUE UNIVERSELLE DAMPFDRUCKFORMEL. UNTERSUCHUNGEN UBER EINE
 ERWEITERUNG DES THEOREMS DER UBEREINSTIMMENDEN ZUSTANDE. I.
 (A NEW UNIVERSAL VAPOR PRESSURE EQUATION. INVESTIGATION CONCERNING THE
 EXTENSION OF THE THEORY OF CORRESPONDING STATES. I.)
 CHEM. ING. TECH. VOL. 26, NO. 2, 83-89 (1954)
 VAPOR PRESSURE, CORRESPONDING STATES THEORY
 OXYGEN, CARBON MONOXIDE, NEON, HYDROGEN, HELIUM
 THEORETICAL - EQUATION, TABLE
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317B. RIEDEL, L.

DIE FLUSSIGKEITSDICHE IM SATTIGUNGZUSTAND. UNTERSUCHUNGEN UBER EINE ERWEITERUNG DES THEOREMS DER UBEREINSTIMMENDEN ZUSTANDE. II.
 (THE LIQUID DENSITY IN SATURATED CONDITION. INVESTIGATION CONCERNING THE EXTENSION OF THE CORRESPONDING STATES THEORY. II.
 CHEM. ING. TECH. VOL. 26, NO. 5, 259-64 (1954)

SATURATED DENSITY (LIQUID) (87 TO 136 DEGREES K, EXPANSION COEFFICIENT (LIQUID) (87 TO 106 DEGREES K), CORRESPONDING STATES THEORY
 METHANE, NITROGEN, HYDROGEN, NEON, CARBON MONOXIDE, HELIUM
 THEORETICAL - EQUATION, TABLES (100 VALUES), GRAPHS
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318. RIEDEL, L.

DIE BERECHNUNG UNBEKANNTER THERMISCHEN DATEN MIT HILFE DES ERWEITERTEN KORRESPONDENZPRINZIPS.
 (THE DETERMINATION OF UNKNOWN THERMAL DATA BY THE MODIFIED THEOREM OF CORRESPONDING STATES.)
 KALTETECHNIK VOL. 9, 127-34 (1957)

COMPRESSIBILITY FACTOR, DENSITY, HEAT OF VAPORIZATION, SURFACE TENSION, COEFFICIENTS FOR VAPOR PRESSURE EQUATION (REDUCED TEMPERATURE OF 0.4 TO 1.0), THEORY OF CORRESPONDING STATES, MELTING AND BOILING POINTS, CRITICAL CONSTANTS, ZERO POINT DENSITY
 OXYGEN, NITROGEN, CARBON MONOXIDE, METHANE AND 55 OTHER COMPOUNDS
 CORRELATION - EQUATIONS, TABLES (600 VALUES), GRAPHS
 GERMAN

319. ROBINSON, D. W.

AN EXPERIMENTAL DETERMINATION OF THE MELTING CURVES OF ARGON AND NITROGEN INTO THE 10,000 ATM REGION.
 PROC. ROY. SOC. (LONDON) VOL. A255, 393-405 (1954)
 C.A. 48, 13315 A

MELTING CURVE (SOLID) (83 TO 234 DEGREES K)
 NITROGEN
 EXPERIMENTAL - TABLE (10 VALUES), GRAPHS
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320. ROBINSON, D. W.

THE VISCOSITY OF ARGON, HELIUM AND NITROGEN AT LOW TEMPERATURES AND HIGH PRESSURES.
 BULL. INST. INTERN. FROID ANNEXE (1955) 329-32
 C.A. 53, 15680 F

VISCOSITY (LIQUID) (90 DEGREES K AND 1 TO 2000 ATM)
 HELIUM, NITROGEN
 EXPERIMENTAL - GRAPH, EQUATION

321. ROEBUCK, J. R. AND OSTERBERG, H.

THE JOULE-TOMSON EFFECT IN ARGON.
 PHYS. REV. VOL. 46, 785-90 (1934)

JOULE-TOMSON COEFFICIENT (GAS) (123 TO 573 DEGREES K AND 1 TO 200 ATM)
 EXPERIMENTAL - TABLE (140 VALUES), GRAPHS

322. ROGOVAYA, I. A. AND KAGANER, M. G.
 COMPRESSIBILITY OF ARGON AT LOW TEMPERATURES UP TO 200 ATMOSPHERES.
 RUSS. J. PHYS. CHEM. VOL. 35, 1049-50 (1961) (ENGLISH TRANSLATION)
 ZHUR. FIZ. KHM. VOL. 35, 2135-6 (1961)
 C.A. 56, 947 G
- ISOETHERMAL COMPRESSIBILITY, DENSITY (GAS)(90 TO 249 DEGREES K AND 24
 TO 194 ATM), VAPOR PRESSURE (143 DEGREES K)
 EXPERIMENTAL - TABLE (80 VALUES)
323. ROSSINI, F. D., WAGMAN, D. D., EVANS, W. R., LEVINE, S. AND JAFFE, I.
 SELECTED VALUES OF CHEMICAL THERMODYNAMIC PROPERTIES.
 NATL. BUR. STANDARDS (U.S.) CIRC. 500, REPRINTED IN 1961 IN TWO PARTS
 AND AVAILABLE FROM U.S. GOVT. PRINTING OFFICE, WASHINGTON 25, D.C.
 PART 1, CATALOG NO. C13.4.500/PT. 1, PART 2, CATALOG NO. C13.4.500/
 PT. 2
- ENTHALPY, FREE ENERGY, HEAT OF FORMATION AT 0 DEGREE K, ENTROPY AND
 HEAT CAPACITY AT 298.16 DEGREES K, ENTHALPY, TEMPERATURE AND ENTROPY
 OF TRANSITION, FUSION AND VAPORIZATION
 REFERENCE WORK
324. ROWLINSON, J. S.
 THE REDUCED EQUATION OF STATE.
 TRANS. FARADAY SOC. VOL. 51, 1317-25 (1955)
 C.A. 50, 6148G
- COMPRESSIBILITY FACTOR (GAS)(REDUCED TEMPERATURE = 0.8 TO 8)
 XENON, NEON
 CORRELATION + TABLE (640 VALUES)
325. ROWLINSON, J. S. AND TOWNLEY, J. R.
 THE APPLICATION OF THE PRINCIPLE OF CORRESPONDING STATES TO THE
 TRANSPORT PROPERTIES OF GASES.
 TRANS. FARADAY SOC. VOL. 49, 20-7 (1953)
- VISCOSITY (GAS)(62 TO 240 DEGREES K)
 NEON, NITROGEN, OXYGEN, METHANE, CARBON MONOXIDE AND 12 OTHER GASES
 CORRELATION - EQUATION, GRAPH
326. RUDENKO, N. S.
 MOLECULAR WEIGHT, DENSITY AND VISCOSITY OF LIQUEFIED GASES.
 ZHUR. TEKH. FIZ. VOL. 18, 1123-6 (1948)
- VISCOSITY (LIQUID)(84 TO 151 DEGREES K)
 OXYGEN, NITROGEN, CARBON MONOXIDE, METHANE, HYDROGEN, DEUTERIUM,
 ETHENE AND CHLORINE
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327. RUDENKO, N. S. AND SCHUBNIKOV, L. W.
 DIE VISKOSITAT VON FLUSSIGEM STICKSTOFF, KOHLENOXYD, ARGON UND
 SAUERSTOFF IN ABHANGIGKEIT VON DER TEMPERATUR.
 (VISCOSITY OF LIQUID NITROGEN, CARBON MONOXIDE, ARGON AND OXYGEN AND
 ITS DEPENDENCE ON TEMPERATURE.)
 PHYSIK. Z. SOWJETUNION VOL. 6, 470-7 (1934)
 C.A. 29, 2040 4
- VISCOSITY (LIQUID)(84 TO 87 DEGREES K)
 NITROGEN, CARBON MONOXIDE, OXYGEN
 EXPERIMENTAL - TABLE (4 VALUES), GRAPH
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 GERMAN
328. RUDORF, G.
 DIE EDELGASE UND DIE ZUSTANDSGEICHUNG.
 (THE NOBLE GASES AND THE EQUATION OF STATE.)
 ANN. PHYSIK VOL. 29, 751-79 (1909)
- P-V-T DATA (GAS)(284.2 AND 510.3 DEGREES K), VAPOR PRESSURE
 (85 DEGREES K), BOILING POINT, CRITICAL POINT, HEAT OF VAPORIZATION,
 EQUATION OF STATE
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 CORRELATION - TABLE (30 VALUES), EQUATIONS
 GERMAN
329. RYABININ, YU. N., MARKEVICH, A. M. AND TAMM, I. I.
 COMPRESSIBILITY OF ARGON UNDER STRONG ADIABATIC COMPRESSION.
 ZHUR. EKSPTL. I TEORET. FIZ. VOL. 24, 107-13 (1953) (IN RUSSIAN)
 C.A. 49, 3589 A
- DENSITY, ADIABATIC COMPRESSIBILITY (GAS)(2000 TO 7000 KG/SQ CM)
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 RUSSIAN
330. RYKOV, V. I.
 TEMPERATURE DEPENDENCE OF HEATS OF VAPORIZATION OF A NON-ASSOCIATED
 LIQUID. (IN RUSSIAN)
 ZHUR. FIZ. KHM. VOL. 34, 1851-55 (1960)
 C.A. 55, 12019 B
- HEATS OF VAPORIZATION (90 TO 140 DEGREES K)
 NITROGEN
 THEORETICAL - EQUATIONS, TABLE (5 VALUES)
 RUSSIAN
331. SAKIADIS, C. AND COATES, J.
 A LITERATURE SURVEY OF THE THERMAL CONDUCTIVITY OF LIQUIDS.
 LOUISIANA STATE UNIV. ENG. EXPT. STA., BATON ROUGE, LA., BULL. NO. 34,
 1-70 (1952)
- THERMAL CONDUCTIVITY (LIQUID)(88 TO 198 DEGREES K)
 NITROGEN, OXYGEN AND 150 OTHER ORGANIC AND INORGANIC COMPOUNDS
 SURVEY - DATA FROM 374

