ROYAL SOCIETY OPEN SCIENCE

rsos.royalsocietypublishing.org

Research



Cite this article: Gao K, Zhang B, Cao Y, Chen X. 2018 Doping induced dielectric anomaly below the Curie temperature in molecular ferroelectric diisopropylammonium bromide. *R. Soc. open sci.* **5**: 181397.

http://dx.doi.org/10.1098/rsos.181397

Received: 29 August 2018 Accepted: 18 October 2018

Subject Category:

Chemistry

Subject Areas: chemical physics/physical chemistry

Keywords:

doping, dielectric anomaly, molecular ferroelectric, diisopropylammonium bromide

Author for correspondence:

Kaige Gao e-mail: kaigegao@qq.com

This article has been edited by the Royal Society of Chemistry, including the commissioning, peer review process and editorial aspects up to the point of acceptance.

Electronic supplementary material is available online at https://dx.doi.org/10.6084/m9.figshare. c.4295993.



THE ROYAL SOCIETY PUBLISHING

Doping induced dielectric anomaly below the Curie temperature in molecular ferroelectric diisopropylammonium bromide

Kaige Gao¹, Binbin Zhang², Yunqing Cao¹ and Xiaobing Chen¹

¹College of Physical Science and Technology, Yangzhou University, Jiangsu 225009, People's Republic of China

²State Key Laboratory of Solidification Processing and Key Laboratory of Radiation Detection Materials and Devices and School of Materials Science and Engineering, Northwestern Polytechnical University, Xi'an 710072, People's Republic of China

🔟 KG, 0000-0002-7634-3400

A dielectric anomaly induced by doping has been observed at about 340 K in chlorine-doped diisopropylammonium bromide. The dielectric anomaly has a switchable behaviour, which indicates potential applications on switches and sensors. Temperature-dependent Raman spectrum, X-ray diffraction and differential scanning calorimetry do not show any anomaly around the dielectric anomaly temperature, which prove that the dielectric anomaly does not come from structure phase transition and has no specific heat variety. It is assumed that this dielectric anomaly can be attributed to the freezing of ferroelectric domain walls induced by the pinning of point defects.

1. Introduction

Piezoelectric materials have been widely applied in the energy conversion, sensing, driving and frequency fields. Ferroelectrics usually feature a switchable spontaneous electric polarization and have good piezoelectric performance [1]. Relaxor-based ferroelectric single crystals, such as $Pb(Zn_{1/3}Nb_{2/3})O_3$ -PbTiO₃ and $Pb(Mg_{1/3}Nb_{2/3})O_3$ -PbTiO₃ attract much attention for their large piezoelectric coefficients up to 2500 pC N⁻¹ and high subsequent strain levels up to 0.6% [2,3]. With proper doping

 \bigcirc 2018 The Authors. Published by the Royal Society under the terms of the Creative Commons Attribution License http://creativecommons.org/licenses/by/4.0/, which permits unrestricted use, provided the original author and source are credited.

2

and phase engineering, the solid solution single crystal of the relaxor ferroelectrics can reach much larger piezoelectric coefficients than that of pure materials and polycrystals. Doping is believed to be responsible for good mechanical properties and piezoelectric of the relaxor-based ferroelectric single crystals. While doping modification of the inorganic ferroelectrics have been widely studied in the past years [2,4–12], there are few researches relevant to the doping of organic ferroelectrics except some doping of organic–inorganic hybrid ferroelectrics [13]. Researches on the doping of organic ferroelectric would help us to understand the organic ferroelectric better and speed up the application of organic ferroelectric materials. It is essential for us to study the effects of doping on organic ferroelectrics in order to modify organic ferroelectrics' mechanical properties and piezoelectricity.

Recently, a series of simple molecular ferroelectrics have been found, such as diisopropylammonium bromide (DIPAB) [14–18], diisopropylammonium perchlorate [19,20], 4-(cyanomethyl)anilinium perchlorate [21], 4-methoxyanilinium tetrafluoroborate 18-crown-6 [22], (4-amino-2-bromopyridinium) (4-amino-2-bromopyridine) tetrafluoroborate [23], pyridin-4-ylmethanaminium perchlorate [24] and so on, which is composed by ammonium cations and acid ions. The organic–inorganic hybrid ferroelectrics also cause much attention for the designable and tunable characteristics of organic and inorganic components. So far, many organic–inorganic hybrid ferroelectrics have been discovered, namely, (pyrrolidinium)MnX₃ (X = Cl,Br) [25,26], (pyrrolidinium)CdCl₃ [27], (cyclopentylammonium)CdCl₃ [28], (3-pyrrolinium)MCl₃ (M = Mn,Cd) [29,30], (benzylammonium)₂PbCl₄ [31] and so on. Therefore, a huge potential in practical applications of molecular ferroelectric can be foreseen in the coming future, when doping of molecular ferroelectrics would be an important part of molecular ferroelectric research.

