

THE

ONTARIO WATER RESOURCES

COMMISSION

INDUSTRIAL WASTES SURVEY

of the

STEEL COMPANY OF CANADA, LIMITED HILTON WORKS

Hamilton, Ontario



1968

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Report On

An Industrial Wastes Survey

of

THE STEEL COMPANY OF CANADA, LIMITED

HILTON WORKS

Hamilton, Ontario

September 9-13, 1968

Division of Industrial Wastes

ONTARIO WATER RESOURCES COMMISSION

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REPORT

Ontario Water Resources Commission

Municipality	Hamilton,	Ontario	Date	of Inspection.	Sept. 9-13, 1968.
Re:	THE STEEL	COMPANY OF CA	NADA, LIMITED	- HILTON	WORKS
Field Inspection	n by P. Kres J. D. I	;in, K. H. Egg Luyt	gers, Repo	т byJ.	D. Luyt

An industrial waste survey was conducted at the Hilton Works of The Steel Company of Canada, Limited, Hamilton, during September, 1968. The survey, carried out by the Division of Industrial Wastes, Ontario Water Resources Commission, was designed to:

- a) evaluate existing waste control facilities and procedures,
- b) locate those sources of water impairment within the plant still requiring corrective action, and
- c) determine or estimate the waste loadings discharged
 to Burlington Bay.

Similar major industrial wastes surveys, the last conducted during March, 1966, have been undertaken in the mast for primarily the same purposes. In addition to these major surveys, several sampling programmes have been conducted at the Hilton Works from time to time in the mast for various reasons, including the estimation of waste loadings exerted on

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Date Sept. 9-13, 1968.

the Bay. No attempt was made during the 1968 survey to evaluate the existing sanitary waste treatment and disposal methods. SUMMARY

The Hilton Works of The Steel Company of Canada, Limited, located on the south shore of Burlington Bay, is a fully integrated steel works producing some 3,800,000 ingot tons of steel per year. Positive steps have been taken by the Company to improve the quality of the waste effluents discharged. However, major problems still remain and are outlined in more detail in the body of this report. The largest single source of contaminating material was the 4-stand cold mill area discharging large quantities of emulsified oils to Burlington Bay. It is recommended that work on improving this and other unacceptable discharges be continued to develop and implement the appropriate waste control and treatment measures, and consequently, provide adequate protection to the receiving body of water. The Bay is in a severe state of impairment as shown by previous water analyses. sediment analyses and biological collections, and if the Commission's objectives are to be met, much tighter control and added waste control facilities and procedures must be instituted by the Company.

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DETAILS OF SURVEY

During the survey a great deal of assistance was given by personnel of The Steel Company of Canada, Limited including:

Mr.	Α.	J. Lafreniere	-	Supervisor, Air and Water Quality Control Group
Mr.	D.	Andrews	-	Area Engineer
Mr.	с.	A. Perl	-	General Duties Engineer

Special acknowledgement is directed to Mr. Perl who contributed much time and effort to the survey.

OWRC personnel participating were:

Mr.	Ρ.	Kr	esin	-	Student Engineer
Mr.	к.	н.	Eggers	-	Technician
Mr.	J.	D.	Luyt	-	Engineer,

all of the Division of Industrial Wastes.

Discussions with Company personnel were held prior to the sampling programme to familiarize OWRC personnel with the plant layout and to obtain up-to-date technical information regarding plant operations and waste control procedures. The sampling was carried out during the week of September 9, 1968.

For purposes of discussion, the Hilton Works may be conveniently divided into two parts, the "West Side" and the "East Side".

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DESCRIPTION OF PLANT AND PLANT OPERATIONS

The Hilton Works of The Steel Company of Canada, Limited is a fully integrated steelworks producing approximately 3,800,000 ignot tons of steel per year. The plant employs in the neighbourhood of 12,000 people and operates 24 hours/day, 7 days/week. Some areas of the plant operate considerably less than 21 shifts/ week, depending on demand.

The basic raw materials in the production of steel are coal, limestone, iron ore and steel scrap. The following amounts of these materials were received (not necessarily used) during 1967:

Iron ore	-	7,300,000	tons
Coal	-	1,950,000	tons
Limestone	-	300,000	tons
Steel scrap	-	200,000	tons plus 1,300,000 tons generated internally.

Other materials used are zinc, tin, sulphuric and hydrochloric acids, dichromates, sodium carbonate, sodium hydroxide, calcium oxide, oils, greases and solvents.

The average volumes of water used during 1967 were:

5.6 MIGPD City water -- 277.7 MIGPD Bay water

For purposes of discussion, the Hilton Works may conveniently be divided into two parts, the "West Side, and the "East Side."

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West Side

Processing facilities on the west side included four batteries of coke ovens, a sintering plant, four blast furnaces, two by-products areas, and a phenol recovery plant.

The coke ovens produced coke by heating bituminous coal in the absence of air. The volatile material was driven off leaving a hardened residue or coke. The volatile materials were recovered and converted into by-products including sodium carbolate, ammonium sulphate, tars and light ends in the By-Product and Phenol Recovery Plants. Coking capacity was approximately 2,000,000 tons of metallurgical coke per year. The capacity of the new #6 battery was 800,000 tons per year.

Pig iron was produced in the blast furnaces. Iron ore, limestone and coke were charged into the tops of the furnaces and heated air blown into the bottoms. Combustion of the coke produced the carbon monoxide and the heat required to achieve the metallurgical reducing reactions of the iron oxides to iron. The limestone formed a fluid slag which carried off the impurities in the ore.

The molten iron and slag were periodically drawn off, the slag being discarded and the iron cast into blocks known as "pigs" for subsequent processing. In the production of one

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ton of pig iron, approximately 2 tons ore, 1 ton coke and 1/2 ton limestone are used and 1/2 ton slag and 5 tons of blast furnace gas are produced.

The Sinter Plant converted fine powdered ores, reclaimed flue dust from the blast furnaces, mill scale, and coke fines from the coke quenching operation into lump form by burning the mixture to a clinker under a forced draft.

The two By-Products Plants and single Phenol Recovery Plant recovered ammonium sulphate, napthalene, light ends containing benzene, sodium carbolate and tars from the coke oven The gas was collected in mains flushed with ammonia liquor gas. and cooled in the primary gas coolers also using ammonia liquor. The ammonia liquor was, in turn, cooled using bay water with no direct contact between the two liquids. The excess ammonia liquor was routed to the phenol recovery and ammonia distillation units before being sewered. The cooled gases after tar removal in tar decanters and precipitators were scrubbed with a 6% sulphuric acid solution to produce ammonium sulphate. The residual gases were passed through tar-bottomed final coolers where the napthalene was entrained in the tar. Water used in the final coolers was recirculated in a closed air-cooled cycle.

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The gas was then passed through the light oil scrubbers where it was scrubbed with a wash oil. The oil stream was steam-stripped to drive off the light fractions which were subsequently condensed in a tube condenser and stored for sale. The steam-stripper bottoms or debenzolized wash oil, after water decantation, cooling and settling, was stored for re-use.

In the phenol extraction plant, the phenol in the excess ammonia liquor was absorbed in a light oil and extracted using a 50% caustic solution to form the product, sodium carbolate. The light oil was directed to the ammonia distillation columns for ammonia recovery, and subsequent processing into ammonium sulphate. The excess waste liquor, after removal of the phenol and ammonia, was passed through a sump to settle out the lime solids added in the process, and was then discharged to the plant sewers and ultimately to the Bay.

East Side

Steel making at The Steel Company of Canada, Limited used both the conventional open-hearth furnaces and the newer electric-arc furnaces. The open hearth furnaces were essentially shallow rectangular basins or hearths enclosed by refractory brick walls and roof. The raw materials were pig

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iron, iron ore, limestone, scrap, hot metal, and various alloying substances. The purpose of the operation was to reduce the impurities present in the scrap and pig iron. The molten product was poured into molds and cooled. The solidified steel castings were called "ingots". Electric-arc furnaces generated the required heat by an electric arc passing from the electrodes to the furnace charge, otherwise the operation was similar to that of the open-hearth furnaces.

The next series of operations may be grouped together as "hot-rolling". The cooled ingots were reheated in "soaking pits" and reduced to blooms (square in cross-section) or slabs (rectangular in cross-section) by compression between the surfaces of two rotating rolls. In the billet mill, the blooms were passed through a series of roll stands, and were converted to billets. In the rod mill, billets were reduced to rods used in Stelco's finishing plants to produce nuts, bolts, nails and screws. The bar mills processed blooms, slabs and billets into merchant bar products. The plate mill converted slabs into plate steel of various sizes. The hot strip mill further reduced rolled plate to a thinner product. In each of the above mills, the raw material was passed through scale breakers where high pressure water jets removed surface scale.

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Prior to steel finishing, the steel surface must be cleaned. The Steel Company of Canada, Limited used both sulphuric acid and hydrochloric acid pickling lines to chemically remove surface oxides and scale. No. 1 pickling line used sulphuric acid but operated intermittently with demand, No. 2 line used sulphuric acid but was to be converted to hydrochloric acid by the end of 1968, (this was done), and No. 3 line employed hydrochloric acid which was regenerated for re-use. Sulphuric acid pickling was also employed in the hot-strip finishing mill.

Spent hydrochloric acid from the No. 3 pickling line was directed to the regeneration plant. The acid was fed to a roaster to convert the iron chloride to iron oxide (Fe₂O₃), a saleable product. A mixture of chlorine gas, water and small amounts of iron chloride and iron oxide were passed through a cyclone to an absorber. The cyclone removed the remaining iron compounds and the adsorber combined the chlorine gas and water to form liquid hydrochloric acid. The exhaust fumes were waterscrubbed and sewered. The over-all regeneration reaction of the plant can be summed as:

 $4FeC1_2 + 0_2 + 4H_20 -- \ge 2Fe_20_3 + 8HC1$

Steel finishing operations included cold rolling, continuous galvanizing and electrolytic tin-plating. The cold

rolling mills, a 4-stand, a 5-stand, and a reversing mill further reduced thin steel strip and produced a smooth, dense surface. To dissipate the heat generated, an oil-water emulsion was directed in jets against the rolls and the steel surface during rolling.

Sources of Liquid Wastes, Treatment and Disposal West Side

In the west side area, waste effluents were contributed by the water scrubbing of blast furnace gas and from the By-Products Plants.