332. SCHAFER, K.
 PROBLEME DER WARMELEITUNG IN GASEN BEI NIEDRIGEM DRUCK UND DER
 ENERGIEUBERTRAGUNG AN FESTEN OBERFLÄCHEN.
 (PROBLEMS CONCERNING THERMAL CONDUCTIVITY IN GASES AT LOW PRESSURES
 AND THE TRANSFER OF ENERGY ON SOLID SURFACES.)
 FORTSCHR. CHEM. FORSCH. VOL. 1, 61-118 (1949)
- THERMAL CONDUCTIVITY (GAS) (276.5 DEGREES K AND 0.1 TO 593.3 MM HG)
 ACCOMMODATION COEFFICIENT (GAS) (195 TO 295 DEGREES K)
 HYDROGEN, AIR, NITROGEN, OXYGEN, CARBON MONOXIDE, HELIUM, NEON
 DISCUSSION - DATA FROM 121, EQUATIONS, GRAPHS
 GERMAN
333. SCHAFER, K.
 TRANSPORT PHENOMENA IN THE TEMPERATURE RANGE UP TO 1100 DEGREES.
 DECHEMA MONOGRAPH VOL. 32, 61-73 (1959)
 C.A. 54, 1002 H
- THERMAL CONDUCTIVITY (GAS) (273 TO 1373 DEGREES K)
 NITROGEN, OXYGEN, CARBON MONOXIDE, METHANE, HYDROGEN, CARBON DIOXIDE
 AND KRYPTON
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 -CODED FROM ABSTRACT-
334. SCHAMES, L.
 UBER DEN DIREKTEN ZUSAMMENHANG VON ZUSTANDSGEICHUNG UND INNERER
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 (DIRECT RELATIONSHIP OF EQUATION OF STATE AND INTERNAL FRICTION.)
 PHYSIK. Z. VOL. 32, 16-20 (1931)
 C.A. 25, 1130 7
- EQUATION OF STATE, VIRIAL COEFFICIENTS (GAS) (173 TO 673 DEGREES K)
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335. SCHIERLOH, J. F.
 UBER DEN KOEFFIZIENTEN DER INNEREN REIBUNG VON REINEM ARGON UND
 REINEM HELIUM.
 (CONCERNING THE COEFFICIENT OF VISCOSITY OF PURE ARGON AND PURE
 HELIUM.)
 VEREINIGTEN FRIEDRICH'S UNIVERSITY, HALLE, PH. D. THESIS (1909)
- VISCOSITY (GAS) (288 DEGREES K AND 1 ATM)
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 EXPERIMENTAL - TABLE (20 VALUES)
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-

336. SCHMITT, K.
 UBER DIE INNERE REIBUNG EINIGER GASE UND GASGEMISCHE BEI
 VERSCHIEDENEN TEMPERATUREN.
 (CONCERNING THE INTERNAL FRICTION OF SOME GASES AND GAS MIXTURES AT
 DIFFERENT TEMPERATURES.)
 ANN. PHYSIK VOL. 30, 393-410 (1909)
- VISCOSITY (GAS) (90 TO 456 DEGREES K AND 73 TO 76 CM HG)
 HELIUM, HYDROGEN, AIR, OXYGEN
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337. SCHMOLKE
 VORRICHTUNGEN ZUR FESTSTELLUNG DER SCHMELZKURVEN NIEDRIGSIEDENDER
 STOFFE.
 (APPARATUS FOR DETERMINING THE FUSION CURVES OF LOW-BOILING
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 DIE WARME VOL. 54, 97-8 (1931)
 C.A. 25, 1708 2
- MELTING CURVE (83 AND 150 DEGREES K)
 HYDROGEN, NITROGEN, NEON
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338. SCHULTZE, H.
 UBER DAS SPEZIFISCHE GEWICHT DES ARGONS.
 (CONCERNING THE SPECIFIC GRAVITY OF ARGON.)
 ANN. PHYSIK. VOL. 48, 269-72 (1915)
- SPECIFIC GRAVITY (GAS) (273 DEGREES K AND 1 ATM). ATOMIC WEIGHT
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 EXPERIMENTAL - TABLE (5 VALUES)
 GERMAN
339. SCHULTZE, H.
 DIE INNERE REIBUNG VON ARGON UND IHRE AENDERUNG MIT DER TEMPERATUR.
 (THE VISCOSITY OF ARGON AND ITS CHANGE WITH TEMPERATURE.)
 ANN. PHYSIK VOL. 5, 140-65 (1901)
- VISCOSITY (GAS) (285 TO 456 DEGREES K AND 75 CM HG)
 EXPERIMENTAL - TABLE (25 VALUES)
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340. SCHWARZE, W.
 UBER DIE WARMELEITUNG DES ARGONS.
 (CONCERNING THE HEAT CONDUCTION OF ARGON.)
 PHYSIK. Z. VOL. 3, 264 (1902)
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K)
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341. SCHWARZE, W.
 UBER DIE WARMELEITUNG VON ARGON U. HELIUM.
 (CONCERNING THE HEAT CONDUCTION OF ARGON AND HELIUM.)
 PHYSIK. Z. VOL. 4, 229 (1903)
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K)
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342. SCHWARZE, W.
 BESTIMMUNG DER WARMELEITUNGSFAHIGKEIT VON ARGON UND HELIUM NACH DER
 METHODE VON SCHLEIERMACHER.
 (DETERMINATION OF THE THERMAL CONDUCTIVITY OF ARGON AND HELIUM BY THE
 METHOD OF SCHLEIERMACHER.)
 ANN. PHYSIK VOL. 11, 303-30 (1903)
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K)
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 GERMAN
343. SHERIF, I. I.
 THE MOLECULAR HEATS OF GASES FROM THE ASPECTS OF HEAT TRANSFER.
 NUOVO CIMENTO VOL. 3, 6-11 (1956)
- ACCOMMODATION COEFFICIENT, MOLECULAR HEAT (GAS) (287 DEGREES K)
 HYDROGEN, NITROGEN, OXYGEN, HELIUM, NEON
 EXPERIMENTAL - TWO TABULAR VALUES
344. SHIMOKAWA, J.
 CONDENSATION THEORY BY CELL METHOD AND CALCULATION OF THE CRITICAL
 TEMPERATURE OF VARIOUS GASES.
 BUSSEIRON KENKYU VOL. 62, 138-51 (1953)
- CRITICAL TEMPERATURE, PARTITION FUNCTION, CELL-SURFACE TENSION,
 CELL-SURFACE FREE ENERGY
 HELIUM, HYDROGEN, NITROGEN, OXYGEN, METHANE
 THEORETICAL - ONE CALCULATED VALUE, EQUATIONS
 JAPANESE
345. SHIMOTAKE, H.
 VISCOSITY-REDUCED STATE CORRELATION FOR THE INERT GASES.
 NORTHWESTERN UNIV., EVANSTON, ILL., MASTER'S THESIS (JUN 1957)
- VISCOSITY (45 TO 15000 DEGREES K), PRINCIPLE OF CORRESPONDING STATES
 NEON, HELIUM, KRYPTON, XENON
 CORRELATION - GRAPH, TABLE
-