Diisopropylammonium chloride (DIPAC) [32] and DIPAB [14–16,33] are good organic ferroelectric materials with large spontaneous polarization (8.9 and 23 μ C cm⁻², respectively) and high Curie temperature, T_c (440 and 426 K, respectively), which possesses ferroelectric properties comparable to those of BaTiO₃. Ferroelectric DIPAC and DIPAB have similar crystal structure (P2₁) and similar structure phase transition (P2₁ \rightarrow P2₁/m). So, the two materials can form solid solution single crystal at any ratio. This is similar to that of relaxor-based ferroelectric single crystals which have large piezoelectric coefficients and large subsequent strain levels. We have studied the chlorine-doped DIPAB (DIPAB-C) single crystal in the earlier work [34]. It is found that the ferroelectric properties, including spontaneous polarization, the phase transition temperatures, the lattice parameters, etc., can be modulated by doping with congeners. Here we will report a dielectric relaxation phase transition induced by the doping of chlorine in DIPAB single crystals. Temperature-dependent Raman spectrum, X-ray diffraction and differential scanning calorimetry (DSC) prove the phase transition does not have structure phase transition and has no specific heat variety. It is assumed that this relaxation process can be attributed to the freezing of ferroelectric domain walls induced by the pinning of point defects. This may help us understand the ferroelectric properties in DIPAB.

2. Material and methods

The single crystal of chlorine-doped DIPAB ($C_6H_{16}NBr_{1-x}Cl_x$, DIPAB-C) was grown by slow evaporation of methanol solution containing 1 mol diisopropylamine, *x* mol hydrochloric acid and (1 - x) mol hydrobromic acid (x = 0-1). The element contents were measured using a CHN elemental analyser (Heraeus CHN-O-Rapid). DSC measurements of single crystals were recorded by using a NETZSCH DSC 200F3 in the temperature range of 300–453 K. The complex permittivity was measured using a Tonghui TH2828A LCR meter. Variable temperature powder X-ray diffraction (PXRD) was performed on a Bruker D8 Advance X-ray diffractometer. Raman spectra were taken using a Horiba Jobin Yvon HR800 spectrometer system with a 488 nm laser line from an air-cooled Ar-ion laser.

3. Results and discussion

Chlorine content in the single crystal of DIPAC was determined by the carbon mass fraction measured by a combustion method using a CHN elemental analyser [34]. The actual chlorine concentration is a little less than the stoichiometric values due to the faster volatility of hydrochloric acid compared with that of hydrobromic acid. As the DIPAB-C samples have similar properties when changing the doping content, here we just choose $C_6H_{16}NBr_{0.76}Cl_{0.24}$ (DIPAB-C1) as an example to analyse in the following.

The structure of ferroelectric DIPAB-C is $P2_1$, similar to that of pure DIPAB. But the structure of the as-grown crystal of DIPAB-C is $P2_12_12_1$, which is non-ferroelectric. Room temperature ferroelectric



Figure 1. Temperature dependence of (*a*) real part (ε') and (*b*) imaginary part (ε'') of complex dielectric constant of DIPAB-C1 in heating process. Inset: the enlarged view of (*a*) real part (ε') and (*b*) imaginary part (ε'').

DIPAB-C could be acquired by heating the samples above the first phase transition temperature at about 420 K due to the irreversible phase transition. The following measurements were all performed on the ferroelectric samples.

The dielectric constant of DIPAB-C1 was measured in the heating cycle with the heating rate of 2 K min^{-1} . Figure 1 shows the temperature-dependent complex dielectric constant ($\varepsilon = \varepsilon' - i\varepsilon''$, where ε'' is the imaginary part) at the frequency range from 500 Hz to 1000 KHz. The real part (ε') of the complex dielectric is shown in figure 1*a* and the imaginary part (ε'') is shown in figure 1*b*. Two dielectric anomalies can be seen in figure 1. A very sharp dielectric constant appears at 412 K (T_c), which belongs to the ferroelectric–paraelectric phase transition. At 412 K, DIPAB-C1 undergoes a structure phase transition from P2₁ to P2₁/m. In the vicinity of T_c, the temperature-dependent dielectric constant along the polar axis follows the Curie–Weiss law of ferroelectric materials parametrized as $\varepsilon' = C/(T - T_0)$, as shown by the linear relationship between reciprocal dielectric constant and temperature in electronic supplementary material, figure S2.