Scrubbing water from furnaces "B", "C" and "D" were passed through the 105-foot diameter Walker thickener, the overflow going to waste and the underflow to a 30-foot thickener near the Sinter Plant. During the survey, "D" furnace, the largest of the three was down for repairs. Scrubbing water from "E" furnace, the newest and largest furnace, passed through a 75-foot thickener with the underflow also directed to the 30-foot thickener and the overflow sewered. The overflow from the 30-foot thickener was routed to the Walker thickener.

No liquid wastes originated at the coke quenching station as all water drained to the breeze basins and was recirculated.

In the By-Products area, liquid wastes were discharged from the lime sump after phenol and ammonia recovery, the naphthalene sump when in use (i.e. when the old final gas cooling system was in use), and the cooling water systems.

All liquid waste effluents from the west side were discharged to Burlington Bay by way of two outfalls, the south "open cut" and the north "trunk" sewer.

East Side

The open hearth furnaces used large volumes of relatively uncontaminated cooling water on a one-through basis. No. 2 open hearth discharged to the East Side Lagoon, while No. 3 open hearth discharged through a separate small lagoon to the Bay.

The hot-rolling mills used large volumes of water both as a roll coolant and for descaling of hot-rolled steel. Water from descaling, containing settleable iron solids and oils, was passed through scale pits to remove the mill scale. The individual overflows from these pits were directed to one of the two lagoon systems (East Side Lagoon and 148" plate mill lagoon) where oil removal devices have been installed. The lagoons also served to remove more settleable solids, although the calculated efficiency was poor. The scale pits were reportedly dredged as required and the sludge processed at the Sinter Plant.

In the cold mills, residual oils and greases on the steel were removed with bay water. Rolling oils were used as the coolant during the rolling operations. The final steel cleaning stages, after rolling, consisted of detergent and water washes.

In the 80" 4-stand cold mill area, all liquid wastes passed through two 40,000 gallon oil separators connected in parallel. These wastes, as described above, consisted of the continuous water washes containing oil, the continuous overflow from the detergent recycle tank and approximately 10,000 gallons of spent rolling oil discharged once per month. In each separator, surface skimmers removed the floating oils and greases to a central trough. Part of the aqueous decant was directed to a pressurized chamber where air was dissolved in the liquid. As the liquid re-entered the separator at atmospheric pressure, the dissolved air, coming out of solution, assisted in separating the soluble oil from the water.

The 5-stand and reversing cold mills did not have this oil-water separation system. Instead, the wastewater from these areas was discharged to the East Side Lagoon where settling and oil removal took place.

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Two pickling lines, one of which was operated only intermittently, used sulphuric acid. No. 2 pickling line operated at about 45,000 tons of steel per month using about 45 lbs of acid per ton. No. 1 pickling line operated at about 5,000 tons of steel per month. Spent pickle liquor from these two lines was directed to the plant sewers. No. 2 pickling line was to be converted to hydrochloric acid by the end of 1968 with the acid regenerated for re-use. No. 3 pickling line employed hydrochloric acid. Hydrochloric acid fumes released at the pickle line were vented to a scrubber where they were dissolved in water and discharged to the plant sewers. Liquid wastes from the acid regeneration plant originated from a scrubber following the absorber. This scrubber removed any excess gases and vapours not picked up in the absorption columm.

At the Hot Strip Finishing Mill about 3,000 gallons per day of spent sulphuric acid pickling liquor were directed to a 15,000 gallon holding tank. The disposal system was designed to allow the acid to drain slowly to the Bay over a period of 10-15 hours, thus achieving dilution both before discharge, by the addition of cooling water and after discharge, in the Bay itself. No attempt at neutralization of the waste was made.

Other waste streams originated from the galvanizing, electrolytic tinning and heavy gauge shear lines.

All liquid effluents from the East Side eventually found their way into Burlington Bay, either through the municipal Depew Street storm sewer or directly by means of a number of outfalls.

Sampling and Analysis

Several visits were made to the Hilton Works to obtain the technical information required in the preparation of this report, to select the sampling locations, and to conduct the sampling programme. All sampling points are shown in Figures 1 and 2. The sampling was carried out during the week of September 9, 1968.

"Triplicate" composite samples were taken at most sample stations. Grab samples of minor waste flows and of unusual flow conditions were also taken. On September 11, a sudden rainstorm occurred at approximately 2:30 p.m. Compositing of the samples was stopped and grab samples of certain key flows were taken.

All samples were returned on the day of collection to the OWRC laboratories for analyses. The analyses were carried

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Ö PQR out for the most part in accordance with the procedures described in "Standard Methods for the Examination of Water and Wastewater". All analytical results are appended to this report.

Waste Volumes

Wastewater flows at most of the locations sampled were supplied by the Company's Utilities Department. The majority of the figures reported below were obtained by flow measurements made during or after the week of the survey.

<u>Location</u>* <u>Description</u> <u>Volume (MIGPD)</u> <u>West Side</u> A South "Open Out" (0.8 would be 50.58

Α	South "Open Cut"	40.8 usually 50-58 MIGP
В	International Harvester Sewer	1.3
С	Blast Furnace Area Sewer	24.2
D	105-foot diameter Thickener	14.9
E	Lime Sump Effluent	1.6
F	North Trunk Sewer	60.0 (approximate)
G	75-foot diameter Thickener	2.8
H	Phenol Recovery Area Sewer	250 IGPD
I	No. 2 By-Products and Phenol Recovery Area Sewer	18.6
J	Cast House, Lime Sump and No. 1 By-Products	not measured
К	Naphthalene Sump Effluent	not measured (1966 figure was 3.05)
	* Refer to figures 1 and 2	1/

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Location*	Description	Volume	(MIGPD)
East Side			
Α	City Storm Sewer (Depew Ave.)		not measured
В	Heavy Gauge Shear Line Sewer	4.8	(est. and not measured)
С	HCl Regeneration Plant, No. 3 Pickle Line and 4-stand Cold Mill	9.2	
D	HCl Regeneration Plant	1.4	(approximate)
E	Hot Strip Finishing Mill		not measured
F	148" Plate Mill Cooling Water	23.5	
G	148" Plate Mill Lagoon Influent	: 14.5	
н	148" Plate Mill Lagoon Effluent	: 14.5	
I	No. 3 Open Hearth Cooling Water	40.0	
J	South Influent to East Side Lagoon	42.3	
к	West Influent to East Side Lagoon	126.5	
L	Effluent from East Side Lagoon	126.5	
М	Continuous Annealing and Electrotinning	18.9	
N	5-stand Cold Mill and No.'s l and 2 Pickling Lines	22.6	
0	No. 2 Bloom Mill	6.3	
P	Low Lift Sump Station	60.2	
Q	Rod, Bar and No. 1 Bloom Mill	21.6	
R	56" Hot Strip Mill	12.0	
* Refe	r to figures 1 and 2		

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Estimated Waste Loadings

In calculating the waste loadings, the flow figures supplied by the Company and the average of the results obtained on the composite samples were used. Both the concentration and loading figures in the following tables have been corrected to show net figures (gross concentrations and loadings less service water concentrations and loadings). The following service water contaminant concentrations were used:

BOD ₅	9	ppm	Chlorides	41	ppm
Susp. Solids	14	ppm	Phenolic Equivalents	1.3	ppb
Diss. Solids	318	ppm	Cyanide	-	
Ether Solubles	0.5	ppm	Total Iron	1.09	ppm
Ammonia	2.5	ppm	Diss. Iron	0.13	ppm
Kjeldahl N	7	ppm	Zinc	0.14	ppm
Phosphates	0.4	ppm	Tin	0	ppm
Sulphates	80	ppm	Chromium	0	ppm

The figures reported below each table in the following sections were calculated as total plant loadings from data for the following sample points: West Side - A, B, and F and East Side - B, C, F, H, I and L. These figures do not necessarily balance with the figure that would be obtained by summing the individual in-plant loading figures obtained and reported.

It should be noted that Blast Furnace "D" was not operating during the West Side sampling programme and that Blast Furnace "E" was not yet operating at full capacity. Data for sample point West Side A have been corrected for the loading contributed by the International Harvester Company of Canada, Limited.

MAJOR SOURCES OF BOD5

Sample Location	Average Concentration (ppm)	Waste Volume <u>(MIGPD)</u>	Waste Loading (1b/day)	1966 Results (1b/day)
West A-Open Cut	19	39.5	7,500	26,750
West E-Lime Sump	611	1.6	9,780	3,220
East C - 4-stand Cold Mill, etc.	143	9.2	13,100	(not sampled)

The total net BOD₅ loading from the Hilton Works was calculated to be 26,600 lbs/day.

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MAJOR SOURCES OF SUSPENDED SOLIDS

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (1b/day)	1966 Results (1b/day)
West A - Open Cut	32	39.5	12,200	23,950
West C - Blast Furn Area	ace 37	24.2	8,950	11,200
West D - 105' Thick	ener 36	14.9	5,360	8,475
West E - Lime Sump	750	1.6	12,000	-
East C - 4-stand Co Mill, Etc.	ld 116	9.2	10,700	(not sampled)
East L - East Side Lagoon Eff	26 luent	126.5	32,900	57,000
East M - 110" Plate	Mill 41	18.9	7,750	2,015
East N - 5-stand Co. Mill	ld 46	22.6	10,400	10,340
East O - No. 2 Bloom Mill	n 66	6.3	4,150	(not sampled)
East Q - Rod, Bar an Bloom Mills	nd 31	21.6	6,700	37,350

The total net suspended solids loading from the plant was calculated to be 59,000 lbs/day.

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MAJOR SOURCES OF DISSOLVED SOLIDS

Sample Location	Average	Waste	Waste
	Concentration	Volume	Loading
	(ppm)	(MIGPD)	(1b/day)
East E-Lime Sump	10,260	1.6	164,000 (may not be representative)

The total net dissolved solids loading from the plant was calculated to be 154,000 lbs/day.

MAJOR SOURCES OF ETHER SOLUBLES

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (1b/day)
East C - 4-stand Cold Mill. etc.	1,240	9.2	144,000

The total net ether solubles loading from the plant was calculated to be 128,000 lbs/day. These figures may not be realistic but certainly do show that large quantities of oil were being sewered.

MAJOR SOURCES OF AMMONIA NITROGEN

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (lb/day)
West A - Open Cut	23.5	39.5	9,600
West E - Lime Sump	2,800	1.6	44,800*
West F - North Trunk	11.5	60.0	6,900

* Sample may not have been representative

The total net ammonia loading from the plant was calculated to be 16,800 lbs/day.