346. SIMON, F. AND KIPPERT, F.
 MESSUNGEN ZUR ZUSTANDSGLEICHUNG DES FESTEN ARGONS.
 (MEASUREMENTS FOR THE EQUATIONS OF STATE OF SOLID ARGON.)
Z. PHYSIK. CHEM. VOL. 135, 113-28 (1928)
- ISOMETRIC SLOPE (SOLID, LIQUID) (72 TO 90 DEGREES K), DENSITY (SOLID)
 (76 DEGREES K)
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347. SIMON, F. E., RUHEMANN, M. AND EDWARDS, W. A. M.
 DIE SCHMELZKURVEN VON WASSERSTOFF, NEON, STICKSTOFF UND ARGON.
 (MELTING-POINT CURVES OF HYDROGEN, NEON, NITROGEN AND ARGON.)
Z. PHYSIK. CHEM. (LEIPZIG) VOL. B6, 331-42 (1930)
 C.A. 24, 3411 2
- MELTING CURVE (SOLID, LIQUID) (83 TO 150 DEGREES K AND 0 TO 3380
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348. SIMON, F. E., RUHEMANN, M. AND EDWARDS, W. A. M.
 DIE SCHMELZKURVEN VON WASSERSTOFF, NEON, STICKSTOFF UND ARGON.
 (BERICHTIGUNG)
 (FUSION CURVES FOR HYDROGEN, NEON, NITROGEN AND ARGON. CORRECTION)
Z. PHYSIK. CHEM. (LEIPZIG) VOL. B7, 80 ONLY (1930)
 C. A. 24, 4681
- MELTING CURVE
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 CORRECTION TO REFERENCE NUMBER 347
 GERMAN
349. SIMON, F. AND VON SIMSON, C.
 DIE KRISTALLSTRUKTUR DES ARGONS.
 (THE CRYSTAL STRUCTURE OF ARGON)
Z. PHYSIK VOL. 25, 160-4 (1924)
- CRYSTAL STRUCTURE
 EXPERIMENTAL
 GERMAN
350. SMITH, B. L.
 REFRACTIVE INDEX OF SOLID KRYPTON AND SOLID ARGON.
PHIL. MAG. VOL. 6, 939-42 (JUL 1961)
 C.A. 56, 9569 H
- DENSITY (SOLID) (77 TO 83 DEGREES K), REFRACTIVE INDEX
 KRYPTON
 EXPERIMENTAL - GRAPH

351. SMITH, B. L. AND PINGS, C. J.
OPTICAL DETERMINATION OF THE COMPRESSIBILITY OF SOLID ARGON.
J. CHEM. PHYS. VOL. 38, 825-7 (FEB 1963)

COEFFICIENT OF THERMAL EXPANSION (SOLID)(78 TO 84 DEGREES K)
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352. SMITH, B. L. AND PINGS, C. J.
REFRACTIVE INDEX OF SOLID ARGON.
PHYSICA VOL. 29, NO. 5, 555-8 (MAY 1963)

REFRACTIVE INDEX (SOLID)(78 TO 84 DEGREES K AND 0 TO 71 ATM)
EXPERIMENTAL - TABLE (40 VALUES), GRAPH, EQUATION

- 352A. SMITH, D. H. AND HARLOW, R. G.
THE VELOCITY OF SOUND IN AIR, NITROGEN AND ARGON.
BRIT. J. APPL. PHYS. VOL. 14, NO. 2, 102-06 (1963)

VELOCITY OF SOUND (GAS)(273 AND 303 DEGREES K)
AIR, NITROGEN
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353. SRIVASTAVA, B. N. AND MADAN, M. P.
THE TEMPERATURE DEPENDENCE OF VISCOSITY OF NON-POLAR GASES.
PROC. NATL. ACAD. SCI. INDIA VOL. 21A, 254-60 (1952)
C.A. 48, 13307 I

INTERMOLECULAR FORCE CONSTANT, COLLISION DIAMETER (GAS)(100 TO 600
DEGREES K)
OXYGEN, NITROGEN, NEON
THEORETICAL - EQUATIONS, TABLE (18 VALUES), GRAPH

354. STANSFIELD, D.
THE SURFACE TENSIONS OF LIQUID ARGON AND NITROGEN.
PROC. PHYS. SOC. (LONDON) VOL. 72, 854-66 (1958)
C.A. 54, 20388 G

SURFACE TENSION (LIQUID)(85 TO 150 DEGREES K)
NITROGEN
EXPERIMENTAL - TABLE (14 VALUES), GRAPH
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355. STAVELEY, L. A. D. AND TUPMAN, W. I.
ENTROPIES OF VAPORIZATION AND INTERNAL ORDER IN LIQUIDS.
J. CHEM. SOC. (LONDON) VOL. 1950, 3597-606 (1950)

ENTROPY OF VAPORIZATION
METHANE, OXYGEN, CARBON MONOXIDE AND 50 OTHER GASES
THEORETICAL - GRAPH, EQUATIONS

356. STEWART, J. W.
 COMPRESSIBILITIES OF SOME SOLIDIFIED GASES AT LOW TEMPERATURES.
 PHYS. REV. VOL. 97, NO. 3, 578-82 (1955)
 C.A. 49, 6676 E
- ISOHERMAL COMPRESSIBILITY (SOLID)(65 AND 77 DEGREES K AND 0 TO
 4000 KG/SQ CM)
 HYDROGEN, DEUTERIUM, NEON, KRYPTON, NITROGEN
 EXPERIMENTAL - TABLE (8 VALUES)
 APPARATUS
357. STEWART, J. W.
 COMPRESSION OF SOLIDIFIED GASES TO 20,000 KG/SQ CM AT LOW TEMPERATURE.
 J. PHYS. CHEM. SOLIDS VOL. 1, 146-58 (1956)
- ISOHERMAL COMPRESSIBILITY (SOLID)(65 AND 77 DEGREES K AND 0 TO
 19000 ATM)
 NITROGEN, HYDROGEN, DEUTERIUM, NEON, HELIUM
 EXPERIMENTAL - TABLE (22 VALUES)
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358. STIEL, L. I. AND THODOS, G.
 LENNARD-JONES FORCE CONSTANTS PREDICTED FROM CRITICAL PROPERTIES.
 CHEM. ENG. DATA VOL. 7, NO. 2, 234-6 (APR 1962)
- THERMAL CONDUCTIVITY, SELF-DIFFUSIVITY, VISCOSITY, LENNARD-JONES FORCE
 CONSTANTS, CRITICAL CONSTANTS
 THEORETICAL - GRAPHS
359. STOGRYN, D. E. AND HIRSCHFELDER, J. O.
 INITIAL PRESSURE DEPENDENCE OF THERMAL CONDUCTIVITY AND VISCOSITY.
 J. CHEM. PHYS. VOL. 31, NO. 6, 1545-54 (1959)
- THERMAL CONDUCTIVITY, VISCOSITY (GAS)(104 TO 273 DEGREES·K)
 OXYGEN, NITROGEN, METHANE
 THEORETICAL - EQUATIONS, TABLE(3 VALUES), GRAPH
360. STRIEDER, F.
 DIE SCHALLGESCHWINDIGKEIT IN ARGON UND DER EINFLUSS DER
 RONTGENSTRAHLEN AUF DIESE.
 (VELOCITY OF SOUND IN ARGON AND THE INFLUENCE OF X-RAYS UPON IT.)
 VERHANDL. DEUT. PHYS. GES. VOL. 16, 615-6 (1914)
- VELOCITY OF SOUND (GAS)(291 DEGREES K)
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361. STRIEDER, F.
 DIE SCHALLGESCHWINDIGKEIT IN ARGON UND IHRE BEEINFLUSSUNG DURCH
 RONTGENSTRAHLEN.
 (SOUND VELOCITY IN ARGON AND THE INFLUENCE OF X RAYS ON IT.)
 UNIVERSITY OF MARBURG, PH. D. THESIS (1915)
- VELOCITY OF SOUND, SPECIFIC HEAT RATIO (GAS)(291 DEGREES K)
 AIR
 EXPERIMENTAL - TABLE (30 VALUES)
 APPARATUS
 GERMAN