It is interesting for us to find another slow step-like dielectric anomaly at about 340 K (T₁), which is not found in the pure DIPAB and has not been reported in the previous literature. The dielectric anomaly has a switchable behaviour, which indicates potential applications on switches and sensors. Both the real part ε' and the imaginary part ε'' of the complex dielectric display slow step-like dielectric anomaly. When heating across T₁, the dielectric frequency dispersion has an abrupt increase, characterizing an abrupt increase of relaxation time phase transition. Furthermore, above T₁, the frequency dispersion of ε' and ε'' enhance with the increase of temperature, indicating the relaxation time decrease with the increasing temperature. Below T₁, the frequency dispersion keeps almost constant, indicating the relaxation time is independent of temperature. Simply speaking, the relaxation time keeps constant below T₁, increases abruptly at T₁ and increases slowly above T₁.

In order to understand the mechanism of the dielectric anomaly happening at T_1 , DSC measurements were performed. The heating and cooling temperature speed during the DSC measurements are both 10 K min⁻¹. The ferroelectric–paraelectric phase could be found easily in figure 2, which displays an endothermic peak at 425.6 K during the heating process and an exothermic peak at 420.4 K during the cooling process. 5 K thermal hysteresis indicates the phase



Figure 2. Temperature dependence of DSC curve at a rate of 10 K min⁻¹ for the compound. The inset is the enlarged DSC curve from 310 to 380 K.



Figure 3. Temperature dependence of PXRD. The inset is the enlarged parts from 24.7° to 25.7°. The short vertical bars indicate the calculated positions of pure DIPAB.

transition is a first-order phase transition. Yet no DSC signal anomaly was found around 340 K, proving the dielectric anomaly at T_1 has no specific heat variety.

Temperature-dependent PXRD measurements of DIPAB-C1 samples were performed from 303 to 373 K to examine the structure change during the dielectric anomaly happening at T₁. All of the PXRD patterns have similar diffraction peaks with the peaks shifting uniformly with increasing temperature (figure 3 inset). The PXRD results prove the structure phase transition does not happen at T₁. Both PXRD and DSC reveal the dielectric anomaly at T₁ is not structure phase transition and has no specific heat variety. According to the shifts of XRD peak of plane (200) and Bragg equation $2d\sin\theta = n^*\lambda$, we can calculate the variance of the plane space *d* along (100) direction against temperature. The calculated coefficient of thermal expansion α is $60 \times 10^{-6} \text{ K}^{-1}$, which is just half of that of polyvinylidene fluoride (PVDF). The large thermal expansion coefficient indicates DIPAB-C1 is as soft as polymer, illustrating DIPAB-C possesses potential electrostriction application [35].

In order to understand the mechanism and dynamics of the phase transitions, we performed Raman measurements as a function of temperature (figure 4), which is an ideal tool for capturing the dynamics and local structural changes from the viewpoint of lattice vibrations [35]. Figure 4*a* shows the Raman spectra measured from 303 to 373 K. The intensity of the Raman peaks in figure 4*a* do not appear to abruptly change. Some typical peaks' positions shifts according to that of 303 K are shown in figure 4*b*. There are no apparent abrupt shifts at about 340 K. The chosen Raman peak positions almost show a linear variation with temperature. The temperature-dependent Raman spectrum proves the lattice vibrations are not affected by the dielectric anomaly at T_1 . Raman spectrum results deepen the conclusion that the dielectric anomaly does not belong to a structure phase transition.



Figure 4. (*a*) Temperature dependence of Raman spectrum. (*b*) The temperature dependence of the Raman peak position shifts with reference to the peak at 303 K.

From DSC, PXRD and Raman measurements, we have reached a conclusion that the dielectric anomaly at T_1 is not structure phase transition. Then what is responsible for the dielectric anomaly? One reasonable explanation may be the freezing of ferroelectric domains at low temperature induced by the defects in DIPAB-C1. It has been reported that ferroelectric domains in some ferroelectric materials normally froze at low temperature [12,36], which can be attributed to the collective pinning of randomly distributed pinning centre to domain walls. Suppose the ferroelectric domains freeze below T_1 , the relaxation time will be long so that it is difficult to establish relaxation polarization and only instantaneous polarization is in operation. This could easily explain why the dielectric constant hardly changes with temperature and have less frequency dispersion below T_1 . Once the ferroelectric domains unfreeze at high temperature, the relaxation time will decrease with the increase of temperature, which behaves as that the frequency dispersion increase with the temperature increasing.