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MAJOR SOURCES OF KJELDAHL NITROGEN

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (1b/day)
West A - Open Cut	26	39.5	10,000
West F - North Trunk	9	60.0	5,400

The total net Kjeldahl nitrogen loading from the plant was calculated to be 15,800 lbs/day.

MAJOR SOURCES OF PHOSPHATES

Sample	Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (lb/day)
East F	- 148" Mill Cooling Water	2.4	23.5	564
East L	- East Side Lago Outlet	on 1.2	126.5	1,520

The total net phosphate loading from the plant was calculated to be 2,344 lbs/day.

MAJOR SOURCES OF PHENOLICS

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (lb/day)	1966 Results (1b/day)
West A - Open Cut	2 500	39.5	1.020	718
West E - Lime Sump	35,000	1.6	560	113
West I - #2 By-Prod	ucts 100	18.6	19	-
East M - 110" Plate	Mill 53.7	18.9	10	45.5

The total net phenolic equivalents loading from the plant was calculated to be 1,030 lbs/day.

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MAJOR SOURCES OF CYANIDE

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (lb/day)	1966 Results (1b/day)
West A - Open Cut	1.45	39.5	590	2,750
West C - Blast Furnace	Area 1.3	24.2	314	420
West E - Lime Sump	18	1.6	288	56

The total net cyanide loading from the plant was calculated to be 630 lbs/day.

MAJOR SOURCES OF SUSPENDED IRON

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (1b/day)	1966 Results (1b/day)
West D - 105' Thickener	19.1	14.9	2,709	1,010*
East L - Lagoon Outlet	26.2	126.5	33,143	-
East N - 5-stand Cold N	(ill 28	22.6	6,100	52,100*
East O - #2 Bloom Mill	45.6	6.3	2,810	(not sample)

* Total Iron

The total net suspended iron loading from the plant was calculated to be 39,070 lbs/day.

MAJOR SOURCES OF DISSOLVED IRON

Sample Location	Average Concentration (ppm)	Waste Volume (MIGPD)	Waste Loading (lb/day)
East L - Lagoon Outles	7.7	126.5	9,740
East N - 5-stand Cold	Mill 162	22.6	36,600

The total net dissolved iron loading from the plant was calculated to be 10,188 lbs/day.

The Steel Company of Canada Page 23 Report re Limited - Hilton Works Date Sept. 9-13, 1968. Batch or Periodic Discharges Location Material Quantity Hot Strip Finishing spent pickling liquor 113,000 lbs H₂SO₄ 117,000 lbs Fés04 20,000 1bs HC1/mo. 131,000 lbs H₂SO₄/mo. No. 1 Pickle Line spent pickling liquor (used intermittently) No.2 Continuous Annealcaustic cleaning solu-3% NaOH solution ling and Cleaning tion dumped to oil containing iron and Line sludge pit dirt; 15,000 ga1/4 mo. 5-stand Cold Mill 4,500 gals twice/week 6% sheet oil and 10% tin oil dumped to and 5,000 gal/week oil sludge pit respectively 0.1% detergent 4,000 gal/week solution 4-stand Cold Mill 4-8% sheet oil to 10,000 gals/mo. separator Reversing Mill 6% sheet oil and 10,000 gals/mo. 9% tin oil Tinning Line 10% H₂SO₄ solution 100 tons/month 1.5% caustic solution 3,000 gals/week 1,600 gals/2 weeks 2.5% sodium bichromate solution 1% chromic acid 1,500 gals/2 weeks 10% HC1 10° - 13° Be ZnNH₄C1 Galvanizing Line 3,100 gals/4 weeks 750 gals/2 weeks $1\% \text{ KCr0}_3 + 1\% \text{ H}_3 \text{PO}_4$ 1,000 gals. periodically Sheet Mill 4 - 5% H₂SO₄ and 2,500 gals/month 12-15% Féso, 12-10 Finishing Mill 5% H2SO4 and 11% FeSO4 3,000 gals/month

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DISCUSSION OF RESULTS

West Side

Coke Ovens A)

The three coke quenching stations have been converted to total recirculation systems without blowdown and with makeup from the oil decanters. Therefore, there was no effluent discharge from the breeze settling basins. This has eliminated one former source of phenolics and suspended solids.

B) Blast Furnaces

Before and during the survey, the new "E" furnace had not reached full capacity. Also, two of the remaining four furnaces were down, "D" furnace temporarily and "A" furnace permanently. Therefore, the flows and loadings presented in this report could be expected to be lower than normal.

The overflow from the new 75-foot diameter thickener (2.8 MIGPD) contained high concentrations of suspended solids (33 ppm) and cyanide (0.5 - 1.1 ppm). Iron at 5.25 ppm was relatively low. At the control point (F), the cyanide concentration had dropped to less than 0.1 ppm but the suspended solids had increased to 50 ppm. Since "E" furnace had not reached full capacity, the thickener efficiency would be expected to decrease.

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The overflow from the 105-foot diameter thickener serving "B", "C" and "D" furnaces also contained high concentrations of suspended solids (50 ppm) and high suspended iron (19 ppm). The main sewer servicing the blast furnace area contained high concentrations of suspended solids (51 ppm) and cyanide (1.3 ppm). The control point (A) contained similar concentrations of these materials as well as other contaminants.

C) Sinter Plant

The 30-foot diameter thickener overflow was routed to the 105-foot diameter thickener thus eliminating a direct discharge to Burlington Bay.

D) By-Products and Phenol Recovery Plants

A number of waste control improvements had been made in the By-Products areas. The refining of light oils had been discontinued. Oil contaminated water from decanters in the light oil stripping section was routed to the coke quenching stations rather than being discharged to the sewers. In 1965, a closed water recirculation system was provided to eliminate the loss of phenols and cyanides from the final gas coolers. Blowdown was directed to the quenching stations. A Phenol Recovery Plant was in operation to recover phenols as sodium phenolate from the excess ammonia liquor.

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> In By-Product area No. 1, the old final gas cooling system and naphthalene settling sump were in operation at the time of the survey. This system is used when the new final gas cooling system is not operating or not operating satisfactorily. A high strength waste in terms of BOD₅ (440 ppm) was generated here. Cyanides and phenols would also be expected to be present but these analyses were not performed on the sample. Also in No. 1 By-Product area, the lime sump used to settle lime solids from the effluent originating at the ammonia stills was not functioning efficiently. The settled solids in the sump had built up to such an extent as to reduce its settling capacity to almost nil. Either the sump is not cleaned as often as is required, or it is greatly undersized. The overflow contained high quantities of suspended solids (764 ppm), BOD₅ (620 ppm), phenols (35,000 ppb), cyanides (18 ppm) and ammonia (2,800 ppm). These contaminants, excepting the suspended solids, would probably not be removed by improved settling. They were also present in excessive amounts downstream in the sewerage system at points J and A (open cut).

In the phenol plant area a very small flow estimated at 250 GPD containing 20,000 ppb phenolics was found (point H). The origin of this waste was not known. Downstream at Point I, 100 ppb phenols was found.

East Side

Page

E) No. 3 Open Hearth

The cooling water effluent from the No. 3 Open Hearth was satisfactory for direct discharge to a natural watercourse during the two sampling periods. However, visible evidence of severe impairment (red discolouration) was observed at another time during the plant survey. This was later reported to be the result of an annual equipment wash-down. This type of discharge must be eliminated or adequately treated before being allowed to gain access to the Bay.

F) 148" Plate Mill

The cooling water effluent from the 148" Plate Mill was satisfactory for direct discharge to a watercourse. The process wastewater was directed through a scale pit and a secondary settling lagoon equipped with a belt-type oil skimmer at the outlet. Visual examination of the facility indicated the scraper mechanism was not functioning efficiently, probably due to a worn-out scraping blade. The ether solubles concentration of the lagoon effluent was not extremely high but was approximately identical to the ether solubles content of the lagoon influent.

The suspended solids concentration was reduced from 48-63 ppm to 26-29 ppm. Examination of the lagoon showed that it was in need of dredging.

Hot Strip Finishing G)

Spent pickle liquor containing 113,000 lbs H₂SO₄, 117,000 1bs FeSO4 and 20,000 1bs HC1 was discharged to the Ottawa Street slip monthly. The discharge of spent pickle liquor at this location was observed and sampled during the evening of September 12, 1968. Disposal facilities consisted of a tank located below the pickling floor. Reportedly, the acid was dumped from the pickling tanks to the disposal tank and allowed to drain to the slip over a period of 8-10 hours (about 3,000 gallons/day). At 10.00 p.m., September 12, the disposal tank was 75% full before receiving the daily batch of spent liquor. The additional spent liquor then caused an overflow directly to the slip.

It was reported that no attempts at neutralization before discharge was carried out.

The discharge of spent pickle liquor at full strength is capable of causing severe impairment of the receiving body of water. Complaints describing periodic low pH water in the slip have been received from non-Stelco personnel in the past.

The practice of discharging spent pickle liquor at a slow rate is also not acceptable, particularly in a relatively quiescent body of water.

H) Acid Regeneration Plant

The liquid waste effluent from the acid regeneration plant was characterized by a reddish-brown discolouration indicative of fine particles of suspended iron. Low pH values were also obtained on one grab sample (3.1-3.5). It was reported that a part of the high solids and iron content could have been caused by a plant wash-down. Further downstream in the discharge sewer (point C), the acidity had been effectively neutralized but the suspended solids and iron concentrations had increased.

4-stand Cold Mill I)

At Point C high concentrations of suspended solids, iron and oil were present. Wastes from the 4-stand cold rolling area were passed through an oil separator using dissolved air flotation but, based on the ether solubles results, the separator was not effective in removing the oil. In addition to continuous oil losses from the rolling area, there were certain batch discharges which would certainly not be effectively treated by the existing facilities. The 4-stand cold rolling mill area was not serviced by the East Side Lagoon. Instead, it discharged directly to the

Page

municipal storm sewer with no additional treatment before discharge to the Bay.

J) Heavy Gauge Shear Line Area (Sample Point B)

The samples collected from the sewer servicing the Heavy Gauge Shear Line which also discharged to the City storm sewer, indicated the waste effluent at the times of sampling was satisfactory for discharge to a storm sewer or watercourse.