362. STULL, D. R.
 VAPOR PRESSURE OF PURE SUBSTANCES. INORGANIC COMPOUNDS.
 IND. ENG. CHEM. VOL. 39, 540-50 (1947)
 VAPOR PRESSURE (SOLID, LIQUID) (55 TO 149 DEGREES K), CRITICAL CONSTANTS
 FLUORINE, HELIUM, HYDROGEN, CARBON MONOXIDE, NEON, NITROGEN, OXYGEN,
 OZONE
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363. STULL, D. R. AND SINKE, G. C.
THERMODYNAMIC PROPERTIES OF THE ELEMENTS.
 ADVANCES IN CHEMISTRY SERIES NO. 18, AMERICAN CHEMICAL SOCIETY,
 WASHINGTON, D. C. (NOV 1956) 233 P.
 IDEAL MONATOMIC AND DIATOMIC GAS PROPERTIES - HEAT CAPACITY
 $(P=CONSTANT)$, ENTHALPY, ENTROPY, GIBBS FUNCTION (298 TO 3000 DEGREES K)
 BOILING AND MELTING POINTS, TRIPLE POINT, CRITICAL TEMPERATURE AND
 PRESSURE, HEATS OF MELTING AND VAPORIZATION
 ARGON AND 91 OTHER ELEMENTS
 REFERENCE BOOK
364. SU, G. AND CHANG, C. H.
 A GENERALIZED VAN DER WAALS EQUATION OF STATE FOR REAL GASES.
 IND. ENG. CHEM. VOL. 38, 800-2 (1946)
 C.A. 40, 5969 2
 EQUATION OF STATE (GAS) (123 TO 673 DEGREES K AND 0 TO 114 ATM)
 HELIUM, NEON, HYDROGEN, NITROGEN, OXYGEN, AIR, METHANE, CARBON
 DIOXIDE AND ETHYL ETHER
 THEORETICAL - EQUATION
- 364A. SU, G-J AND CHANG, C-H.
 GENERALIZED BEATTIE-BRIDGEMAN EQUATION OF STATE FOR REAL GASES.
 J. AM. CHEM. SOC. VOL. 68, 1080-83 (1946)
 C.A. 40, 5312
 EQUATION OF STATE (GAS) (123 TO 673 DEGREES K)
 NEON, OXYGEN, NITROGEN, AIR, METHANE, HYDROGEN, HELIUM
 THEORETICAL - EQUATION, TABLE
365. SU, G. AND CHANG, C.
 GENERALIZED EQUATION OF STATE FOR REAL GASES.
 IND. ENG. CHEM. VOL. 38, 802-3 (1946)
 C.A. 40, 5969 5
 EQUATION OF STATE (GAS) (123 TO 673 DEGREES K AND 0 TO 114 ATM)
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 THEORETICAL - EQUATION, GRAPH

- 365A. TAUSZ, J. AND GORLACHER, H.
 UBER DIE LICHTBRECHUNG IN WASSERSTOFF, SAUERSTOFF, ARGON UND
 STICKSTOFF. I.
 (CONCERNING THE REFRACTIONS OF LIGHT IN HYDROGEN, OXYGEN, ARGON AND
 NITROGEN. I.)
 Z. TECH. PHYSIK VOL. 12, 19-24 (1931)
 INDEX OF REFRACTION (GAS) (289 TO 296 DEGREES K)
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- 365B. TAUSZ, J. AND GORLACHER, H.
 UBER DIE LICHTBRECHUNG IN WASSERSTOFF, SAUERSTOFF, ARGON UND
 STICKSTOFF. II.
 (THE REFRACTION OF LIGHT IN HYDROGEN, OXYGEN, ARGON AND NITROGEN. II.)
 Z. TECH. PHYSIK VOL. 12, 123-25 (1931)
 INDEX OF REFRACTION (GAS) (289 TO 296 DEGREES K)
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 THEORETICAL - EQUATIONS, TABLE (4 VALUES)
 GERMAN
366. THERMOPHYSICAL PROPERTIES RESEARCH CENTER
 DATA BOOK, VOL. 2, NONMETALLIC ELEMENTS AND THEIR COMPOUNDS (GASEOUS
 AND LIQUID STATES).
 PURDUE UNIV., LAFAYETTE, IND., SCHOOL OF MECHANICAL ENGINEERING (1961)
 150 P.
- VISCOSITY (LIQUID, GAS) (54 TO 2253 DEGREES K)
 AIR, CARBON MONOXIDE, DEUTERIUM, FLUORINE, METHANE, NEON, NITROGEN,
 OXYGEN, AND 45 OTHER LIQUIDS AND GASES
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- 366A. THODOS, G.
 ON THE REDUCED FROST-KALKWARF VAPOR PRESSURE EQUATION.
 IND. ENG. CHEM. FUNDAMENTALS VOL. 2, NO. 1, 80 (1963)
 C.A. 59, 42E
 VAPOR PRESSURE (LIQUID)
 DISCUSSION - EQUATION
- 366B. THOMAES, G. AND VAN ITTERBEEK, J.
 APPLICATION OF THE PRINCIPLE OF CORRESPONDING STATES TO THE VISCOSITY
 AND DIFFUSION OF PURE LIQUIDS AND MIXTURES.
 MOL. PHYS. VOL. 2, NO. 4, 372-78 (OCT 1959)
 CORRESPONDING STATES, VISCOSITY
 NITROGEN, CARBON MONOXIDE, OXYGEN, METHANE
 THEORETICAL - EQUATIONS

367. TRAUTZ, M. AND BINKELE, H. E.
 DIE REIBUNG, WARMELEITUNG UND DIFFUSION IN GASMISCHUNGEN. VIII. DIE
 REIBUNG DES HELIUM, WASSERSTOFF, NEON, ARGON UND IHRER BINAREN
 GEMISCHE.
 (THE VISCOSITY, HEAT CONDUCTION AND DIFFUSION IN GAS MIXTURES. VIII.
 THE VISCOSITY OF HYDROGEN, HELIUM, NEON, ARGON AND THEIR BINARY
 MIXTURES.)
 ANN. PHYSIK VOL. 5, 561-80 (1930)
- VISCOSITY (GAS) (293 TO 523 DEGREES K)
 AIR, HYDROGEN, HELIUM, NEON
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 GERMAN
368. TRAUTZ, M. AND KAUFMAN, F.
 KRITIK DER ELEKTRISCHEN DIFFERENTIALMETHODE ZUR MESSUNG VON C
 (V=KONSTANT) AN GASEN. IV. MESSUNGEN. DIE NORMIERUNG MIT ARGON.
 (CRITICISM OF THE ELECTRICAL DIFFERENTIAL METHOD OF MEASURING C
 (V=CONSTANT) IN GASES. IV. MEASUREMENTS. THE STANDARDIZATION WITH
 ARGON.)
 ANN. PHYSIK (5) VOL. 5, 581-605 (1930)
 C.A. 25, 10 3
- SPECIFIC HEAT (V=CONSTANT) (GAS) (293 DEGREES K)
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369. TRAUTZ, M. AND REICHLE, A.
 KRITIK DER ELEKTRISCHEN DIFFERENTIALMETHODE ZUR MESSUNG VON C (V= KONSTANTE) AN GASEN. VI. DIE SPEZIFISCHE WARME VON ARGON UND LUFT.
 (CRITICISM OF THE ELECTRICAL DIFFERENTIAL METHOD OF MEASURING SPECIFIC
 HEAT AT CONSTANT VOLUME OF GASES. VI. THE SPECIFIC HEATS OF ARGON AND
 AIR.)
 ANN. PHYSIK VOL. 22, 513-24 (1935)
 C.A. 29, 4227 7
- SPECIFIC HEAT (V=CONSTANT) (GAS) (289 DEGREES K)
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 GERMAN
370. TREPP, C.
 DIE HARTE VON STICKSTOFF, SAUERSTOFF, ARGON UND VON STICKSTOFF-
 SAUERSTOFF GEMISCHEN.
 (THE HARDNESS OF NITROGEN, OXYGEN, ARGON AND OF NITROGEN-OXYGEN
 MIXTURES.)
 SCHWEIZ. ARCH. ANGEW. WISS. U TECH. VOL. 24, 191-200 (1958)
- HARDNESS (SOLID) (3 TO 78 DEGREES K)
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 EXPERIMENTAL - GRAPH
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371. TREPP, C.
 DIE HARTE VON STICKSTOFF, SAUERSTOFF, ARGON UND VON STICKSTOFF-SAUERSTOFF GEMISCHEN.
 (THE HARDNESS OF NITROGEN, OXYGEN, ARGON AND OF NITROGEN-OXYGEN MIXTURES.)
 SCHWEIZ. ARCH. ANGEW. WISS. U TECH. VOL. 24, 230-9 (1958)
 HARDNESS (SOLID)
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 THEORETICAL - EQUATIONS, DIAGRAMS
 GERMAN
372. TSEDERBERG, N. W., POPOV, V. N. AND MOROZOVA, N. A.
 EXPERIMENTAL INVESTIGATION OF THE THERMAL CONDUCTIVITY OF ARGON.
 (IN RUSSIAN)
 TEPLOENERGETIKA VOL. 7, NO. 6, 82-6 (1960)
 C.A. 54, 23540 I
 THERMAL CONDUCTIVITY (GAS) (200 TO 673 DEGREES K AND 1 TO 500 KG/SQ CM)
 EXPERIMENTAL - GRAPH, TABLES (100 VALUES)
 RUSSIAN
373. TSIEN, H. S.
 THE PROPERTIES OF PURE LIQUIDS.
 J. AM. ROCKET SOC. VOL. 23, 17-24, 35 (1953)
 SPECIFIC HEAT (P=CONSTANT) (LIQUID) (85 TO 87 DEGREES K)
 CARBON MONOXIDE, NITROGEN, OXYGEN, METHANE AND 25 OTHER LIQUIDS
 THEORETICAL - EQUATIONS, 1 TABULAR VALUE
374. UHLIR, A.
 THERMAL CONDUCTIVITY OF FLUID ARGON AND NITROGEN.
 J. CHEM. PHYS. VOL. 20, 463-72 (1952)
 C.A. 46, 7385 F
 THERMAL CONDUCTIVITY (GAS) (90 TO 191 DEGREES K AND 1 TO 96 ATM),
 THERMAL CONDUCTIVITY (LIQUID) (87 TO 141 DEGREES K)
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375. VAN AGT, F. P. G. A. J.
 ISOTHERMS OF DI-ATOMIC SUBSTANCES AND THEIR BINARY MIXTURES XXXII.
 ON THE BEHAVIOR OF HYDROGEN ACCORDING TO THE LAW OF CORRESPONDING STATES.
 COMMUNS. PHYS. LAB. UNIV. LEIDEN NO. 176C (1925)
 VERSLAG. GEWONE VERGADER. WIS-EN NATUURK. AFDEEL. VOL. 34, 638-40
 (JUN 1925)
 LAW OF CORRESPONDING STATES
 NITROGEN, NEON, OXYGEN, HYDROGEN
 CORRELATION - GRAPH
-