In addition, when the doping chlorine content *x* is low, the change ratio of the dielectric constant is small around T_1 . Electronic supplementary material, figures S3–S6 show the dielectric constant curves when x = 0.024, 0.062, 0.41 and 0.60. The dielectric constant increases at 500 Hz by 7.6%, 25%, 521%, 428% and 160% in the vicinity of T_1 when x = 0.024, 0.062, 0.24, 0.41 and 0.60, respectively. Therefore, the dielectric anomaly is closely related with the doping content. At low doping level, the defects density will increase with the doping content increase. The increased doping level will make freezing process more clear. It is noting that defects exist in pure DIPAB single crystals unavoidably, so the freezing process of ferroelectric domains exists unavoidably. This may explain that it is hard for the spontaneous polarization in DIPAB to flip through eternal applied electric field at low temperature.

4. Conclusion

A dielectric anomaly was found at 340 K in the chlorine-doped DIPAB, which does not appear in pure DIPAB. Relaxation modes are different near the dielectric anomaly temperature. Above the dielectric anomaly temperature, the dielectric frequency dispersion is larger, and below the dielectric anomaly temperature, the dielectric frequency dispersion is less. DSC, PXRD and Raman spectrum measurements show the dielectric anomaly does not come from structure phase transition and has no specific heat variety. The freezing of ferroelectric domains at low temperature may be responsible for the dielectric anomaly. The dielectric anomaly was induced by doping in DIPAB.

Data accessibility. Our data are deposited at Dryad Digital Repository: https://doi.org/10.5061/dryad.rs39j28 [37]. Authors' contributions. K.G. grew the single crystals, and took the dielectric and DSC measurements. Y.C. took the Raman spectrum measurements; B.Z. took the XRD measurements. K.G. and X.C. designed the experiments and wrote the paper. All authors participated in the discussion of the article.

Competing interests. We have no competing interests.

Funding. This work was supported by Natural Science Foundation of Jiangsu Province, China (BK20170482 and BK20170514), NSFC (no. 61704148).

rsos.royalsocietypublishing.org R. Soc. open sci. 5: 181397

6

References

- You Y-M et al. 2017 An organic-inorganic perovskite ferroelectric with large piezoelectric response. Science 357, 306–309. (doi:10.1126/ science.aai8535)
- Park SE, Shrout TR. 1997 Ultrahigh strain and piezoelectric behavior in relaxor based ferroelectric single crystals. J. Appl. Phys. 82, 1804–1811. (doi:10.1063/1.365983)
- Saito Y, Takao H, Tani T, Nonoyama T, Takatori K, Homma T, Nagaya T, Nakamura M. 2004 Lead-free piezoceramics. *Nature* 432, 84–87. (doi:10.1038/nature03028)
- Kuwata J Uchino K, Nomura S. 1982 Dielectric and piezoelectric properties of 0.91Pb(Zn_{1/3}Nb_{2/3})0₃-0.09PbTiO₃ single-crystals. *Jpn. J. Appl. Phys.* **21**, 1298 – 1302.
- Cox DE, Noheda B, Shirane G, Uesu Y, Fujishiro K, Yamada Y. 2001 Universal phase diagram for high-piezoelectric perovskite systems. *Appl. Phys. Lett.* **79**, 400–402. (doi:10.1063/1. 1384475)
- La-Orauttapong D, Noheda B, Ye Z-G, Gehring PM, Toulouse J, Cox DE, Shirane G. 2002 Phase diagram of the relaxor ferroelectric (1-x)Pb(Zn_{1/3}Nb_{2/3})0₃-xPbTi0₃. *Phys. Rev. B* 65, 144101. (doi:10.1103/PhysRevB.65.144101)
- Viehland D, Amin A, Li JF. 2001 Piezoelectric instability in (011)-oriented Pb(B¹_{1/2}B¹_{2/3})0₃ – PbTi0₃ crystals. *Appl. Phys. Lett.* 79, 1006–1008. (doi:10.1063/1.1392307)
- Wang HF, Cao WW. 2002 Determination of full set material constants of piezoceramics from phase velocities. J. Appl. Phys. 92, 4578–4583. (doi:10.1063/1.1505998)
- Park SE, Mulvihill ML, Risch G, Shrout TR. 1997 The effect of growth conditions on the dielectric properties of Pb(Zn_{1/3}Nb_{2/3})O₃ single crystals. *Jpn. J. Appl. Phys.* **36**, 1154–1158. (doi:10. 1143/JJAP.36.1154)
- Tu CS, Chao FC, Yeh CH, Tsai CL, Schmidt VH. 1999 Hypersonic and dielectric anomalies of (Pb(Zn_{1/3}Nb_{2/3})0₃)_{0.905}(PbTi0₃)_{0.095} single crystal. *Phys. Rev. B* **60**, 6348–6351. (doi:10. 1103/PhysRevB.60.6348)
- Samara GA, Venturini EL, Schmidt VH. 2001 Dielectric properties and phase transitions of (Pb(Zn_{1/3}Nb_{2/3})0₃)_{0.905}(PbTi0₃)_{0.095}: influence of pressure. *Phys. Rev. B* 63, 184104. (doi:10.1103/ PhysRevB.63.184104)
- Bao P, Yan F, Dai Y, Zhu J, Wang Y, Luo H. 2004 Dielectric relaxation in 91% Pb(Zn_{1/3}Nb_{2/3})0₃ – 9% PbTiO₃ single crystal at low temperature. *Appl. Phys. Lett.* **84**, 5317–5319. (doi:10.1063/ 1.1766081)
- Ye H-Y, Liao W-Q, Hu C-L, Zhang Y, You Y-M, Mao J-G, Li P-F, Xiong R-G. 2016 Bandgap engineering of lead-halide perovskite-type ferroelectrics. *Adv. Mater.* 28, 2579–2586. (doi:10.1002/adma.201505224)
- Fu D-W et al. 2013 Diisopropylammonium bromide is a high-temperature molecular ferroelectric crystal. Science 339, 425–428.