K) East Side Lagoon System

The East Side Lagoon had two influents, termed the south inlet and the west inlet. The south inlet served a large part of the rolling and finishing operations, and was characterized by high suspended solids (largely iron), high ether solubles and phenol concentrations and by intermittent low pH's. The pH of the lagoon outlet, although still acidic, was not seriously affected.

The west inlet contained quantities of suspended solids and, at times, oil.

The lagoon was equipped with a baffle at the outlet to collect floating oil and debris. The oil was pumped from the lagoon to an oil sludge pit located on high ground on the bank of the channel flowing into Burlington Bay. Water was allowed to drain back to the lagoon. The location of the
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sludge pit was poor. Evidence of overflows of oil from the pit to the water channel downstream of the lagoon outlet was observed. The efficiency of the lagoon was calculated to be:

> Suspended solids: -5% Iron (total) : 26% Ether solubles : 70%

These efficiencies were calculated based on the two sets of composite samples collected on September 10 and 11. At approximately 2.30 p.m., September 11, a rainstorm occurred which would have altered the characteristics of the wastewater and therefore, sampling was stopped. The efficiencies for September 12, the day after the rainstorm were:

> Suspended solids: -25% Iron (total) : 31% Ether solubles : trace of oil reported in lagoon outlet

At the time of the survey, the first half of the lagoon was being dredged. The second half was in need of dredging as "islands" of solid material were present above the surface of the water.

Although the ether solubles content of the samples collected at the lagoon outlet were relatively low, ranging from

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Date. Sept. 9-13, 1968.

traces to 19 ppm, oily patches were seen on the surface of the water channel downstream of the outlet at all times of observation.

L) Accidental Spillages and Emergency Discharges

Accidental spillages must be reported to the Commission through the Division of Industrial Wastes as soon as they are discovered. According to the Ontario Water Resources Commission Act, Section 27(1), the discharge of any material, whether deliberate or accidental, which has the ability and potential to impair water quality is an offence. A critical examination of the processing facilities at Stelco's Hilton Works, as at any other industrial plant, would contribute towards minimizing impairment of the receiving body of water resulting from accidents, errors or negligence. The aim of such a survey should be the development and implementation of measures for containing spills and eliminating operating errors.

In a large industrial complex such as the Hilton Works, there are procedures that are carried out intermittently and which result in the discharge of contaminating material. Examples include equipment wash-downs and de-scaling of cooling jackets in blast furnaces. These are planned events and therefore sufficient time to develop procedures to adequately handle the

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Report re Limited - Hilton Works Date Sep

Date Sept. 9-13, 1968.

waste effluents should be available. Prior to the event, the Commission should be informed of the nature of the discharge expected and the procedures planned to prevent water impairment. Since the characteristics of the waste waters would vary widely, it is recommended that the Division of Industrial Wastes be consulted during the planning stages to determine the objectives to be achieved in each instance.

M) Oils and Greases

In the Hamilton Harbour area, impairment of the water by oil is a major cause for concern. Oil is not only aesthetically objectionable when seen floating on water or along shorelines, but it is also harmful to aquatic life and imparts tastes and odours to the water. It is felt that emphasis should be placed on the control of oil from all sources in the area. Of particular importance in this respect was the discharge from the 4-stand Cold Mill building containing 1,240 ppm ether solubles. This concentration may or may not be representative but it does show that greatly improved control was required.

N) Iron

The 1964-65 OWRC biological survey of Burlington Bay revealed that there was an area of several hundred acres near the steel mills devoid of aquatic life, including sludgeworms.

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The report stated the area of absence of sludgeworms in the Bay coincided very closely with the area where the iron content of the sediments exceeded 25% and suggested the iron had a toxic Therefore, the need to prevent further effect on the worms. accumulations of iron is apparent and if a balanced biological habitat is to be achieved, the iron already in the Bay must be removed.

The east side lagoon outflow contained unacceptable quantities of iron in terms of both concentration and loading. A concentration of 26.2 ppm suspended iron and 7.7 ppm dissolved iron in a flow of 126.5 MIGPD represents a loss of some 42,800 lbs. of iron per day.

CONCLUSIONS

Although positive steps have been taken by Stelco to reduce the discharge of contaminants to Burlington Bay from its Hilton Works, the implementation of additional waste control procedures are necessary.

The tables included in this report indicate the areas that were the major sources of the individual contaminants. Suspended solids, phenolics, cyanide, ammonia, iron and oil are the waste components of concern.

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> A comparison of waste loadings calculated during this latest survey and those reported after previous surveys is made below:

	1964	1966	1968
Production	less than 3,000,000 ingot tons	-	3,800,000 ingot tons
Waste Flow (MIGPD)	173	217.4	280
BOD ₅ (1bs/day)	31,000	38,000	26,600
Susp. Solids (lbs/day)	107,000	98,000	59,000
Iron (1bs/day)	80,000	-	49,200
Phenol (1bs/day)	1,050	-	1,030
Cyanide (1bs/d a y)	3,250	2,750	630
Ether Solubles (lbs/day)	-	-	128,000*
Ammonia Nitrogen (1bs/day)	-	-	16,800
Phosphates (1bs/day)	-	-	2,344
	" ACCULACY YUESL	Ullapte	

RECOMMENDATIONS

Items considered to require immediate attention to improve existing equipment or to develop and install adequate waste control procedures and facilities are listed in an

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approximate order of priority as follows:

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1) The waste effluent discharge from the 4-stand Cold Mill area. Greatly improved control of the oil-bearing wastes was required. The average ether solubles concentration found was 1,240 ppm in a flow of 9.2 MIGPD. This flow was discharged directly to a municipal storm sewer with no additional treatment.

2) The existing methods of treating and disposing of spent pickling liquor. At the time of the survey, No. 3 pickling line was on hydrochloric acid and acid regeneration. At the 1968 year-end, No. 2 line was also on hydrochloric acid and acid regeneration. Conversion to hydrochloric acid and acid regeneration or alternate adequate treatment and disposal methods should be found for No. 1 pickling line and the Hot Strip Finishing pickling line.

3) The operation of the final coolers at the coke mlant. The use of these coolers was intended to eliminate the direct contact cooling of ammonia liquor. During the survey, direct cooling of the liquor was being carried out, at least martially, resulting in the loss of quantities of cyanide, phenols and organic material through the maphthalene sump. This situation should be corrected.

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4) The settling efficiency of the East Side lagoon. The efficiencies calculated for its settling characteristics showed it had almost no function as a settling basin. The lagoon has been constructed in such a way as to generate strong currents with the practical absence of quiescence to promote good settling. The lagoon was being dredged at the time of the survey. However, the lagoon has consistently been in excess of OWRC objectives for many years with respect to both suspended solids and iron.

During 1966, the lagoon was divided into two compartments to enable new trackwork to be constructed. At that time it was understood that the existing lagoon would be relocated within a few years. It is recommended that this relocation to provide a new and larger terminal settling and oil removal basin now be undertaken. This would not only act as a final polishing unit following the primary treatment units within the plant, but would also act as a device capable of providing emergency treatment in the event of acid spillages, oil losses or similar discharges due to equipment failure or other causes. As an example, acid losses could be discharged over a period of time to the lagoon while simultaneously manually adding neutralizing The solids of neutralization, including suspended iron, agents. would then be contained in the lagoon.

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Studies into the following problem areas should be continued or initiated and the appropriate treatment methods developed:

The reduction or elimination of the phenolics, 5) cyanides and ammonia in the waste stream discharged from the lime sump.

The reduction or elimination of the cyanide in the 6) waste flow from the blast furnace area.

Improvement of the settling efficiencies of the 7) thickeners, particularly the 105-foot thickener.

The treatment or containment of slag quench water. 8) Although this aspect of the Company's operations was not examined, it is expected that an unacceptable waste discharge could originate at this source.

The following items are those requiring improved maintenance procedures or modifications to existing operating procedures:

Regular and more frequent dredging of the East Side 9) lagoon be instituted.

Improved maintenance of the lime sump at the ammonia 10) plant to improve its operation. More frequent dredging or

Page 39 Report re Limited - Hilton Works Date Sept. 9-13, 1968. providing increased settling capacity is required.

11) Discharges of contaminated waste water through the No. 3 Open Hearth cooling water sewer be discontinued or adequately treated before discharge to the Bay.

12) Regular maintenance of the belt-type oil skimmer at the 148" mill lagoon be provided.

Contingency plans should be developed to adequately handle accidental oil losses to Burlington Bay. It is felt that the availability of a portable floating oil boom would be of value in minimizing the effects of any possible accidental oil loss to the Bay.

Periodic reports giving the results of the Company's waste effluent sampling programme should be submitted from time to time. Details of the sampling programme have already been discussed with Company personnel.

The Steel Company of Canada, Limited should submit a report to the Commission outlining the Company's future waste control programme. The report should include both short and longterm projects designed to provide the required protection of the receiving waters as outlined in the Commission's "Objectives for Water Quality Control".

GENERAL COMMENTS

One of the purposes of this survey was to point out unsatisfactory areas in the plant's existing waste disposal

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methods. The nature of the survey does not allow the formulation of a complete set of recommendations to rectify all objectionable situations. The tables included in this report indicate the more serious sources of water impairment.

Some rather difficult problems have presented themselves, some of which are:

- the elimination of cyanides from the Blast furnace and coke plant wastes
- the very efficient oil removal system required to comply with the Commission's objective that not more than a faint irridescence result in the receiving water
- the removal of the dissolved iron, and
- the removal of the nutrients, nitrogen, and phosphorus from the waste discharges.

However, it is believed that an intensive effort on the part of the Company would result in a major improvement in the quality of the waste water discharged to Burlington Bay.