376. VAN ITTERBEEK, A.
 VELOCITY AND ATTENUATION OF SOUND AT LOW TEMPERATURES.
P. 355-80 IN PROGRESS IN LOW TEMPERATURE PHYSICS, VOL. 1,
 NORTH HOLLAND PUBLISHING CO., AMSTERDAM, THE NETHERLANDS (1955)
- VELOCITY OF SOUND, ISOTHERMAL AND ADIABATIC COMPRESSIBILITY
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 OXYGEN, NITROGEN, METHANE, HYDROGEN, HELIUM
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377. VAN ITTERBEEK, A., GREVENDONK, W., VAN DAEL, W. AND FORREZ, G.
 SOUND VELOCITY MEASUREMENTS IN LIQUID ARGON UNDER HIGH PRESSURE.
PHYSICA VOL. 25, 1255-8 (1959)
C.A. 54, 20390 E
- VELOCITY OF SOUND (LIQUID)(84 TO 90 DEGREES K AND 0 TO 75 KG/SQ CM),
 DENSITY, ADIABATIC AND ISOTHERMAL COMPRESSIBILITY COEFFICIENT,
 SPECIFIC HEAT RATIO (LIQUID)(90 DEGREES K AND 5 TO 75 KG/SQ CM)
 EXPERIMENTAL - TABLES (50 VALUES), GRAPHS, EQUATIONS
378. VAN ITTERBEEK, A. AND VAN DAEL, W.
 THE VELOCITY OF SOUND IN LIQUID ARGON AND LIQUID NITROGEN AT HIGH
 PRESSURES.
CRYOGENICS VOL. 1, NO. 4, 226-8 (1961)
- VELOCITY OF SOUND (LIQUID)(87.3 AND 90.3 DEGREES K AND 3 TO
 201 KG/SQ CM)
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 EXPERIMENTAL - TABLE (25 VALUES). GRAPH
 APPARATUS
379. VAN ITTERBEEK, A., VAN DAEL, W. AND FORREZ, G.
 MEASUREMENTS ON THE VELOCITY OF SOUND IN FLUIDS.
BULL. INST. INTERN. FROID ANNEXE 1961-5, 167-77 (1961)
- VELOCITY OF SOUND (LIQUID)(77 TO 90 DEGREES K AND 0 TO 200 KG/SQ CM)
 HELIUM, OXYGEN, HYDROGEN, NITROGEN
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380. VAN ITTERBEEK, A., VAN DAEL, W. AND GREVENDONK, W.
 MEASUREMENTS ON THE VELOCITY OF SOUND IN ARGON UNDER HIGH PRESSURE.
PHYSICA VOL. 25, 640-4 (1959)
C.A. 54, 2857 C
- VELOCITY OF SOUND (LIQUID,GAS)(90 TO 299 DEGREES K AND 2 TO 72 ATM)
 EXPERIMENTAL - TABLES (80 VALUES), GRAPH
381. VAN ITTERBEEK, A. AND VAN PAEMEL, O.
 MEASUREMENTS ON THE VELOCITY OF SOUND IN GASEOUS ARGON AND DEUTERIUM
 RESPECTIVELY AT LIQUID OXYGEN AND HYDROGEN TEMPERATURES. CALCULATION
 AND DISCUSSION OF THE SECOND VIRIAL COEFFICIENT OF ARGON.
PHYSICA VOL. 5, NO. 9, 845-53 (1938)
C.A. 33, 1561 9
- VELOCITY OF SOUND (GAS)(79 TO 90 DEGREES K AND 0.05 TO 0.69 ATM),
 SECOND VIRIAL COEFFICIENT (GAS)(80 TO 250 DEGREES K)
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382. VAN ITTERBEEK, A. AND VAN PAEMEL, O.
 MEASUREMENTS ON THE VISCOSITY OF ARGON GAS AT ROOM TEMPERATURE AND
 BETWEEN 90 AND 55 DEGREES K.
 PHYSICA VOL. 5, 1009-12 (1938)
 C.A. 33, 2006 9
- VISCOSITY (GAS)(58 TO 90 DEGREES AND 0.2 TO 10.4 CM HG)
 (CALCULATED VALUES OF VISCOSITY FOR ARGON GAS FROM 55 TO
 293 DEGREES K)
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383. VAN ITTERBEEK, A. AND VERBEKE, O.
 DENSITY OF LIQUID NITROGEN AND ARGON AS A FUNCTION OF PRESSURE AND
 TEMPERATURE.
 PHYSICA VOL. 26, NO. 11, 931-8 (1960)
 C.A. 55, 18226 B
- SATURATED DENSITY (LIQUID)(86 TO 90 DEGREES K AND 13 TO 150 KG/SQ CM)
 (CALCULATED VALUES FOR SPECIFIC HEAT ($P=CONSTANT$), VELOCITY OF SOUND,
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384. VAN ITTERBEEK, A. AND VERBEKE, O.
 DENSITY VARIATION OF LIQUEFIED GASES (HYDROGEN, OXYGEN, NITROGEN,
 ARGON) WITH PRESSURE UP TO 850 KG/SQ CM.
 BULL. INST. INTERN. FROID ANNEXE 1961-5, 179-89 (1961)
- DENSITY (LIQUID)(86 TO 90 DEGREES K)
 NITROGEN, HYDROGEN, OXYGEN
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385. VAN ITTERBEEK, A., VERBEKE, O. AND STAES, K.
 MEASUREMENTS ON THE EQUATION OF STATE OF LIQUID ARGON AND METHANE UP
 TO 300 KG/SQ CM AT LOW TEMPERATURES.
 PHYSICA VOL. 29, 742-54 (1963)
- P-V-T DATA (LIQUID)(90 TO 148 DEGREES K AND 10 TO 300 KG/SQ CM), VAPOR
 PRESSURE (138 TO 141 DEGREES K), EQUATION OF STATE, HEAT CAPACITY ($V=CONSTANT$)
 ($P=CONSTANT$), SPECIFIC HEAT RATIO, ISOTHERMAL
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 VELOCITY OF SOUND (LIQUID)(90 TO 130 DEGREES K)
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386. VAN ITTERBEEK, A. AND VERHAGEN, L.
 MEASUREMENTS OF THE VELOCITY OF SOUND IN LIQUID ARGON AND LIQUID
 METHANE.
 PROC. PHYS. SOC. (LONDON) VOL. B62, 800-4 (1949)
 C.A. 44, 1771 B
- VELOCITY OF SOUND (LIQUID)(84 TO 87 DEGREES K AND 525 TO 767 MM HG)
 (CALCULATED VALUES FOR SPECIFIC HEAT RATIO, ISOTHERMAL AND ADIABATIC
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387. VAN ITTERBEEK, A., ZINK, H. AND VAN PAEMEL, O.
VISCOSITY MEASUREMENTS IN LIQUEFIED GASES.
CRYOGENICS VOL. 2, NO. 4, 210-11 (1962)