- Gao K, Liu C, Cui Z, Zhu J, Cai H-L, Wu XS. 2015 Room-temperature growth of ferroelectric diisopropylammonium bromide with 12-crown-4 addition. CrystEngComm 17, 2429–2432. (doi:10.1039/C4CE02567A)
- Gao K, Xu C, Cui Z, Liu C, Gao L, Li C, Wu D, Cai H-L, Wu XS. 2016 The growth mechanism and ferroelectric domains of diisopropylammonium bromide films synthesized via 12-crown-4 addition at room temperature. *Phys. Chem. Chem. Phys.* 18, 7626–7631. (doi:10.1039/ C6CP00568C)
- Jiang C, Lin H, Luo C, Zhang Y, Yang J, Peng H, Duan C-G. 2016 Crystal growth and dynamic ferroelectric hysteresis scaling behavior of molecular ferroelectric diisopropylammonium bromide. J. Cryst. Growth 438, 25–30. (doi:10. 1016/j.jcrysgro.2015.12.019)
- Piecha A, Gagor A, Jakubas R, Szklarz P. 2013 Room-temperature ferroelectricity in diisopropylammonium bromide. *CrystEngComm* 15, 940–944. (doi:10.1039/C2CE26580J)
- Gao K et al. 2014 Above-room-temperature molecular ferroelectric and fast switchable dielectric of diisopropylammonium perchlorate. J. Mater. Chem. C 2, 9957–9963. (doi:10.1039/ C4TC01882F)
- Gao K, Cui Z, Liu C, Zhu J, Cai H-L, Wu X. 2015 Two reversible ferroelectric phase transitions in diisopropylammonium perchlorate. *RSC Adv.* 5, 62 647 – 62 651. (doi:10.1039/C5RA09971D)
- Cai H-L, Zhang W, Ge J-Z, Zhang Y, Awaga K, Nakamura T, Xiong R-G. 2011 4-(Cyanomethyl)anilinium perchlorate: a new displacive-type molecular ferroelectric. *Phys. Rev. Lett.* **107**, 147601. (doi:10.1103/ PhysRevLett.107.147601)
- Fu D-W, Gao K, Cui Z, Gao L, Fu D-W, Cai H-L, Wu XS. 2013 4-Methoxyanilinium perrhenate 18-crown-6: a new ferroelectric with order originating in swinglike motion slowing down. *Phys. Rev. Lett.* **110**, 257601. (doi:10.1103/ PhysRevLett.110.257601)
- Liu C, Gao K, Cui Z, Gao L, Fu D-W, Cai H-L, Wu XS. 2016 New molecular ferroelectrics accompanied by ultrahigh second-harmonic generation. J. Phys. Chem. Lett. 7, 1756–1762. (doi:10.1021/acs.jpclett.6b00618)
- Cui Z, Gao K, Liu C, Yin Y, Fu D-W, Cai H-L, Wu XS. 2016 Molecular ferroelectric pyridin-4ylmethanaminium perchlorate undergoes paraelectric-ferroelectric and ferroelectricferroelectric phase transitions. J. Phys. Chem. C 120, 2925–2931. (doi:10.1021/acs.jpcc. 5b12348)
- Zhang Y, Liao W-Q, Fu D-W, Ye H-Y, Liu C-M, Chen Z-N, Xiong R-G. 2015 The first organicinorganic hybrid luminescent multiferroic: (pyrrolidinium)MnBr₃. Adv. Mater. 27, 3942–3946. (doi:10.1002/adma.201501026)
- 26. Zhang Y, Liao W-Q, Fu D-W, Ye H-Y, Chen Z-N, Xiong R-G. 2015 Highly efficient red-light