Prepared by:

G. D. Lugt

J. D. Luyt, P. Eng., Division of Industrial Wastes.

Approved by:

F.C. Stewart

R. C. Stewart, P. Eng., District Engineer, Division of Industrial Wastes.

JDL/maf

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in

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1 p.p.m. = 1 mgm. / litre

p.m. unless	otherwise i	ndicated									= 1	5./100,000	Imp. Gals
Municipa	lity: Har	ilton		Repo	rt to:	J. Luyt *				c.c.	Ches.	Lab. +	
Source:	Stelco		K.H	gers									
Date San	npled: Sep	t. 9/68	by: J.Lu;	yt									(rj)
Lab.	5-Day		Solida		COD	pli at	Phenols	Tet	AS Te	Zinc	rree n:onia	Tin	Phospi
No.	B.O.D.	Total	Susp.	Diss.		Lab.	in ppb	106.	VI 36.	22 VU	as 1	ES 30	
T-2500	75	474	59	415	330	6.9	30	4.33	0.90	C.25	5.9	Tr	7.0
1-2907	14	31 8	10	306	49	7.2	0	C.80	0.10	C.12		Ĩr	
T-2908	3.6	412	34	278	11	6.8	20	12.40	0.39	0.17		îr	
1-2909	3.6	402	33	369	11	6.5	15	12.78	0.40	0.12		Ir	
T-2910	2.4	450	7	443	16	3.1	20	1.06		0.04		0.0	
1-2911	7.6	228	7	221	22	7.5	3	0.44	c.1c	0.08	3.1	0.0	2.5
1-2912	5.0	534	1	533	36	2.7	4	**		0.01	1.5	0.0	U.3
1-2913	2.0	202	2	200	16	7.6	3	0.66	0.45	0.05	0.49	0.0	c.1
						Tr -	less than	L.1					
						** 5	erple extu	usted, To	st coul	d not	e perf	red.	
T-2906	A1	City S	Storm Sew	er	Compo	site 10.3	C AM - 4.3	O PN					
1-2907	B1	licary	Gege Line	e		r.	¥.						
1-2902	11	HCL I	Regenerat	ion Plant	**	10.3	0 AH - 12.	00					
1-2909	D2	-15		۶.	•	1.0	C PK - 4.3	C P#					
1-2510	D3	No. 3	Pickle L	inc - Fin	se 'dater	Grab 1	C.30 AF						
1-2011	1	Het St	trip Finis	shing Mis	1 - Morth	Fewer	Grab 9.3	O AF					
1-2010	22				South	Sewer No	.1 C	rab 2.00	PX				
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The Steel Limited -Company of Canada, Hilton Works

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Sept. 9-13, 1968

INDUSTRIAL WASTE ANALYSIS

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All analyses except pH reported in n n m unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

ource: ate San	stelco npled: Sei	1. 3/64	J. 1 by: ^{N.H.}	ecers										(r.)
Lab.	5-Day		Solids		Atter Scipples	Chloride as Cl	Chrosius as Cr							
No.	B.O.D.	Total	Susp.	Diss.									Ŧ.	
1-2-06					45	76	0.00							
-2:07					5	91	0.00							
-2908						70	0.00							
1-2909						74	0.00					80		
7-2910						169	0.00			•				
1-2911					5	35	0.00							
1-2912					**	7 6	0.00							
1-2913						34	0.00							
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						++ Saag	le estaus	ted, Te	it could	DOL	Dey	ericr	. D9	
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		sie oni	R PAGE P	ua desch	Lr:10a									
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		see oni	R PAGE P	ur disch	17:10:									
		see oni	E PAGE P	ur descr	17:10:									
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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

Report to: J. Layt .

1 p.p.m. == 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Ches. Lab. *

c.c.

Municipality: Stelce Source:

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Date Sampled: Sept. 9/66 by: P.E

Haz11ton

			Golida		ma	Phenols	1805 L	Te .	110	101.	[Ether	Chier	199ee	
Lab.	5-Day BOD		Solids	Die		In ppo	707.	DISS.	as SE		Solubl	es ĉi	Aneca	a
1-7614	A.A.	Total	Susp.	D188.	36	0	2.45	A 114	0.00					
- 0/3 E	4.0			744	20		6.9 7	V.20	0.30				-	
1-2917	4.0	212	У	203	10	•	1.06	0.10	0.00	0.00	5	44	3-2	도급
T-2 516	3.6	326	29	297	11	0	1.19	0.26	C.00	0.00	7	43	1.9	E. e
1-2917	4.0	324	15	509	16	3	2.30	0-14	0.00	0.00	3	45	3.1	ted
	pil at Lob.	Total Kieldahl	Phosphol as PQ	28	Zinc as 48					-				- HI Co
T-2914	7.7	-			-									
1-2915	7.5	3.6	2.0		0.12									ă ă
1-2910	7.8	2.1	0.8		0.08									Woo
1-2917	7.7	0.0	C.2		0.09									rks
I-2714	C]	Lagoos (14tr*	Influent Plant Hil	i C	mposite 5	9-40 AB - 4	1.40 PF	1		<u></u>	1	1		lada,
T-2915	n	148.1	till Cooli	ng inte	r Cos	posite 9.4	0 - 4.40	PK						
7-291(E1	Lagoos (14:"	Plant H1	Com	posite 9.4 n)	IG AM - 4-4	IC PN							Sept
1-2917	11	~3 Cm	n Bearth	Couling	water -	at cutfall	l into bag	Cos	posite		- 4.30	PH		
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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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Municipality: Hamilton

Report to: J. Luyt *

c.c. Chem. Lab.

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(rj)

Source: Stelco

Date Sampled: Sept. 10.68 by: K.H. Egger

6 P

Lab	5-Day		Solids		COD	pH at	Phenols	Ether	Iron	as Fe	Zinc as Zn.	Tin as Sn	Chromi as Cr	um
No.	B.O.D.	Total	Susp.	Diss.		Lab.	IN pps	DOLUDIES	Tot.	Diss.				1
T - 2922	85	464	78	386	270	6.8	12	68	5.75	1.35	0.35	0.00	0.00	
T-2923	26	424	23	401	40	7.1	12	5	1.00	0.20	0.08	0.10	0.00	Line
	Free Ammonia as NH _e	Phosphate as PO4	e Chlorid as Cl	e Total Kjeldahl as N										ited - H
T -2922	4.2	5.5	82	4.7										
T-2923	2.3	0.9	46											l n n
														Woj
														ike c
											L			
T - 2922	A2	City St	torm Sewe	r	Comp.	9.00 AM -	4.30 PM							ļ,
T-2923	∼ B2	Heavy (Sauge Lin	e		11	н							
														Sel
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Municipal	ity: He	erilton		Repo	ort to: J. 1	layt *				c.c.				4
Source:	Stelco S.pt	. 10/68	K.F.										(rĵ)	
Date Sam	npled:		by:P.K.	J.L.	pH at		Phenols	Iron a	s Fe	Lther	S Zinc	Tin	Chromi	un
Lab.	5-Day		Solids		lab.	COD	in ppb	TOT.	DISS.	201001	as Zn	88 51	as Cr	
No.	B.O.D.	Total	Susp.	Diss.	2001						0.11	0.00	0.00	$\frac{1}{1}$
1-2924	8.8	314	9	305	7.2	16	10	1.38	0,10	2	0.11	0.00	0,00	
r-2925	12.	368	63	305	7.3	31	o	16.50	0,20	22				
1-2926	11.	330	36	294	7.3	23	3	12.3	0.11	43				
1-2927	17.	516	76	440	5.7	36	10	0.03	30.0	67				
		-												
				*										
	Po	1461	Plata +11		g Water	(Comp. 9	.OC AM - 4	. 30 PM						
1-2924	12	140.	Tiate Mill	Lecon	Influent		n	4						
1-2925	G2	148.1	Place Mill	n - West	Inlet	· 10	.30 - 4.30)						
T- 2926	K1	Last :	side mague	Sout	h Inlet		H H							
1-2927	J-1			5041										
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INDUSTRIAL WASTE ANALYSIS

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All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipa	lity: Ha	milton		Repor	t to:	J. Luyt *	•			c.c.	Chem. 1	ab.	
Source: Date Sar	Stelco mpled: ^{Se} p	t. 10/68	P.K. k.i.	å J.L.							-		(r j)
			Solida		pli at	COD	Phenols in ppb	Iron at	He DICC	Zine se Zn	Tin as Sn	Chrosi as Cr	
Lab. No.	5-Day B.O.D.	Total	Susp.	Diss.	Lab.			101.	D122.	ab 611			
T - 29 3 0	13	384	5 9	325	6.7	36	12	27.5	1.60	0.11	0.6	0.00	
	Phosph as Po	orus	Frec Ammenia as NH _a	Chloride a s Cl	Total Kjeldahl as N	Ether Solubles							
1-2930	1.9		2.0	43	8 .C	19					-		
-2930	L- 1	E st	Side Lagoo	n iffluer	nt Comp.	10.30 0	4.30				1	1	

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100.000 Imp. Gals.

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	Han	ilton			J	Luyt *							
Municipa Source:	ity: Stelco			Repor	t to:					c.c.			
Date San	Sept. npled:	. 10/68	P.K. by: J	K.E. .L.	pH at		Phenols	Iron as	Fe	Zinc	Tin	Chromium	
Lab.	5-Day		Solids		Lab.	COD	in ppb	Tot.	Diss.	as Zn	as Sn	as Cr	
T-2931	6.0	Total 306	Susp.	Diss. 295	7.2	16	0	2.50	0.20	0.20	0.00	0.00	
T-2932	3.0	326	9	317	7.5	9	0	0.95	0.08	0.04	0.00	0.00	
	Phosph as PO ₄	orus	Free Ammonia as NH ₅	Chloride as Cl	Total Kjeldahl as B								
T- 2931	0.6	-	4.0	45	6.0				-				
T -293 2	0.3		1.9	45	8.0					-			
										1			
1-2931	1		West	Bay Water	Intake (P	unphouse "	at No. 2	DOCK)	uomp.	. 1.00 -	- 4. <u></u> 01	14	
T-29 3 2	2		Last	Day water	Intake		10.1						

INDUSTRIAL WASTE ANALYSIS

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All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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(rj)

Chess. Lab.

c.c.

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Report to: J. Luyt *

Source: Stelco

Municipality:

Date Sampled: S pt./0/68 by: P.K.

hamilton

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					nliat		Phenols	IRO		Zinc	Tin	Ether	
Lab.	5-Day		Solids	1	Lab	CLD	in ppb	107.	DISS.	as Zn.	as Sn	Solubles	
N0.	B.O.D.	Total	Susp.	Diss.									
T-2933	9.6	344	26	318	7.5	31	C	10.5	0.15	0.09	0.00	20	
T- 29 34	7.6	332	15	317	7.1	16	o	3.40	0.15	0.12	0.00	18	
	Chromium as Cr	Phospho: as PO ₄	rus	Free Annonia as NH ₂	Chlorid as Cl	8							The Stee Limited
T-2933	0.00	0.2		1.7	45								=
2-2934	0.00	C.3		1.8	48				÷			- K	11 tomp
													ion
													Woj
													rks C
-2933	H2	148'	Plate M	ill Lagoon	Effluent	Conn	. 9.30 AM	- 4.30 PM					anao
1-2934	12	NO. 3	Open H	earth Cocl	ing Water		н	"					la,
					0								s
													ept.
													9-1
													ω,
													19
													68.