VISCOSITY (LIQUID)(84 TO 87 DEGREES K)
HYDROGEN, OXYGEN, NITROGEN
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- 387A. VAN LIERDE, J.
MEASUREMENTS OF THERMAL DIFFUSION AND VISCOSITY OF CERTAIN GAS
MIXTURES AT LOW AND VERY LOW TEMPERATURES. (IN DUTCH)
VERHANDEL. KONINKL. VLAAM. ACAD. WETENSCHAP. BELG. KL. WETENSCHAP.
VOL. 9, NO. 24, 7-78 (1947)
C.A. 43, 5658 H

VISCOSITY (GAS)(286.8 AND 290.7 DEGREES K AND 0.01 TO 50 MM HG)
HYDROGEN, NITROGEN, OXYGEN, NEON
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388. VAN PAEMEL, O.
METINGEN EN THEORETISCHE BESCHOUWINGEN IN VERBAND MET DE VISCOSITEIT
VAN GASSEN EN GECONDENSEerde GASSEN.
(MEASUREMENTS AND THEORETICAL CONSIDERATION OF THE VISCOSITY OF GASES
AND CONDENSED GASES.)
VERHANDEL. KONINKL. VLAAM. ACAD. WETENSCHAP. BFLG. (3) VOL. 3, 3-59
(1941)
C.A. 37, 6170 4

VISCOSITY (GAS)(58 TO 293 DEGREES K AND 0.22 TO 7.4 CM HG)
NEON, HELIUM, HYDROGEN, DEUTERIUM, OXYGEN
EXPERIMENTAL - TABLE (15 VALUES), EQUATION
APPARATUS
DUTCH

389. VASILESCO, V.
RECHERCHES EXPERIMENTALES SUR LA VISCOSITE DES GAZ AUX TEMPERATURES
ELEVEES.
(EXPERIMENTAL RESEARCH ON VISCOSITY OF GAS AND HIGH TEMPERATURES.)
UNIVERSITY OF PARIS, PH. D. THESIS (1940)

VISCOSITY (GAS)(273 TO 1868 DEGREES K)
EXPERIMENTAL - TABLE (50 VALUES), GRAPH
FRENCH

390. VERHAEGEN, L.
METINGEN OVER DE VOORTPLANTINGSSNELHEID VAN HET GELUID IN ENKELE
VLOEIBAAR GEMAakte GASSEN.
(MEASUREMENTS OF THE SPEED OF PROPAGATION OF SOUND IN SOME LIQUEFIED
GASES.)
VERHANDEL. KONINKL. VLAAM. ACAD. WETENSCHAP. BELGIE, KL. WETENSCHAP.
NO. 38 (1952) 65 P.

VELOCITY OF SOUND (LIQUID)(84 TO 87 DEGREES K AND 525 TO 767 MM HG)
METHANE, HYDROGEN, OXYGEN, NITROGEN
EXPERIMENTAL - TABLE (7 VALUES), GRAPHS
APPARATUS
DUTCH

391. VERSCHAFFELT, J. F.
 LA TENSION DE VAPEUR ET LA CHALEUR DE VAPORISATION AUX BASSES
 TEMPERATURES.
 (VAPOR TENSION AND HEAT OF VAPORIZATION AT LOW TEMPERATURES.)
 ARCH. NEERLAND SCI. (IIIA) VOL. 8, 109-35 (1924)
 COMMUNS. PHYS. LAB. UNIV. LEIDEN SUPPL. NO. 49 (1924)
 PROC. 4TH INTERN. CONGRESS OF REFRIGERATION VOL. 1, 13-4 A (1924)
 C.A. 20, 863 2
- VAPOR PRESSURE, HEAT OF SUBLIMATION (SOLID) (0 TO 83 DEGREES K)
 HYDROGEN, HELIUM, MERCURY
 THEORETICAL - EQUATIONS, TABLE (20 VALUES)
 FRENCH
392. VORONEL, A. V.
 THE HEAT VAPORIZATION AND THE LAW OF CORRESPONDING STATES.
 SOVIET PHYS. TECH. PHYS. VOL. 4, 270-2 (1959)
 TRANSLATED FROM ZHUR. TECH. FIZ. VOL. 29, 304-6 (1959) (IN RUSSIAN)
 HEAT OF VAPORIZATION, LAW OF CORRESPONDING STATES
 NEON, NITROGEN, OXYGEN, CARBON MONOXIDE, KRYPTON AND XENON
 THEORETICAL - EQUATION
393. WAELBROECK, F., LAFLEUR, S. AND PRIGOGINE, I.
 CONDUCTIBILITE THERMIQUE DES GAZ REELS.
 (THERMAL CONDUCTIVITY OF REAL GASES.)
 PHYSICA VOL. 21, 667-75 (1955)
 THERMAL CONDUCTIVITY EXCESS DUE TO TEMPERATURE GRADIENT DIFFUSION
 EFFECTS (90 TO 400 DEGREES K AND 1 TO 56 ATM)
 THEORETICAL - EQUATIONS, TABLE (19 VALUES)
 FRENCH
394. WALKER, P. A.
 THE EQUATION OF STATE AND THE SPECIFIC HEAT OF LIQUID ARGON.
 UNIVERSITY OF LONDON, PH. D. THESIS (1956) 234 P.
 P-V-T DATA (GAS) (90 TO 200 DEGREES K AND 0 TO 500 ATM), SPECIFIC HEAT
 (P=CONSTANT) (LIQUID, GAS) (92 TO 175 DEGREES K AND 10 TO 50 ATM),
 SPECIFIC HEAT (V=CONSTANT) (LIQUID, GAS) (99 TO 196 DEGREES K AND 51 TO
 492 ATM), SPECIFIC HEAT (SATURATION) (135 TO 149 DEGREES K), EQUATION
 OF STATE
 EXPERIMENTAL - TABLE (350 VALUES), GRAPHS, EQUATIONS
 APPARATUS
395. WALTER, J. AND EYRING, H.
 A PARTITION FUNCTION FOR NORMAL LIQUIDS.
 J. CHEM. PHYS. VOL. 9, 393-7 (MAY 1941)
 SPECIFIC HEATS (V=CONSTANT) (P=CONSTANT), VAPOR PRESSURE (84 TO 148
 DEGREES K), VIRIAL COEFFICIENT (150 TO 800 DEGREES K), CRITICAL
 CONSTANTS, MELTING POINT
 NITROGEN, BENZENE
 THEORETICAL - EQUATIONS, TABLES (11 VALUES), GRAPHS

