

PERIODIC TABLE OF THE ELEMENTS

for
Medium Energy Ion Scattering

I II III IV V VI VII

hydrogen	1 H 1.0 14.01 hcp 3.75 (6.12) 1
	3 Li 6.9 452 440 bcc 3.491 7.6
lithium	4 Be 9.0 1550 1031 hcp 2.27 (3.59) 11
	11 Na 22.9 1042 155 bcc 4.225 23
sodium	12 Mg 24.3 924 330 hcp 3.21 (5.21) 24,26,25
	19 K 39.0 336.8 100 bcc 5.225 39,41
potassium	20 Ca 40.0 1120 230 fcc 5.58 40
	21 Sc 44.9 1812 476 hcp 5.27 (5.27) 45
rubidium	37 Rb 85.4 2520 59 bcc 5.585 85,87
	38 Sr 87.6 1042 148 fcc 6.08 88,86,87
caesium	55 Cs 132.9 1120 43 bcc 6.045 133
	56 Ba 137.3 1000 116 hcp 5.02 138,137,136,135
frankium	87 Fr [223.0] 1585 ?
	88 Ra [226.0] 970 ?

Element name
atomic number
atomic symbol
atomic weight
melting point (K)
Debye temperature (K)
@ 300 K
crystal type
a Ao (c Ao)
stable isotopes >1%
(in decreasing abundance)

Debye temperatures and thermal vibrations

The root mean square (rms) vibration amplitude is:

$$\sqrt{\langle u^2 \rangle} = \left(\frac{\hbar}{\theta_D} \right) \sqrt{\frac{3T}{mk}} \quad (\text{m})$$

where:

$m \equiv$ atomic mass (weight) (x 1.673 10⁻²⁷ kg) {approx.}

$\hbar = \frac{h}{2\pi} \equiv 1.05 \times 10^{-34}$ (m² kg s⁻¹) {Plank's constant}

$T \equiv$ absolute temperature (K)

$k \equiv$ Boltzman's constant (m² kg s⁻² K⁻¹)

$\theta_D \equiv$ Debye temperature (K) {at 300 K}

* Lanthanides	22 Ti 47.8 1948 380 hcp 2.95 (4.68) 48,46,47,49,50	23 V 50.9 2160 390 bcc 3.03 51	24 Cr 51.9 2160 423 bcc 2.88 52,53,50,54	25 Mn 54.8 924 363 cubic complex 55	26 Fe 55.8 1808 373 bcc 2.87 56,54,57	27 Co 58.9 1765 386 hcp 2.51 (4.07) 59	28 Ni 58.6 1726 345 fcc 3.52 58,60,62,61,64	29 Cu 63.5 1356 310 fcc 3.61 63,65	30 Zn 65.3 692.6 237 hcp 2.66 (4.95) 64,66,68,67	31 Ga 69.7 302.9 240 complex 69,71	32 Ge 72.6 1210.5 403 diamond 5.658 74,72,70,73,76	33 As 74.9 1090 (28 at) 275 rhomb 75	34 Se 78.9 490 90 hex. Chains 80,78,82,76,77	35 Br 79.9 265.9 complex 79,81	36 Kr 83.8 116.5 fcc 5.64 84,86,82,83,80														
	39 Y 88.9 1768 214 hcp 3.65 (5.73) 89	40 Zr 91.2 2125 250 hcp 3.65 (5.15) 90,94,92,91,96	41 Nb 92.9 2741 260 bcc 3.30 95	42 Mo 95.9 2880 377 bcc 3.15 98,96,92,95,100	43 Tc [98.9] 2500 422 hcp 2.74 (4.40) none	44 Ru 101.0 2520 415 hcp 2.71 (4.28) 102,104,101-98	45 Rh 102.9 2230 350 fcc 3.80 103	46 Pd 106.4 1825 275 fcc 3.89 106,105,110,104	47 Ag 107.8 1234 221 fcc 4.09 107,109	48 Cd 112.4 594.2 220 hcp 2.98 (5.62) 114,112,111,110,113,116	49 In 114.8 42929 108 tetr. 3.25 115,113	50 Sn 118.7 505.1 254 complex 120,118,116,6.49	51 Sb 121.7 903.7 200 rhomb 121,123	52 Te 127.6 722.6 153 hex chains 128,126,125,124	53 I 126.9 386.6 complex 127	54 Xe 131.2 161.2 fcc 6.13 132,129,131,134,136,130													
** Actinides	58 Ce 140.1 1070 138 fcc 5.16 140,142	59 Pr 140.9 1208 100 hex 3.67 (ABAC) 141	60 Nd 144.2 1297 148 hex 142,144,146, 145,143,148,150	61 Pm 144.9 1308 complex 152,154,147, 149,148,150,144	62 Sm 150.3 1345 184 complex 153,151	63 Eu 151.2 1100 118 bcc 4.58 158,156,157,155..	64 Gd 157.2 1585 155 hcp 3.63 (5.78) 159	65 Tb 158.9 1629 158 hcp 3.60 (5.70) 164,162,163,161..	66 Dy 162.9 1680 158 hcp 3.59 (5.65) 165	67 Ho 164.9 1734 161 hcp 3.58 (5.62) 166,168,167,170..	68 Er 167.2 1770 163 hcp 3.56 (5.59) 169	69 Tm 168.9 1818 167 hcp 3.54 (5.56) 174,172,173,171,176,170	70 Yb 173.0 1097 120 fcc 5.48 175,176	71 Lu 174.9 1925 116 hex. 3.50 (5.55) 177	72 Hf 178.4 178,177,179,176, 181	73 Ta 180.9 3269 225 bcc 3.30 184,186,182,183	74 W 183.8 3650 312 bcc 3.16 187,185	75 Re 186.2 3450 275 hcp 2.76 (4.46) 192,190,189,188	76 Os 190.2 3300 400 hcp 2.74 (4.32) 195,191	77 Ir 192.2 2716 129 fcc 3.84 195,194,196,198	78 Pt 195.0 2042 225 fcc 3.92 197	79 Au 196.9 1336.1 178 fcc 4.08 199,201,198,204	80 Hg 200.5 234.3 92 rhomb. 202,200, 199,201,198,204	81 Tl 204.3 576.6 96 hcp 3.46 (5.52) 205,203	82 Pb 207.2 600.4 87 fcc 4.95 208,206,207	83 Bi 208.9 544.4 116 rhomb 109 (alpha)	84 Po [208.9] 527 sc 3.34 none	85 At [209.9] 520 none	86 Rn [222.0] 202 none
	90 Th 232.0 2000 100 fcc 5.08	91 Pa 231.0 1500 138 tetr. 3.92 (3.24)	92 U 238.0 1405.4 300 complex 238 alpha	93 Np 237.0 ? 163 complex	94 Pu 244.0 ? hex 3.64 (ABAC)	95 Am 243.0 ? ?	96 Cm 247.0 ? ?	97 Bk 247.0 ? ?	98 Cf 251.0 ? ?	99 Es 252.0 ? ?	100 Fm 257.0 ? ?	101 Md 258.0 ? ?	102 No 259.1 ? ?	103 Lr [262.1] ? hcp 3.77 (ABAC)	The Debye temperatures are mostly from – www.knowledgedoor.com with references from – J. Phys and Chem. Ref. Data, vol. 8 and, of course, – Kittle C. Intro. to Sol. St. Phys., 8th ed. and also – CRC Handbook of Chemistry and Physics														