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipa	ality:			Rep	ort to:				c.c.		4
Source:											9
Date Sa	mpled :		by:								
Lab.	5-Day		Solids		I_RON	AS F	<u>.</u>				1
No.	B.O.D.	Total	Susp.	Diss.	Tot.	Diss				 	1
1 2943					9,38	0.10					
T 2044					115	94					돈埥
T 2945					31.5	2.00					E.e.
1 2946					21.4	0.30					ted
T 2947			1		2.30	0.25					e1
T 2948					19.0	0.24					HC
T 2949					6.05	0.13					1 to
T 2950					113.	64.8					ny
T 2951	h.				27.8	0.32					Nor
T 2952					20.5	0.35					ks C
T 2953					206.	122.5					ana
T 2954					128.	0.60					da
T 2955					14.6	0.62					"
T 2956					17.0	1.20					
T 2957					1.05	0.25					
T 2958					9,50	2.15					Sel
T 2959					160.	130					ř.
T 2060					72.0	1.25					9-1
T 2961					0.50	0.08					μ
T 2962					0,50	0.06					15
											89(

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. \equiv 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipa	lity: Ham i	lton		Repo	rt to: J	Luyt				c.c.	Chem.L	ab.	
Source: Date Sar	Stelco	pt. 11/68	by: <u>R</u> .va	nS. & P.K	resin								86
Lab.	5-Day		Solids		Chloride	sAmmonia	Chromium	COD	Phenol	spH	Zinc	Tin og Sn	Sther
No.	B.O.D.	Total	Susp.	Diss.	as u	B. 5 M	es or		ru ppe	at hav	Zn	19 41	
2943	n.	33 8	23	315				43	30	7.2	-	-	7
2944	34	766	67	699		-	-	122	50	3.2			22
2945	10	422	32	390	42	2.6	0.45	20	3	5.8	0.18	0.00	trac
2946	6.4	37 6	52	324				20	0	6.6	-		2
i 2947	6.4	310	12	298		-		20	4	7.1	-	-	trac
			Phospate as PO_	5		Trace =	less than	2 ppm					
T 2945			0.5						,				
T 2943	K- 2	Vest	Influent	to East S	ide Lagoo	on Com	p. 9:00) - 2130		J	L	1	_
T 2944	j_ 2	South	Influent	to sast	Side Lag	oon Com	p. 9:00) - 2 :30					
T 2945	L- 2	Sast	Side Lag	oon Afflu	ient	Co	mp. 9:00) - 2 :3 0					
T 2946	Q-1	Rod,	Bar, & Bl	coming Mi	11 s	Com	p. 10:00	0 - 2:30					
T 2947	R-1	56 "	llot S	trip Mill		Com	p. 10:00	0 - 2:30					
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The Steel Limited -

Company of Canada, Hilton Works

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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Municip	ality: Hamil	ton Repor	rt to:	T Turet				c.c.	Chem.L.	ab.*	
Source:	Stelco			0.100y (
Date Sa	mpled: Sep	t. 11/68 by: P.K./R.S./J.L	./								ge
Lab.	5-Day	Solids	000	Phenols	Sther	Chromium	pH	Zinc	Tin	Chlori	les

NO.	B.O.D.	Total	Susp.	Diss.	005	In ppo	soluties	as or	at Lao	as 6n	as on	as u	1
T 2948		43 6	79	357	31	0			7.8			-	
T 2949		394	52	342	23	6			7.6				EB
T 2950		892	130	76 2	62	80		-	3.5			-	ni t
T 2951		398	45	353	40	15			6.4	-	-		ed
T 2952		526	124	402	71	12		-	7.0		-		· ¤
T 2953		1126	86	1040	56	35			2.7	-	-	-	Eg
T 2954		586	238	348	51	0	-		6.4			66	to
T 2955	84	464	56	406	420	20		0.00	6.6	0.27	0.00	89	n y
T 2956	8.0	388	33	355	36	6	6		7.4	-	-		or
T 2957	6.4	340	29	311	20	3	3		7.4	-		-	ks
													nad
T 2948	P-2	Low Lif	t Sump						Grab	3130 PM			a,
T 2949	K-3	West I	nfluent t	o dist S	ide Lagoor	1			Grab	3:30 PM	l		
T 2950	J-3	South *	-										
T 2951	L-3	Effluen	t from						Grab	3145 PM	l		Se
T 2052	M-2	110 "Pl a	te Mill C	ontinuou	s Annealli	ing Election	rotiming		Grab	3:30 PM			pt.
T 2053	N-2	5-stand	Cold Mil	1 # 1 &	2 pickler	reversing	g cold mil	1	•				9
T 2954	·C-2	4-stand	Cold Mil	l Acid R	legeneratio	on, # 3 pi	ickler		Grab	4:00 PM	l		L.
T 2055	Cel	4-stand	Cold Hil	1 Acid R	legenaratio	on, 1 3 p	ickler		Comp.	10:30	- 2:30		-
T 2056	0-1	2 Dlo	oming Mil	1					Comp.	0:30 /	14-3:30	PM	196
T ::957	P-1	Low Lif	t sump st	ation					Comp.	D:30 A	m-2:30	Pn	8

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ONTARIO WATER RESOURCES COMMISSION CHEMICAL LABORATORIES INDUSTRIAL WASTE ANALYSIS

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All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Date Sa	mpled: Sep	t. 11/68	by: R.van	s.									ge
Lab.	5-Day		Solids		Chronius	Sulphate	Ether	Zinc	Tin	рH	COD	Phenol	\$
No.	B.O.D.	Total	Susp.	Diss.	as cr	85 304	SOLUDICE	as 2n	as Sn	at Lat		an ppo	
T 295 8	18	35 6	77	279	2.5	81	4	0.61	0.60	7.1	43	100	
2959	26	1542	78	1464	0.17	480	10	0.04	0.00	2.9	83	12	
T 2960		5 92	158	434	-							-	
T 2961	12.	328	17	311	-	91	2	0.25	0.00	7.2	31	0	
T 296 2	11.	390	26	364		70	0	0.09	0.00	7.4	11	2	
	Phosphate as PO4	s Ammonia as N	Chloride as Cl	8									
T 2961	0.2	3.2	40										
T 296 2	0.3	2.1	40										
T 2958	N-1	110 - F	Plate Mill	, Contin	nous Annea	alling, El	ectrotimi	ng (Com	p. 9100	- 2130)			
T 2959	N-1	5-stand	cold mil	1, / 1 3	e # 2 pick]	Len revers	ing cold	mill "		•			
T 2960	1	Hess -	Von_Bullo	w Efflue	ent to Basi	t Lagoon (omp. 10:3	0 - 2130					
T 2961	1 /	- // 1 Bay	wa ter Pum	phouse (ves terly) Comp 10:	30 - 2:30	• 3					
T 2962	2	i 2 Bay	water Pum	phouse (e asterly)) Comp 101	30 - 2130						
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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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Municipa	ality:	amilton		Repo	rt to: J	. Luyt *	×			c.c.	Chem.	Lab.*	/rd	53
Source:	3	telco												
Date Sar	mpled: _{Sep}	t. 12/68	by: J.L											
Lab.	5-Day		Solids		pH at		Phenols	Iron	as Fe	• Phos-	Ammonia	Chlori	des Eth	er
No.	B.O. D.	Total	Susp.	Diss.	Lab.	C.O.D.	ppb	Tot.	Diss.	as PO4	N.	a.s.	Solubl	es
T-2965	16	328	16	312	7.6	32	2	1.59	0.17	0.6	2.5	39	tr.	
T-2966	5.0	314	7	307	7.8	11	4	0.49	0.16	0.20	1.5	38	tr.	Line
														fited
														- H
														L1 tor
														N WO
														rks
														Inada
Í-1905	1.	· . 1 ·	laywater.	Pumphouse	- Comp.	10:30	- 12:00 n	oon. (t	riplic.	te)				
1-1945	<u></u>	lo, 2	- Bayanet	er Rumpho	use - Jo	mp. 10:	30 - 12 . 3	0 noon (triplic	ate)				Sept
														9
														-13,
														19
														68.

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

npled: Sep 5-Day	iteleo t. 12/68	by: J.L.										
pled: Sep	ot. 12/68	by: J.L.										
5-Day			P.K.				.					
BOD		Solids		pH at		Phenols	Iron	as Fe.	Ether	Zinc	Tin as	
D .0. D .	Total	Susp.	Diss.	Lab.	C.O.D.	in ppb	Tot.	Diss.	Solubl	es Zn.	Sn.	
220	470	203	267	7.0	15,000	20	9•3	3.5	1240.	0.15	0.00	
				÷								
						5. 1						
∖ 0-3	HCL	regenerat	ion plant	, #3 pic)	kler, 4.	- stand co	old mill ·	- Comp.	9:00 -	· 4:30	(tripl	icate)
	3-3	220 470 0-3 HOL	d-3 ROL regenerat	d-3 HCl regeneration plant	220 470 209 207 7.0 3-7 HCl regeneration plant, #3 pick	220 470 203 207 7.0 13,000 J-J HOl regeneration plant, #3 pickler, 4	3-3 KOl regeneration plant, #3 pickler, 4 - stand co	220 470 205 207 7.0 15,000 20 9.5 0-5 KOl regeneration plant, #3 pickler, 4 - stand cold mill -	220 478 205 207 7.0 15.000 20 9.5 5.5 0-3 X01 rejeneration plant, #3 pickler, 4 - stand cold mill - Comp.	220 470 203 207 7.0 15,000 20 9.3 5.3 1240.	220 440 205 207 7.0 15,000 20 9.5 5.5 1240. 0.15 0.5 301 rejeneration plant, #3 pickler, 4 - stand cold mill - Comp. 9100 - 4130	220 470 205 207 7.0 15,000 20 9.5 5.5 1240. 0.15 0.00 3.5 3.5 1240. 0.15 0.00 3.5 5.5 1240. 0.