396. WASHBURN, E. W. (EDITOR)
INTERNATIONAL CRITICAL TABLES
MC GRAW-HILL BOOK CO., INC., NEW YORK (1928)
- P-V-T, DENSITY, THERMAL CONDUCTIVITY, HEAT CAPACITY, VAPOR PRESSURE,
COMPRESSIBILITY, VISCOSITY, TRIPLE POINT, TRANSITION POINT, MELTING
POINT, BOILING POINT, SUBLIMATION POINT
BOOK
397. WEBER, S.
UNTERSUCHUNGEN UBER DIE WARMELEITFAHIGKEIT DER GASE. II.
(RESEARCHES ON THE THERMAL CONDUCTIVITY OF GAS. II.)
ANN. PHYSIK VOL. 54, 437-62 (1917)
- THERMAL CONDUCTIVITY (GAS)
HYDROGEN, NEON, HELIUM, NITROGEN, OXYGEN, METHANE, CARBON MONOXIDE,
CARBON DIOXIDE
CORRELATION OF CONDUCTIVITY PARAMETERS WITH REDUCED TEMPERATURE
GERMAN
398. WEBER, S.
UBER DIE WARMELEITFAHIGKEIT DER GASE.
(THE THERMAL CONDUCTIVITY OF GAS.)
ANN. PHYSIK VOL. 82, 479-503 (1927)
- THERMAL CONDUCTIVITY (GAS) (273 DEGREES K)
AIR, NEON, HYDROGEN, CARBON DIOXIDE
REVIEW - DATA FROM 397
GERMAN
399. WENTORF, R. H., BUEHLER, R. J., HIRSCHFELDER, J. O. AND CURTISS, C. F.
LENNARD-JONES AND DEVONSHIRE EQUATION OF STATE OF COMPRESSED GASES
AND LIQUIDS.
J. CHEM. PHYS. VOL. 18, 1484-1500 (1950)
C.A. 45, 3672 G
- EQUATION OF STATE, COMPRESSIBILITY FACTOR, SPECIFIC HEAT, ENTROPY,
INTERNAL ENERGY (GAS, LIQUID)
NITROGEN, HYDROGEN
THEORETICAL - EQUATIONS, TABLES (1600 VALUES)
400. WHALLEY, E.
THE VISCOSITY OF GASES AND THE THEORY OF CORRESPONDING STATES.
CAN. J. CHEM. VOL. 32, 485-91 (1954)
- VISCOSITY (GAS) (90 TO 1800 DEGREES K)
OXYGEN, NITROGEN, CARBON MONOXIDE, METHANE AND 3 OTHER GASES
CORRELATION - EQUATION, GRAPH
401. WHALLEY, E.
THERMODYNAMIC PROPERTIES OF ARGON IN THE TEMPERATURE RANGE -100 TO
600 DEGREES C. AND PRESSURE RANGE 0 TO 80 ATMOSPHERES.
CAN. J. TECHNOL. VOL. 33, NO. 2, 111-6 (1955)
- P-V-T DATA (GAS) (173 TO 873 DEGREES K AND 0 TO 80 ATM)
SPECIFIC HEAT (P=CONSTANT) (V=CONSTANT), ENTHALPY, ENTROPY,
JOULE-THOMSON COEFFICIENT (GAS) (273 TO 873 DEGREES K AND 0 TO 80 ATM)
CORRELATION - TABLES (900 CALCULATED VALUES), GRAPH, EQUATION

402. WHALLEY, E., LUPIEN, Y. AND SCHNEIDER, W. G.
 THE COMPRESSIBILITY OF GASES. VII. ARGON IN THE TEMPERATURE RANGE
 0 TO 600 DEGREES C AND THE PRESSURE RANGE 10 TO 80 ATMOSPHERES.
 CAN. J. CHEM. VOL. 31, 722-33 (1953)
 C.A. 48, 3086 H
- VIRIAL COEFFICIENTS (GAS)(273 TO 873 DEGREES K)
 EXPERIMENTAL - TABLE (27 VALUES), GRAPH
403. WHALLEY, E. AND SCHNEIDER, W. G.
 THE LENNARD-JONES 12.6 POTENTIAL AND THE VISCOSITY OF GASES.
 J. CHEM. PHYS. VOL. 20, 657-61 (1952)
- VISCOSITY (GAS)
 HELIUM, HYDROGEN, DEUTERIUM, FLUORINE, NITROGEN, OXYGEN, NEON
 THEORETICAL - EQUATION, GRAPH
404. WHALLEY, E. AND SCHNEIDER, W. G.
 INTERMOLECULAR POTENTIALS OF ARGON, KRYPTON AND XENON.
 J. CHEM. PHYS. VOL. 23, 1644 (1955)
- INTERMOLECULAR POTENTIAL CALCULATED FROM 2ND VIRIAL COEFFICIENT
 FROM REFERENCE NUMBER 254
 KRYPTON, XENON
 THEORETICAL - TABLE (3 VALUES)
405. WHEELER, T. S.
 THEORY OF LIQUIDS.
 CURRENT SCI. VOL. 3, 347-8 (1935)
 C.A. 29, 3890 3
- DENSITY, SURFACE TENSION (84 DEGREES K), INTERNAL LATENT HEAT
 HELIUM, HYDROGEN, NEON, NITROGEN
 THEORETICAL CALCULATION
 -CODED FROM ABSTRACT-
406. WHITE, G. K. AND WOODS, S. B.
 THERMAL CONDUCTIVITY OF SOLID ARGON AT LOW TEMPERATURES.
 NATURE (LONDON) VOL. 177, 851-2 (1956)
 C.A. 50, 11794 I
- THERMAL CONDUCTIVITY (SOLID)(2 TO 25 DEGREES K)
 EXPERIMENTAL - GRAPH
407. WHITE, G. K. AND WOODS, S. B.
 THERMAL CONDUCTIVITY OF THE SOLIDIFIED INERT GASES. ARGON, NEON AND
 KRYPTON.
 PHIL. MAG. VOL. 3, NO. 32, 785-97 (1958)
- THERMAL CONDUCTIVITY (SOLID)(2 TO 80 DEGREES K)
 NEON, KRYPTON
 EXPERIMENTAL - GRAPH, EQUATION
 APPARATUS
-

408. WHITE, G. K., WOODS, S. R. AND MAC DONALD, K. D. C.
 THERMAL CONDUCTIVITY IN ELEMENTAL SEMI-CONDUCTORS AND SOLID ARGON.
 BULL. INST. INTERN. FROID, ANNEXE 1956-2, 91-5
 C.A. 52, 886 I
- THERMAL CONDUCTIVITY (SOLID)(1 TO 25 DEGREES K)
 SILICON, GERMANIUM
 EXPERIMENTAL - GRAPH
409. WINN, E. B.
 THE TEMPERATURE DEPENDENCE OF THE SELF-DIFFUSION COEFFICIENTS OF
 ARGON, NEON, NITROGEN, OXYGEN, CARBON DIOXIDE AND METHANE.
 PHYS. REV. VOL. 80, 1024-7 (1950)
- SELF-DIFFUSION COEFFICIENT (GAS)(78 TO 353 DEGREES K AND 1 ATM)
 METHANE, OXYGEN, NEON, NITROGEN, CARBON DIOXIDE
 EXPERIMENTAL - TABLE (6 VALUES)
 APPARATUS
410. WINTER, E. R. S.
 DIFFUSION OF PROPERTIES OF GASES. III. THE DIFFUSION AND THERMAL
 DIFFUSION COEFFICIENTS FOR ISOTOPIC GASES AND GAS MIXTURES.
 TRANS. FARADAY SOC. VOL. 46, 81-92 (1950)
- THERMAL DIFFUSION COEFFICIENTS (GAS), SELF-DIFFUSION COEFFICIENTS
 (GAS)(295 DEGREES K)
 HELIUM, NEON, HYDROGEN, NITROGEN, METHANE, CARBON MONOXIDE, OXYGEN,
 CARBON DIOXIDE
 CORRELATION - TABLES (6 VALUES), EQUATIONS
411. WIRTH, H. AND KLEMENCIC, A.
 DIE WARMELEITFAHIGKEIT VON GASEN BEI NIEDEREN DRUCKEN.
 (THE THERMAL CONDUCTIVITIES OF GASES AT LOW PRESSURES.)
 MONATSH. CHEM. VOL. 83, 879-82 (1952)
 C.A. 48, 13308 D
- THERMAL CONDUCTIVITY (GAS)(273 DEGREES K AND 1 ATM)
 EXPERIMENTAL - ONE TABULAR VALUE, EQUATION
 GERMAN
412. WOOLLEY, H. W.
 NOTE ON THE RESONANCE METHOD OF MEASURING THE RATIO OF THE SPECIFIC
 HEATS OF A GAS.
 CAN. J. PHYS. VOL. 31, 604-12 (1953)
- SPECIFIC HEAT RATIO (GAS)(297 DEGREES K AND 1 TO 20 ATM)
 NITROGEN, CARBON DIOXIDE, NITROUS OXIDE, SULFUR DIOXIDE
 RECALCULATION OF EXPERIMENTAL DATA FROM 68
413. WOOLLEY, H. W. AND BENEDICT, W. S.
 GENERALIZED TABLES OF CORRECTIONS TO THERMODYNAMIC PROPERTIES FOR
 NON-POLAR GASES.
 NATL. ADVISORY COMM. AERONAUT. TECH. NOTE NO. 3272 (1956) 62 P.
- COMPRESSIBILITY FACTOR, INTERNAL ENERGY, ENTHALPY, ENTROPY, SPECIFIC
 HEAT (V=CONSTANT)(P=CONSTANT), SPECIFIC HEAT RATIO, VELOCITY OF SOUND,
 EXPANSION COEFFICIENT (SOLID, LIQUID, GAS)(18 TO 1220 DEGREES K)
 THEORETICAL - EQUATIONS, TABLES OF CORRECTIONS BASED ON LENNARD-JONES
 6-12 POTENTIAL