emission in an organic-inorganic hybrid ferroelectric: (pyrrolidinium)MnCl₃. *J. Am. Chem. Soc.* **137**, 4928–4931. (doi:10.1021/jacs. 5b01680)

- Xu W-J et al. 2016 Molecular dynamics of flexible polar cations in a variable confined space: toward exceptional two-step nonlinear optical switches. Adv. Mater. 28, 5886-5890. (doi:10.1002/adma.201600895)
- Zhang Y, Ye H-Y, Zhang W, Xiong R-G. 2014 Room-temperature ABX(3)-typed molecular ferroelectric: C₅H₉-NH₃ CdCl₃. *Inorg. Chem. Front.* 1, 118–123. (doi:10.1039/C3qi00058c)
- Ye H-Y, Zhang Y, Fu D-W, Xiong R-G. 2014 An above-room-temperature ferroelectric organometal halide perovskite: (3-pyrrolinium)(CdCl₃). *Angew. Chem. Int. Edit.* 53, 11 242–11 247. (doi:10.1002/anie.201406810)
- Ye H-Y *et al.* 2015 High-temperature ferroelectricity and photoluminescence in a hybrid organic-inorganic compound: (3-pyrrolinium)MnCl₃. *J. Am. Chem. Soc.* 137, 13 148 – 13 154. (doi:10.1021/jacs.5b08290)
- Liao W-Q, Zhang Y, Hu CL, Mao JG, Ye HY, Li PF, Huang SD, Xiong RG. 2015 A lead-halide perovskite molecular ferroelectric semiconductor. *Nat. Comm.* 6, 7338. (doi:10. 1038/ncomms8338)
- Fu D-W, Zhang W, Cai HL, Ge JZ, Zhang Y, Xiong RG. 2011 Diisopropylammonium chloride: a ferroelectric organic salt with a high phase transition temperature and practical utilization level of spontaneous polarization. *Adv. Mater.* 23, 5658–5662. (doi:10.1002/adma. 201102938)
- Sui Y, Chen WT, Ma JJ, Hu RH, Liu DS. 2016 Enhanced dielectric and ferroelectric properties in PVDF composite flexible films through doping with diisopropylammonium bromide. *RSC Adv.* 6, 7364–7369. (doi:10.1039/C5RA25371C)
- Gao K, Liu C, Cui Z, Xu C, Zhu J, Gao L, Cai H-L, Wu XS. 2016 A series of high-temperature molecular ferroelectric crystals: chlorine doped diisopropylammonium bromide. J. Mater. Chem. C 4, 1959–1963. (doi:10.1039/ C5TC04398K)
- Zhang QM, Bharti V, Zhao X. 1998 Giant electrostriction and relaxor ferroelectric behavior in electron-irradiated poly(vinylidene fluoridetrifluoroethylene) copolymer. *Science* 280, 2101–2104.
- Huang YN, Li X, Ding Y, Wang YN, Shen HM, Zhang ZF, Fang CS, Zhuo SH, Fung PCW. 1997 Domain freezing in potassium dihydrogen phosphate, triglycine sulfate, and CuAlZnNi. *Phys. Rev. B* 55, 16 159–16 167. (doi:10.1103/ PhysRevB.55.16159)
- Gao K, Zhang B, Cao Y, Chen X. 2018 Data from: Doping induced dielectric anomaly below the Curie temperature in molecular ferroelectric diisopropylammonium bromide. Dryad Digital Repository. (doi:10.5061/dryad.rs39j28)