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipa Source :	ality: H S	amilton telco		Repor	rt to:	J. Luyt * P. Diosad	y *			c.c.	Chem.	Lab*		55
Date Sal	mpled: S	ept. 12/68	by: J.L.		I	H. Vanesc	he *						br	
Lab. No.	5-Day B.O.D.	Total	Solids Susp.	Diss.	Ether Solubles									
T 2968	The eth examina A chrom semi gr fuel oi	er soluble tion of th atographic ease. The 1 and (2)	s consist is oil ga separati se two ma tallow on	ed of a l we a spector on of the terials w a simila	1360 light brow trogram s extracts ere chars ar animal	n oil ha suggestin yielded acterized fat.	ving a fa g a mixtur an appro: by infra:	tty odour. re of petr mimately f red spect	Infra oleum o 0:50 mi roscopy	ared spe ails and xture o as (1)	ctrosco l natura of an oi a light	pic l fats. l and bunker		Limited - Hilton Works
т 2968	√ C-4 .	H Cl r	egenerati	ion plant,	,#3 pickle	es, 4 - s	tand cold	mill — G	rab 10:4	10 a.m.				Sept. 9-13, 1968.

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

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1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Source:		Stelco									
Date Sar	npled: _{Sep}	t. 12/68	by: J. L	uyt							
Lab.	5-Day		Solids		pH at	Iron	as Fe.	Chloride	Acidity		
No.	B.O.D.	Total	Susp.	Diss.	Lab.	Tot.	Diss.	Cl.	as CaCO ₃	 	
-2969		462	89	373	3.5	21.5	2.00	116	100.0		
											÷
£.							4				
									÷.		
]					_L	1		 1	
-2969	► D-3	Acid R	egenerati	on Plant	- Comp.	10:30 -	2:00 p.m.	(duplic	ate)		

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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Date Sat	mpled: Sep	t. 12/68	by: J.L.										
Lab.	5-Day		Solids	F	pH at		Phenol	Iron	as To.	Zinc	Tin	Ether	
NO.	B.O.D.	Total	Susp.	Diss.	Lab,	C.O.D,	ipb	Tot.	Mes.	Zn.	Sn.	Soluble	.
297 0	30	1030	22	1008	3.1	68	70	173	159	0.19	0.4	20	
F-2971	13	356	25	331	7.2	28	2	9.10	0.13	0.10	0.1	2	
T -297 2	18	488	30	458	5.9	337	2	425	19.4	0.15	0.00	tr.	
Lab. No.	hlorine l as Cl. /	ree monia as I.	Total H Rjeldahl 85 H.	as FO ₄		_							
-2970													
-2971													
	44	<u> </u>	5.4	2.4				_		1		1	
-2970	J - 4	South	Influent i	to East i	de Lago	on - Comp.	9:30 -	4:30 (ta	riplicate)			
-2971	K - 4	Vest I	nfluent to	Bast Sid	le Legooi	n - Comp.	9:30 -	4:30 (ta	riplicate)			
-2972	L - 4	Efflue	nt from Ea	st Side L	acoon -	Comp. 9:	130 - 4130) (trij	plicate)				

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipal	lity:	muilton		Repo	rt to:	J.layt *				c.c.	Chen.	iad.®	/mi
Source: Date San	npled :	itel:0	by:	1									
Lab. No.	5-Day B.O.D.	Total	Solids	Diss	pil at		Phonole in	Iron	as Fe.	2 inc 86	Tin as	Ether	Chron
T-2973	14	360	3 2	336	7.0	C.O. ₽. 40	10	8.95	0.90	0.11	1.2	tr.	0.6
T-2974	50	1236	43	1193	2.9	90	15	220	194	0.19	0.00	9	0.0
					a.								
					4								
						¥.							
"_2272		Contin	teus Ann	cellin: 2	Dector	untra Cre	an. 0.80 -	4.30 (• - []] []				
7-202/		5 Ston	d Cold S	411. ² 1.			- 0.20	- 4.20	er alice				
1-6714		<i>)</i>		AAA CI C		ue	<i>w</i>	- 4150 (ortheres				

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

	36	pt. 12/00	Solida	Eggers		pH	Iron	Ra Fe.		Dthem	24	9 4	
Lab. No.	5-Day B.O.D.	Total	Susp.	Diss.	C.O.D.	at Lab.	Tot.	Dise.	in ppb	Solubl	2111C 8.6 96 20.	as Sa.	
T-2975	18	412	127	285	112	7.7	75.8	0.35	0	12	0.10	0.00	
T-2976	10	372	29	343	23	7.7	9.39	0.05	6	24	0.11	0.00	
-297 5	02	#2 Blo	oming Mil	1	Composi	lte 9130	a.m 41	30 p.m.	(triplic	cato)			
5-2976	P3	Low 11:	ft sump s	tat1on	Composi	lte 9:30	a.m 41	30 p.m.	(triplic	zate)			
[-297 5 [-2976	02 P3	#2 Blo Low lis	oming Mill ft sump s	l tation	Composit	lte 9130	a.m 41 a.m 41	30 р.н. 30 р.н.	(t ripli c)	xate) xate)			

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

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Lab.	5-Day		Solids			PH	Iron	as Fe.	Phenols	Ether	Tin	21nc
No.	B.O.D.	Total	Susp.	Diss.	C.O.D.	Lab.	Tot.	Dise.	ppb.	Soluble	a \$n.	Zn.
r -297 7	12	390	3 8	35 2	32	7.7	18.1	0.30	o	34	0.00	0.11
T-297 8	16	356	16	340	40	7.7	2.85	0.15	٥	40	0.00	0,12
-2977	e 2	Rod. Be	r. #1 Blo	oom 111	Compos	aite 0:3/) a m _ /					
0070						52.06 717		+•) • p •m•				
-2910	n2	50 110	t Strip I	1 11	Compos	site 9:3) a.g 4	130 p.m.				