414. WYLIE, L. M.
 THE VAPOR PRESSURE OF SOLID ARGON, CARBON MONOXIDE, METHANE, NITROGEN, AND OXYGEN FROM THEIR TRIPLE POINTS TO THE BOILING POINT OF HYDROGEN. GEORGIA INST. TECHNOL., ATLANTA, MASTER THESIS (1958) 90 P.
 VAPOR PRESSURE (SOLID)
 CARBON MONOXIDE, OXYGEN, METHANE AND NITROGEN
 COMPILATION
 (A COPY OF THIS THESIS IS NOT AVAILABLE FOR CODING)
415. YOUNGLOVE, B. A. AND DILLER, D. E.
 THE SPECIFIC HEAT OF SATURATED LIQUID PARA-HYDROGEN FROM 15 TO 32 DEGREES K.
 CRYOGENICS VOL. 3, 283-7 (SEPT 1962)
 SPECIFIC HEAT (SOLID) (22 TO 59 DEGREES K)
 HYDROGEN
 EXPERIMENTAL - TABLE (12 VALUES), GRAPH
 APPARATUS
416. ZAITSEVA, L. S.
 AN EXPERIMENTAL INVESTIGATION OF THE HEAT CONDUCTIVITY OF MONATOMIC GASES OVER WIDE TEMPERATURE INTERVALS.
 SOVIET PHYS. - TECH. PHYS. VOL. 4, 444-50 (1959)
 ZHUR. TEKH. FIZ. VOL. 29, 497-505 (1959)
 C.A. 53, 19551 H
 HEAT CONDUCTIVITY (GAS) (273 TO 773 DEGREES K)
 HELIUM, NEON, KRYPTON, XENON, MERCURY
 EXPERIMENTAL - GRAPH, EQUATION.
417. ZHDANOVA, N. F.
 TEMPERATURE DEPENDENCE OF VISCOSITY OF LIQUID ARGON.
 SOVIET PHYS. JETP VOL. 4, NO. 5, 749-50 (1957)
 ZHUR. EKSPTL. I TEORET. FIZ. VOL. 31, 724-5 (1956) (IN RUSSIAN)
 VISCOSITY (LIQUID) (90 TO 149 DEGREES K)
 EXPERIMENTAL - TABLE (10 VALUES), GRAPH
 LETTER TO EDITOR.
418. ZIEBLAND, H.
 DIE WARMELEITFAHIGKEIT VON STICKSTOFF, SAUERSTOFF UND ARGON IN FLUSSIGEM UND GASFORMIGEM ZUSTAND.
 (THE THERMAL CONDUCTIVITY OF NITROGEN, OXYGEN AND ARGON IN THE LIQUID AND GASEOUS STATES.
 DECHHEMA MONOGRAPHIEN (GERMANY) VOL. 32, 74-82 (1959)
 THERMAL CONDUCTIVITY (GAS) (90 TO 200 DEGREES K AND 24 TO 120 ATM)
 NITROGEN, OXYGEN
 REVIEW - GRAPHS
 GERMAN

419. ZIEBLAND, H. AND BURTON, J. T. A.
 THE THERMAL CONDUCTIVITY OF NITROGEN AND ARGON IN THE LIQUID AND
 GASEOUS STATES.
 BRIT. J. APPL. PHYS. VOL. 9, 52-9 (1958)
 C.A. 52, 10704 G
- THERMAL CONDUCTIVITY (LIQUID, GAS) (85 TO 200 DEGREES K AND 1 TO
 135 ATM)
 NITROGEN
 EXPERIMENTAL - TABLE (175 VALUES), GRAPH, EQUATION
420. ZIEGLER, W. T., MULLINS, J. C. AND KIRK, B. S.
 CALCULATION OF THE VAPOR PRESSURE AND HEATS OF VAPORIZATION AND
 SUBLIMATION OF LIQUIDS AND SOLIDS, ESPECIALLY BELOW ONE ATMOSPHERE
 PRESSURE. II. ARGON.
 GEORGIA INST. TECH., ENG. EXP. STA., TECH. REPT. NO. 2 (JUN 1962)
- VAPOR PRESSURE, HEATS OF VAPORIZATION AND SUBLIMATION (20 TO 88
 DEGREES K)
 CORRELATION - EQUATIONS, TABLES (90 VALUES)
421. ZIEMAN, C. M.
 DIELECTRIC CONSTANT OF ARGON AT 9700 MEGACYCLES.
 J. APPL. PHYS. VOL. 24, 110 (1953)
 C.A. 47, 4674 A
- DIELECTRIC CONSTANT (273 DEGREES K AND 1 ATM)
 EXPERIMENTAL - ONE TABULAR VALUE
 'LETTER TO EDITOR'
422. ZUCKER, I. J.
 ANHARMONIC EFFECTS IN THE THEORY OF SOLID ARGON.
 PHIL. MAG. VOL. 3, 987-98 (1958)
- ISOTHERMAL COMPRESSIBILITY, EXPANSIVITY, SPECIFIC HEAT ($V=CONSTANT$) ($P=CONSTANT$) (SOLID) (0 TO 80 DEGREES K)
 THEORETICAL - GRAPHS, EQUATIONS, TABLE (40 VALUES)
423. ZUCKER, I. J.
 THE REDUCED EQUATION OF STATE OF INERT GAS SOLIDS AT ABSOLUTE ZERO.
 PROC. PHYS. SOC. (LONDON) VOL. 77, NO. 4, 889-900 (1961)
- EQUATION OF STATE (SOLID) (0 DEGREES K), LAW OF CORRESPONDING STATES,
 QUANTUM MECHANICAL CONSTANTS, QUANTUM EFFECTS ON INERT GASES
 NEON, DEUTERIUM, HYDROGEN, HELIUM, XENON, KRYPTON
 THEORETICAL - EQUATION
424. ZWANZIG, R. W., KIRKWOOD, J. G., OPPENHEIM, I. AND ALDER, B. J.
 STATISTICAL MECHANICAL THEORY OF TRANSPORT PROCESSES. VII. THE
 COEFFICIENT OF THERMAL CONDUCTIVITY OF MONATOMIC LIQUIDS.
 J. CHEM. PHYS. VOL. 22, 783-90 (1954)
 C.A. 48, 11862 G
- THERMAL CONDUCTIVITY (LIQUID)
 COMPARISON OF THEORY WITH EXPERIMENTAL AT NORMAL BOILING POINT.
 THEORETICAL - EQUATIONS

425. ZWANZIG, R. W., KIRKWOOD, J. G., STRIPP, K. F. AND OPPENHEIM, I.
RADIAL DISTRIBUTION FUNCTIONS AND THE EQUATION OF STATE OF MONATOMIC
FLUIDS.
J. CHEM. PHYS. VOL. 21, 1268-71 (1953)

EQUATION OF STATE, P-V-T DATA, INTERNAL ENERGY ISOTHERMAL
COMPRESSIBILITY (100 TO 600 DEGREES K)
THEORETICAL - EQUATIONS, TABLES
426. ZWANZIG, R. W., KIRKWOOD, J. G., STRIPP, K. F. AND OPPENHEIM, I.
ERRATA. RADIAL DISTRIBUTION FUNCTIONS AND THE EQUATION OF STATE OF
MONATOMIC FLUIDS.
J. CHEM. PHYS. VOL. 22, 1625 (1954)
C.A. 49, 22 F

ISOTHERMAL COMPRESSIBILITY (LIQUID)(87 DEGREES K)
CORRECTION TO REFERENCE NUMBER 425