10M-60-22690

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INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municipa	lity:	Hamilton	1	Repor	t to:	J. Luyt	*			c.c.	Chem. 1	Lab.* /	rd
Source:		Stelco											
Date San	npled: _{Se}	pt. 12/68	by: R.V.	s.									
Lab.	5-Day		Solids		pH at	Acidity	Sulphate	s_Ir	o n				
No.	B.O.D.	Total	Susp.	Diss.	Lab.	CaCO ₃	SO4	Tot.	Diss.				
T-3042		140,608	**	**	0.5	156,800	134,000	3,460	3,460				
T-3043		90,704	**	**	0.8	75,200	59,000	1,540	1,540				
				** v	ery acid:	ic samples	attack f	ilter pa	per.				
												1	
									-				
						÷							
												-34)	
T-3042	►-2	Not St	trip Fini	shing Mil	1 - South	h Outfall	- Grab 1	.0:30 p.m	•				
т-3043	► E-3	Hot St	trip Fini	shing Mil	1 - South	n Outfall	- Grab 1	1:15 p.m	•				
			(both same	les taker	n durina d	lisposal c	of spent	sulphurid	e acid	pickli	ng solui	tion.)
			```	nee naaroot ann - onooronaeus ቸ		ý	<b>_</b>				•		

10M-60-22690

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All analyses p.p.m. unles	except pH rep s otherwise i	orted in ndicated			Indust	rial Wast	es			$\begin{array}{c}1 \text{ p.r}\\=1\end{array}$	m. = 1 lb./100,0	mgm. / li 000 Imp. Ga
Municip	ality: H	lamilton		Repor	rt to: ,	J. Luyt *				c.oghem. La	b. *	/rd
Source:	5	Stelco										
Date Sa	mpled : Se	pt. 13/68	by:	L.							_	
Lab.	5 Day BOD	S O L	I D	S	pH at Lab		Phenols in .	Cyanide	Free Ammonia			
	5.0.2.	100.	ousp.	2100,	Dab.	C.O.D.	ррр	as HCN	as N.		1	-
-3011	620	11342	764	10578	10.1	3025	35,000	18.0	2800.			
-3012			<del>-</del>				40,000					
<b>r-301</b> 3								10.2*				
					* Test	performe	on pres	rved sam	ple.	-		L U P
												4
			•									
5-3011	E-1	Lime	Sump Ef:	fluent	Grab	10:45 a.	<b>m</b> .					
<b>F-301</b> 2	E-2	Lime	Sump Ef:	fluent	Grab	2:00 p.m	•					
<b>F-301</b> 3	E- 3	Lime	Sump Ef:	fluent	Grab	2:00 p.m	•					
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The Steel Company of Canada,

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Municipality Source: Date Sample Lab. No. B C-3014 C-3015 C-3016 C-3017	y: Ha St led: Sept B.O.D. 29 	umilton celco c. 13/68 <u>S 0</u> Tot. 492	by: <u>ID</u> <u>Susp.</u> 46 	Report S Diss. 446	pH at Lab. 7.4	J. Luyt	* Fluoride as F.	Phenols in pub	Iron	c.c. as Fe.	Chem.	Lab.*	/rd
Source: Date Sample Lab. No. B C-3014 C-3015 C-3016 C-3017	St led: Sept B.O.D. 29 	elco 13/68 <u>5 0 1</u> Tot. 492	by: <u>ID</u> <u>Susp.</u> 46 	<u>S</u> Diss. 446	pH at Lab. 7.4	C.O.D.	Fluoride as F.	Phenols in pub	Iron	as Fe.	Ether		/rd
Date Sample Lab. No. B 2-3014 2-3015 2-3016 2-3017	B.O.D. 29 	2. 13/68 S 0 1 Tot. 492	by: <u>ID</u> <u>Susp.</u> 46 	<u>S</u> Di <b>ss.</b> 446	pH at Lab. 7.4	C.O.D.	Fluoride as F.	Phenols in pub	Iron	as Fe.	Ether		
Lab. No. B 2-3014 2-3015 2-3016 2-3017	B.O.D. 29 	<u>5</u> 0; Tot. 492	<u>J</u> I D Susp. 46	<u>S</u> Diss. 446	pH at Lab. 7.4	C.O.D.	Fluoride as F.	in ppb	Iron	as Fe.	Ether		
No. B -3014 -3015 -3016 -3017	в.о.р. 29  	Tot. 492 	<u>Susp.</u> 46 	Diss. 446	Lab. 7.4	C.O.D.	as F.	in ppb					
2-3014 2-3015 2-3016 2-3017	29  	<b>49</b> 2 	46 	446	7.4				Tot.	Diss.	Sol.		
r-3015 r-3016 r-3017						81	2.0	2500	6.45	0.38	2		
2-3016 2-3017													
-3017								2500*					
	1.8	354	15	339	7.4	43	0.9	4	4.20	0.40	Tr.		
Cy: Lab. as	yanide s HCN A	Free Pr mmonia as N.	osphates as PO ₄	Kjeldahl Nitrogen as N.			2	* Tes	perfo	med on	preserv	ed sam	le.
-3014	1.45	26	1.0	33.									
-3015 :	1.20*												
-3016													
-3017	0.06	14.	0.4	16.									
-3014		A-1	West	Side Main	South C	Outfall - (	Comp. 9:	30 - 4:30					
-3015		<b>A-</b> 2	West S	Side Main	South (	Outfall - (	Comp. 1:0	00 - 4:30					
-3016		A-3	West S	Side Main	South	Outfall -	Comp. 1.0	00 - 4:30	)				
-3017		F-1	Main (	outfall of	"E" FL	irnace Area	a Comp.	9:30 - 4:	30				

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Municipa	ality:	Hamilton		Re	port to:	J	Luyt *			c.c.	Chem. 1	Lab.*	/rd
Source:		Stelco											
Date Sa	mpled: S	ept. 13/68	by: J	.L.	P.K.	K.E.	R.P.						
Lab.	Cyanide	Phenols											
No.	as HCN	in ppb								 			
T-3018	0.50												1
T-3019		50											
			Test j	erforme	d on pr	eserv	ed sample	•					
<b>.</b>										 	1		
-3018	F-2	Main	Outfall	of "E"	Furnac	e Are	a Comp.	1:30 - 4	: 30				
T-3019	F-3	Main	Outfall	of "E"	Furnac	e Are	a - Comp	. 1:30	- 4:30				

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Municip	ality: Ha	milton		Repo	rt to:	J.Luyt *				c.c.	Chem.	Lab.*	/rd	] 。
Source:	St	elco												
Lab.	5 Day	s 0	by: R.F	D S	pH at		Fhenol in	Iron a	s Fe.	Ether	Cyanide as	Free F Ammonia	hosph- ates a	je] s
No.	B.O.D.	Tot.	Susp.	Diss.	Lab.	C.O.D.	ppb	Tot.	Diss.	Soluble	s hicn	as N.	PO4	1
-3020	19	<b>56</b> 2	68	494	6.5	121	20	10.5	0.15	o	0.05	1.9	2.0	6
-3021											0.01*			-
-3022							20-							-
														1
				* Tea	st perform	aed on pre	served sa	maple.						
							v	-					•	1
			÷											
-3020	B-1	Interna	tional Ha	rvester S	Sewer - Co	omp. 9:30	- 4:30							
-3021	B-2	Interna	tional Ha	rvester S	Sewer - Co	omp. 1:00	- 4:30							
-3022	B-3				C	omp. 1:00	- 4:30							
														1-1-1
														1

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1 p.p.m. = 1 mgm. / litre All analyses except pH reported in = 1 lb./100,000 Imp. Gals. p.p.m. unless otherwise indicated INDUSTRIAL WASTES c.c. Chem. Lab.* Report to: Municipality: J. Luyt * 66 Hamilton /rd Source: Stelco Date Sampled: by: 13/68 Sout, Chlorides Total Diag. Acidity Sulphates pH Lab. Iron Iron 88 at 8.8 25 No. Guilly 804 <del>n</del>. as Pe. as Pe. Lab. 8.60 9.00 376 336 T-3023 2.5 27 The Steel Limited -1570 264 495. 491. T-3024 1.8 340 Hilton Works Hot Strip Finishing Mill - South Large Outlet - Grab 9:45 a.m. E-5 T-3023 Hot Strip Finishing Mill - South Small Outlet - Grab 9:45 a.m. 5-6 T-3024 Sept. 9-13, 67 1968

Company of

Canada,

5M-60 12969

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ll analyses .p.m. unless	except pH rep s otherwise i	ported in ndicated			INI	USTRIAL	WASTES				1 p.p.m = 1 lb.	= 1  mg /100,000	m. / litre Imp. Gals.
Municipa	ality: Ha	amilton		Rep	ort to: J.	Luyt *				c.c.	Chem.	Lab.*	
Source:	S	telco										/rd	
Date Sa	mpled: Sej	pt. 13/68	by: J	.L.							1		
Lab.	5 Day	<u>S 0</u>	L	I D	S pH at		Phenol in	Iron a	s Fe.	Ether	Cyanide		
No.	B.O.D.	Tot.	Susp.	Diss.	Lab.	C.O.D.	ppb	Tot.	Diss.	Sol.	as HCN		
-3025	49	468	38	430	8.1	140	5000	3.30	0.30	2	4.90		
-3026	4.0	444	51	393	7.3	40	4	2.28	0.17	tr.	1.30		
					trace =	less th	an 2 ppm						
-3025	J - 1	Cast h	ose, Lim	e Sump a	and By-Produ	uct #1 -	Comp. 9:3	0 - 4:30.					
- 30 26	C - 1	Blast	furnace	area	- Comp	. 9:30 -	4:30.						
<i></i>					-								

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ll analyses .p.m. unles	except pH re s otherwise	ported in indicated			I	NDUSTRIA	L WASTES				1 p.p.r = 1 lb	n. = 1 m 0./100,000	gm. / litr Imp. Gal
Municipa	ality: Ha	milton		Repor	t to: J	. Luyt *				c.c.	Chem.	Lab.*	/rd
Source:	St	elco											
Date Sa	mpled: Sep	t. 13/68	by:										
Lab.	5 Day	<u>S</u> 0	L I	D S	pH at		Phenols	Iron a	s Fe.	Ether	Cyanid	e Fluor	ide
No.	B.O.D.	Tot.	Susp.	Diss.	Lab.	¢.0.D.	ppb	Tot.	Diss.	Sol.	as HCN	as F.	
							(0)	r 07	1.05		0.50	4.0	
-3027	1.6	478	33	445	7.3	20	60	5.25	1.23	Ur.	0.50	4.0	
-3028											1.10		
				* Test	perform	ied on pro	eserved sa	mple.					
r_3027	G=1	75° thic	kener ove	rflow - C	omposite	- 9:30 -	4:30.						
- 3027	0-1		kener eve		omposito	- 1.30	- 4:30						
- 30 20	G=2	75 thi	Kener ove	11104 - 0	omposite	- 1.0	4.)01						

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l analyses p.m. unless	except pH rep s otherwise i	ndicated				INDUSTRIA	L WASTES				1  p.p.m = 1 lb.	/100,00	0 Imp. Ga	.la.
Municipa	ality: S	telco -	Hamilton	Repor	t to:	J. Luyt	×			c.c.	Chem.	Lab.	* /rd	
Source:														
Date Sa	mpled:	1 3-1/0	by:	-										
	5 Day	pt. 13/68 S 0	LI	D S	рH		Phenols	Free	Total					1
Lab. No.	B.O.D.	Tot.	Susp.	Diss.	at Lab.	C.O.D.	in ppb	Ammonia as N.	Kjeldah Nitroge	1 n as N				
T-3029	440	356	10	346	6.8	810		3.0	6.0					
<b>F - 30</b> 30							20,000							
F-3031				-			20,000*							
20.0														
				* Test	performe	d on pres	erved samp	le.						
			8											
								1						-
m ~0.00		M4 h	- <b>1</b>	- 0	. Cmah	10.30 2								
r-3029	K-1	Napth	alene Sum	p Overiio	W GIAD	10:50 a.	, 141 p							
<b>F-3030</b>	H-1	Pheno	1 Recover	y Area -	Grab 11:	30 a.m.								
T-3031	H-2	Pheno	l Recover	y Area -	Grab -	11:30 a.m	n.							
														~

Municipa	lity:	Hamilton		Repor	t to:	. Luyt *				c.c.	Chem.	Lab.* /	rd
Source:		Stelco				•							
Date Sar	mpled: 13	/9/68	by: P.K.	& K.E.						1			
Lab.	5 Day	5 O L	II	S	pH at		Phenols	Ir o	n	Ether	Cyanide	Fluorid as	es
No.	BOD	Tot.	Susp.	Diss.	Lab.	C.O.D.	in ppb	Tot.	Diss.	Sol.	HCN	F.	
<b>-303</b> 2	5.0	334	12	322	7.4	28		3.38	1,13	tr.		0.6	
-3033											0.09		
- 30 34							100*						
<b>r-</b> 3032	Il	#2 Bj	-Products	s & Phenol	Plant A	rea - (C	omposite	9:30 a.m.	- 4:30	p.m.)			
-3033	I 2	₽ <b>#</b> 2 B <b>y</b> -	Products	& Phenol	Plant A	rea (Comp	osite 1:3	0 a.m 4	4:30 p.1	n.			
r-3034	IB	5 #2 By-	-Products	& Phenol	Plant Ar	ea - (Co	mposite l	:30 p.m. ·	- 4:30 ]	p.m.)			

Steel Company of Canada,

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### INDUSTRIAL WASTE ANALYSIS

All analyses except pH reported in p.p.m. unless otherwise indicated

1 p.p.m. = 1 mgm. / litre = 1 lb./100,000 Imp. Gals.

Municip	ality:	Hamilton		Repor	t to:	. Luyt *	ŧ			c.c.	Chem.	Lab.*	/rd	71
Source:		Stelco												
Date Sa	ampled: Se	pt. 13/68	by: R.F	•				_		. °		,		
Lab.	5-Day		Solids		pH at		Phenol	Iron	as Fe.	Cyanid	e Fluor as	ides		
No.	B.O.D.	Total	Susp.	Diss.	Lab.	¢.0.D.	ppb	Tot.	Diss.	HCN	F.			-
T-3035		456	50	406	7.3	32	12	21.0	1.88	0.03	5.0			
T-3036										0.01*				imi
														ted
				* T	est perfo	med on	preserved	sample.						- H
							1							ilto
														W
														ork
														- "
T-3035	D-1	10	5° thicke	ner overf	low Co	mp. 9:3	30 - 4:30							
T-3036	D-2	2 10	5' thicke	ner overf	low Co	mp. 1:0	00 - 4:30							Sel
														1.
														1-6
														3,
														1908



DATI	E DUE	

#### MOE/HAM/IND/ASZQ

Ontario Water Resources Co Industrial wastes survey of the Steel Co. aszq of Canada 4td.c.1 a aa



#### Environment Ontario

Laboratory Library 125 Resources Rd. Etobicoke, Ontario M9P 3VC Cenada